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# REDOX BIOINTERACTOME OF CYTOCHROMES IN RESPIRATION AND PHOTOSYNTHESIS 

Trabajo presentado por el Licenciado D. José Blas Moreno Beltrán para optar al título de Doctor en Bioquímica por la Universidad de Sevilla con la acreditación de Mención Internacional

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A mis abuelos

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## I. FOREWORD

## I. FOREWORD

The current document has been prepared by following the guidelines required by the University of Seville to submit the PhD thesis as a collection of journal papers and book chapters. It consists of the following sections:
II. The list of Abbreviations.
III. The list of Publications and merits of the PhD candidate.
IV. A brief Summary of the PhD thesis.
V. An Introduction and the state-of-the-art.
VI. A section covering the Objectives of the PhD thesis.
VII. A brief description of the Results and Discussion, including main achievements and outcomes.
VIII. The global Conclusions.
IX. The list of References which have been cited in previous sections.
X. The Appendix I, containing the published journal papers and an accepted book chapter, on which the PhD thesis is based. All papers were published in scientific journals indexed in Journal Citation Reports (JCR) database.
XI. The Appendix II, containing an additional manuscript in preparation focused on the structure and dynamics of a phosphomimetic variant of cytochrome $c$.
XII. The Appendix III, including three datasets deposited in the Biological Magnetic Resonance Data Bank (BMRB) and the Protein Data Bank (PDB).

## II. ABBREVIATIONS

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ADP Adenosine 5'-diphosphate
AIRs Ambiguous Interaction Restraints
ATP Adenosine 5'-triphosphate
$b_{6} f \quad$ Cytochrome $b_{6} f$ complex
BMRB Biological Magnetic Resonance Data Bank
BN-PAGE Blue Native Polyacrylamide Gel Electrophoresis
C11orf83 Chromosome 11 open reading frame 83
Cbc $c_{1} \quad$ Cytochrome $b c_{1}$ complex
Cc Cytochrome $c$
$\mathrm{C} c_{1} \quad$ Cytochrome $c_{1}$
C $c_{6} \quad$ Cytochrome $c_{6}$
$\mathrm{C} c \mathrm{O} \quad$ Cytochrome $c$ oxidase
CSP Chemical-Shift Perturbations
cI complex I
cII complex II
cIII complex III
cIV complex IV
cV complex V
Em Mid-point redox potential
ET Electron Transfer
Fd Ferredoxin
FNR Ferredoxin-NADP Reductase
Fv Flavodoxin
GALDH L-GAlactono-1,4-Lactone DeHydrogenase
HADDOCK High Ambiguity Driven biomolecular DOCKing

| HetNOE | Heteronuclear NOE |
| :--- | :--- |
| HIG1 | Hypoxia-Inducible Genes |
| HSQC | Heteronuclear Single-Quantum Correlation |
| ITC | Isothermal Titration Calorimetry |
| JCR | Journal Citation Reports |
| $K_{\text {D }}$ | Equilibrium dissociation constant |
| $k_{\text {off }}$ | Dissociation rate constant |
| $k_{\text {on }}$ | Association rate constants |
| LHCI | PSI light-harvesting antenna complexes |
| LHCII | PSII light-harvesting antenna complexes |
| n | stoichiometry |
| NADH | Reduced Nicotinamide Adenine Dinucleotide |
| NADPH | Reduced Nicotinamide Adenine Dinucleotide Phosphate |
| NDH | NADH dehydrogenase |
| NMR | Nuclear Magnetic Resonance |
| NOE | Nuclear Overhauser Effect |
| NOESY | NOE spectroscopy |
| OXPHOS | Oxidative phosphorylation |
| Pc | Plastocyanin |
| PCD | Programmed Cell Death |
| $p$ CMF | $p$-carboxymethyl-L-phenylalanine |
| PCS | Pseudo-Contact Shifts |
| PDB | Protein Data Bank |
| PQ | Plastoquinone |
| PRE | Paramagnetic Relaxation Enhancement |
| PSI | Photosystem I |
| PSII | Photosystem II |


| Q | Coenzyme $\mathrm{Q}_{10}$ |
| :--- | :--- |
| $\mathrm{R}_{1}$ | Longitudinal relaxation rate constant |
| $\mathrm{R}_{2}$ | Transverse relaxation rate constant |
| RDC | Residual Dipolar Couplings |
| RMD | Restrained Molecular Dynamics |
| RNOS | Reactive Nitrogen and Oxygen Species |
| SC | respiratory SuperComplex |
| SPR | Surface Plasmon Resonance |
| $\Delta \mathrm{G}$ | Gibbs energy |
| $\Delta \mathrm{H}$ | Binding enthalpy |
| $\Delta \mathrm{S}$ | Binding entropy |
| $\Delta \delta_{\text {avg }}$ | Average CSP values |

## III. LIST OF PUBLICATIONS

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In the following list, the publications and achievements of the PhD candidate are enumerated as follows:

## 1. Journal Papers:

- González-Arzola, K, Díaz-Moreno, I, Cano-González, A, DíazQuintana, A, Velázquez-Campoy, A, Moreno-Beltrán, B, LópezRivas, A and De la Rosa, MA. (2015) Structural basis of SET/TAFI $\beta$ inhibition by cytochrome $c$. Proc. Natl. Acad. Sci. U.S.A., under review.
- Moreno-Beltrán, B, Díaz-Moreno, I, González-Arzola, K, GuerraCastellano, A, Velázquez-Campoy, A, De la Rosa, MA and DíazQuintana, A. (2015) Respiratory complexes III and IV can each bind two molecules of cytochrome $c$ at low ionic strength. FEBS Lett., 589, 476-483.
- Moreno-Beltrán, B, Díaz-Quintana, A, González-Arzola, K, Velázquez-Campoy, A, De la Rosa, MA and Díaz-Moreno, I. (2014) Cytochrome $c_{1}$ exhibits two binding sites for cytochrome $c$ in plants. Biochim. Biophys. Acta - Bioenergetics, 1837, 1717-1729.
- Díaz-Moreno, I, Hulsker, R, Skubak, P, Foerster, JM, Cavazzini, D, Finiguerra, MG, Díaz-Quintana, A, Moreno-Beltrán, B, Rossi, G, Ullmann, GM, Pannu, NS, De la Rosa, MA and Ubbink, M. (2014) The dynamic complex of cytochrome $c_{6}$ and cytochrome $f$ studied
with paramagnetic NMR spectroscopy. Biochim. Biophys. Acta Bioenergetics, 1837, 1305-1315.
- Olloqui-Sariego, JL, Moreno-Beltrán, B, Díaz-Quintana, A, De la Rosa, MA, Calvente, JJ and Andreu, R. (2014) Temperature-driven changeover in the electron-transfer mechanism of a thermophilic plastocyanin. J. Phys. Chem. Lett., 5, 910-914.
- Hervás, M, Bashir, Q, Leferink, NG, Ferreira, P, Moreno-Beltrán, B, Westphal, AH, Díaz-Moreno, I, Medina, M, De la Rosa, MA, Ubbink, M, Navarro, JA and van Berkel, WJ. (2013) Communication between (L)-galactono-1,4-lactone dehydrogenase and cytochrome $c$. FEBS J., 280, 1830-1840.


## 2. Book Chapter:

- Moreno-Beltrán, B*, González-Arzola, K*, Martínez-Fábregas, J, Díaz-Moreno, I and De la Rosa, MA (2015). Cytochrome c-based signalosome. In Redox proteins in supercomplexes and signalosomes, Editors: R.O. Louro and I. Díaz-Moreno. Taylor and Francis Editorial Group. ISBN: 978-1-4822-5110-4.
*These authors have equally contributed.


## 3. Oral Communications:

- A 'Floating Boat Bridge' of cytochrome $c$ molecules in plant respirasome.
Moreno-Beltrán, B, González-Arzola, K, Díaz-Quintana, A, Velázquez-Campoy, A, De la Rosa, MA, Díaz-Moreno, I.
$14^{\text {th }}$ FEBS Young Scientists' Forum. August $27^{\text {th }}-30^{\text {th }}, 2014$, Paris (France).
- NMR as a tool to analyze the structural basis for the different functional properties of phosphorylated cytochrome $c$.


## Moreno-Beltrán, B.

$3^{\text {rd }}$ Bio-NMR Annual User-Meeting: NMR in protein dynamics and structural biology. June $10^{\text {th }}-13^{\text {th }}, 2013$, Budapest (Hungary).

## 4. International Research Training:

- Oxford Protein Production Facility (OPPF), Division of Structural Biology, University of Oxford, United Kingdom. 2 weeks, 2015.

Supervisor: Dr. Louise Bird.

- Magnetic Resonance Center (CERM), Department of Chemistry, University of Florence, Italy. 16 weeks, 2012-2014.

Supervisors: Prof. Paola Turano and Dr. Rebecca Del Conte

- Leiden Institute of Chemistry, Department of Protein Chemistry, University of Leiden, The Netherlands. 16 weeks, 2011.

Supervisor: Prof. Marcellus Ubbink

## 5. Awards:

- The Finn Wold Travel Award of The Protein Society. 2015.
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- $14^{\text {th }}$ FEBS Young Scientists' Forum Award. 2014.
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## IV. SUMMARY

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Protein complex formation is at least a two-step process in which the formation of a final, well-defined complex entails the initial formation of a dynamic encounter complex. Highly transient complexes, with lifetimes in the order of microseconds-milliseconds, exhibit moderate or low binding affinities, with dissociation constants in the micromolarmillimolar range. Electron transfer reactions mediated by soluble redox proteins exchanging electrons between large membrane complexes in respiration and photosynthesis are excellent examples of transient interactions.

Here, experimental approaches based on diamagnetic and paramagnetic Nuclear Magnetic Resonance (NMR) spectroscopy and/or Isothermal Titration Calorimetry, combined with computational methods, have been used to study the molecular recognition processes of particular redox complexes involved in respiration and photosynthesis.

The studies presented in this PhD thesis go into detail about the structural and biophysical basis of the following redox complexes: cytochrome $c$-cytochrome $c_{1}$, cytochrome $c$-cytochrome $c$ oxidase and cytochrome $c$-galactonolactone dehydrogenase interactions in respiration and the cytochrome $c_{6}$-cytochrome $f$ adduct in photosynthesis. All these ET ensembles exhibit proper coupling between the redox centers although they differ in their dynamic behavior, which can be ascribed to its distinct functionality depending on the organism and its biological context. Moreover, post-translational regulation can alter the usual mechanisms of such proteins. In this work, the structure determination of
a phosphomimetic mutant of cytochrome $c$ has also been addressed by means of solution NMR.

Needless to say, such a multidisciplinary methodology, combining experimental and computational methods, opens new perspectives in our understanding of the dynamic, transient adducts formed between proteins beyond the model systems herein analyzed.

## V. INTRODUCTION

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Electron transfer (ET) processes are of great importance in many of the metabolic pathways of living organisms. They are essential for, among other processes, cellular respiration and photosynthesis, in which biochemical energy obtained from the oxidation of nutrients or from capturing light is converted into adenosine 5'-triphosphate (ATP). In these two processes, small and soluble redox proteins facilitate ET between large membrane-embedded complexes via transient contacts (Hervás et al., 2003; Lenaz and Genova, 2010; Hasan and Cramer, 2012).

General aspects of transient protein-protein redox interactions, cellular respiration and photosynthesis are presented in this section, as well as the fundamentals of the more relevant experimental and computational methodologies used.

## V. 1 TRANSIENT PROTEIN-PROTEIN REDOX INTERACTIONS

Transient inter-protein complexes exhibit high a turnover and usually involve proteins that interact with several partners: e.g. ET chains in cellular respiration and photosynthesis or signal transduction cascades (Bashir et al., 2011; Díaz-Moreno and De la Rosa, 2011; Schilder and Ubbink, 2013). Due to the interest in intermolecular ET, studies of interactions between redox proteins have provided a large amount of knowledge on transient complexes (Crowley and Ubbink, 2003; Prudêncio and Ubbink, 2004; Díaz-Quintana et al., 2015).

To sustain the high turnover rates required for several biological processes, some proteins need to interact transiently. This phenomenon is achieved by a high dissociation rate constant ( $k_{\text {off }} \geq 10^{3} \mathrm{~s}^{-1}$ ). The association rate constants $\left(k_{\mathrm{on}}\right)$ for such transient protein-protein complexes often vary from $10^{7}$ to $10^{9} \mathrm{M}^{-1} \cdot \mathrm{~s}^{-1}$. The binding affinity between two proteins that interact reversibly is defined by the equilibrium dissociation constant ( $K_{\mathrm{D}}$ ), which is calculated as the ratio between $k_{\text {off }}$ and $k_{\text {on }}$. Thus, the $K_{\mathrm{D}}$ values are generally in the $\mu \mathrm{M}-\mathrm{mM}$ range for transient contacts. The length of such interactions is significantly short, yielding complex lifetimes in the microsecondmillisecond time-scale. The lifetime of such complexes is calculated as the ratio between Ln 2 and the $k_{\text {off }}$ value (Crowley and Ubbink, 2003; Prudêncio and Ubbink, 2004; Kastritis and Bonvin, 2013; Díaz-Quintana et al., 2015).

## V. 2 CELLULAR RESPIRATION

In eukaryotes, cellular respiration is based on a series of energy transformations and metabolic processes that take place in the cytoplasmic and mitochondrial compartments to convert biochemical energy from nutrients into ATP.

Mitochondrial or nuclear genetic defects involving enzymes acting in this process impair cellular respiration. In particular, tissues with a high energy demand are vulnerable to deficiencies in or damage to cellular respiration. Such mitochondrial dysfunctions are implicated in several diseases in humans such as Parkinson's disease, Alzheimer's disease, diabetes and cancer (DiMauro, 2004; Koopman et al., 2012; Pinto and Moraes, 2014).

The following topics about eukaryotic cellular respiration are commented on in-depth below: oxidative phosphorylation, solid and fluid models of the mitochondrial electron chain, cytochrome $c$ signalosome, respiratory complexes III and IV, along with the galactono-lactone dehydrogenase.

## V.2.1 OXIDATIVE PHOSPHORYLATION

Oxidative phosphorylation (OXPHOS) is the metabolic culmination of aerobic cellular respiration in the mitochondrial electron transport chain, in which proton translocations are coupled to ET reactions. The resulting electrochemical proton gradient is used for the generation of ATP molecules from adenosine 5 '-diphosphate (ADP) and inorganic phosphate (Figure 1).

First, metabolites are oxidized in the Krebs cycle in the mitochondrial matrix to produce electrons with high transfer potential. Then, this so generated electron-motive force is converted into a proton-motive force in the mitochondrial electron transport chain and, finally, transformed into phosphoryl transfer potential, generating ATP molecules.

The transformation of the electron-motive force into the proton-motive force is performed by three respiratory multisubunit enzymes that act as electron-driven proton pumps: complex I (cI, reduced nicotinamide adenine dinucleotide (NADH) oxidoreductase); complex III, (cIII, cytochrome $c$ oxidoreductase, also called cytochrome $b c_{1}$ complex; $\mathrm{Cb} c_{1}$ ); and complex IV (cIV, cytochrome $c$ oxidase; $\mathrm{C} c \mathrm{O}$ ). These large transmembrane complexes contain multiple redox centers, including
hemes, iron-sulfur clusters, copper ions and flavins (Lenaz and Genova, 2010).

In addition, coenzyme $\mathrm{Q}_{10}(\mathrm{Q})$ and cytochrome $c(\mathrm{C} c)$ work as freely diffusible electron carriers in the mitochondrial electron chain, being required for ET between transmembrane complexes. Whereas Q is a lipid-soluble carrier, $\mathrm{C} c$ is a water-soluble protein.

Electrons from NADH are transferred to cI, which, in turn, transfers electrons to cIII by means of Q molecules. The Q pool also receives electrons from complex II (cII, succinate dehydrogenase). In addition, $\mathrm{C} c$ molecules are involved in the ET from cIII to cIV. Electrons are finally transferred from the latter cIV to molecular oxygen and, with the addition of protons, water molecules are formed. The final phase of OXPHOS is carried out by complex V (cV, ATP synthase), an ATP-synthesizing assembly that is driven by the electrochemical proton gradient (Figure 1). The OXPHOS pathway is a highly efficient way of releasing ATP molecules, compared to alternative fermentation processes, such as anaerobic glycolysis (Lenaz and Genova, 2010).

The traditional scheme of OXPHOS that includes the core protontranslocating complexes (cI, cIII and cIV), the auxiliary cII and the ATPsynthesizing cV , can be complemented with other auxiliary enzymes of the respiratory chain, such as the electron-transferring flavoprotein Q oxidoreductase, glycerophosphate dehydrogenase and dihydroorotate dehydrogenase (Lenaz and Genova, 2010). It is worth noting that the organization and dynamics of the respiratory complexes in the inner mitochondrial membrane are currently a matter of intense debate, for
which two opposite models have been proposed: the fluid and solid models (Acín-Perez and Enriquez, 2014).


