



Research article

Active proton efflux, nutrient retention and boron-bridging of pectin are related to greater tolerance of proton toxicity in the roots of two *Erica* species



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ABSTRACT

Background and aims: Tolerance to soil acidity was studied in two species of Ericaceae that grow in mine-contaminated soils (S Portugal, SW Spain) to find out if there are interspecific variations in H⁺ tolerance which might be related to their particular location.

Methods: Tolerance to H⁺ toxicity was tested in nutrient solutions using seeds collected in SW Spain. Plant growth and nutrient contents in leaves, stems and roots were determined. Viability tests and proton exchange were studied in roots exposed, short-term, to acidic conditions. Membrane ATPase activity and the cell-wall pectic polysaccharide domain rhamnogalacturonan-II (RG-II) were analysed to find out interspecific differences. **Results:** Variation in survival, growth and mineral composition was found between species. The H⁺-tolerant species (*Erica andevalensis*) showed greater concentration of nutrients than *E. australis*. Very low pH (pH 2) produced a significant loss of root nutrients (K, P, Mg) in the sensitive species. Root ATPase activity was slightly higher in the tolerant species with a correspondingly greater H⁺ efflux capacity. In both species, the great majority of the RG-II domains were in their boron-bridged dimeric form. However, shifting to a medium of pH 2 caused some of the boron bridges to break in the sensitive species.

Conclusions: Variation in elements linked to the cell wall-membrane complex and the stability of their components (RG-II, H⁺-ATPases) are crucial for acid stress tolerance. Thus, by maintaining root cell structure, active proton efflux avoided toxic H⁺ build-up in the cytoplasm and supported greater nutrient acquisition in H⁺-tolerant species.

1. Introduction

The heathers *Erica andevalensis* Cabezudo & Rivera and *E. australis* L. (Ericaceae) are common species in the Riotinto (SW Spain) and São Domingo (S Portugal) areas where they grow under very harsh climatic conditions and in soils contaminated by past mining activities. The soils present very low nutrient availability, low pH (pH 3.3–4.9) and high concentration of potentially phytotoxic elements like As, Cu, Fe, Pb and Zn (Soldevilla et al., 1992; Rodríguez et al., 2007; Monaci et al., 2011). Elements like Al and Mn are present at low or normal concentrations (Rodríguez et al., 2007; Monaci et al., 2011). Both species behave as excluders of metals like Cu, Zn and Pb but accumulate significant

concentrations of Mn (Soldevilla et al., 1992; Monaci et al., 2011). Tolerance to high heavy metal concentration is quite similar in both species (Rossini Oliva et al., 2009a, 2012; Trigueros Vera, 2011; Pérez-López et al., 2014). They grow in mixed communities or in mono-specific patches in heavily polluted mining spots of the Iberian Pyrite Belt (Abreu et al., 2008; Monaci et al., 2011). However, the endemic species (*E. andevalensis*) seems to thrive best in the vicinity of water courses carrying highly acidic waters (pH 2.3) (Rodríguez et al., 2007; Rossini Oliva et al., 2009b; Monaci et al., 2011). Soil pH was a distinctive trait associated with the distribution of the species in South Portugal (*E. australis* not found at pH values below 3.5) (Rodríguez et al., 2007; Abreu et al., 2008). However, in a different survey in the SW of Spain

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no significant association was found between soil acidity and species distribution (Monaci et al., 2011). Might tolerance to proton toxicity explain their differential distribution? To our knowledge, no attempt to determine the pH tolerance in both species has been carried out. The tolerance to acid soils and high concentration of metals makes both species ideal for re-vegetation of contaminated areas and phyto-stabilization of contaminated soils (Abreu et al., 2008; Rodríguez et al., 2007; Rossini Oliva et al., 2009a; Pérez-López et al., 2014). The remarkable adaptation of these species to soils with very low nutrient contents (Rodríguez et al., 2007; Monaci et al., 2011) makes these Ericaceae interesting subjects of research to provide information on nutrient acquisition under severe acidity and metal toxicity stresses.

Plant tolerance to acid soils involves a series of traits to cope with H^+ and Al toxicity, deficiency of main nutrients either by low availability or inhibition of uptake (e.g. Ca, Mg, K), or decreased solubility (P, Mo) (Marschner, 1991). Interestingly, primary limitations to plant growth in acid soils have emphasized Al and Mn toxicities and low P levels (Kochian et al., 2004). However, H^+ toxicity may be equally crucial for the adaptation of some species to particular acid soils (Kidd and Proctor, 2001). Both *Erica andevalensis* and *E. australis* tolerate Al and acidity (Abreu et al., 2008; Monaci et al., 2011) and grow at very low soil contents of Ca and P (Soldevilla et al., 1992; Monaci et al., 2011). Apart from the observation of a possible soil pH-based discrimination on predominance of one or another species (Abreu et al., 2008), no attempt has been made to test such hypothesis. The acidity tolerance might rely on physiological (acclimation) mechanisms leading to the avoidance of H^+ toxicity and/or on adaptation to nutrient limitations or metal excesses (Marschner, 1991). Proton toxicity may directly inhibit root (and plant) growth by complex mechanisms affecting cell wall and plasma membrane Ca, inhibiting K uptake and disturbing cytosol pH regulation (Schubert et al., 1990; Yan et al., 1992; Kinraide, 1998; Koyama et al., 2001). Growth adaptation to low pH leads to an increase in plant root H^+ ATPase (Yan et al., 1998; Zhu et al., 2009) but when external pH falls, H^+ release may be inhibited by an increased membrane H^+ permeability (Yan et al., 1998). Frequently, Ca, Mg and K deficiencies develop at acidic pH and the main symptoms in plants are root growth inhibition and water stress (Marschner, 1991). Additional effects of H^+ toxicity are the reduction of apoplastic Ca (Kinraide, 1998) and the disruption in the structure and stability of cell wall polysaccharides (Koyama et al., 2001). Low pH levels are also associated with inhibited cation uptake, whereas anion uptake may be only slightly influenced or unaffected (Marschner, 1995).

Plants utilize a variety of mechanisms to cope with the adverse soil factors and successful adaptation will depend on the plasticity of the species. Mechanical and biochemical malfunctioning of root cell walls, plasma membranes and wall–membrane attachment caused by acid pH might be attributable to a failure of polysaccharide or glycolipid fractions to bind boron and/or calcium. In particular, the cell-wall pectic polysaccharide domain rhamnogalacturonan-II (RG-II) is noted for participating in boron bridges (Ishii et al., 2001; O'Neill et al., 2001; Chormova et al., 2014), and glycolipids of the glycosylinositol phosphorylceramide (GIPC) class may bond through boron bridges to RG-II (Voxeur and Fry, 2014).