Figure 1. Mitochondrial electron transport chain. Respiratory complexes I, III, IV and V constitute the basis for eukaryotic OXPHOS. Water- and lipidsoluble redox carriers are necessary for the ET between respiratory complexes. The proton translocations are coupled to the ET reactions, and the resulting proton gradient is used for energy production by cV. PDB coordinates were taken from X-ray structures of the mentioned complexes (PDB entries: 3M9S and 3 RKO for cI , 1 KYO for cIII, 10 CC for cIV and 2 XND for cV ) and the NMR structure of human $\mathrm{C} c$ (PDB entry 1J3S). This figure was generated with Chimera (Pettersen et al., 2004).

## V.2.2 FLUID AND SOLID MODELS OF THE MITOCHONDRIAL RESPIRATORY CHAIN

In the fluid model, all membrane proteins and redox components catalyzing electron transport and ATP synthesis in the mitochondrial respiratory chain are in constant and independent diffusional motion (Hackenbrock et al., 1986). By contrast, the solid model of the membrane is based on specific interactions between individual respiratory
complexes to form stable assemblies of them, named supercomplexes (SCs; Chance and Williams, 1955; Acín-Pérez et al., 2008).

The proposal of the fluid model, also known as the random collision model, is based on the fact that all redox components are independent lateral diffusants, and the ET reactions between large respiratory complexes occur during multicollisional, long-range kinetic processes (Hackenbrock et al., 1986). Such respiratory enzymes would not be in close contact with each other, excluding any solid state organization of the respiratory chain components. Thus, this model states that each complex would act as an individual entity. These observations are mainly supported by the existence of lateral diffusion of respiratory complexes and by the occupancy of only one half of the membrane surface by proteins (Sowers and Hackenbrock, 1981; Gupte et al., 1984).

By contrast, the solid model has received strong support since the discovery of detergent-based strategies for the isolation and solubilization of respiratory SCs. The finding of distinct compositions and stoichiometries of SCs has come mainly from studies of comigrations in Blue native polyacrylamide gel electrophoresis (BNPAGE) and co-purifications by sucrose gradient centrifugations (Schägger and Pfeiffer, 2000; Eubel et al., 2004; Krause et al., 2004).

Notably, plant and mammalian SCs comprising cI, cIII and cIV, have been shown to autonomously transfer electrons from NADH to molecular oxygen in the presence of Q and $\mathrm{C} c$, the so-called respirasome (Schägger and Pfeiffer, 2000; Acín-Pérez et al., 2008). Such respirasomes are SCs that contain the necessary respiratory chain complexes to accomplish reduced equivalent ET to molecular oxygen. More evidence suggesting
functional supramolecular associations in the OXPHOS systems have also been provided by flux control analysis, questioning the hypothesis of the random diffusion model (Bianchi et al., 2004; Trouillard et al., 2011).

The fundamental features of the supramolecular organization of the respiratory chain have been revealed in several eukaryotic organisms, being the best characterized those from plant and mammalian mitochondria (Schägger and Pfeiffer, 2000; Eubel et al., 2003). These assemblies are present in multiple forms, and four copies have even been found of cIV (Schägger and Pfeiffer, 2000). Thus, cI, cIII and cIV assemble into $\mathrm{cI}+\mathrm{cIII} 2, \mathrm{cI}+\mathrm{cIII}_{2}+\mathrm{cIV}_{1-4}$ and $\mathrm{cIII}_{2}+\mathrm{cIV}_{1-2}$ SCs (Krause et al., 2004; Bultema et al., 2009). In addition to them, several modules of these larger assemblies and other different SCs organizations have also been described in plant and mammalian mitochondria (Genova and Lenaz, 2014). For example, cII has been found to be associated in SCs only in the mitochondria of mice. This finding could be explained due to the participation of cII in the Krebs cycle (Acín-Pérez et al., 2008; Genova and Lenaz, 2014).

The first structural insight into a $\mathrm{cI}_{1}+\mathrm{cIII} 2+\mathrm{cIV}_{1}$ respirasome was carried out by transmission electron microscopy from mammalian mitochondria (Schäfer et al., 2007). Since then, more refined investigations have been performed not only in mammals but also in plants and other organisms, as with recent studies by single particle cryo-electron microscopy and tomography (Figure 2; Bultema et al., 2009; Althoff et al., 2011; Dudkina et al., 2011). From all these studies, it can be inferred that cytochrome $b$ and Rieske $\mathrm{Fe}-\mathrm{S}$ protein subunits of cIII are in close proximity to subunits III, VIa and VIIa of cIV, while the productive soluble $\mathrm{C} c$ binding sites on both complexes appear to face each other at a
distance of 11 nm , facilitating a putative electron pathway or gliding mechanism between both complexes, as has been proposed in plants and in humans (Genova and Lenaz, 2013; De March et al., 2014).


Figure 2. Electron cryomicroscopy structure of the respirasome $\mathbf{c I}_{1}+\mathbf{+} \mathbf{I I I}_{2}+\mathbf{+} \mathbf{I V}_{1}$. a) Side view of the fitting of the high- and medium-resolution structures of cI, cIII and cIV to the cryo-electron microscopy map. White arrowhead points to flavoproteins of cI and the olive one to the position of detergent micelles. Scale bar stands for 10 nm . b) A second side view of the respirasome, arrows point to some subunits of cIII. White and olive arrowheads point to flavoproteins of cI and the position of detergent micelles, respectively. c) Above view of the SC, showing the gaps between the complexes. This figure has been adapted from Dudkina et al. (2011).

Due to the observed variations in the amount and composition of SCs between different cell types, a new scenario for the membrane called the plasticity model has been proposed, wherein these SCs can likely co-exist with free complexes and that a variable combination of SCs and free complexes seems to exist under diverse physiological conditions, as a consequence of the cell type or the physiological state encountered (Figure 3; Lapuente-Brun et al., 2013; Genova and Lenaz, 2014). In accordance with this model, SC composition in plant and mammalian mitochondria has been shown to vary due to several factors, such as pH
and oxygen availability (Ramírez-Aguilar et al., 2011). Such diversity in the arrangement of the respiratory complexes may modulate the capability of cells to respond to diverse environmental conditions as the respirasome could quickly drive electrons from NADH to molecular oxygen in the presence of Q and $\mathrm{C} c$ (Lapuente-Brun et al., 2013).

The organization in SCs is predicted to provide functional advantages in the mitochondrial respiratory function (Lapuente-Brun et al., 2013; Genova and Lenaz, 2014). In addition, the tighter organization in the SCs may protect auto-oxidizable prosthetic groups, hampering their reaction with molecular oxygen and protecting them from reactive nitrogen and oxygen species (RNOS) damage (Panov et al., 2007; Maranzana et al. 2013). At the same time, cI seems to be stabilized within the SCs to preserve its structural integrity and activity (Schägger and Pfeiffer, 2001). Nowadays, the role of several integral membrane proteins involved in the SC assembly, stability and regulation of its activity is being investigated in-depth. Such extra membrane proteins have been named as SC assembly factors and can be critical for the optimization of the electron flow and the minimization of ROS production under stress conditions. For example, yeast respiratory supercomplex factors 1 and 2 , known as Rcf1 and Rcf2, have been reported to be relevant for $\mathrm{cIII}_{2}+\mathrm{cIV}_{1}$ SC assembly and for the optimal activity of cIV (Chen et al., 2012; Strogolova et al., 2012; Vukotic et al., 2012). Rcf proteins are highly conserved in eukaryotes and belong to the hypoxia-inducible genes family (HIG1 family). Their mammalian homologues, known as Rcf or HIG1 factors, have been also described to be critical for the assembly and function of SCs (Chen et al., 2012; Hayashi et al., 2015). Proteins from the HIG1 family are helical bundles composed of two conserved
transmembrane crossing segments, which can be preceded by an additional, amphiphilic N -terminal $\alpha$-helix (Klammt et al., 2012). In addition, another SC assembly factor, named chromosome 11 open reading frame 83 (C11orf83), has been reported in mammals (Desmurs et al., 2015). C11orf83 directly interacts with cIII and is involved in the stabilization of cIII-containing SCs, especially the $\mathrm{cIII}_{2}+\mathrm{cIV}$ SC. The sequence of C11orf83 is highly conserved in mammals, presenting an N terminal $\alpha$-helix which is anchored to the membrane (Desmurs et al., 2015).


Figure 3. Plasticity model of the mitochondrial electron chain. Respiratory complexes and SCs coexist in the membrane. Fluid and solid models can be considered as extremes of a more dynamic situation in which the respirasome cI+cIII ${ }_{2}+\mathrm{cIV}_{1}$, other SCs (cI+cIII ${ }_{2}$ and $\mathrm{cIII}_{2}+\mathrm{cIV}_{1}$ ) and free respiratory complex populations cohabit. PDB coordinates were taken from the sources mentioned in the Figure 1 legend. This figure was generated with Chimera (Pettersen et al., 2004).

## V.2.3 CYTOCHROME C SIGNALOSOME

$\mathrm{C} c$ provides a clear example of the connections that exist between supercomplexes of redox proteins that may be called respirasomes and those that form the so-called signalosomes. Indeed, Cc roles in the mitochondrial electron chain and the beginning of apoptosis, which form part of cell life and cell death signalosomes, respectively, are directly connected with the working and disruption of SCs.
$\mathrm{C} c$ is a positively-charged heme-protein located in the intermembrane mitochondrial space under homeostatic conditions in eukaryotes. $\mathrm{C} c$ has a nearly spherical shape and its overall fold is mainly $\alpha$-helical, with several regions with a coiled structure located between the helices. Cc contains four helices: the N-terminal, 50's helix, 60's helix and Cterminal. The buried hydrophobic residues making up the hydrophobic core, as well as most of the polar residues in contact with the porphyrin moiety, are conserved within highly evolved organisms. The structure is anchored through an interaction between the N - and C-terminal helices via conserved side chains present in each of the helices. The heme binding motif is essential for proper 3D folding, which is stabilized by hydrophobic and aromatic interactions with the heme. Two cysteine residues are required for the covalent attachment of the heme cofactor to the amino acid chain. The heme contains an iron ion that has two physiologically relevant low-spin oxidation states, $\mathrm{Fe}(\mathrm{II})$ and $\mathrm{Fe}(\mathrm{III})$. The heme is diamagnetic in the reduced form and paramagnetic in the oxidized state. In addition, histidine and methionine residues act as axial ligands of the iron, being the heme iron six-coordinated (Figure 4; Keilin and Hartree, 1955; Baistrocchi et al., 1996; Banci et al., 1997).


Figure 4. Three-dimensional structure of human cytochrome c. A) Overall fold of the protein. The heme is shown in sticks. Protein ends are indicated by the labels. B) His18 and Met80 axial ligands of the iron are shown (bold labels). Cys14 and Cys17 covalently bound to the heme group are also indicated in the figure, as well as the propionates of the heme. PDB coordinates were taken from the NMR structure of human $\mathrm{C} c$ (PDB entry 1J3S). This figure was generated with Chimera (Pettersen et al., 2004).

The cellular role of $\mathrm{C} c$, which was first described as a long-distance redox carrier from cIII to cIV (Keilin and Hartree, 1955), has been continuously revisited. In the mitochondrial electron chain, $\mathrm{C} c$ receives electrons from the cytochrome $c_{1}\left(\mathrm{C} c_{1}\right)$ of the cIII via a heme-to-heme ET reaction. The globular, water-soluble domain of $\mathrm{C} c_{1}$ is responsible for the ET to $\mathrm{C} c . \mathrm{C} c_{1}$ is bound to the membrane through a C-terminal $\alpha$ helix. Once $\mathrm{C} c$ is reduced, electrons are transferred from $\mathrm{C} c$ to the binuclear copper center of cIV , called $\mathrm{Cu}_{\mathrm{A}}$ (Lange and Hunte, 2002; Sakamoto et al., 2011; Díaz-Moreno et al., 2011).

Since the discovery of the respiratory function of Cc in the mitochondrial electron transport chain, several other mitochondrial $\mathrm{C} c$ partners have been reported, including the flavoenzyme L-galactono-1,4lactone dehydrogenase (GALDH) in plants. GALDH is required for the
correct assembly of cI and catalyzes the terminal step of L-ascorbate biosynthesis (Leferink et al., 2008).

In the context of SCs, it has been proposed that $\mathrm{C} c$ glides between cIII and cIV to shuttle electrons instead of carrying them by long-distance random diffusion, which is in agreement with the metabolic channeling model proposed by Kholodenko and Westerhoff (1993). Within this framework, the channeling would imply gliding mechanisms of $\mathrm{C} c$ molecules on the surface of respiratory complexes, impairing its release to the bulk phase but providing a diffusion path between its redox partners (Genova and Lenaz, 2013; De March et al., 2014). This idea demands a detailed structural and functional analysis of the interactions between $\mathrm{C} c$ and its respiratory partners cIII and cIV, whose behavior in plant and mammalian SCs remains obscure from a structural point of view.

In addition, it has been demonstrated that $\mathrm{C} c$ is capable of interacting with several protein and/or lipid targets not only under homeostatic conditions but also in the cytoplasm and even in the nucleus under programmed cell death conditions (Díaz-Moreno et al., 2011; MartínezFábregas et al., 2013; Martínez-Fábregas et al., 2014a). Notably, Ccmediated peroxidation of cardiolipin is accepted as one of the early signals of apoptosis in the cell death signalosome. Such oxidative modification of cardiolipin leads to the disruption of mitochondrial supercomplexes and consequent increase in RNOS formation (Lenaz et al., 2010). Thus, the impairing of supercomplexes can be linked with the participation of $\mathrm{C} c$ in the cell death signalosome.

A further step of complexity in these networks of interconnected molecules is added by post-translational modifications of eukaryotic $\mathrm{C} c$, which drastically alter the binding patterns and, then regulate cell signaling mechanisms of Cc. Post-translational modifications, such as phosphorylation, of mammalian $\mathrm{C} c$ could open up another level of regulation of $\mathrm{C} c$ activity. Indeed, phosphorylated $\mathrm{C} c$ has been shown to cause partial inhibition of the oxidative phosphorylation process, and in turn, has been linked to some pathological situations, such as ischemia or reperfusion injury (Yu et al., 2008; Pecina et al., 2010; Hüttemann et al., 2012a). On the other hand, experiments with a phosphomimetic substitution of Tyr48 in human $\mathrm{C} c$ have suggested functional effects on programmed cell death (PCD; Hüttemann et al., 2011; Hüttemann et al., 2012b; Sanderson et al., 2013). The possibility that phosphorylation of human $\mathrm{C} c$ regulates PCD has potentially important therapeutic implications for diseases like cancer, in which PCD is inhibited (Hüttemann et al., 2012b).

## V.2.4 RESPIRATORY COMPLEX III

Respiratory cIII is a homodimeric multisubunit membrane enzyme with a molecular mass close to 500 kDa . Each monomer contains three redoxactive subunits: cytochrome $b$ with two $b$-type heme groups, $\mathrm{C} c_{1}$ with a $c$-type heme, and the Rieske protein containing a [2Fe-2S] cluster (Figure 5). The enzyme operates via a mechanism called the Q cycle, in which it couples ET from Q to $\mathrm{C} c$, with the net translocation of protons across the membrane. Key features of the mechanism are: the bifurcated ET upon Q oxidation at the Qo catalytic site, a spatially separated second catalytic site for Q reduction ( Qi site), and the large-scale domain
movement of the Rieske protein, which facilitates ET from the Qo site to subunit $\mathrm{C} c_{1}$. $\mathrm{C} c$ docks on the latter subunit to accept the electron (Lange and Hunte, 2002; Lenaz and Genova, 2010).


Figure 5. Structural monomer of eukaryotic respiratory complex III. Model of the structural monomer of cIII based on the dimeric structure of this complex from yeast (PDB entry 1 KYO ). The cytochrome $b$ subunit is colored in blue, the Rieske subunit in purple and the cytochrome $c_{1}$ subunit in red. Other yeast cIII subunits are colored in tan. Heme cofactors are in green and the iron-sulfur cluster is in cyan. This figure was generated with Chimera (Pettersen et al., 2004).

## V.2.5 RESPIRATORY COMPLEX IV

Respiratory cIV is another multisubunit enzyme that is usually present in its monomeric form, with a molecular mass close to 200 kDa . This enzyme contains several metallic cofactors required for ET reactions including low-spin hemes and copper centers (Figure 6). ET through cIV occurs from $\mathrm{C} c$ to the $\mathrm{Cu}_{\mathrm{A}}$ center (which acts as a single-electron receptor). Then, ET is conducted to the heme $a$, and in turn, onto the bimetallic center heme $a_{3}-\mathrm{Cu}_{\mathrm{B}}$. Finally, electrons and protons are transferred to oxygen bound to heme $a_{3}$, generating water molecules (Lenaz and Genova, 2010; Sakamoto et al., 2011).


Figure 6. Structural monomer of eukaryotic respiratory complex IV. Model of the structural monomer of cIV based on the dimeric structure of this complex from bovine (PDB entry 1OCC). The cytochrome $c$ oxidase subunit containing the binuclear copper site is colored in red, whereas the subunit that contains the
$a$-type hemes and the mononuclear copper center is in pink. Other bovine cIV subunits are colored in tan. Heme cofactors are in green and the copper atoms are in blue. This figure was generated with Chimera (Pettersen et al., 2004).