The objective of the present work was to study both species under controlled conditions to test their ability to grow at different pH values and to determine the possible mechanisms of acidity tolerance.

2. Materials and methods

2.1. Plant materials

Seeds of *Erica andevalensis* and *E. australis* were collected in the mining area of Riotinto (Huelva) and sown in Rockwool for germination and further cultivation in fully aerated nutrient solutions at pH 4 (Rossini Oliva et al., 2009b). Basically, the nutrient solution contains (in mM): K, 4 mM; Ca, 2 mM; Mg, 1 mM; NH_4^+ , 1 mM; NO_3^- , 5; SO_4^{2-} ,

1; $H_2PO_4^-$, 1; and (in μ M): B, 50; Fe, 100; Mn, 10; Cu, 1; Zn, 1; Mo, 0.5. Plants used for root membrane extraction and isolation of root cell wall components were cultivated in a 1/10th diluted version (0.1x) of the same nutrient solution.

2.2. Determination of pH tolerance

When the plantlets reached approximately 5 cm height were transferred to nutrient solutions at different pH (1.5, 2, 3, 4 and 6). The different pH treatments were obtained by adding NaOH (1 M) or diluted H_2SO_4 (1:10) into the nutrient solution. The experiment was carried out in a growth chamber with cycles of 26–22 °C (day–night temperature) and 16 h light/8 h darkness during 2 months. The solution was continuously aerated with an aquarium air pump and changed every week to maintain nutrient concentration relatively constant.

An additional test of tolerance was performed assaying root cell viability (Koyama et al., 2001) by transferring seedlings growing in control solution (pH 4) to a nutrient solution pH 2 and sampling roots after 0, 2, 6 and 12 h of treatment. Roots were then stained with a plant cell viability assay kit (Sigma, Missouri, USA). Fluorescence images were obtained with a Zeiss fluorescence microscope (Axioskop 2) equipped with filter cubes for green fluorescence (Chroma 41017 Endow GFP (exciter HQ470/40; dichroic Q495LP; emitter HQ525/50) and red fluorescence (HQ-Filterset for Texas Red (exciter HQ560/55; dichroic Q595LP; emitter HQ645/75), acquired with AxioVision 4 and merged with Image J.

2.3. Effect of Ca and Mg in pH tolerance

To test the effect of increasing Ca and Mg in the tolerance to low pH (Marschner, 1995), an experiment was carried out by adding additional Ca (10 or 20 mM Ca as $CaCl_2$) or Mg (20 mM as $MgSO_4$) to nutrient solutions with pH 2. Every 10 days, the culture medium was renewed and the fresh weight of plants was recorded. After two months, the plants were harvested and processed for chemical analysis as described below.

2.4. Effect of low pH on root H^+ release or uptake and nutrients leakage

Proton release by roots was measured in intact plants similarly as described by Yan et al. (1998) using a pH meter with continuous recording of pH changes (pH 301, Hanna Instruments) in a solution containing 1 mM K (KCl), 1 mM Ca ($CaCl_2$) and 1 mM Na (NaCl). Roots from plants cultivated at pH 3 were rinsed with the solution and set into the measuring system providing continuous aeration and temperature control. When the pH of the bathing solution was stable, its pH was lowered to pH 2 by adding concentrated H_2SO_4 and the pH changes were recorded during 6 h.

The effect of acidity on root nutrient leakage was tested using plants set in distilled H_2O with pH adjusted to 2.0 with concentrated H_2SO_4 and a control (distilled H_2O , pH 6.0). At different time intervals (0, 1, 2, 4, 8, 16 and 24 h), samples were taken from the solution for the analysis of roots released elements by inductively coupled plasma atomic emission spectroscopy (ICP/AES). Root chemical contents before and after acidic treatment were analysed as described below.

2.5. ATPase activity in root membranes

Isolation of root membranes was as described by Vera-Estrella et al. (1999) and microsomes composed of plasma membranes and tonoplasts were used for studying ATP hydrolytic activities. Approximately 5–10 g roots were chopped, placed in a pre-cooled glass jar and homogenized in 100–200 ml of an ice-cold medium containing 400 mM mannitol, 10% (v/v) glycerol, 5% (w/v) polyvinylpyrrolidone (PVP-10), 0.5% (w/v) BSA, 1 mM phenylmethylsulfonyl fluoride (PMSF), 30 mM Tris, 2 mM DTT, 5 mM EGTA, 5 mM $MgSO_4$, 0.5 mM butylated

hydroxytoluene, 0.25 mM dibucaine, 1 mM benzamidine, and 26 mM K^+ -metabisulfite, adjusted to pH 7.5 with H_2SO_4 . The homogenate was filtered through four layers of Miracloth and centrifuged at 10,000g 20 min and the pellet discarded. The supernatants were centrifuged at 100,000g for 45 min. The pellet was carefully resuspended in a buffer containing 400 mM mannitol, 10% glycerol, 6 mM Tris/Mes pH 7.5 and 2 mM DTT. All operations were performed at 4 °C. Root tonoplast (V) and plasma membrane (PM) ATPases were measured by the release of Pi according to the method of Ames (1966) as previously described by Vera-Estrella et al. (1999). Estimates of V-ATPase and PM-ATPase were made by assaying Pi release using buffers of different pH (pH 8 and 6.5 respectively) and inclusion of specific inhibitors (bafilomycin, vanadate, and Na azide). The values are presented as $\mu\text{mol Pi released (mg membrane protein)}^{-1} \text{ h}^{-1}$.

2.6. Analysis of rhamnogalacturonan-II from roots

The preparation of alcohol-insoluble residue (AIR; i.e. cell wall-enriched fraction) from roots was performed essentially as described by Chormova et al. (2014). Roots were rinsed with deionized water, chopped and left stirring in 75% ethanol for 6 h and the process was repeated twice. Then the AIR was saponified with 1 M Na_2CO_3 for 16 h at 4 °C, neutralised with acetic acid, rinsed with deionized water to remove salts, and freeze-dried. An aliquot of root AIR was used for ICP measurement of mineral elements as indicated below. Another aliquot, used for RG-II analysis, was digested with endopolygalacturonase (EPG; Megazyme, Ireland) or Driselase (Chormova et al., 2014). One milligram of AIR was incubated for 16 h at room temperature with 100 μl EPG solution (2.25 U/ml) in 50 mM acetate buffer (Na^+ , pH 4.8) supplemented with 0.5% (w/v) chlorobutanol. For the Driselase digestion, AIR was incubated at 37 °C for 16 h in 0.02% enzyme (purified as described by Fry (2000)) in the same buffer/chlorobutanol mixture. In both cases, the enzymic digest was analysed by polyacrylamide gel electrophoresis as described by Chormova et al. (2014). Authentic monomeric and (boron-bridged) dimeric RG-II for use as markers were obtained from Rosa cell-suspension cultures (Chormova et al., 2014).