## II.2.6 L-GALACTONO-1,4-LACTONE DEHYDROGENASE

L-galactono-1,4-lactone dehydrogenase, namely GALDH, catalyzes the oxidation of L-galactono-1,4-lactone to L-ascorbate with the concomitant reduction of C . GALDH is presumed to be an integral membrane protein of the inner mitochondrial membrane where it shuttles electrons into the mitochondrial electron transport chain via $\mathrm{C} c$. Although GALDH has been isolated from the mitochondria of several plants, it has been poorly characterized. Most aldonolactone oxidoreductases contain a covalently bound FAD, but plant GALDH binds the FAD cofactor in a non-covalent manner. Little information is known about the active site and 3D structures of plant GALDH are not available to date (Leferink et al., 2008).

## V. 3 PHOTOSYNTHESIS

Photosynthesis takes place on the thylakoid membranes in plants, algae and cyanobacteria. It is a fundamental biochemical process, through which light energy is converted into chemical energy, stored in ATP and reduced nicotinamide adenine dinucleotide phosphate (NADPH) molecules. These compounds provide the energy and reducing power for biosynthetic pathways, including carbon assimilation. The production of ATP using the energy of sunlight is known as photophosphorylation.

Molecular oxygen is produced as a waste product (Hasan and Cramer, 2012).

Cyanobacterial thylakoids are topologically rather different from those of algae and plant chloroplasts. Thylakoids of the latter organisms contain prominent structures known as grana, which are formed by multiple stacked layers of thylakoid disks linked by intergrana thylakoids, also known as stroma thylakoids or lamellae. Cyanobacterial thylakoid membranes lack grana and they are much more homogeneous than chloroplast thylakoids. Notably, cyanobacteria carry out both respiration and photosynthesis in the thylakoid membranes. In the absence of light, the respiratory consumption of stored photosynthesis products is essential for maintaining cell functions. In contrast to algae and plants, the majority of respiratory electron transport in cyanobacteria occurs in the thylakoid membrane, in close proximity to the photosynthetic apparatus. The proximity of both photosynthetic and respiratory systems opens up several possibilities for "hybrid" modes of electron transport involving electron exchange between photosynthetic and respiratory complexes (Bailey et al., 2008; Lea-Smith et al., 2013).

The following topics about photosynthesis are commented on in detail below: photosynthesis electron chain, photosynthetic supercomplexes, and cytochrome $c_{6}$ and plastocyanin.

## V.3.1 PHOTOSYNTHETIC ELECTRON CHAIN

The photosynthetic electron chain machinery consists of membranebound protein complexes and mobile partners, which accomplish the ET process coupled with proton translocations in the thylakoid space. The fundamentals of the photosynthesis process present a high degree of
similarity with that described for OXPHOS systems, in which proton translocations are coupled to ET steps and soluble carriers transfer electrons between large multi-subunit enzymes (Figure 7). These large transmembrane complexes contain multiple redox centers, including hemes, iron-sulfur clusters, manganese centers, phylloquinones, pheophytins, chlorophylls and carotenoids (Hasan and Cramer, 2012).

The photosynthetic ET transport in oxygen-evolving photosynthesis is mediated by two photochemical reactions, carried out by photosystem II (PSII) and photosystem I (PSI). When light photoexcites the P680 reaction center of PSII (directly or indirectly via the light-harvesting antenna complexes, which are presented in the following section), PSII converts this energy into the electrochemical potential energy required for ET reactions through PSII redox centers to plastoquinone (PQ) molecules, energizing the electrons coming from the water splitting. The P680 reaction center is composed of a chlorophyll dimer. The mechanism of water splitting is carried out by the manganese center of PSII and generates molecular oxygen and proton molecules in the thylakoid space. This mechanism and the basic subunit structure of the PSII core are highly conserved from cyanobacteria to flowering plants, constituting the source of nearly all the oxygen in the atmosphere (Umena et al., 2011). However, some peripheral subunits of PSII differ among oxygenic photosynthetic organisms (Ifuku et al., 2011).

From PSII, electrons are transferred through the membrane to the cytochrome $b_{6} f$ complex $\left(b_{6} f\right)$ with the help of the lipid-soluble PQ. The Q cycle is a mechanism for coupling the ET through $b_{6} f$ with proton translocation across the thylakoid membrane. The membrane proton gradient is the driving force for ATP production. Then, a mobile water-
soluble carrier named plastocyanin (Pc), carries out the ET from $b_{6} f$ to PSI at the lumen side. Finally, electrons are transferred from PSI to ferredoxin-NADP reductase (FNR) via ferredoxin ( Fd ) on the stromal side for the production of NADPH. The latter step culminates the lineal photosynthetic ET from water splitting to NADPH production by FNR. Photoexcitation on the P700 reaction center of PSI, which contains two chlorophyll molecules, is required to generate the electrochemical potential energy required for ET reactions towards Fd. NADPH and ATP are the first stable products of photosynthesis and are used in $\mathrm{CO}_{2}$ fixation (Hasan and Cramer, 2012).


Figure 7. Oxygen-evolving photosynthetic electron transport chain. Formation of the transmembrane proton gradient by the linear electron transport chain, which extends from the water oxidation site of PSII to the reduction of NADP molecules in the chloroplast stroma or cyanobacterial cytoplasm. Waterand lipid-soluble redox carriers are necessary for the ET between multi-subunit enzymes. A proton gradient is formed by the action of the PSII reaction center and $b_{6} f$. The ATP synthase use this gradient for the formation of energy. PDB coordinates were taken from X-ray structures of mentioned complexes (PDB entries: 1S5L for PSII, 2ZT9 for $b_{6} f, 1 \mathrm{PCS}$ for Pc, 4GYD for $\mathrm{C} c_{6}, 1 \mathrm{JB} 0$ for PSI,

2W5J and 1FX0 for ATP synthase and 2XND for FNR and Fd). This figure was generated with Chimera (Pettersen et al., 2004).

In addition, cyclic ET can occur in photophosphorylation. In this process, electrons are transferred from Fd to the PQ pool. Flavodoxin (Fv) plays the same ET role as Fd. The goal of the cyclic photosynthetic ET is ATP production, whereas the lineal ET generates ATP and NADPH molecules (Hasan and Cramer, 2012).

An excess of reducing power generated by ET induces the production of ROS. Light intensity fluctuations are normal under natural conditions. To avoid photoinhibition by excessive light energy, the most efficient strategy is the dissipation of absorbed light energy from PSII safely as heat, but there are other strategies that can involve inactive state transitions or disassemblies of the components of the photosynthetic electron chain. The dissipation of this excess light energy can be monitored as non-photochemical quenching of chlorophyll fluorescence (Horton et al., 1996; Murata et al., 2007; Tikkanen and Aro, 2014).

In addition, to sustain an efficient performance and satisfy the light reactions, photosynthetic protein complexes alter their structure and activity dynamically, forming photosynthetic supercomplexes.

## V.3.2 PHOTOSYNTHETIC SUPERCOMPLEXES

The traditional models of the photosynthetic electron chain are incomplete, giving us an inadequate understanding of the ability of photosynthetic organisms to adapt to fluctuating light environments. However, technical innovations have enabled us to reconsider the
molecular mechanisms for the regulation of photosynthesis as the dynamic interaction of thylakoid membrane complexes.

In plants and green algae, the PSII core complex is associated with the membrane-embedded light-harvesting antenna complexes, named LHCII, containing multiple pigments, e.g. carotenoides and chlorophyls. LHCII forms large macromolecular complexes, namely the PSII-LHCII supercomplexes (Croce and van Amerongen, 2014). Such an organization of supercomplexes has been visualized by electron microscopy and single particle analysis (Dekker and Boekema, 2005; Caffarri et al., 2009; Pagliano et al., 2014). The LHCII trimers associated with the PSII core are categorized into three types on the basis of their affinity with the PSII supercomplex, namely those that have a strong (S), moderate (M), or loose (L) association with the PSII core (C). These LHCII trimers consist of different combinations of three LHCII proteins (Caffarri et al., 2009).

In addition, the PSI core complex in the mentioned organisms is also associated with the membrane-embedded light-harvesting antenna complexes, known as LHCI. The structure of the PSI-LHCI complex depends on the organism (Busch and Hippler, 2011). For example, in flowering plants, the PSI-LHCI supercomplex consists of a PSI core and one copy each of four major LHCI molecules (Amunts et al., 2007; Amunts et al., 2010). In contrast, in Chlamydomonas, it is composed of a single PSI core complex and nine LHCI proteins (Drop et al., 2011). In addition to these major LHCI molecules, some organisms encode minor LHCI molecules, but they are necessary in order for the organism to function properly (Wientjes et al., 2009).

Nowadays, the assembly of reaction centers forming antenna complexes in photosynthetic supercomplexes is well accepted. However, the composition of such photosynthetic supercomplexes may involve further proteins, such as a large multisubunit complex named NADH dehydrogenase (NDH, NDH complex), which is quite similar to the respiratory cI (Shikanai, 2007). The latter enzyme can be distinguished from respiratory cI according to structural and functional differences. In fact, in flowering plants, NDH forms a supercomplex with PSI and LHCI components (Shikanai, 2014). A single particle analysis has allowed for the identification of the supercomplex consisting of a single NDH sandwiched by two copies of the PSI-LHCI supercomplexes, although several versions of such NDH-PSI-LHCI supercomplexes have been recognized (Kouril et al., 2014). Other proteins have been discovered to be essential for supercomplex associations, such as the PGR5 and PGRL1 proteins, which promote the cyclic ET around PSI in the LHCI-LHCII-FNR- $b_{6} f$-PGRL1-PGR5 supercomplex (Iwai et al., 2010). Notably, the formation and dissociation of the latter supercomplex not only switches the mode of photosynthetic electron flow, but also controls the energy balance of the two photosystems (Iwai et al., 2010).

The variability of photosynthetic supercomplexes is quite high among distinct organisms and conditions, in which distinct supercomplex associations and additional proteins can be involved, such as the PGR5/PGRL1 machinery (Shikanai, 2014). Despite the conservation of the plant and green algae core subunits of PSII and PSI, the lightharvesting systems are rather divergent in distinct organisms, suggesting different strategies for adapting to different light environments. Notably, there is no strong evidence to date for electron transport supercomplexes
in cyanobacteria, in which the role of the soluble electron carriers seems to be critical to understand the adequate working of the ET transport.

## V.3.3 PLASTOCYANIN AND CYTOCHROME $\boldsymbol{c}_{6}$

In some cyanobacteria and green algae, the copper protein Pc can be substituted by the hemeprotein cytochrome $c_{6}\left(\mathrm{C} c_{6}\right)$ to perform the same function. Although the two small and soluble proteins have no structural similarities, their interaction surfaces seem to be evolutionarily correlated. Such surfaces are proposed to be iso-functional, presenting a conserved hydrophobic region around the redox centers, as well as a relevant contribution of charge residues (Hervás et al., 1995). In addition, their heme mid-point redox potential (Em) and isoelectric points are quite similar when comparing proteins from the same organism, varying their values among distinct species (Howe et al., 2006; Worrall et al., 2008).

Pc is a blue copper protein with a type-I redox center, showing the immunoglobulin fold composed of eight $\beta$-strands along with a small $\alpha$ helix. The metal atom is buried in a hydrophobic pocket close to the helix, and it is coordinated by two histidines, one methionine and one cysteine. The copper binding site features a distorted trigonal pyramid. One of the histidine residues, called HisC, is the only copper ligand that is solvent-exposed, thus making this group the most likely physiological ET port to its partners (Canters and Gilardi, 1993; Gong et al., 2000).

Opposite to Pc , the structure of $\mathrm{C} c_{6}$ consists mainly of $\alpha$-helices, typical of a $c$-type cytochrome (Figure 8). The heme is thus covalently linked by two thioether bonds to two cysteine residues, which are part of a conserved CXXCH motif near the N -terminus. A histidine and a
methionine are coordinating the iron atom of $\mathrm{C} c_{6}$ (Worrall et al., 2007; Bialek et al., 2009).

It has been proposed that the alternative expression of the genes coding for the Pc and $\mathrm{C} c_{6}$ proteins depends on the relative environmental abundance of copper and iron (Hervás et al., 1995). In this way, cyanobacteria and green algae are able to adapt to chemically different environments. In recent years, a number of new $c$-type cytochromes have been identified in plants (cytochrome $c_{6 \mathrm{~A}}$ ), cyanobacteria (cytochrome $c_{6 B}$ ) and algae (cytochrome $c_{6 C}$ ). They have structural features similar to $\mathrm{C} c_{6}$ but their function has not been yet elucidated, although their low Em excludes a possible functional analogy between $\mathrm{C} c_{6}$ and Pc (Mason et al., 2012).


Figure 8. Three-dimensional structure of cytochrome $\boldsymbol{c}_{6}$ from Nostoc. A) Overall fold of the protein. The heme group is shown as sticks. Protein ends are indicated by the labels. B) The axial ligands are shown (bold labels) as well as the Cys residues covalently bound to the heme and the heme propionates. PDB coordinates were taken from the X-ray structure of Nostoc $\mathrm{Cc}_{6}$. (PDB entry 4GYD). This figure was generated with Chimera (Pettersen et al., 2004).

The possibility of inter-exchange between Pc and $\mathrm{C} c_{6}$, depending on the copper availability in the medium, within the photosynthetic chain in some cyanobacteria and green algae underlines the importance of the transient and dynamic nature of the complexes formed by Pc and $\mathrm{C} c_{6}$ with both $b_{6} f$ and PSI.

## V.3.4 CYTOCHROME $\boldsymbol{b}_{\boldsymbol{6}} \boldsymbol{f}$

The multisubunit enzyme $b_{6} f$, known as cytochrome $b_{6} f$ complex, is composed of several catalytic subunits (Figure 9). The Rieske protein and the $\mathrm{C} f$ form the so-called high-potential path, whereas the two hemes of cytochrome $b_{6}$, located on the opposite side of the membrane, form the low-potential path. Two Q binding sites are located on the two sides of the membrane. This complex takes up two electrons from a Q bound at the lumenal Qo site and results in the release of two protons in the aqueous lumen. $\mathrm{C} f$ is anchored in the thylakoid membrane by hydrophobic residues. Interestingly, the structure of $\mathrm{C} f$ has an elongated shape and is mainly made up of $\beta$-sheets. It is composed of a small and a large domain joined by a hinge region, the latter domain houses the heme group. A histidine acts as the fifth ligand to the iron, whereas the N terminal Tyr1 is the sixth ligand.

A comparison of the X-ray structures of $b_{6} f$ from diverse organisms with the reported respiratory cIII, the mitochondrial homologue of the $b_{6} f$, suggest that the transmembrane parts are quite similar, confirming the structural homology inferred from sequence analysis (Breyton, 2000; Baniulis et al., 2008). Indeed, the same type of large-scale domain movement of the Rieske protein is observed in the $b_{6} f$ as it is in the mitochondrial cIII upon the binding of the quinol analogue stigmatellin
(Breyton, 2000). X-ray data also suggest other specific movements of the transmembrane domains of the $b_{6} f$ required for intermediate states in the ET reaction (Breyton, 2000; Baniulis et al., 2008). On the other hand, the soluble domains of the $b_{6} f$ exhibit significant divergences between distinct species that can be correlated with an evolutionary adaptation for efficient ET in different photosynthetic organisms (Baniulis et al., 2008).


Figure 9. Structure of the cytochrome $\boldsymbol{b}_{\boldsymbol{f}} \boldsymbol{f}$ complex monomer from Nostoc cyanobacterium.
Model of the structural monomer of cytochrome $b_{6} f$ based on the dimeric structure of this complex from Nostoc (PDB entry 2ZT9). The cytochrome $b_{6}$ subunit is colored in blue, the Rieske subunit in purple and the cytochrome $f$ subunit in red. Other Nostoc cytochrome $b_{6} f$ subunits are colored in tan. Heme cofactors are in green and the iron-sulfur cluster is in cyan. This figure was generated with Chimera (Pettersen et al., 2004).

## V. 4 METHODS BASED ON SOLUTION NUCLEAR MAGNETIC RESONANCE TO STUDY TRANSIENT PROTEIN-PROTEIN INTERACTIONS DETERMINATION

Solution NMR spectroscopy is a powerful technique for the characterization and structural determination of protein-protein transient complexes, as well as for the elucidation of protein structure and dynamics in solution, with a resolution at the atomic level.

The NMR parameters measured in this PhD thesis were, mainly, diamagnetic chemical-shift perturbations (CSP) and paramagnetic relaxation enhancement (PRE). All of them were included in NMRdriven docking models. Besides this, Nuclear Overhauser Effect (NOE) techniques for protein structure and dynamics characterizations in solution have also been tackled in another section.