2.7. Plant analyses

At the end of the experiment, plants were harvested and separated into roots, stems and leaves, weighed and the chemical analysis performed on each organ separately. To assess plant growth, changes in biomass weight every 10 days were used and biomass production was calculated as the difference between initial and final fresh weight. Water content in shoots and roots was calculated as $100 \times (\text{fresh weight} - \text{dry weight}) / \text{fresh weight}$. Plants samples were dried in oven at 70 °C for 48 h, powered and digested with HNO_3 and H_2O_2 . Concentration of elements was measured by ICP/AES and accuracy of digestion and analytical procedures was checked by routine determination of element concentration in standard reference materials (Monaci et al., 2011).

2.8. Data analysis

All treatments had four replicates and the experiments were run two months. The statistical significance of the differences between the treatments and species was tested using a factorial design and if the F-value showed significant differences ($p < 0.05$), the means were compared by Tukey post-hoc test. When variables did not fit a normal distribution, means were compared by Kruskal-Wallis and Mann-Whitney non-parametric tests. Data were analyzed using a statistical package (Statsoft package v6.12).

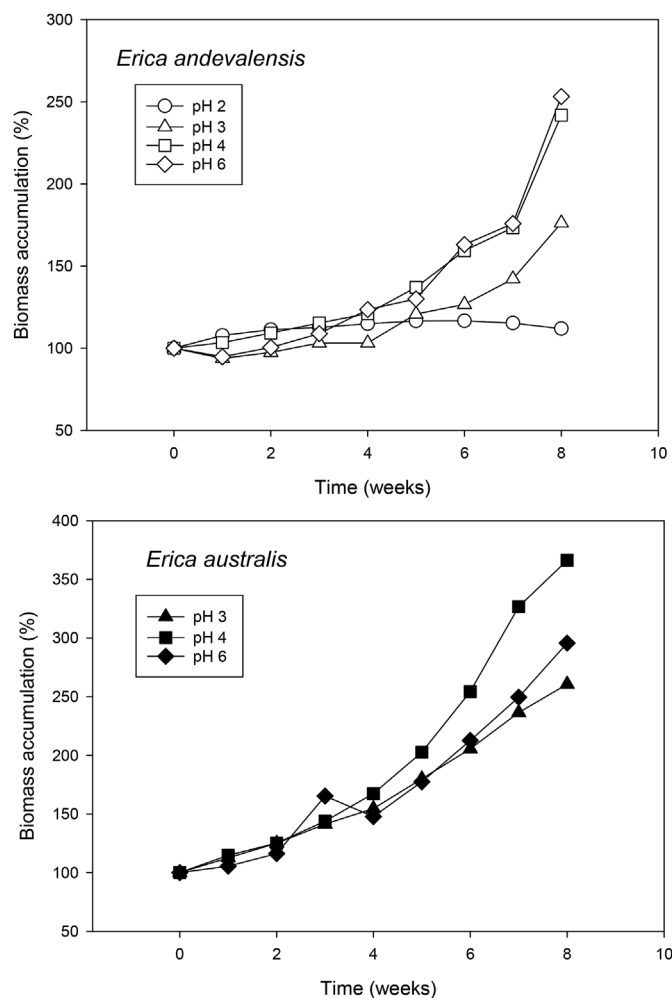


Fig. 1. Change in plant biomass (shoots + roots) of *Erica andevalensis* and *E. australis* grown in nutrient solutions with different pH treatments (relative values based in initial fresh weight, 100%). Values at pH 2 for *E. australis* are not shown as plants were dead at the third week of treatment.

3. Results

3.1. Effect of low pH in plant growth

Neither species survived at pH 1.5 but *Erica australis* died even at pH 2, quickly developing initial toxicity symptoms (shoot wilting) and reaching 100% mortality after 3 weeks. The species *Erica andevalensis* recovered from an initial wilting and survived at pH 2 but remained quiescent (Fig. 1). The low pH also led to a reduction in shoot and root water content in comparison with higher pHs (Table 1). No significant differences in growth were found for *E. andevalensis* among pH treatments above pH 2. However, *E. australis* showed a significantly greater growth rate at pH 4 than at lower or higher pH ($p < 0.05$, Table 1). In *E. andevalensis*, the low pH (pH 2) also modified the shoot/root ratio ($p = 0.001$) as result of the significant inhibition of root growth (Table 1). No differences were observed for these parameters at higher pH ($p > 0.05$).

The differential tolerance to H^+ toxicity between species was also found when plants were transferred to acidic solution (pH 2) for a short-term treatment. The species *E. australis* showed greater root sensitivity to acidity than *E. andevalensis* with root cell death in the elongation zone and mature roots (Fig. S1).

The capacity of root cell membranes to adapt to high H^+ concentration was assayed by measuring net proton exchange. Significant differences were found between roots of both species exposed to low pH

Table 1

Shoot and root biomass, root/shoot ratio and shoot water contents in two species of *Erica* after 60 days of growth in nutrient solutions of different pH. Different letters following means \pm standard error indicate statistically significant differences in data group ($p < 0.05$).

| Species | pH | Shoot biomass (g dry weight) | Root biomass (g dry weight) | Shoot/root ratio | Shoot water content (%) |
|------------------------|----------------|------------------------------|-----------------------------|------------------|-------------------------|
| <i>E. andevalensis</i> | 2 | 0.71 \pm 0.20a | 0.30 \pm 0.08a | 2.34 \pm 0.58a | 61.6 \pm 2.8a |
| | 3 | 0.88 \pm 0.37a | 1.05 \pm 0.46b | 0.89 \pm 0.25b | 79.9 \pm 7.0b |
| | 4 | 0.91 \pm 0.40a | 0.77 \pm 0.40b | 1.23 \pm 0.17b | 80.7 \pm 5.3b |
| | 6 | 1.13 \pm 0.58a | 0.91 \pm 0.44b | 1.21 \pm 0.29b | 76.4 \pm 8.0b |
| <i>E. australis</i> | 2 ¹ | – | – | – | – |
| | 3 | 3.03 \pm 1.16a | 1.58 \pm 0.56a | 1.97 \pm 0.54a | 72.4 \pm 1.2ab |
| | 4 | 6.17 \pm 1.70b | 4.44 \pm 1.60b | 1.45 \pm 0.32a | 70.0 \pm 1.0ab |
| | 6 | 3.02 \pm 1.44a | 1.63 \pm 0.90a | 1.90 \pm 0.51a | 73.9 \pm 2.0a |

¹, All *E. australis* plants died at pH 2.