## V.4.1 DIAMAGNETIC CHEMICAL-SHIFT PERTURBATIONS AND SURFACE MAPPING

Protein interface mapping by CSP is a well-established procedure to study weak complexes by NMR. ${ }^{15} \mathrm{~N}$ labeling of proteins allows us to perform 2D $\left[{ }^{15} \mathrm{~N}-{ }^{1} \mathrm{H}\right] \quad$ Heteronuclear Single-Quantum Correlation (HSQC) experiments. The $\left[{ }^{15} \mathrm{~N}-{ }^{1} \mathrm{H}\right]$ HSQC spectrum gives a fingerprint of protein structure. Indeed, this spectrum provides precise information about the folding state of the protein, the presence of disordered regions,
and even degradation or aggregation phenomena. Each $\left[{ }^{15} \mathrm{~N}-{ }^{1} \mathrm{H}\right]$ HSQC cross-peak represents an amide group, correlating the chemical-shift of the amide proton with the chemical-shift of the attached amide nitrogen of a particular ${ }^{15} \mathrm{~N}$ labeled residue (Figure 10).


Figure 10. $\left[{ }^{15} \mathrm{~N}-{ }^{1} \mathrm{H}\right]$ HSQC spectra of reduced and oxidized human C . Reduced $\mathrm{C} c$ amide groups are colored in blue, whereas oxidized $\mathrm{C} c$ resonances are in red.

The assignment of the $\left[{ }^{15} \mathrm{~N}-{ }^{1} \mathrm{H}\right]$ HSQC spectrum requires other experiments, ideally triple resonance experiments with ${ }^{15} \mathrm{~N}$ and ${ }^{13} \mathrm{C}$ labeled proteins that provide sequential connectivities between residues. Thus, amide groups can be linked to particular residues and sequentially assigned. The assignment of the spectrum is essential for a meaningful interpretation of more advanced NMR experiments such as binding experiments, structure determination and relaxation analysis (Crowley and Ubbink, 2003; Prudêncio and Ubbink, 2004; Kastritis and Bonvin, 2013).

The binding site for a biomolecule-biomolecule interaction can be determined by comparison of the spectra of the biomolecule for its free and bound states when the resonance assignments are known. The position and intensities of the mentioned cross-peaks can change upon interaction with one or more partners. CSP occurs as a result of changes in the chemical environment of the observed nucleus.

CSP analyses are commonly performed on [ $\left[{ }^{15} \mathrm{~N}-{ }^{1} \mathrm{H}\right]$ HSQC spectra of the ${ }^{15} \mathrm{~N}$ labeled molecule recorded in the absence and presence of increasing amounts of the unlabeled partner. Thus, it is possible to obtain a map of the residues involved in the interaction and visualize the binding sites. The analysis of NMR spectra can be performed using the proper NMR analysis software, such as Sparky (Goddar and Kneller, University of California). The chemical-shift differences between free and bound spectra can be easily followed and calculated. With ${ }^{15} \mathrm{~N}-{ }^{1} \mathrm{H}$ correlation spectra, average CSP values ( $\Delta \delta_{\text {avg }}$ ) from nitrogen and hydrogen nuclei ( $\Delta \delta^{\mathrm{N}}$ or $\Delta \delta^{\mathrm{H}}$ ) are calculated according to the following equation (Williamson, 2013):

$$
\begin{equation*}
\Delta \delta_{\mathrm{avg}}=\frac{\sqrt{\left(\Delta \delta^{N} / 5\right)^{2}+\left(\Delta \delta^{H}\right)^{2}}}{2} \tag{1}
\end{equation*}
$$

Further, the CSP mapping and analysis can also provide information about the time scale of association and dissociation events, the stoichiometry of the binding reaction and the affinity constant of the binding (Crowley and Ubbink, 2003; Prudêncio and Ubbink, 2004; Kastritis and Bonvin, 2013).

In comprehensive terms, the average size of the CSP of backbone amide resonances provides information on the degree of dynamics of the
complex, and thus the population of the encounter complex. Large CSP indicate well-defined, short-range interactions, whereas small CSP denote high dynamics and weaker interactions. Mainly hydrophobic, short-range interactions are involved in the formation of the final complex, which is stabilized in a single orientation through a network of hydrogen bonds, salt bridges and van der Waals interactions. Water exclusion also takes place. Average CSP values can reflect the population of the encounter state and, therefore, can also be used as a reliable diagnostic tool for the dynamics within transient inter-protein complexes (Worrall et al., 2001; Worrall et al., 2002; Worrall et al., 2003; Hulsker et al., 2008; Xu et al., 2008; Bashir et al., 2011; Kastritis and Bonvin, 2013; Schilder and Ubbink, 2013).

Moreover, in inter-protein interactions, the increase of molecular weight due to the complex formation leads to significant changes in the line widths of resonances, which are associated with increased transverse relaxation rates. If the binding kinetics of a complex are slow-tointermediate in the NMR timescale, line broadening can also be observed. By analyzing the line-width differences between the free and bound states in the $\left[{ }^{15} \mathrm{~N}-{ }^{1} \mathrm{H}\right]$ correlation spectra mentioned before, the binding interface can also be mapped (Madl and Sattler, 2012).

## V.4.2 PARAMAGNETIC RELAXATION ENHANCEMENT

The information obtained from CSP analyses is not enough to study lowly populated and dynamic states like the encounter ensembles. For this reason, other NMR methods, such as paramagnetic NMR, are
required for detailed structure determination, providing valuable information about the dynamics in biomolecule complexes (Bashir et al., 2011; Schilder and Ubbink, 2013).

The presence of a paramagnetic center causes distance-dependent effects on the NMR signals of observed nuclei, which can provide very precise structural information. The paramagnetic source, which contains an unpaired electron, can be intrinsic to the molecule, for example, a metal of the protein, or can be inserted into it via site-directed spin labeling (Figure 11). In the latter case, a paramagnetic probe is linked to a cysteine residue that, in many cases, is introduced by site-directed mutagenesis. Paramagnetic tags are attached at different places on the target surface to facilitate a proper evaluation of complex orientations (Bashir et al., 2011; Kastritis and Bonvin, 2013; Hass and Ubbink, 2014; Schilder and Ubbink, 2013; Camacho-Zarco et al., 2015).

In general, when a complex is studied by NMR PRE, one of the interaction partners contains the paramagnetic center, whereas the other one is isotopically labeled, so as to monitor its signals in NMR spectra. The properties of the paramagnetic center determine the nature of the effects on the NMR spectrum. In the case of PRE, the unpaired electron causes signal line broadening of the proximal nuclei. This PRE effect is proportional to $r^{-6}$ ( r is the distance between the paramagnetic center and the observed nucleus), which is analogous to NOE. The peaks corresponding to residues close to the paramagnetic center show a decrease in peak height or disappear entirely from the spectrum. In fact, PREs can be very strong and can affect nuclei located up to $10-40 \AA$ from the paramagnetic center. This distance dependence makes PRE particularly sensitive to lowly populated states of proteins orientations
(Iwahara et al., 2004; Tang et al., 2006; Volkov et al., 2006; Tang et al., 2008; Bashir et al., 2011; Kastritis and Bonvin, 2013; Schilder and Ubbink, 2013).

For each observed amide proton of a ${ }^{15} \mathrm{~N}$-labeled protein, a paramagnetic center-induced PRE can be calculated according to equation 2 (Volkov et al., 2006; Bashir et al., 2011):

$$
\begin{equation*}
\frac{I_{\text {para }}}{I_{\text {dia }}}=\frac{R_{2, \text { dia }} \exp \left(-t R_{2, \text { para }}\right)}{R_{2, \text { dia }}+R_{2, \text { para }}} \tag{2}
\end{equation*}
$$

Where $I_{\text {para }}$ and $I_{\text {dia }}$ are measured intensities of HSQC peaks for the ${ }^{15} \mathrm{~N}$ labeled protein probe in the complex with a particular protein target in the presence of a paramagnetic or diamagnetic (control) spin label; $R_{2 \text {,dia }}$ is the transverse relaxation rate of amide protons of the protein probe in the complex with the diamagnetic spin label; $R_{2, \text { para }}$ is the paramagnetic contribution to the relaxation rate (PRE); and $t$ is the total polarization transfer time of the HSQC. For the residues whose resonances disappear in the paramagnetic spectrum, $I_{\mathrm{para}}$ was estimated from the noise level of the spectrum.

Calculated PRE rates can be converted into distances as follows (Volkov et al., 2006; Bashir et al., 2011):

$$
\begin{equation*}
r=\sqrt[6]{\frac{\gamma^{2} g^{2} \beta^{2} \tau_{c}}{20 R_{2, \mathrm{para}}}\left(4+\frac{3}{1+\omega_{h}^{2} \tau_{c}^{2}}\right)} \tag{3}
\end{equation*}
$$

Where $r$ is the distance between the unpaired electron of the paramagnetic spin label and a given amide proton of the protein probe; $\tau_{c}$ is the rotational correlation time of the electron-nucleus vector; $\omega_{h}$ and $\gamma$
are the proton Larmor frequency and the gyromagnetic ratio, respectively; $g$ is the electronic $g$ factor, and $\beta$ is the Bohr magneton.


Figure 11. Site-directed spin labeling of $\mathbf{C}$. Several spin labels are colored in blue at distinct positions on $\mathrm{C} f$, colored in red. The heme group of $\mathrm{C} f$ is colored in green. PDB coordinates were taken from the crystal structure of Nostoc $\mathrm{C} f$ (PDB entry 2ZT9). The structure was modified to introduce Cys residues and spin labels at the following positions: Gln7, Ala63, Asn71, Gln104 and Ser192. This figure was generated with Chimera (Pettersen et al., 2004).

To study encounter ensembles, other paramagnetic effects can be measured, such as Pseudo-Contact Shifts (PCS). PCS provide long-range restraints optimal for structure determination. Most lanthanoids have an anisotropic magnetic susceptibility that gives rise to PCS. The PCS is usually measured as the difference in the observed NMR shift between a paramagnetic sample and a diamagnetic analog. Ideally, the diamagnetic analog should be identical to the paramagnetic sample except for the absence of paramagnetism. Due to novel advances in developed tags,

PCS can be observed over very large distances ( $1-15 \mathrm{~nm}$ ) and measured with great accuracy (Hass and Ubbink, 2014).

## V.4.3 NUCLEAR MAGNETIC RESONANCE RESTRAINED-DRIVEN DOCKING COMPUTATIONS

Several methods based on docking computations have been developed to study protein-protein complexes over the last few decades. Most of these approaches are, however, not driven by experimental data but based on combination of energetics and shape complementarity using distinct algorithms. These traditional algorithms generally carry out $a b$ initio docking simulations followed by a posteriori scoring based on the experimental information. However, the use of a priori restrained driving docking computations, such as High Ambiguity Driven biomolecular DOCKing (HADDOCK) or in-home, XPLOR-NIH scripts-based docking computations, has significantly improved the reliability and accuracy of such docking models (van Dijk et al., 2005; Kastritis and Bonvin, 2013).

HADDOCK is a docking approach that makes use of biophysical data, such as CSP resulting from NMR titration experiments. This information is introduced as Ambiguous Interaction Restraints (AIRs) to drive the docking process. An AIR is defined as an ambiguous distance between all residues shown to be involved in the interaction. HADDOCK protocol includes several essential steps for the proper performance of docking simulations: generation of topologies for cofactors, rigid body energy minimization, semi-flexible simulated annealing and flexible refinement in water. In addition, the specifications of several parameters are critical,
such as the number of structures to generate and refine, the histidine protonation states, the definition of flexible segments, electrostatic treatment, scoring scheme or solvated docking options. The accuracy of the HADDOCK approach has been tested with many transient complexes, being applied to a large variety of them (Díaz-Moreno et al., 2005a; Domínguez et al., 2003; van Dijk et al., 2005; van Dijk and Bonvin, 2006; de Vries et al., 2007; Kastritis and Bonvin, 2013; DíazMoreno et al., 2014; Moreno-Beltrán et al., 2014).

NMR-restrained docking calculations can be also run by XPLOR-NIH rigid body docking scripts, which allow introducing manually those docking steps that can be critical for the correct modeling of particular interactions, as happens for ensemble dockings. For ensemble determination, XPLOR-NIH docking scripts based on PRE restraints constitute an excellent and better alternative to standard methods for structural elucidation of transient complexes. The PRE restraints are transformed into distances, being classified in several categories according to the PRE values. In some cases, a first docking protocol is launched before the ensemble docking. Those distance restraints satisfied by the single docking are then subtracted from all available PRE constraints. These differences serve as input data for the ensemble docking. In other cases, the all available PRE constraints are used to calculate the ensemble docking (Clore and Schwieters, 2003; Scanu et al., 2013; Kastritis and Bonvin, 2013; Díaz-Moreno et al., 2014).

## V.4.4 NUCLEAR MAGNETIC RESONANCE STRUCTURE CALCULATION

NMR spectroscopy is a fundamental technique for understanding the behavior of proteins, especially highly dynamic and small proteins that adopt several conformations in solution. Sophisticated 2D, 3D and even 4D NMR experiments and the development of molecular biology tools for the production of sufficient quantities of the appropriately labeled samples have all played important roles in establishing NMR at the vanguard of modern structural biology. However, protein structures determined from NMR spectroscopy data only constitute about $10 \%$ of the PDB collection (Bonvin and Brünger, 1995; Kay, 2005; Berman et al., 2000; Kleckner and Foster, 2011).

In particular, NMR has emerged as a powerful probe for the study of protein structure and dynamics. The goal is to relate function to dynamics and to study structures of weakly populated protein states that are thought to play an important role in biology (Güntert, 1997; Kleckner and Foster, 2011).

The backbone chemical-shift assignments serve as a starting point for studies of structure and dynamics of proteins. In fact, triple resonance experiments with ${ }^{15} \mathrm{~N}$ and ${ }^{13} \mathrm{C}$-labeled proteins are usually carried out for this purpose. In addition, the side chain chemical-shift assignments are necessary for structural determination. However, to extract distance constraints for further structure determination calculations, the conformation data are mainly derived from 2D and 3D NOE spectroscopy (NOESY) spectra. Most of NOESY cross-peaks have to be
assigned, by hand or automated modes, using the assigned chemicalshifts lists. In addition to the problem of peak overlap, some difficulties arise from spectral artifacts, noise, fast relaxation or conformational exchange. Once NOE assignments are carried out, peak volumes are integrated and introduced as input data for structure calculations. The evaluation of NOESY spectra is generally the time-limiting step in protein structure determination by NMR (Wüthrich, 1986; Güntert, 1997; Güntert et al., 1997; Kleckner and Foster, 2011; Maslennikov and Choe, 2013; Oxenoid and Chou, 2013; Gautier, 2014). Dihedral-angle constraints are also used as input data in these computations. TALOS software is habitually used to predict protein backbone torsion angles from NMR chemical-shifts (Shen et al., 2009). Other restraints derived from NMR data, such as pseudo-contact shifts (PCS) and residual dipolar couplings (RDC), can be included during a structure calculation to better restrict molecular conformations to those which are consistent with the observed data.

In the program CYANA, iterative cycles of calculation and refinement are required to perform the structure calculation, in which NMR-derived structural data is transformed into PDB coordinates. Some of the critical steps in this process are: calibration of distance constraints, elimination of spurious NOESY cross-peaks and re-assignment of incorrect NOE cross-peaks (Güntert et al., 1997). In addition, NMR-restrained molecular dynamics (RMD) simulations are performed to minimize the structure (Berndt et al., 1996; Marion, 2013). Finally, NMR data and structures have to be deposited in the Protein Data Bank (PDB, http://www.rcsb.org/pdb/) and the Biological Magnetic Resonance Data Bank (BMRB, http://www.bmrb.wisc.edu/).

## V. 5 ISOTHERMAL CALORIMETRY EXPERIMENTS <br> TITRATION

Isothermal Titration Calorimetry (ITC) is the most quantitative technique available for measuring the stoichiometry and thermodynamic properties of a biomolecule-biomolecule interaction. ITC measures the heat uptake or release during an inter-molecule interaction in solution. An ITC titration experiment consists of successive additions of a biomolecule target to a solution of a protein contained in a reaction cell. Each addition leads to a specific amount of complex formation and can be monitored as the heat released or absorbed. ITC can precisely determine the equilibrium dissociation constant ( $K_{\mathrm{D}}$ ), the binding enthalpy $(\Delta \mathrm{H})$ and the stoichiometry of the reaction (n). In addition, the Gibbs energy ( $\Delta \mathrm{G}$ ) and binding entropy ( $\Delta \mathrm{S}$ ) parameters can be determined from the $K_{\mathrm{D}}$ values. Notably, ITC is extensively used for transient interactions (Velázquez-Campoy et al., 2015). Analysis of the ITC data can be performed using distinct binding models that consider independent binding sites or cooperative effects, among other binding parameters (Velázquez-Campoy et al., 2004; Freire et al., 2009).

## VI. OBJECTIVES

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The main objectives of this PhD thesis are as follows:

1. To determine the biophysical and structural features of the transient complex between plant cytochrome $c$ and the globular domain of plant cytochrome $c_{1}$, by means of Nuclear Magnetic Resonance in solution and restrained-docking methodologies.
2. To understand the structural aspects of the transient interaction of respiratory cytochrome $c$ with complexes III and IV in solution by using heterologous proteins, namely human cytochrome $c$, the globular domain of plant cytochrome $c_{1}$ and bovine cytochrome $c$ oxidase.
3. To search for the structural basis for the role of cytochrome $c$ transferring electrons into the mitochondrial electron transport chain in plants and animals.
4. To solve the Nuclear Magnetic Resonance Structure and Dynamics of a phosphomimetic variant of cytochrome $c$, built using transfer RNA-based technologies.

In addition, secondary objectives are listed below:

1. To model the transient encounter complex between cyanobacterial cytochrome $c_{6}$ and the soluble domain of cyanobacterial cytochrome $f$ by means of ensemble restraineddocking calculations.
2. To provide further insight into the biophysical characterization of the transient communication between plant cytochrome $c$ and
plant L-galactono-1,4-lactone dehydrogenase, by Nuclear
Magnetic Resonance in solution.

## VII. RESULTS AND DISCUSSION

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In this section, a quick overview at the Results and Discussion of published works is presented. Several subsections have been included to address the following topics: short-lived complexes, promiscuity and binding surfaces, binding affinity and stoichiometry, transient complex formation, electron transfer pathways within supercomplexes, variability in binding mode and post-translational modifications. Please, refer to the journal publications in Appendix I for further details.

## VII.1. ELECTRON TRANSFER COMPLEXES AS EXAMPLES OF SHORT-LIVED COMPLEXES

Inter-protein complexes analyzed in this PhD thesis involve ET reactions that take place during cellular respiration (e.g., $\mathrm{C} c-\mathrm{C} c_{1}$ or $\mathrm{C} c$ $\mathrm{C} c \mathrm{O}$ ), photosynthesis (e.g., $\mathrm{C} c_{6}-\mathrm{C} f$ or $\mathrm{Pc}-\mathrm{C} f$ ) and vitamin C synthesis (e.g., $\mathrm{C} c-\mathrm{GALDH})$. All these complexes are excellent examples of shortlived interactions, in which soluble proteins exchange electrons with membrane-proteins via transient contacts (Lenaz and Genova, 2010; Díaz-Quintana et al., 2015; Moreno-Beltrán et al., 2015a). The molecular interactions within any short-lived complex must be able to balance high dissociation rates with binding specificity (Bashir et al., 2011).

The binding affinities of the above mentioned redox complexes are weak, with $K_{\mathrm{D}}$ values within the micromolar-minimolar range (Hervás et al., 2013; Díaz-Moreno et al., 2014; Moreno-Beltrán et al., 2014; Moreno-Beltrán et al., 2015b), with the exception of the $\mathrm{Cc}-\mathrm{CcO}$ complex (Moreno-Beltrán et al., 2015b). The higher binding affinity of
the latter complex could be ascribed to the use of the entire membranebound Cc O in contrast to the use of soluble domains, as previously suggested for cyanobacterial respiratory complexes (Navarro et al., 2005).

The intermediate/fast exchange regimes observed by NMR for $\mathrm{C} c-\mathrm{C} c_{1}$, $\mathrm{C} c-\mathrm{C} c \mathrm{O}, \mathrm{C} c-\mathrm{GALDH}$ and $\mathrm{C} c_{6}-\mathrm{C} f$ complexes are consistent with $k_{\text {off }}$ values greater than or equal to $10^{3} \mathrm{~s}^{-1}$ and lifetimes in the microsecondmillisecond range. This guarantees the balance between specificity and turnover, which is critical for ET (Sakamoto et al., 2011; Hervás et al., 2013; Scanu et al., 2013; Díaz-Moreno et al., 2014; Moreno-Beltrán et al., 2014; Moreno-Beltrán et al., 2015b).

## VII.2. PROMISCUITY AND BINDING SURFACES

In addition to a high turnover, a certain amount of promiscuity in the electron carriers $\mathrm{C} c, \mathrm{C} c_{6}$ and Pc is needed, since their surfaces are often optimized to recognize and interact with multiple redox partners (Schreiber and Keating, 2011; Hervás et al., 2013; Díaz-Moreno et al., 2014; Moreno-Beltrán et al., 2014; Moreno-Beltrán et al., 2015a; Moreno-Beltrán et al., 2015b). In general, redox centers of the electron carriers are usually surrounded by a hydrophobic patch and a charged rim where the interacting partners bind (Williams et al., 1995).

NMR CSP analyses have been performed for the identification of the inter-protein binding surfaces of the $\mathrm{C} c-\mathrm{C} c_{1}, \mathrm{C} c-\mathrm{C} c \mathrm{O}, \mathrm{C} c$-GALDH, $\mathrm{C}_{6}$ $\mathrm{C} f$ and $\mathrm{Pc}-\mathrm{C} f$ complexes in solution (Sakamoto et al., 2011; Hervás et al., 2013; Scanu et al., 2013; Díaz-Moreno et al., 2014; Moreno-Beltrán et
al., 2014; Moreno-Beltrán et al., 2015b). Such CSP maps provide valuable data about key residues involved in ET.

For the case of Cc -involving complexes, the participation of the surface surrounding the heme crevice of $\mathrm{C} c$ was found not only to be involved in its binding with the mitochondrial redox targets, such as $\mathrm{C}_{1}$ and CcO (Moreno-Beltrán et al., 2014; Moreno-Beltrán et al., 2015b), but also with their pro-apoptotic targets in the cytosol and even in the nuclei (Martínez-Fábregas et al., 2013; Martínez-Fábregas et al., 2014a; Moreno-Beltrán et al., 2015a). However, in contrast to the high turnover rate of the mitochondrial $\mathrm{C} c$ redox adducts, those occurring under apoptosis lead to the formation of stable complexes, as inferred mainly from Surface Plasmon Resonance (SPR) measurements (MartínezFábregas et al., 2013; Martínez-Fábregas et al., 2014a). Such diversity in the binding mode can be understood due to the distinct contexts of the various interactions of $\mathrm{C} c$-involving complexes. Whereas respiration is governed by highly transient interactions of $\mathrm{C} c$ required for an efficient ET within the mitochondria, the amazingly stable nucleocytoplasmic adducts of $\mathrm{C} c$ could both block survival pathways and unlock proapoptotic signals (Martínez-Fábregas et al., 2013; Martínez-Fábregas et al., 2014a; Martínez-Fábregas et al., 2014b; Moreno-Beltrán et al., 2015a).

The variability of binding modes in complexes involving heterologous transient partners can be determined by CSP analysis (Díaz-Moreno et al., 2005b; Hervás et al., 2013; Moreno-Beltrán et al., 2015b). Indeed, such chemical-shift maps can be indicative of how dynamic or unspecific a complex is, as has been observed in the case of Cc-GALDH and CcC $c_{1}$ heterologous complexes (Hervás et al., 2013; Moreno-Beltrán et al.,

2015b). The diffusion properties of orthologous proteins may not differ substantially, due to their nearly identical sizes and shapes. However, the binding and ET rates of heterologous complexes usually differ from physiological ones, insofar as the ET may be optimum within the physiological complexes (Moreno-Beltrán et al., 2015b). Such differences can be rationalized not only in terms of the total net charge of the proteins, but also in subtle changes arising from specific charge localizations on the surfaces of the two molecules, as was described for the heterologous complexes $\mathrm{C} c$-GALDH and $\mathrm{C} c$ - $c_{1}$ (Hervás et al., 2013; Moreno-Beltrán et al., 2015b).

## VII.3. STOICHIOMETRY AND BINDING AFFINITY

To transport an electron from a donor protein to a remote acceptor one, the carrier must specifically recognize the two partners, along with being dissociated quickly from them (Bashir et al., 2011; Díaz-Quintana et al., 2015; Moreno-Beltrán et al., 2015a). The binding affinity, as well as the stoichiometry of the interaction, is essential to properly understand the biological system (Moreno-Beltrán et al., 2015a). Both reaction parameters can be determined by CSP using distinct mathematical binding models, as is done for $\mathrm{Cc}-\mathrm{C} c_{1}$ complexes (Moreno-Beltrán et al., 2014; Moreno-Beltrán et al., 2015b).

In addition, binding parameters are strongly modulated by the ionic strength since the short-lived ET complexes are electrostatically driven (Hervás et al., 2013; Moreno-Beltrán et al., 2014; Moreno-Beltrán et al., 2015b). Indeed, the two principal interaction forces leading to the
formation of such adducts are mainly electrostatics and hydrophobics, as concluded from theoretical and experimental data for many transient complexes (Moreno-Beltrán et al., 2014; Moreno-Beltrán et al., 2015a).

As a complement to solution NMR, ITC binding experiments have been used to accurately determine the binding affinity and stoichiometry of such transient complexes in solution (Hervás et al., 2013; MorenoBeltrán et al., 2014; Moreno-Beltrán et al., 2015b). The differences in dissociation constants obtained in some cases by ITC and NMR can be explained by the distinct physical phenomena measured and by differences in the experimental conditions. However, there are no significant discrepancies in the results obtained from the biological systems that we have analyzed in this PhD thesis.

It is important to note here that most NMR or ITC analyses are performed at low ionic strength. Under such conditions, the non-specific long-range interactions are favored (Díaz-Quintana et al., 2015; MorenoBeltrán et al., 2015b). A full comprehension of the reaction mechanism for ET complexes demands the consideration of the diverse available data, including the cross-linking approaches, direct binding or kinetics experiments (Díaz-Quintana et al., 2015; Moreno-Beltrán et al., 2015a). For instance, in addition to the NMR and ITC assays, laser-flash measurements were carried out to monitor the oxidation of GALDH by $\mathrm{C} c$. These experiments showed a nonlinear dependence on $\mathrm{C} c$ concentration at low ionic strength, suggesting the formation of a transient complex prior to the ET step. Notably, in contrast to what happens at low ionic strength, the ET rate of this interaction increases at high ionic strength and is consistent with a second-order collision process (Hervás et al., 2013).

## VII.4. FROM THE ENCOUNTER ENSEMBLE TO THE WELL-DEFINED COMPLEX

The formation of protein-protein complexes is thought to be a two-step process in which the final complex is preceded by an intermediate state, or a so-called encounter complex (Bashir et al., 2011; Schilder and Ubbink, 2013; Hass and Ubbink, 2014). The binding event is first described by a coupled diffusion model where a protein binds nonspecifically to its partner's surface - forming the so-called encounter complex - and then diffuses along the partner's surface (i.e., slides on it) until it either finds an active conformation or dissociates from its partner. At this stage, it is worth noting that the population of complexes capable of ET reactions may correspond to either a single conformation or a subset of an ensemble of conformers.

On the one hand, paramagnetic NMR experiments have shown that cyanobacterial $\mathrm{C} c_{6}$ slides along a particular region on the surface of $\mathrm{C} f$, generating an encounter complex in which the achievement of the productive conformation is practically guaranteed during the lifetime of the complex (Díaz-Moreno et al., 2014). Although previous docking models have pointed to a well-defined $\mathrm{C}_{6}$ - $\mathrm{C} f$ complex (Díaz-Moreno et al., 2005a), PRE NMR data are not compatible with this (Díaz-Moreno et al., 2014). The complex is best described by a highly dynamic ensemble, first formed by electrostatic pre-orientation and then stabilized mainly by hydrophobic contacts.

The binding affinity in this complex is lower in comparison with those calculated for $\mathrm{C} c$-involving complexes at low ionic strength (Díaz-

Moreno et al., 2014; Moreno-Beltrán et al., 2014; Martínez-Fábregas et al., 2014a), but comparable with those performed at a moderate or high ionic strength (Hervás et al., 2013; Moreno-Beltrán et al., 2015b). It is relevant to mention that the high off-rate is as significant as the binding affinity for the success of the transient complexes (Díaz-Quintana et al., 2015). In fact, given the high local concentration of $\mathrm{C} c_{6}$ and $\mathrm{C} f$ in the lumen, the low affinity does not imply a strict requirement for this complex formation.

PRE NMR profiles indicate a major encounter population in which electrostatic forces promote the establishment of hydrophobic forces, but do not take part in the activation barrier. This activation barrier might be low due to the gradual resolvation proposed for this interaction (DíazMoreno et al., 2014). This coincides with the finding that complexes involving Nostoc $\mathrm{C} c_{6}$ do not show the enthalpy-entropy compensation in activation energy characteristic of other ones and exhibits a rate constant for electron donation to PSI that is independent of temperature (Hervás et al., 1996).

The idea that multiple orientations of an encounter ensemble are suitable for ET was also described for the myoglobin-cytochrome $b_{5}$ complex, although the latter complex appears to be dominated by chargecharge interactions (Liang et al., 2002; Worrall et al., 2002). Notably, the encounter complexes cannot be fully analyzed by X-ray diffraction or standard NMR methodologies like diamagnetic CSP, as they are not sensitive to populations of less than $10 \%$ of the molecules in question. They can be studied, however, using paramagnetic NMR methodologies.

On the other hand, the X-ray diffraction structure of yeast $\mathrm{C} c$ bound to the $\mathrm{C} b c_{1}$ complex can serve as an example of a transient complex that adopts a single, well-defined ET conformation resulting from multiple encounters of $\mathrm{C} c$ on the $\mathrm{Cbc} c_{1}$ surface (Lange and Hunte, 2002; Heinemeyer et al., 2007). This productive conformation of $\mathrm{C} c$ with respect to the surface of plant $\mathrm{C} c_{1}$ has also been detected in solution by NMR and ITC. Notably, a second non-ET conformation of $\mathrm{C} c$ was also identified by combining NMR and ITC with computational methods (Moreno-Beltrán et al., 2014). The docking of $\mathrm{C} c$ at a negatively-charged pocket of plant $\mathrm{C} c_{1}$, named as the distal binding site, is fully compatible with the complex III structure (Moreno-Beltrán et al., 2014). The finding of this distal site matches previous cross-linking and functional assays (Speck and Margoliash, 1984; Stonehuerner et al., 1985; Devanathan et al., 2007; Genova and Lenaz, 2013). Both the distal and the productive site, named as the proximal site, differ in the heme-to-heme distance according to the ab-initio and NMR-restrained docking calculations (Moreno-Beltrán et al., 2014). The visualization of the proximal, but not the distal site in X-ray coordinates, could be explained by assuming a weaker nature of the distal binding site, which could be, on the contrary, stable enough to be detected by NMR. Indeed, $\mathrm{C} c$ interacts with both the proximal and distal sites of $\mathrm{C} c_{1}$ with different dissociation constants. The differences in affinity between both binding sites are moderate. This latter finding is compatible with the linear behavior of the CSP observed for $\mathrm{C} c$ amide groups and with a similar chemical environment on the surface of plant $\mathrm{C} c_{1}$, which is negatively charged (Moreno-Beltrán et al., 2014; Moreno-Beltrán et al., 2015b).

## VII.5. ELECTRON TRANSFER PATHWAYS WITHIN RESPIRATORY SUPERCOMPLEXES

The concept of substrate channeling has been applied in enzymology to explain how small molecules undergo sequential transformations without diffusing to the bulk phase (Kholodenko and Westerhoff, 1993; Brown et al., 1996). Recently, a channeling of plant $\mathrm{C} c$ molecules was proposed to account for the ET between the $\mathrm{C} b c_{1}$ and $\mathrm{C} c \mathrm{O}$ complexes (Genova and Lenaz, 2013; Moreno-Beltrán et al., 2014). According to our results, a novel, extra distal site of plant $\mathrm{C} c_{1}$ detected in solution may provide a path for diffusion of the $\mathrm{C} c$ molecules towards $\mathrm{C} c \mathrm{O}$, facilitating the functional shuttle of electrons (Moreno-Beltrán et al., 2014).

The presence of additional binding sites for $\mathrm{C} c$ has also been detected in bovine $\mathrm{C} c \mathrm{O}$ in equilibrium conditions (Moreno-Beltrán et al., 2015b) and in gel filtration studies (Osheroff, et al., 1983). These new sites of Cc on $\mathrm{C} c_{1}$ and $\mathrm{C} c \mathrm{O}$ open up new scenarios within the mitochondrial electron chain, which could explain the "sliding" dissociation pathway hypothesis in mammals (De March et al., 2014; Moreno-Beltrán et al., 2015b). This model suggests that the $\mathrm{C} c$ glides along the surface of $\mathrm{C} c_{1}$ during their functional binding instead of a direct dissociation from the $\mathrm{C}_{1}$ surface.

## VII.6. POST-TRANSLATIONAL REGULATION

$\mathrm{C} c$ is a target of several post-translational modifications such as nitration (Rodríguez-Roldán et al., 2008; García-Heredia et al., 2010; Díaz-Moreno et al., 2011b; García-Heredia et al., 2012) and phosphorylation (García-Heredia et al., 2011), which have an effect on
the conformation of the heme moiety and its functionality as an electron carrier and apoptotic protein (Lee et al., 2006; Yu et al., 2008; Pecina et al., 2010; Hüttemann et al., 2011; Díaz-Moreno et al., 2011; GarcíaHeredia et al., 2011; García-Heredia et al., 2012; Ly et al., 2012).