(Fig. 2). The roots of *E. australis* showed a net uptake of H⁺ while those from *E. andevalensis* presented a continuous H⁺ efflux during the short-treatment in acidic conditions (Fig. 2). The activity of root plasma membrane ATPases, responsible of H⁺ efflux, showed small differences between species (Table S1). However, the ATPase activity in the tolerant species (*E. andevalensis*) was consistently higher and correlated with its greater capacity for H⁺-efflux in acidic conditions (Fig. 2). The species presented higher ATP hydrolytic activity when plants were grown at two different pHs (Table S1).

In both of these *Erica* spp., the pH of the medium influenced the monomer:dimer ratio of the pectic polysaccharide domain, RG-II (Fig. S2). At pH 4.0, both species showed more monomer and slightly less dimer than at pH 3.0. On a purely chemical basis, this was unexpected as a low pH tends to monomerise dimeric RG-II (O'Neill et al., 1996) [but for complete monomerisation in the laboratory, pH 1.0 is usually employed]. The more acid-tolerant *E. andevalensis* appears less capable of keeping its RG-II in the dimeric form, exhibiting a higher proportion of RG-II monomer than the acid-susceptible *E. australis*; this result was also unexpected as it is generally assumed that maintaining RG-II–boron–RG-II bridges is beneficial for plants. The above observations were made on plants grown in media containing low (5 μ M) boric acid. On the other hand, supplementation of the pH 4.0 medium with 1 mM boric acid converted most of the RG-II to its (boron-bridged) dimer.

Similar behaviour was observed, especially in gels of EPG digests, in plants grown at pH 3.0 and then transferred for 6 h to pH 2.0 (both media containing a low boric acid concentration, 5 μ M; Fig. 3). Prior to the pH shift, *E. andevalensis* roots exhibited a higher monomer:dimer ratio in their RG-II than did *E. australis*, as observed in Fig. S2. However, after the shift to an extremely acidic medium, *E. australis* increased in monomer:dimer ratio, indicating a partial cleavage of boron bridges, whereas *E. andevalensis* did not (Fig. 3). Since it is generally assumed that RG-II dimerisation (i.e. maintenance of boron bridging) is beneficial for plants (Brown et al., 2002; Brown and Hu, 1997; Dell and Huang, 1997; Findeklee and Goldbach, 1996; Fleischer et al., 1999; Ghanati et al., 2001; Goldbach, 1997; Holdaway-Clarke et al., 2003; Hu and Brown, 1994; Ishii et al., 2001; Loomis and Durst, 1992; Matoh et al., 2000; O'Neill et al., 2001; Reiter et al., 1997; Ryden et al., 2003),

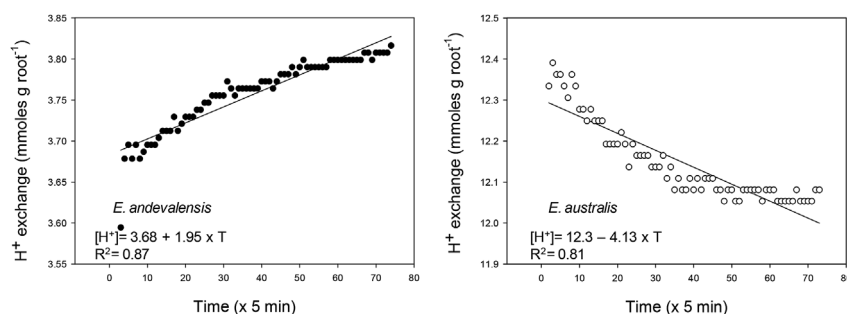


Fig. 2. Proton exchange by roots of *Erica andevalensis* and *E. australis* when incubated in minimal medium (1 K: 1 Na: 1 Ca, in mM) at low pH (pH = 2) during 6 h.

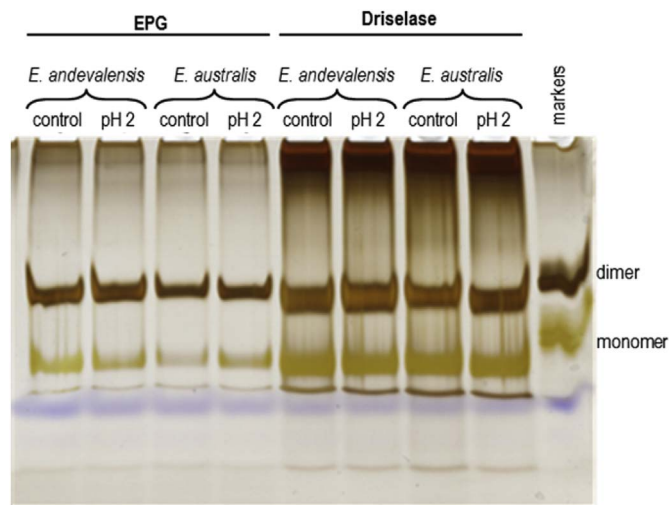


Fig. 3. Effect of a highly acidified medium on the boron-bridging of endogenous RG-II in roots of two *Erica* spp. Electrophoresis of root RG-II from *Erica andevalensis* and *E. australis* after digestion with endo-polygalacturonase (EPG; left panel) or Driselase (right panel). Plants were grown in 0.1 \times nutrient solutions (i.e. containing 5 μ M boric acid) at pH 3.0 (control) and then in some cases transferred to a highly acidified medium (pH 2) for 6 h.

this observation tallies with the observed acid-susceptibility of *E. australis*.

The alcohol insoluble residues (AIR) used for the isolation of RG-II species showed a variation in the concentration of B and Ca between species (Table S2) that may be correlated with their actual concentration in whole roots (Figs. 4 and 5).

Maintenance of root cell membrane selectivity under acidic conditions was also determined by measuring leakage of nutrients. When roots were treated with acidified water a massive leakage of K and P was recorded in the roots from *E. australis* (Fig. S3) while in roots from *E. andevalensis* nutrient losses were lower. Meanwhile, the release of Fe and Ca induced by H⁺-stress was significant but similar between species (Fig. S3). The release of other nutrients into the incubation medium

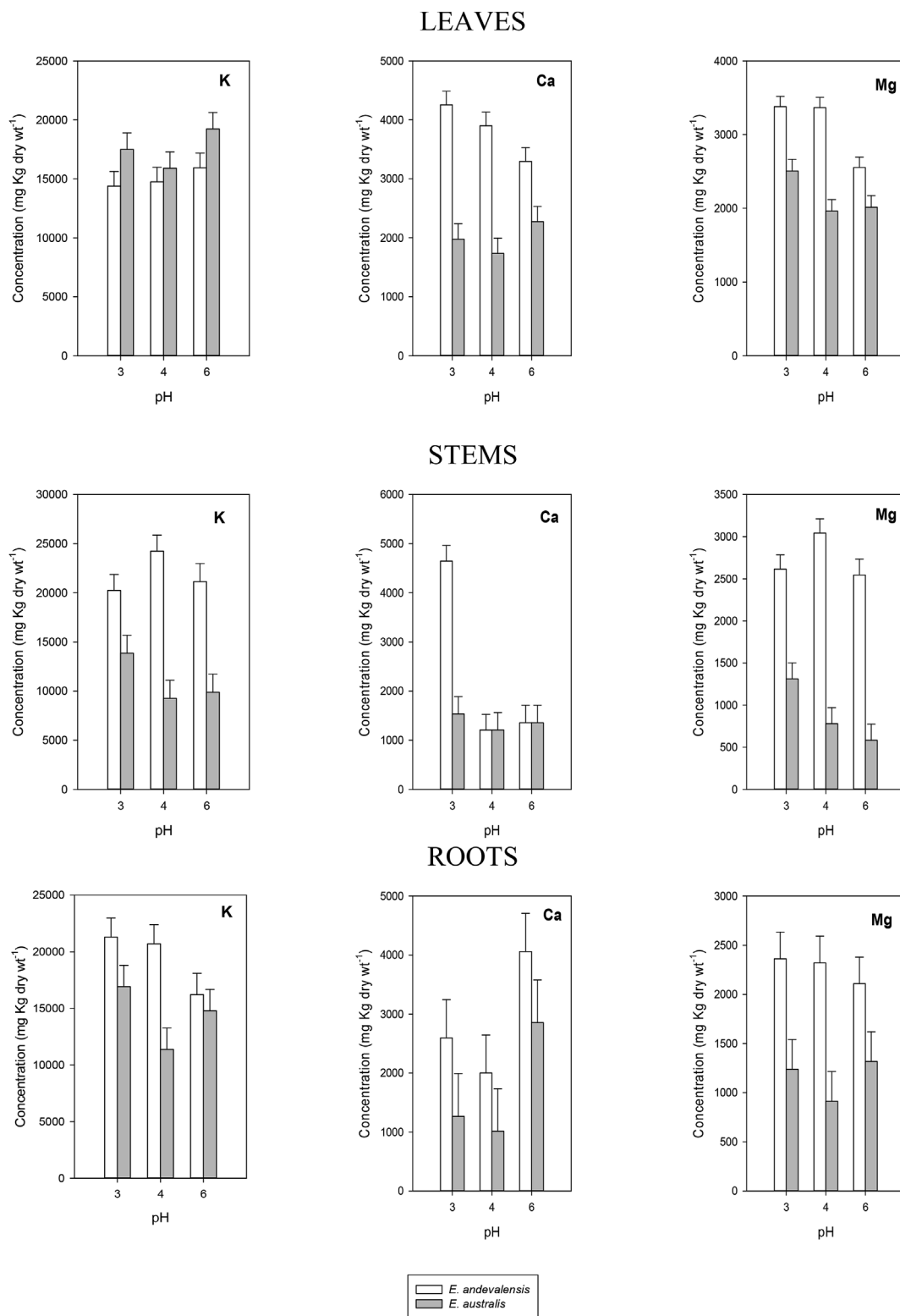


Fig. 4. Concentration of K, Ca and Mg in leaves, stems and roots of *Erica andevalensis* and *E. australis* grown in full nutrient solutions at different pH. Bars on columns indicate standard error of the mean.

remained fairly stable or they were in the detection limit for ICP system used (data not shown). The analysis of roots after the acid treatment revealed greater retention of most nutrients (K, Ca, Mg, B, Mn, Zn) in *E. andevalensis* than in *E. australis* with the exception of Fe (Table S3).

3.2. Plant nutrient contents as affected by pH and species

The solution pH affected the root concentration of Ca, P, Fe, Cu and Zn as well as the concentration of Mg and Mn in the leaves and the accumulation of S, B and Fe in the stems (Table 2). When comparing both species significant differences were found in the concentration in