However, many functional and structural aspects of the biointeractomics of $\mathrm{C} c$ inside the mitochondria upon these modifications still remain obscure. Further research is now in progress to unveil the role of phosphorylated human $\mathrm{C} c$ at Tyr48 in cellular respiration and PCD. Unfortunately, the specific $\mathrm{C} c$-phosphorylating kinase remains unknown and due to the technical difficulties of obtaining enough phosphorylated $\mathrm{C} c$ from cell extracts, it is usually mimicked by using Tyr-by-Glu mutations (García-Heredia et al., 2011). To improve the traditional mimicking of Tyr48 phosphorylation, Tyr48 residue can be replaced by the non-canonical amino acid p-carboxymethyl-L-phenylalanine ( $p \mathrm{CMF}$ ). $p \mathrm{CMF}$ better emulates the volume of a phosphorylated tyrosine residue than a glutamic residue (Xie et al., 2007). Thus, the Tyr48encoding triplet of the human $\mathrm{C} c$ gene was substituted by an amber stop codon to replace the original residue with the non-canonical amino acid by the evolved tRNA technique. This innovative approach allows for the efficient incorporation of $p \mathrm{CMF}$ at a site-specific position as a result of the action of an orthogonal tRNA that recognizes the amber codon.

The structure and dynamics of this novel phosphomimetic Cc at position 48 have been solved by using NMR in solution. Notably, the structure of phosphomimetic $\mathrm{C} c$ reveals significant changes at the loop containing the $p \mathrm{CMF}$ and its 3D proximal environment, especially the 50's helix and the omega-loop including the methionine ligand of the iron. In addition, the NMR relaxation measurements point to a highly
dynamic behavior of the loop containing the mutation, which displays a high flexibility in the $\mu \mathrm{s}-\mathrm{ns}$ timescale as can be inferred from heteronuclear NOE (HetNOE) and longitudinal relaxation rate $\left(\mathrm{R}_{1}\right)$ measurements. In addition, conformational exchanges in the ms timescale are also detected, as determined by transverse relaxation rate $\left(\mathrm{R}_{2}\right)$ parameters and a careful evaluation of NOESY spectra. These structural data strongly support its biological functionality, in which this post-translationally modified $\mathrm{C} c$ impairs both mitochondrial respiration and programmed cell death events (Moreno-Beltrán et al., unpublished data).

## VIII. CONCLUSIONS

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The main conclusions of this PhD thesis inferred from the distinct redox systems are:

1. Additional binding sites in equilibrium conditions for respiratory cytochrome $c$ in both mitochondrial complexes III and IV have been elucidated in vitro. These extra sites could facilitate the turnover and sliding mechanisms of cytochrome $c$ molecules between complexes III and IV within plant and mammalian supercomplexes. In this context, the traditional and simple view of cytochrome $c$ as a long-distance carrier in the membrane would be replaced by a new and innovative one, in which cytochrome $c$ molecules would glide on respiratory complexes III and IV surfaces without entering into the mitochondrial bulk phase. Such diffusion paths might have physiological significance in the electron flow within supercomplexes so as to optimize the use of available substrates.
2. The Nuclear Magnetic Resonance structure of phosphomimetic cytochrome $c$ shows significant changes at the region containing the analogue of the phosphorylated tyrosine and its environment, affecting proximal areas of cytochrome $c$, such as the 50 's helix and the omega-loop containing the methionine axial ligand. The Nuclear Magnetic Resonance relaxation measurements point to a
highly dynamic behavior of the phosphomimetic protein. These structural data can support its functional behavior, in which this post-translationally modified cytochrome $c$ impairs both mitochondrial respiration and programmed cell death events.
3. The structure of the cyanobacterial complex between cytochrome $f$ and cytochrome $c_{6}$ reveals that the encounter ensemble is essential to molecular recognition events among photosynthetic electron transfer proteins. Experimental evidence for the role of hydrophobic interactions in the encounter complex is cumulative, blurring the distinction between encounter complex and stereospecific complex. This finding becomes more evident in electron transfer complexes, where a short distance between the metal ions is all that is required for their electron transfer activity. Then, there is no reason for the assembly of a single, active orientation within the complex.
4. Plant L-galactono-1,4-lactone dehydrogenase forms a transient low affinity complex with plant and yeast cytochrome $c$. This relatively non-specific interaction involves protein-protein dynamic motions and does not preclude rapid electron transfer within the complex, since sufficient electron transfer permissible conformations are apparently sampled in this dynamic complex.

## IX. GENERAL REFERENCES

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## X. APPENDIX I

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## JOURNAL PAPERS:

1. Moreno-Beltrán, B, Díaz-Quintana, A, González-Arzola, K, Velázquez-Campoy, A, De la Rosa, MA and Díaz-Moreno, I. (2014) Cytochrome $c_{1}$ exhibits two binding sites for cytochrome $c$ in plants. Biochim. Biophys. Acta - Bioenergetics 1837, 1717-1729.
2. Moreno-Beltrán, B, Díaz-Moreno, I, González-Arzola, K, GuerraCastellano, A, Velázquez-Campoy, A, De la Rosa, MA and DíazQuintana, A. (2015) Respiratory complexes III and IV can each bind two molecules of cytochrome $c$ at low ionic strength. FEBS Lett. 589, 476-483.
3. Díaz-Moreno, I, Hulsker, R, Skubak, P, Foerster, JM, Cavazzini, D, Finiguerra, MG, Díaz-Quintana, A, Moreno-Beltrán, B, Rossi, G, Ullmann, GM, Pannu, NS, De la Rosa, MA and Ubbink, M. (2014) The dynamic complex of cytochrome $c_{6}$ and cytochrome $f$ studied with paramagnetic NMR spectroscopy. Biochim. Biophys. Acta Bioenergetics 1837, 1305-1315.
4. Hervás, M, Bashir, Q, Leferink, NG, Ferreira, P, Moreno-Beltrán, B, Westphal, AH, Díaz-Moreno, I, Medina, M, De la Rosa, MA, Ubbink, M, Navarro, JA and van Berkel, WJ. (2013) Communication between (L)-galactono-1,4-lactone dehydrogenase and cytochrome c. FEBS J. 280, 1830-1840.

## BOOK CHAPTER:

1. Moreno-Beltrán, $\mathbf{B}^{*}$, González-Arzola, $\mathrm{K}^{*}$, Martínez-Fábregas, J, Díaz-Moreno, I and De la Rosa, MA (2015). Cytochrome c-based signalosome. In Redox proteins in supercomplexes and signalosomes, Editors: R.O. Louro and I. Díaz-Moreno. Taylor and Francis Editorial Group. ISBN: 978-1-4822-5110-4.

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## Journal Paper

1. Moreno-Beltrán, B, Díaz-Quintana, A, González-Arzola, K, Velázquez-Campoy, A, De la Rosa, MA and Díaz-Moreno, I. (2014) Cytochrome $c_{1}$ exhibits two binding sites for cytochrome $c$ in plants. Biochim. Biophys. Acta - Bioenergetics 1837, 1717-1729.

## Journal Paper

2. Moreno-Beltrán, B, Díaz-Moreno, I, González-Arzola, K, GuerraCastellano, A, Velázquez-Campoy, A, De la Rosa, MA and DíazQuintana, A. (2015) Respiratory complexes III and IV can each bind two molecules of cytochrome $c$ at low ionic strength. FEBS Lett. 589, 476-483.

## Journal Paper

3. Díaz-Moreno, I, Hulsker, R, Skubak, P, Foerster, JM, Cavazzini, D, Finiguerra, MG, Díaz-Quintana, A, Moreno-Beltrán, B, Rossi, G, Ullmann, GM, Pannu, NS, De la Rosa, MA and Ubbink, M. (2014) The dynamic complex of cytochrome $c_{6}$ and cytochrome $f$ studied with paramagnetic NMR spectroscopy. Biochim. Biophys. Acta Bioenergetics 1837, 1305-1315.

PhD candidate contribution:
Ensemble docking computations and validation of PRE NMR data were performed by Blas Moreno-Beltrán in the laboratory of Prof. Dr. Marcellus Ubbink (Leiden University, Leiden, The Netherlands).

## Journal Paper

4. Hervás, M, Bashir, Q, Leferink, NG, Ferreira, P, Moreno-Beltrán, B, Westphal, AH, Díaz-Moreno, I, Medina, M, De la Rosa, MA, Ubbink, M, Navarro, JA and van Berkel, WJ. (2013) Communication between (L)-galactono-1,4-lactone dehydrogenase and cytochrome c. FEBS J. 280, 1830-1840.

PhD candidate contribution:
Expression and purification of the ${ }^{15} \mathrm{~N}$-labeled form of reduced plant cytochrome $c$ and the NMR assignment of its backbone amide resonances were performed by Blas Moreno-Beltrán in the Biointeractomics Unit (Universidad de Sevilla - CSIC, Sevilla, España). For this sequence-specific assignment, a $2 \mathrm{D}\left[{ }^{1} \mathrm{H},{ }^{15} \mathrm{~N}\right] \mathrm{HSQC}, 3 \mathrm{D}\left[{ }^{1} \mathrm{H}\right.$, $\left.{ }^{15} \mathrm{~N}\right]$ NOESY-HSQC and 3D $\left[{ }^{1} \mathrm{H},{ }^{15} \mathrm{~N}\right]$ TOCSY-HSQC spectra of ${ }^{15} \mathrm{~N}$ labeled form of reduced plant cytochrome $c$ were recorded at CERM (University of Florence, Italy). The form of Access to the Research Infrastructure was funded by the BioNMR project.

## Book Chapter

1. Moreno-Beltrán, $\mathbf{B}^{*}$, González-Arzola, $\mathrm{K}^{*}$, Martínez-Fábregas, J, Díaz-Moreno, I and De la Rosa, MA (2015). Cytochrome $c$-based signalosome. In Redox proteins in supercomplexes and signalosomes, Editors: R.O. Louro and I. Díaz-Moreno. Taylor and Francis Editorial Group. ISBN: 978-1-4822-5110-4.

* These authors have equally contributed.


## XI. APPENDIX II

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# Structure and Dynamics of the Y48pCMF Variant of Human Cytochrome $c$ 

This work was performed in collaboration with
Prof. Paola Turano
(CERM, University of Florence, Italy).

## 1. Introduction

Post-translational modifications (PTMs) of proteins, such as phosphorylation, nitration, glycosylation or acetylation, regulate a large number of processes involved in cell metabolism. PTMs are common mechanisms for controlling the metabolic role of proteins, e.g. activating or inactivating their function. Among all PTMs, reversible phosphorylation is the most frequently reported, being detected in both, homeostatic and stress-induced conditions (Khoury et al., 2011; Song et al., 2014). Phosphorylation allows rapid modification of protein function, adapting the protein to cellular changes. Kinases and phosphatases form part of the machinery responsible for the modulation of the reversible phosphorylation and dephosphorylation events respectively, which in turn are regulated by multiple signals (Corcoran and Cotter, 2013).

Human cytochrome $c(\mathrm{hCc})$ is a small, soluble and globular hemeprotein. Despite its small size, it comprises five well-known foldons or folding units (Krishna et al., 2003; Maity et al., 2005). hCc performs a double function in respiration and programmed cell death (PCD). Under homeostatic conditions, hCc shuttles electrons between respiratory complexes III and IV in the mitochondrial electron transport chain (DíazMoreno et al., 2011; Moreno-Beltrán et al., 2015a; Moreno-Beltrán et al., 2015b). However, under PCD stimuli, hCc is released from the mitochondria to the cytosol and even to the nucleus, acting as a PCD inductor (Díaz-Moreno et al., 2011; Martínez-Fábregas et al., 2014a; Martínez-Fábregas et al., 2014b). Both $\mathrm{hC} c$ functions are regulated by post-translational modifications, such as phosphorylation or nitration of tyrosine residues (García-Heredia et al., 2010; Diaz-Moreno et al., 2011; García-Heredia et al., 2011; García-Heredia et al., 2012; Hüttemann et
al., 2012a; Hüttemann et al., 2012b). Mammalian Cc can be phosphorylated in vivo in two different tyrosines located at positions 48 and 97 (Lee et al., 2006; Yu et al., 2008), and recently, novel phosphorylation sites have been reported at positions 28 and 47 (Zhao et al., 2011).
Notably, phosphorylation of mammalian $\mathrm{C} c$ at Tyr48 causes partial inhibition of the oxidative phosphorylation process. This PTM is associated with certain pathological situations, such as ischemia or reperfusion injury (Yu et al., 2008; Pecina et al., 2010; Hüttemann et al., 2012a). At the same time, experiments with a phosphomimetic mutant of $\mathrm{hC} c$ at Tyr48 (Y48E mutant) have suggested the functional relevance of this PTM on PCD (Pecina et al., 2010; García-Heredia et al., 2011). The possibility that Tyr48-phosphorylation of $\mathrm{hC} c$ regulates PCD has potentially important therapeutic implications for diseases like cancer, in which PCD is inhibited (Hüttemann et al., 2012b).
In general, tyrosine phosphorylation of $\mathrm{hC} c$ is mimicked by glutamic residues due to the technical difficulties of obtaining enough physiological phosphorylated protein from tissues and the fact that the specific Cc-phosphorylating kinase still remains unknown (Pecina et al., 2010; García-Heredia et al., 2011). However, the substitution of a tyrosine by glutamic results in a considerable decrease in the volume and surface of the replaced residue, whereas a tyrosine increases its volume upon phosphorylation.
To solve this problem, Tyr residues can be substituted by the noncanonical amino acid $p$-carboxymethyl-L-phenylalanine ( $p \mathrm{CMF}$ ), whose side-chain volume is closer to that of the phosphotyrosine residue (Xie et al., 2007) and whose negative-charge resembles the phosphate
group. Hence, we have replaced the Tyr48-encoding triplet of the hCc gene with an amber stop codon (TAG) to substitute this residue with the phosphorylation mimic compound using the evolved tRNA technique. This approach allows the selective incorporation of $p \mathrm{CMF}$ assisted by an orthogonal tRNA that recognizes the amber stop codon (Figure 1; Xie et al., 2007). The resulting phosphomimetic $\mathrm{hC} c$ at Tyr 48 , named $\mathrm{Y} 48 p \mathrm{CMF} \mathrm{hC} c$, contains a non-canonical $p \mathrm{CMF}$ amino acid at position 48 that better emulates the effect of phosphorylation in comparison with the traditional Tyr-by-Glu substitution.
Notably, the overlap of the $2 \mathrm{D}\left[{ }^{1} \mathrm{H},{ }^{15} \mathrm{~N}\right]$ HSQC spectra of uniformly ${ }^{15} \mathrm{~N}$-labeled forms of wild-type and $\mathrm{Y} 48 p \mathrm{CMF} \mathrm{hC} c$, acquired under the same experimental conditions, showed significant chemical shifts in many amide proton signals, as well as a higher number of exchangeable amide protons for the $\mathrm{Y} 48 \mathrm{pCMF} \mathrm{hC} c$ (Figure 2). The environmental changes of the amide protons evidenced a significant change in the overall structure of $\mathrm{Y} 48 p \mathrm{CMF} \mathrm{hC} c$, whereas the larger number of exchangeable amide protons pointed to differences in its dynamism.
We report here the NMR solution structure of $\mathrm{Y} 48 p \mathrm{CMF} \mathrm{hC} c$, as well as the NMR relaxation data-based dynamic properties of this phosphomimetic mutant with regards to the wild-type form. The ultimate aim of our research on $\mathrm{Y} 48 p \mathrm{CMF} \mathrm{hC} c$ has been to obtain evidence of structural and dynamical differences between this mutant and the wildtype form that can explain its distinct functional implications.

## 2. Materials and methods

### 2.1. Construct design

Site-directed mutagenesis was performed, using pBTR1 vector (Olteanu et al., 2003) as a template and the QuikChange II method (Stratagene), to replace the TAT triplet corresponding to Tyr 48 of $\mathrm{hC} c$ with the amber stop codon TAG. pBTR1 vector contains hCc gen (GenBank ID: M22877.1). The final stop codon of the $\mathrm{hC} c$ insert is the opal stop codon TGA. The primers for mutagenesis were Y48amber_fw (5-CTACAGCTAGACGGCGGCGA-3) and Y48amber_rv (5-TCGCCGCCGTCTAGCTGTAG-3). The resulting construct, named $\mathrm{pC} c$ Y48AMBER, was verified by automated sequencing.