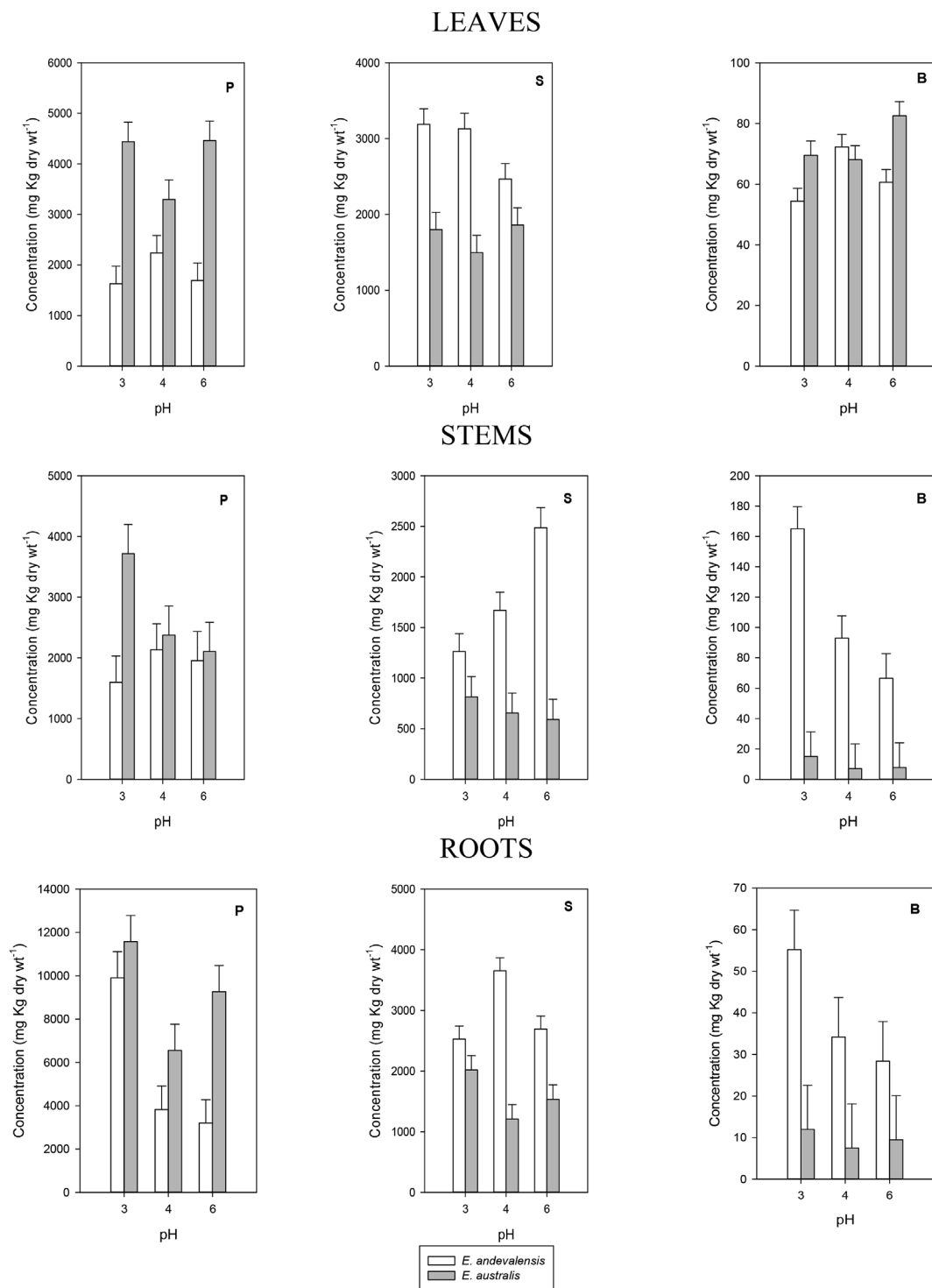


Fig. 5. Concentration of P, S and B in leaves, stems and roots of *Erica andevalensis* and *E. australis* grown in full nutrient solutions at different pH. Bars on columns indicate standard error of the mean.

the mineral contents of their leaves, stems and roots (Table 2, Figs. 4–6). The acid-tolerant species (*E. andevalensis*) showed significantly greater concentration of K, Ca, Mg, S, B and Mn in the roots. A significant inter-specific variation also existed in the macro- and micronutrient concentration of leaves and stems (Table 2). While the roots of *E. andevalensis* displayed greater concentration in the elements indicated above (Figs. 4–6), the roots and leaves from *E. australis* showed significantly greater concentration of P (Fig. 4). Among other differential features of *E. andevalensis* was the greater leaf concentration of macronutrients like Ca, Mg, S (Figs. 4 and 5) and micronutrients as

Mn (Fig. 6).

3.3. Effect of Ca and Mg availability on pH tolerance

Increasing Ca^{2+} levels in the nutrient solution did not improve the survival *E. australis* at low pH and all plants of this species had died by the end of the experiment. Additional Mg^{2+} in the nutrient solution did not enhance acid tolerance either in *E. australis* and a similar mortality (100%) as in the control treatment was recorded. In *E. andevalensis*, growth was not improved at low pH either by additional Ca^{2+} or Mg^{2+}

Table 2Results of 2-way ANOVA for nutrient concentration in leaves, stems and roots from *E. andevalensis* and *E. australis* plants grown in nutrients solutions at pH 3, 4 and 6.

| | K | Ca | Mg | P | S | B | Fe | Mn | Cu | Zn |
|---------------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|
| Leaves | | | | | | | | | | |
| pH (P) | ns | ns | P < 0.001 | ns | ns | ns | ns | P < 0.05 | ns | ns |
| Species (S) | P < 0.05 | P < 0.001 | P < 0.001 | P < 0.001 | P < 0.001 | P < 0.001 | P < 0.001 | P < 0.001 | P < 0.001 | P < 0.001 |
| P x S | ns | P < 0.05 | P < 0.05 | P < 0.05 | ns | P < 0.05 | ns | P < 0.01 | ns | P < 0.01 |
| Stems | | | | | | | | | | |
| pH (P) | ns | ns | ns | ns | P < 0.05 | P < 0.01 | P < 0.01 | ns | ns | ns |
| Species (S) | P < 0.001 | P < 0.001 | P < 0.001 | P < 0.05 | P < 0.001 | P < 0.001 | P < 0.001 | P < 0.001 | P < 0.001 | P < 0.001 |
| P x S | ns | ns | P < 0.05 | ns | P < 0.01 | P < 0.05 | ns | P < 0.05 | ns | ns |
| Roots | | | | | | | | | | |
| pH (P) | ns | P < 0.05 | ns | P < 0.001 | ns | ns | P < 0.001 | ns | P < 0.01 | P < 0.001 |
| Species (S) | P < 0.01 | P < 0.05 | P < 0.001 | P < 0.001 | P < 0.001 | P < 0.01 | P < 0.05 | P < 0.001 | ns | P < 0.001 |
| P x S | ns | ns | ns | ns | P < 0.001 | ns | ns | ns | P < 0.01 | P < 0.001 |

(Table S4) and plants showed similar growth rates in the control solution and in the solutions supplemented with bivalent cations. However, the addition of Ca^{2+} significantly increased B and K concentration (and depressed Cu concentration) in *E. andevalensis* roots at low pH (Table S5). Increasing Mg^{2+} concentration had no significant effect either in K or B concentration in roots but significantly inhibited the concentration of Cu and Zn (Table S5).

4. Discussion

Very acidic conditions produced an early shoot wilting from which only plants of *E. andevalensis* recovered but showed stunted growth afterwards. Meanwhile, *E. australis*, more sensitive to low pH, died at pH 2 but it showed higher growth rates at higher pH. Both species showed marked phenotypic responses in nutrient accumulation at different pH. Root cells of the sensitive species showed damage soon after exposure to low pH stress as the membranes became disrupted by external H^+ and leaked nutrients required for cell structure, turgor and metabolism.

The differential behaviour between species recorded under controlled conditions in the present report is in agreement with the one observed in highly acid soils of South Portugal (Abreu et al., 2008). The survival in highly acidic media was not improved by increasing the concentration of divalent cations (Ca^{2+} , Mg^{2+}) even though it was reported these ions may improve root growth in acidity-sensitive species when affected by low pH stress (Tan et al., 1992; Marschner, 1995). In acid soils, a major constrain to plant growth is the reduced concentration of metal cations like Ca, Mg or K (Marschner, 1991). In fact, low nutrient concentration in the rhizosphere was one of the factors associated with mono-specific patches of *E. andevalensis* (Monaci et al., 2011). We found significant inter-specific differences in the concentration of macro and micronutrients in all plant organs (Table 2). The effects of solution pH were significant only for Mg and Mn in leaves, for S, B and Fe in stems, and for Ca, P, Fe, Cu and Zn in roots (Table 2). With few exceptions (e.g. P and Fe) the acid tolerant species (*E. andevalensis*) showed greater root nutrient concentration than the sensitive one (Figs. 4–6). No significant differences were found in the root concentration of Cu between species (Table 2) but leaf and stem Cu concentration was significantly higher in *E. andevalensis* (Table 2, Fig. 6). Particularly distinctive traits between species were the concentrations of B and Mn in roots and stems which were significantly greater in *E. andevalensis* than in *E. australis* (Figs. 5 and 6). Fine roots of both species grown in acidic soils of Riotinto similarly showed significant differences in B concentration (F. Monaci, personal communication) with *E. andevalensis* holding the higher root B concentration.