### 2.2. Protein expression and purification

Uniformly ${ }^{13} \mathrm{C},{ }^{15} \mathrm{~N}$-labeled, ${ }^{15} \mathrm{~N}$-labeled and unlabeled samples of Y48pCMF hCc were expressed using the evolved tRNA-based strategy (Xie et al., 2007). E.coli BL21 (DE3) cells were co-transformed with the plasmids pBTR1-Y48AMBER and pEVOL/pCMF/tRNA. The latter vector encodes an orthogonal amber suppressor tRNA/aminoacyl-tRNA synthetase (aaRS) pair (Xie et al., 2007). The cells were grown for 20 h at 150 rpm and $30{ }^{\circ} \mathrm{C}$ in minimal media M9 supplemented with ampicillin and chloramphenicol and induced at OD 0.6 with $0.02 \%$ arabinose and 1 mM IPTG. ${ }^{15} \mathrm{~N}$-labeled ammonium chloride and ${ }^{13} \mathrm{C}$ glucose were added to minimal media to express the ${ }^{13} \mathrm{C},{ }^{15} \mathrm{~N}$-labeled samples, whereas only ${ }^{15} \mathrm{~N}$-labeled ammonium chloride was added to media for ${ }^{15} \mathrm{~N}$-labeled samples. The non-canonical amino acid $p \mathrm{CMF}$ and
$\delta$-aminolevulinic acid were immediately added after induction with IPTG at 1 mM and 0.1 mM final concentrations, respectively. The latter two compounds were unlabeled. Cells were collected by centrifugation, and then resuspended in 1.5 mM borate buffer pH 8.5 , supplemented with 1 mM phenylmethylsulfonyl fluoride (PMSF), $0.02 \mathrm{mg} \mathrm{mL}^{-1}$ DNase and $0.2 \mathrm{mg} \mathrm{mL}^{-1}$ lysozyme. The cytoplasmic fraction was obtained by sonication and centrifuged at $20,000 \mathrm{~g}$ for 15 min . Then, the supernatant was loaded onto a cationic exchange column. The purification protocol was performed as previously reported for wild-type hCc (Moreno-Beltrán et al., 2015a). Tryptic digestion and MALDI-TOF analyses confirmed the molecular mass and the tyrosine substitution by $p$ CMF. Protein concentration was determined by Vis spectrophotometry, using extinction coefficients of $29 \mathrm{mM}^{-1} \mathrm{~cm}^{-1}$ for reduced Y48pCMF hCc. Sodium ascorbate was used to reduce the sample. Pure fractions were dialyzed against 10 mM sodium phosphate pH 6.3 . Finally, protein samples were concentrated in Millipore 3 K Nominal Molecular Weight Limit (NMWL) centricons to a final concentration of 0.7 mM .

### 2.3. Circular Dichroism experiments

Circular dichroism (CD) spectra were recorded in the UV range (190250 nm ) at 298 K in a J-815 spectropolarimeter, equipped with a Peltier temperature control system, using a $1-\mathrm{mm}$ quartz cuvette. Samples contained $3 \mu \mathrm{M}$ protein in 5 mM sodium phosphate buffer ( pH 6.3 ). Twenty scans were averaged out for each sample. Secondary structure analysis was performed using CDPRO software (Sreerama and Woody, 2000; Sreerama and Woody, 2004). CLSTR was used as reference database.

### 2.4. NMR experiments

NMR spectra of fresh $0.6 \mathrm{mM}{ }^{13} \mathrm{C},{ }^{15} \mathrm{~N}$-labeled Y48pCMF hCc samples, in $90 \%$ buffer and $10 \% \mathrm{D}_{2} \mathrm{O}$, were recorded at 298 K on Bruker Avance spectrometers operating at 950,700 and $500 \mathrm{MHz}{ }^{1} \mathrm{H}$ frequencies. A standard set of triple resonance experiments, necessary for the full assignment of backbone and side chain resonances, were acquired at 700 $\mathrm{MHz}{ }^{1} \mathrm{H}$ frequency, whereas 2D and 3D NOESY experiments required for structure determination were acquired at $950 \mathrm{MHz}{ }^{1} \mathrm{H}$ frequency. Recorded NMR experiments for determination of backbone resonances were 2D $\left[{ }^{1} \mathrm{H},{ }^{15} \mathrm{~N}\right]$ HSQC, 2D $\left[{ }^{1} \mathrm{H},{ }^{13} \mathrm{C}\right]$ HSQC, 3D HNCA, 3D HNCACB, 3D CACB(CO)NH, 3D HN(CA)CO and 3D HNCO, whereas specific experiments for determination of side-chain resonances were 3D HCCHTOCSY and 3D HBHA(CO)NH (Grzesiek and Bax, 1993; Zuiderweg and Fesik, 1988; Kay et al., 1993). Additional 2D COSY and aromatic 2D $\left[{ }^{1} \mathrm{H},{ }^{13} \mathrm{C}\right] \mathrm{HSQC}$ on an unlabeled sample of the $\mathrm{Y} 48 p \mathrm{CMF}$ $\mathrm{hC} c$ were acquired for the assignment of aromatic residues. Therefore, the $\left[{ }^{1} \mathrm{H},{ }^{13} \mathrm{C}\right]$ HSQC spectrum was recorded in a natural abundance of ${ }^{13} \mathrm{C}$. Water suppression was achieved in all mentioned spectra by WATERGATE (Piotto et al., 1992). 1D ${ }^{1} \mathrm{H}$ spectra were launched before and after each spectrum to check the state of the sample, especially the redox state of the hemeprotein.
The following NOESY experiments were acquired: 2D $\left[{ }^{1} \mathrm{H},{ }^{15} \mathrm{~N}\right]$ NOESY, 3D $\left[{ }^{1} \mathrm{H},{ }^{15} \mathrm{~N}\right]$ NOESY-HSQC and 3D $\left[{ }^{1} \mathrm{H},{ }^{13} \mathrm{C}\right]$ NOESY-HSQC spectra in the aliphatic region (Zuiderweg and Fesik, 1989; Marion et al., 1989a; Marion et al., 1989b). Mixing times were 100 ms for all NOESY experiments, recorded on a ${ }^{15} \mathrm{~N}$-labeled sample, with the exception of the 3D $\left[{ }^{1} \mathrm{H},{ }^{13} \mathrm{C}\right]$ NOESY-HSQC that was acquired on a ${ }^{13} \mathrm{C},{ }^{15} \mathrm{~N}$-labeled
sample. An additional $2 \mathrm{D}\left[{ }^{1} \mathrm{H},{ }^{15} \mathrm{~N}\right]$ EXSY spectrum was launched for assignment of heme resonances, by using a partially oxidized ${ }^{15} \mathrm{~N}$-labeled sample. WATERGATE suppression was used. $1 \mathrm{D}{ }^{1} \mathrm{H}$ spectra were again launched to check the state of the samples.
${ }^{15} \mathrm{~N}$ relaxation $\mathrm{R}_{1}\left(=1 / \mathrm{T}_{1}\right), \mathrm{R}_{2}\left(=1 / \mathrm{T}_{2}\right)$ and $\left\{{ }^{1} \mathrm{H}\right\}-{ }^{15} \mathrm{~N}$ NOE parameters were obtained from standard experiments recorded at $500 \mathrm{MHz}{ }^{1} \mathrm{H}$ frequency and 298 K (Kay et al., 1989) on ${ }^{15} \mathrm{~N}$-labeled samples of the Y48pCMF and wild-type hCc species. $\left\{{ }^{1} \mathrm{H}\right\}-{ }^{-15} \mathrm{~N}$ NOE relaxation parameter is also known as HetNOE.
The NMR data processing was carried out using the Bruker Topspin software package. The assignments of 2D and 3D spectra were carried out manually with the help of the CARA and SPARKY software packages (Keller, 2004; Goddar and Kneller, SPARKY 3, University of California). A list of NOEs of the reduced yeast $\mathrm{C} c$ was used as a reference (Baistrocchi et al., 1996). Final reviews of peak assignments and integrations of peak volumes were carried out by XEASY (Bartels et al., 1995). ${ }^{15} \mathrm{~N}$ relaxation parameters were analyzed using CARA routines (Keller, 2004). $\mathrm{R}_{1} / \mathrm{R}_{2}$ ratios of residues in well-defined regions were used to estimate the rotational correlation times ( $\tau_{\mathrm{c}}$ ) of the protein constructs (Dosset et al., 2000).

### 2.5. Distance Geometry Calculations

The volumes of the cross-peaks between assigned resonances were obtained using the integration routines present in the program XEASY. Elliptical integration was applied. NOESY cross-peak intensities were converted into upper limits of inter-atomic distances by CYANA (Güntert et al., 1997). Upper and lower distance limits were imposed to
build up the heme. Upper $(1.90 \AA)$ and lower $(1.70 \AA)$ distance limits from the $\alpha$-carbons of thioethers 2 and 4 of the heme moiety, to the $S_{\gamma}$ of cysteines 14 and 17, respectively, were used in the computations to covalently link the heme moiety to the cysteine residues. An upper distance limit of $2.50 \AA$ and a lower distance limit of $2.20 \AA$ between the $\mathrm{S}_{\delta}$ of the Met80 and the iron of the heme were introduced, too. The orientation of Met80 and His18 side-chains was defined only by the experimental NOE constraints.
A residue containing the heme moiety was added to the standard CYANA library. In addition, the non-standard amino acid $p$ CMF was built and added to the CYANA library. Several cycles of the structure calculation were carried out in order to recalibrate the NOE distance constraints. Indeed, CYANA calculations were performed following the procedure and with the parameters used for the determination of other $c$ type cytochromes (Baistrocchi et al., 1996; Banci et al., 1997; Assfalg et al., 2002). Initially, 200 structures were calculated. In each calculation, violated constraints were analyzed for the best 20 structures with respect to the target function. After consecutive rounds of review and refinement of violated constraints, a final CYANA computation was performed in which no consistent violations were determined. The final value of the target function was equal to 0.73 .

### 2.6. Molecular Dynamics Simulations

NMR-restrained Molecular Dynamics (RMD) computations were performed by the AMBER 12.1 package and using the AMBER-2003 force field on a selection of the best 20 structures derived from the CYANA calculations (Duan et al., 2003; Case et al., 2006). Distance
constraints were introduced by the DIS_RST module of Amber 12.1. Simulations were performed under periodic boundary conditions using an orthorhombic cell geometry (the minimum distance between protein and cell faces was initially set to $10 \AA$ ) and PME electrostatics with a Ewald summation cut off of $9 \AA$. The structures were solvated with SPC water molecules, and $\mathrm{Cl}^{-}$counterions were added to neutralize the net charge of the full systems. Afterwards, solvent and counter-ions were subjected to 2500 steps of steepest descent minimization followed by 500 ps NPTMD computations using isotropic molecule position scaling and a pressure relaxation time of 2 ps at 298 K . Once the systems were NMRrestrained energy minimized (REM), these works were submitted to RMD computations for 5 ns at 298 K . Temperature was regulated using a Langevin thermostat with a collision frequency of $5 \mathrm{ps}^{-1}$ (Andersen, 1980). Finally, structures from RMD were energy minimized for 5000 steps. The SHAKE algorithm was used to constrain bonds involving hydrogen atoms (Ryckaert et al., 1977). The PTRAJ module of AMBER was used for trajectory analyses. Force field parameters for the heme group were taken from a previous work (Autenrieth et al., 2004). The Met80 residue was non-bonded to the iron and a constraint was applied instead. Molecular graphics were performed with UCSF Chimera (Pettersen et al., 2004). Validation of final minimized structures was performed by PSVS software (Bhattacharya et al., 2007). In addition, a final non-restrained MD computation of 20 ns was launched to check the stability of the resulting conformers.

## 3. Results and Discussion

### 3.1. Sequence-Specific Assignment and Secondary Structure Elements

We have performed an extensive assignment of the resonances of reduced $\mathrm{Y} 48 p \mathrm{CMF} \mathrm{hC} c$ by means of triple resonance NMR experiments. Backbone amide groups and their sequential NH-NH connectivities have been observed for most residues of Y48pCMF hCc. However, the amide resonances of 9 residues (Gly1, Glu21, Thr28, Thr49, Ala51, Gly56, Ile57, Lys79 and Gly84) could not be located. In addition, the ${ }^{15} \mathrm{NH}$ resonance of $p$ CMF48 was undetectable since the residue lacked labeling. Four prolines (Pro30, Pro44, Pro71 and Pro76) also broke the sequential NH-NH connectivities. Notably, the intensities of backbone amide groups of Asn31, Gly45 and Ser47 were significantly low in comparison with the rest of the signals. Triple resonance experiments were also critical to properly assign most side-chain protons. Additional two-dimensional experiments were carried out for the correct assignment of proton chemical-shifts from the heme group and from the side-chains of aromatic residues. Finally, a $96 \%$ completeness of the assignment of proton chemical shifts was achieved, and only 24 proton chemical-shifts remained missing.

The assignments of NOESY spectra, based on the previous proton chemical-shift assignment of backbone and side chain-resonances, allowed us to identify helical structures, which are characterized by strong $\mathrm{NH}-\mathrm{NH}$ and medium range $\mathrm{H} \alpha-\mathrm{NH}(\mathrm{i}, \mathrm{i}+3)$ and $\mathrm{H} \alpha-\mathrm{NH}(\mathrm{i}, \mathrm{i}+4)$ NOEs. The stretches Val3-Lys13, Ala50-Asn54, Glu61-Asn70, Lys72Ile75 and Lys88-Thr102 adopt helical structures. These segments, known
as $\alpha_{1}, \alpha_{2}, \alpha_{3}, \alpha_{4}$ and $\alpha_{5}$, are almost identical to those present in the NMR structure of wild-type reduced hCc (Jeng et al., 2002). These data are in agreement with the Circular Dichroism (CD) spectra, in which Y48pCMF hCc displayed a similar CD pattern and almost the same percentage of secondary structure elements in comparison with the wildtype form (Figure 3a,b).

### 3.2. NMR ${ }^{15}$ N-Relaxation Data-Based Dynamic Properties

Even though the secondary structure elements were conserved between both the wild-type and $\mathrm{Y} 48 p \mathrm{CMF} \mathrm{hC} c$ forms, $\mathrm{R}_{1}, \mathrm{R}_{2}$ and HetNOE relaxation measurements indicated that phosphomimetic $\mathrm{hC} c$ showed a highly dynamic behavior in distinct time scales, as compared to wild-type $\mathrm{hC} c$. In particular, $\mathrm{Y} 48 p \mathrm{CMF} \mathrm{hC} c$ contained a highly flexible area that included the mutation loop, whereas several segments of the protein were involved in conformational exchanges. Both features were also in agreement with CD data of the reduced $\mathrm{hC} c$ species, in which the percentage of disordered regions was higher in the $\mathrm{Y} 48 \mathrm{CMF} \mathrm{hC} c$ with regards to the wild-type form (Figure 3b).

First, the segment Gly41-Lys55, which contains the loop of the Tyr48 and corresponds to the foldon 5 of $\mathrm{hC} c$, shows a larger mobility in the Y48pCMF hCc than in the wild-type form, according to the drastic change of HetNOE values for this region (Figure 4a). These data are in consonance with the differences in R1 values of the stretch Tyr46-Lys55, confirming that the $p$ CMF-containing loop, along with the $\alpha_{2}$ helix, exhibited a high mobility in the $\mu \mathrm{s}$-ns time scale (Figure 4b). In addition, other residues, such as Lys27, Asp62, Lys72 and Ile85, showed a significant variation of the HetNOE values.

Second, the analyses of the differences between the $\mathrm{R}_{2}$ parameters among the wild-type and $\mathrm{Y} 48 \mathrm{pCMF} \mathrm{hC} c$ species evidenced the existence of three regions that exhibited conformational exchanges (Figure 4c). These regions were the following ones: His26-Thr28, Thr40-Trp59 and Ile75-Thr78. These data were consistent with a substantial decrease in the intensity of the signals and the existence of secondary conformations in mentioned segments. Indeed, such second species were identified in the stretches: Val20-Asn31 (foldon 2); Thr40-Trp59 (foldons 3 and 5) and Ile75-Glu90 (the omega loop of Met80, also known as foldon 4). Notably, residues His26, Pro30, Asn31, Tyr46, pCMF48, Trp59, Gly77, Thr78, Lys79 and Met80 showed multiple conformations.

Thus, the existence of new exchangeable amide protons in the phosphomimetic mutant ( $p$ CMF48, Thr49, Ala51, Ile57 and Lys79) could be ascribed to its highly dynamic behavior, mainly located in particular regions, such as the surroundings of the $p \mathrm{CMF}$ or the omega loop of the distal axial ligand. The overall analyses of the relaxation parameters determined that $\mathrm{R}_{1}$ and $\mathrm{R}_{2}$ parameters were also affected (Table 1). Remarkably, the rotational correlation time of the phosphomimetic mutant was higher than that of the wild-type form, evidencing not only specific local changes on the protein dynamics but also global changes in the protein motion as an entity upon phosphorylation (Table 1).

### 3.3. NMR solution structure

Experimental NOE constraints were transformed into upper distance limits to elucidate the solution structure of $\mathrm{Y} 48 p \mathrm{CMF} \mathrm{hC} c$. A total of 2,167 NOEs were introduced into CYANA (Table 2). The number of
experimental NOE constraints per residue was shown in Figure 5. It corresponds to an average of 20.8 experimental NOEs per residue. The heme group, the axial ligands and the two Cys residues covalently linked to the porphyrin were treated as in previous computations (Banci et al., 1995). Two hundred structures were initially calculated and twenty structures with the lowest target function were selected. The target function of the final computation was 0.73 . Selected structures were then energy minimized by the AMBER software package.
The overall fold of the molecule was maintained with respect to the wild-type $\mathrm{hC} c$, with the exception of the mutation-containing loop, which harbored the main differences in terms of protein folding and structure (Figure 6). The side-chain of $p$ CMF48 was exposed to the solvent and displayed multiple conformations, which suggests a highly dynamic behavior. Nevertheless, other regions of the $\mathrm{Y} 48 p \mathrm{CMF} \mathrm{hC} c$ structure were also significantly altered with regard to the wild-type $\mathrm{hC} c$, such as the 20 's loop (part of foldon 2), the 50 's helix ( $\alpha_{2}$; foldon 5) and the $\Omega$-loop of the Met80 (foldon 4).