Species in very poor soils grow slowly and have a low nutrient uptake rate but maximize nutrient intake by developing a greater root system (a lower shoot/root ratio) (Chapin, 1987; Lambers and Pooter, 2004). Thus, a greater concentration of nutrients in the roots in the

acid-tolerant species might be explained by differences in growth rate and shoot/root ratio (Table 1, Table S6). When the nutrient demand for growth is low, intake exceeds the demand and nutrients are accumulated. The differences in requirements for growth might result in the greater accumulation of many nutrients in *E. andevalensis* (Figs. 4–6). In fact, other reports with data from field-grown plants show differences between both species in leaf concentration of Ca, Mg, Mn and P (Soldevilla et al., 1992; Monaci et al., 2011; Trigueros Vera, 2011) showing the same tendency as the observed in the present report. Our results provide the first evidence of the differential plant nutrient concentration between these species when the elements are at the same concentration. One might speculate that long term adaptation to nutrient poor and acidic soils pH led to the recorded variation in element concentration as a strategy for survival.

A short-term root treatment with very low pH (pH 2), which might reflect an inherent acid tolerance, affected roots differently in both species. The sensitive species showed greater damage in its roots with loss of viability in the elongation zone and mature roots (Fig. S1). Additionally, the pH-sensitive species (*E. australis*) showed a net root proton uptake while the tolerant one (*E. andevalensis*) was able to maintain a continuous root H^+ efflux (Fig. 2). In maize, the adaptation to low pH resulted in the increase in plasmalemma H^+ -ATPase providing a greater proton extrusion ability (Yan et al., 1998). The ATP hydrolytic activity of root membranes was slightly greater in the H^+ tolerant species and it might be related to its greater H^+ extrusion. A very low medium pH (pH 2) stunted plant growth and inhibited root growth (Table 1).

Nutrients like Ca, Mg and B are linked to cell wall stability and maintenance of membrane selective properties (Wimmer and Goldbach, 1999; Koyama et al., 2001; Brown et al., 2002). Boron is the required element for the formation of cross-links between RG-II domains, providing stability to plant cell walls and their biophysical properties (Ishii et al., 2001; O'Neill et al., 2001; Chormova et al., 2014). We have shown in the present work that the RG-II–B–RG-II bridges are more resistant to extremely acidic growth media (pH 2) in *E. andevalensis* than in *E. australis*. We suggest that this correlation may be related to the enhanced ability of *E. andevalensis* to survive in highly acidic soils. The bridging of RG-II domains through a boron atom is well established to be necessary for the normal functioning of the primary cell walls in healthy vascular plants, especially in terms of wall growth, strength and porosity (Hirsch and Torrey, 1980; Hu and Brown, 1994; Findeklee and Goldbach, 1996; Fleischer et al., 1999; Ishii et al., 2001). The differential capacity to keep greater contents of B and Ca in the H^+ -tolerant species was observed even in the alcohol insoluble residues used for RG-II isolation (Table S2).

Decrease in root growth is one of the main symptoms of proton toxicity occurring during adaptation (Schubert et al., 1990; Yan et al., 1992; Kidd and Proctor, 2001; Shavrukov and Hirai, 2016). In the short-term, membranes are depolarized and proton pumps work for re-

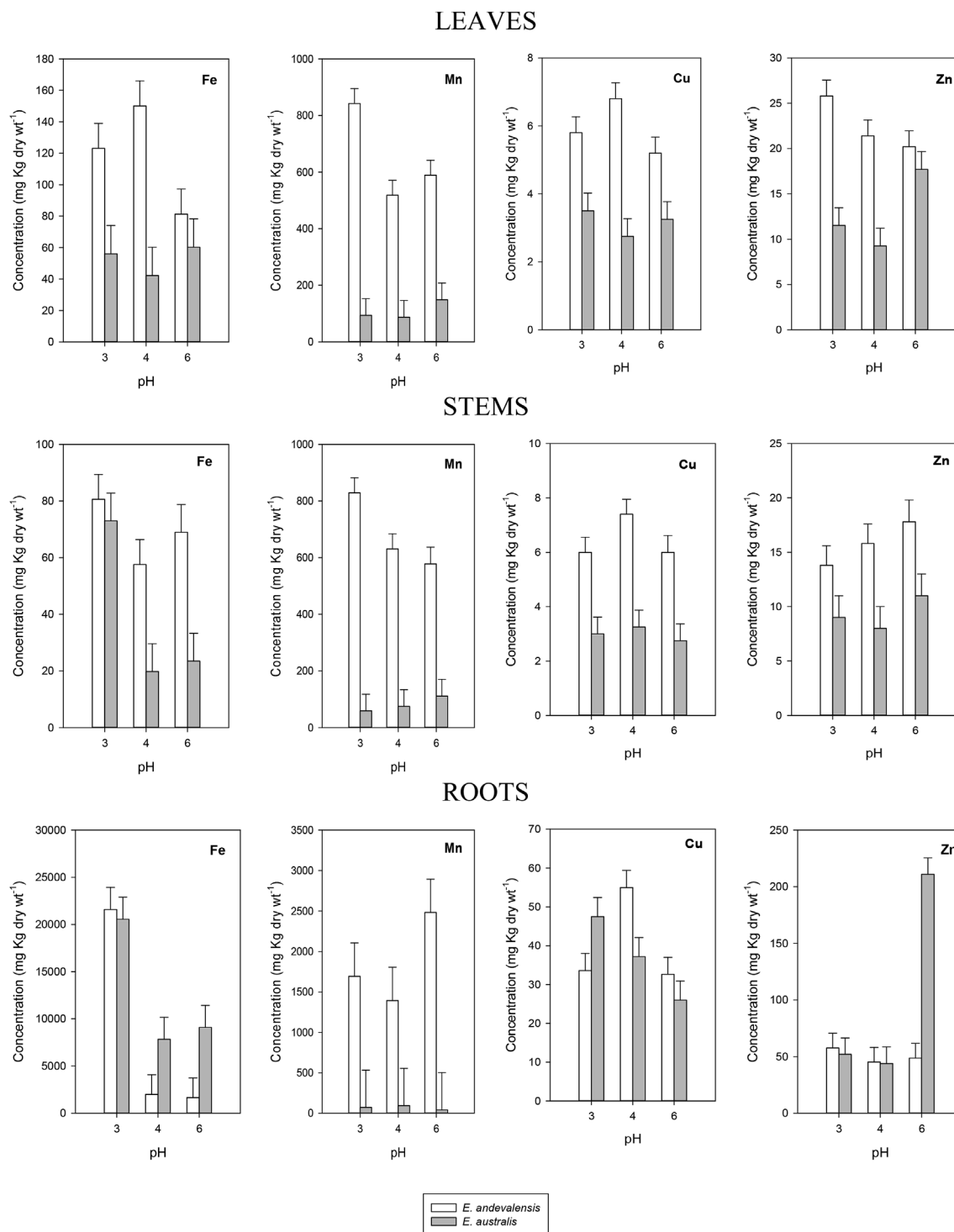


Fig. 6. Variation in the concentration of Fe, Mn, Cu and Zn in leaves, stems and roots of *Erica andevalensis* and *E. australis* grown in full nutrient solutions at different pH. Bars on columns indicate standard error of the mean.

establishing gradients to permit nutrient uptake (Shavrukov and Hirai, 2016). Some species, root medium acidity produces irreversible root damage which is reduced by increasing Ca and B concentration (Koyama et al., 2001). However, increasing the concentration of bivalent cations (Ca^{2+} , Mg^{2+}) at low pH neither increased survival in *E. australis* nor diminished the acid-induced growth inhibition in *E. andevalensis* (see Table S4). Additional supply of Ca^{2+} and Mg^{2+} in acidic solutions (pH 2) only partially recovered nutrient uptake in the tolerant species (Table S5).

The significant loss of nutrients induced after treating roots with

very low pH (Fig. S3, Table S3) may be associated to the disruption of membrane selective properties and cell wall structure induced by high H^+ concentrations. The differential amount of nutrients retained after a short-term acid treatment (Table S3) may be among the particular features providing the H^+ -tolerant species (*E. andevalensis*) with a greater tolerance to acid stress. The sensitive species (*E. australis*) showed a greater loss of nutrients in acidified water (Table S3). The greater root concentration of Ca, Mg and B in the acid-tolerant species (Figs. 4–5) might buffer the acidity-induced loss of structural cell wall elements and thereby keep root cell viability for a longer term. Among

the rapid root responses to low pH, the loss of K and P were the most significant changes (Fig. S3). The roots of acid-sensitive species (*E. australis*) released both nutrients at a much greater rate than the tolerant one (Fig. S3). Massive K leakage from roots is a general response to a variety of abiotic and biotic stresses (Demidchik et al., 2014) and has been described in roots under H⁺ toxicity long time ago (Marschner et al., 1966). The loss of root K, a main osmoticum for cell turgor and root osmotic water uptake (Marschner, 1995), may have led to the rapid dehydration of shoots observed after acidic treatment. Shoot wilting produced by soil acidity has been described in other H⁺ sensitive-species (Kidd and Proctor, 2001). While *E. andevalensis* recovered from wilting and maintained a quiescent state, *E. australis* was unable to re-establish root functionality and restore shoot water contents and the plants died shortly after. It is noteworthy that *E. andevalensis* stems also hold significantly greater B concentration and it might be an additional feature providing greater tolerance in the species.

Proton-impermeable membranes, channels of reduced pores, efficient proton-efflux systems and uptake of cations in excess to inhibit H⁺ influx are among adaptations developed by acidophile organisms, to grow at pH less than 3 (Baker-Austin and Dopson, 2007). Not being an acidophile *sensu stricto* but an acid-tolerant species, *E. andevalensis* might have evolved acquiring plasma membrane properties to keep H⁺ at bay (Messerli et al., 2005), complex cell wall structures to tolerate low pH (Baker-Austin and Dopson, 2007) and greater cation uptake (Fig. 5). Adaptation to the harsh conditions in acid root environment by significant accumulation of nutrients like Ca, Mg and B might provide *E. andevalensis* roots with a greater resistance to the structural disturbing effects of H⁺ excess on cell walls and membranes. The greater tolerance of *E. andevalensis* to H⁺ toxicity found in this work may explain its adaptation to grow in very acid soils and the formation of monospecific communities (Abreu et al., 2008; Monaci et al., 2011). The extremely low soil pH and periodic flooding with the acid river waters might be too hostile for *E. australis* establishment. Considering the significant differences in root nutrient concentration, the species might also cope with the poor nutrient availability found in the acid soils (Monaci et al., 2011). Other factors like the intrinsic lower growth rate in the tolerant species or interactions with the soil microbiota (Marschner, 1991) which also might play a role in the plant adaptation to this extreme environment.

The maintenance of root cell-wall–membrane structure in the tolerant species through a greater stability of the RG-II dimers kept a functional plasma membrane H⁺-ATPase leading to cytoplasmic H⁺ toxicity avoidance by active H⁺ efflux and providing the driving force for nutrient uptake.

5. Conclusions

The significant differences in pH tolerance between *Erica andevalensis* and *E. australis* were mostly associated with major variation in the concentration of nutrients like Ca, Mg and B required for the maintenance of cell wall structure and membrane selective properties. Indeed, differences in boron bridging of the pectic polysaccharide domain, RG-II, were noted. The interspecific variation in pH sensitivity might certainly explain the differential distribution of the *Erica* species in the contaminated acid soils and margins of highly acid rivers in South Portugal and SW Spain.

In the soils affected by past mining activities (e.g. Riotinto, SW Spain), complex environments evolved and at present are occupied by species highly adapted to situations of heavy metal contamination and low nutrient availability. However, in conditions of very low pH some species like *E. andevalensis* may thrive more successfully than others because of their intrinsic root tolerance to H⁺ toxicity.

Author contributions

S.R.O. and E.O. Leidi, performed and designed research; M.D.

Mingorance, performed chemical analyses; D. Sanhueza and S.C. Fry, performed research on RGII; E.O.L. and S.C.F. wrote the paper.

Compliance with ethical standards

The authors comply with the Ethical Standards in the COPE statement and declare that they have no conflict of interest.

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Appendix A. Supplementary data

Supplementary data related to this article can be found at <http://dx.doi.org/10.1016/j.plaphy.2018.02.029>.

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