Thus, the replacement of the Tyr48 by the $p$ CMF non-canonical amino acid might lead to a local destabilization of the mutation-containing loop and its surroundings, which explain the differences in the dynamic behavior and the worse definition of several segments. Actually, the structure of $\mathrm{Y} 48 p \mathrm{CMF} \mathrm{hC} c$ had a relatively high accuracy when the stretch of the mutation was excluded. The RMSD of the backbone to the mean was $0.87 \pm 0.16 \AA$ considering the whole primary sequence, whereas this value decreases to $0.53 \pm 0.12 \AA$ if the Thr 40 -Ile 57 segment is not considered. The RMSD values per residue were shown in Figure 7.

To the contrary, other regions of the NMR structure of $\mathrm{Y} 48 p \mathrm{CMF} \mathrm{hC} c$ exhibited high RMSD values. In particular the two segments Val20Gly29 and Thr40-Ile57. The high RMSD values found in the segment Val20-Gly29 have also been described in other cytochromes (Banci et al., 1995; Banci et al., 1997; Banci et al., 1999). This finding also correlates well with the existence of second conformations in this region, which were detected for His26 and Pro30 residues. A drastic decrease in the intensity of the amide group of Asn31 was also detected, which could be linked to the described phenomena. All mentioned residues are in direct contact with the mutation-containing loop, which is spatially close.

Residues at the Thr40-Ile57 stretch, which include the loop of $p$ CMF48 and $\alpha_{2}$, were strongly affected by two distinct phenomena: a drastic reduction of the NOE cross-peaks and the existence of conformational exchanges. Thus, the high RMSD values of backbone and heavy atoms, which are the highest values considering the whole primary sequence (Figure 7), perfectly match these two findings. The conformational exchanges in this area were first evidenced by high R2 values and then confirmed by direct evaluation of NOESY spectra. Double conformations were observed for Tyr46, pCMF48 and Trp59 in the phosphomimetic mutant. In addition, the loss of amide proton signals or the severe decrease in their intensities were specifically detected for Ser47, $p$ CMF48, Thr49, Ala51, Gly56 and Ile57 of $p$ CMF48 hC $c$ in comparison with the wild-type form (Jeng et al., 2002).

Finally, the $\Omega$-loop also encloses some residues with high RMSD values. The high values of the segment Lys72-Met80 could be attributed to the same findings explained previously: a significant decrease in NOE cross-peaks and conformational exchanges. Notably, the high $\mathrm{R}_{2}$ values
of residues placed at the beginning of the $\Omega$-loop (Ile75-Thr78) matched their second conformation. A second conformation was clearly identified for the Pro76-Met80 segment. Moreover, the high RMSD values of the heavy atoms at the end of this $\Omega$-loop have also been described in other cytochromes (Banci et al., 1995; Banci et al., 1997; Baistrocchi et al., 1996).

## 4. Conclusions

In summary, the structural model of $\mathrm{Y} 48 p \mathrm{CMF} \mathrm{hC} c$ herein present is in agreement with relaxation measurements and highlights how the phosphomimetic $\mathrm{C} c$ becomes more dynamic than the wild-type species, with potential consequences for protein function. In particular, relevant structural and dynamic changes in the $p$ CMF-containing loop and its surroundings, along with the $\Omega$-loop of the distal axial ligand, may have direct implications in the two $\mathrm{C} c$ functions, aerobic respiration and PCD.

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Figure 1. Comparison between the molecular pathways that incorporate the canonical amino acid Tyr (left panel) and the non-canonical $p \mathrm{CMF}$ residue (right panel) during protein synthesis. The biological inclusion of $p$ CMF is achieved by using an evolved suppressor tRNA $_{\text {AUC }}$ ( $p \mathrm{CMF}$ tRNA) that acts as an AMBER suppressor tRNA along with an orthogonal aminoacyl-tRNA synthetase. The Figure shows the punctual mutation of a specific Tyr encoding mRNA triplet (UAC) by the AMBER stop codon (UAG).


Figure 2. Superimposition of the $\left[{ }^{1} \mathrm{H},{ }^{15} \mathrm{~N}\right]$ HSQC spectra of uniformly ${ }^{15} \mathrm{~N}$-labeled forms of the wild-type and $\mathrm{Y} 48 p \mathrm{CMF} \mathrm{hC} c$ species. Wildtype $\mathrm{hC} c$ is colored in blue, whereas $\mathrm{Y} 48 p \mathrm{CMF} \mathrm{hC} c$ in red. Backbone amide resonances of wild-type hCc are displayed (Jeng et al., 2002).


Figure 3. Comparison of CD spectra and secondary structure elements between the reduced forms of the wild-type and $\mathrm{Y} 48 p \mathrm{CMF} \mathrm{hC} c$ species.
a) Superimposition of the Far-UV spectra. Wild-type hCc is colored in blue, whereas $\mathrm{Y} 48 p \mathrm{CMF} \mathrm{hC} c$ in red. b) Analysis of the diverse secondary structure elements for the wild-type and Y48pCMF hCc species by CD Pro Software Package.


Figure 4. The differences between the experimental values of the relaxation rates $\mathrm{R}_{1}$ (a), $\mathrm{R}_{2}$ (b) and NOE (c) between the reduced forms of the wild-type and $\mathrm{Y} 48 p \mathrm{CMF} \mathrm{hC} c$ species, plotted as a function of the residue number. Gaps in the data result from overlapping resonances, broadened resonances beyond the detection limit and unassigned resonances. More affected regions are highlighted.


Figure 5. Number of experimental meaningful NOEs per residue used for the structure calculation of $\mathrm{Y} 48 p \mathrm{CMF} \mathrm{hC} c$. White bars indicate intraresidue NOEs; light gray, sequential NOEs; dark gray and black, medium and long range NOEs, respectively. Residue 105 corresponds to the heme group.


Figure 6. NMR solution structure of $\mathrm{Y} 48 p \mathrm{CMF} \mathrm{hC}$ c. a) Ribbon representation of the best 20 conformers of $\mathrm{Y} 48 p \mathrm{CMF} \mathrm{hC} c$. The heme group and the side chain of the pCMF 48 is shown for all conformers. Ribbons are colored in red and atoms from the heme group and $p$ CMF by CPK color scheme. The five helices are indicated, as well as the mutation-containing loop. b) A close-up of the mutation-containing loop. The non-canonical $p$ CMF48 adopts multiple orientations. C) Comparison between the NMR solution structures of the wild-type (Jeng et al., 2002) and $\mathrm{Y} 48 p \mathrm{CMF} \mathrm{hC} c$ species. The ribbon of the wild-type form is colored in blue. Arrows indicate the regions with substantial structural changes. Protein structures were represented by Chimera (Pettersen et al., 2004).


Figure 7. The RMSD values per residue of the best 20 conformers of the $\mathrm{Y} 48 p \mathrm{CMF} \mathrm{hC} c$ mutant to the mean structure. Backbone atoms are represented as gray rhombs and heavy atoms, as black circles. The RMSD calculations and the generation of the mean structure were performed by Molmol (Koradi et al., 1996).

Table 1. Comprehensive ${ }^{15} \mathrm{~N}$-relaxation parameters of the wild-type and $\mathrm{Y} 48 p \mathrm{CMF} \mathrm{hC} c$ species.

|  | wild-type <br> $\mathbf{h C} \boldsymbol{c}$ | Y48pCMF hC $\boldsymbol{c}$ | Difference |
| :---: | :---: | :---: | :---: |
| $\mathrm{R}_{1}\left(\mathrm{~s}^{-1}\right)$ | $2.26 \pm 0.08$ | $2.15 \pm 0.07$ | 0.11 |
| $\mathrm{R}_{2}\left(\mathrm{~s}^{-1}\right)$ | $8.53 \pm 0.46$ | $9.14 \pm 0.48$ | 0.61 |
| $\mathrm{R}_{2} / \mathrm{R}_{1}\left(\mathrm{~s}^{-1}\right)$ | $3.77 \pm 0.23$ | $4.25 \pm 0.23$ | 0.48 |
| NOE ratio $^{0.80} \pm 0.03$ | $0.80 \pm 0.02$ | 0.00 |  |
| $\tau_{\mathrm{c}}(\mathrm{ns})$ | $6.33 \pm 0.28$ | $6.89 \pm 0.25$ | 0.56 |

Table 2. NMR statistics of the Y48pCMF hCc structure

NMR distance and dihedral constraints

| Distance constraints | 2,176 |
| :--- | :---: |
| Total meaningful NOE | 362 |
| Intra-residue | 1,814 |
| Inter-residue | 769 |
| Sequential $(\|i-j\|=1)$ | 562 |
| Medium-range $(\|i-j\| \leq 5)$ | 483 |
| Long-range $(\|i-j\|>5)$ |  |
| Total dihedral angle restraints | 71 |
| $\phi$ | 71 |
| $\psi$ |  |

## Structure statistics

| Violations |  |
| :--- | :---: |
| Target function | $0.73 \pm 0.18$ |
| RMSD of upper limits $(\AA)$ | $0.0059 \pm 0.0014$ |
| RMSD of torsion angles $\left({ }^{\circ}\right)$ | $0.3336 \pm 0.0973$ |
| Max. distance constraint violation $(\AA)$ | 3.79 |
| Max. dihedral angle violation $\left({ }^{\circ}\right)$ |  |
| RMSD of minimized 20 conformers to the |  |
| mean $(\AA)$ | $0.87 \pm 0.16$ |
| Backbone | $1.33 \pm 0.24$ |
| Heavy atoms |  |

## XII. APPENDIX III

## Dataset

1. Moreno-Beltrán, B et al. (2012) NMR Assignment of Arabidopsis thaliana cytochrome $c$ in its reduced state. BMRB Entry 18828 (Released).
```
save_assigned_chem_shift_list_1
    _\overline{A}signed_\overline{chem_\overline{shift_\}\ist.\overline{S}f_category assigned_chemical_shifts}
    Assigned_chem_shift_list.Sf_framecode assigned_chem_shift_list_1
    _-Assigned_chem_shift_list.EnĒry_ID 18828
    _Assigned_chem_shift_list.ID
1
    _Assigned_chem_shift_list.Sample_condition_list_ID 1
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stop_
save_

## Dataset

2. Moreno-Beltrán, B et al. (2015) NMR Assignment of Homo sapiens cytochrome $c$ in its oxidized state. BMRB Entry 26578 (Entry on Hold).
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|  | 106 | 1 | 1 | 54 | 54 | ASN | HD21 | H | 1 | 6.897 | 0.001 |  | 1 |
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| 54 | ASN | HD21 | . | . | . . | . | . . | . |  | . . . | 26578 | 1 |  |
|  | 107 | 1 | 1 | 54 | 54 | ASN | HD22 | H | 1 | 7.427 | 0.001 |  | 1 |
| 54 | ASN | HD22 | . | . | . . | . | . . | . | . | . . . | 26578 | 1 |  |
|  | 108 | 1 | 1 | 54 | 54 | ASN | N | N | 15 | 113.081 | 0.001 |  | 1 |
| 54 | ASN | N | . | . | . . | . | . . | . | . | . . . | 26578 | 1 |  |
|  | 109 | 1 | 1 | 54 | 54 | ASN | ND2 | N | 15 | 112.161 | 0.001 |  | 1 |
| 54 | ASN | ND2 | . | . | . . | . | . . | . | . | . . . | 26578 | 1 |  |
|  | 110 | 1 | 1 | 55 | 55 | LYS | H | H | 1 | 6.987 | 0.001 |  | 1 |
| 55 | LYS | H | . | . | . . | . | . . | . | . | . . . | 26578 | 1 |  |
|  | 111 | . 1 | 1 | 55 | 55 | LYS | N | N | 15 | 121.530 | 0.001 |  | 1 |
| 55 | LYS | N | . | . | . . | . | . . | . | . | . . . | 26578 | 1 |  |
|  | 112 | 1 | 1 | 56 | 56 | GLY | H | H | 1 | 7.752 | 0.001 |  | 1 |
| 56 | GLY | H | . | . | . . | . | . . | . | . | . . . | 26578 | 1 |  |
|  | 113 | 1 | 1 | 56 | 56 | GLY | N | N | 15 | 103.167 | 0.001 |  | 1 |
| 56 | GLY | N |  | . | . . | . | . . | . | . | . . . | 26578 | 1 |  |
|  | 114 | . 1 | 1 | 57 | 57 | ILE | H | H | 1 | 6.463 | 0.001 |  | 1 |
| 57 | ILE | H |  | . | . . | . | . . | . |  | . . . | 26578 | 1 |  |
|  | 115 | . 1 | 1 | 57 | 57 | ILE | N | N | 15 | 110.660 | 0.001 |  | 1 |
| 57 | ILE | N | . | . | . . | . | . . | . | . | . . . | 26578 | 1 |  |
|  | 116 | . 1 | 1 | 58 | 58 | ILE | H | H | 1 | 8.164 | 0.001 |  | 1 |
| 58 | ILE | H | . | . | . . | . | . . | . | . | . . . | 26578 | 1 |  |
|  | 117 | . 1 | 1 | 58 | 58 | ILE | N | N | 15 | 119.269 | 0.001 |  | 1 |
| 58 | ILE | N | . | . | . . | . | . . | . | . | . | 26578 | 1 |  |
|  | 118 | . 1 | 1 | 59 | 59 | TRP | H | H | 1 | 8.643 | 0.001 |  | 1 |
| 59 | TRP | H | . | . | . . | . | . . | . | . | . . . | 26578 | 1 |  |
|  | 119 | . 1 | 1 | 59 | 59 | TRP | HE1 | H | 1 | 9.497 | 0.001 |  | 1 |
| 59 | TRP | HE1 |  | . | . . | . | . . | . |  | . . . | 26578 | 1 |  |
|  | 120 | . 1 | 1 | 59 | 59 | TRP | N | N | 15 | 130.596 | 0.001 |  | 1 |
| 59 | TRP | N | . | . | . . | . | . . | . |  | . . . | 26578 | 1 |  |
|  | 121 | . 1 | 1 | 59 | 59 | TRP | NE1 | N | 15 | 125.473 | 0.001 |  | 1 |
| 59 | TRP | NE1 | . | . | . . | . | . . | . |  | . . . | 26578 | 1 |  |
|  | 122 | . 1 | 1 | 60 | 60 | GLY | H | H | 1 | 7.682 | 0.001 |  | 1 |
| 60 | GLY | H | . | . | . . | . | . . | . | . | . . . | 26578 | 1 |  |
|  | 123 | . 1 | 1 | 60 | 60 | GLY | N | N | 15 | 111.288 | 0.001 |  | 1 |
| 60 | GLY | N | . | . | . . | . | . . | . | . | . . . | 26578 | 1 |  |
|  | 124 | . 1 | 1 | 61 | 61 | GLU | H | H | 1 | 9.540 | 0.001 |  | 1 |
| 61 | GLU | H | . | . | . . | . | . . | . | . | . . . | 26578 | 1 |  |
|  | 125 | . 1 | 1 | 61 | 61 | GLU | N | N | 15 | 120.579 | 0.001 |  | 1 |
| 61 | GLU | N | . | . | . . | . | . . | . | . | . . . | 26578 | 1 |  |
|  | 126 | . 1 | 1 | 62 | 62 | ASP | H | H | 1 | 8.335 | 0.001 |  | 1 |
| 62 | ASP | H | . | . | . . | . | . . | . | . | . . . | 26578 | 1 |  |
|  | 127 | . 1 | 1 | 62 | 62 | ASP | N | N | 15 | 115.286 | 0.001 |  | 1 |
| 62 | ASP | N | . | . | . . | . | . . | . | . | . . . | 26578 | 1 |  |
|  | 128 | . 1 | 1 | 63 | 63 | THR | H | H | 1 | 8.227 | 0.001 |  | 1 |
| 63 | THR | H | . | . | . . | . | . . | . | . | . . | 26578 | 1 |  |
|  | 129 | . 1 | 1 | 63 | 63 | THR | N | N | 15 | 113.073 | 0.001 |  | 1 |
| 63 | THR | N | . | . | . . | . | . . | . | . | . | 26578 | 1 |  |
|  | 130 | . 1 | 1 | 64 | 64 | LEU | H | H | 1 | 8.513 | 0.001 |  | 1 |
| 64 | LEU | H | . | . | . | . | . . | . | . | . ${ }^{\text {. }}$ | 26578 | 1 |  |
|  | 131 | . 1 | 1 | 64 | 64 | LEU | N | N | 15 | 120.582 | 0.001 |  | 1 |
| 64 | LEU | N | . | . | . . | . | . . | . | . | . | 26578 | 1 |  |
|  | 132 | . 1 | 1 | 65 | 65 | MET | H | H | 1 | 7.668 | 0.001 |  | 1 |
| 65 | MET | H | . | . | . | . | . . | . | . | . . | 26578 | 1 |  |
|  | 133 | . 1 | 1 | 65 | 65 | MET | N | N | 15 | 118.953 | 0.001 |  | 1 |
| 65 | MET | N | . | . | . . | . | . . | . | . | . ${ }^{\text {a }}$ | 26578 | 1 |  |
|  | 134 | . 1 | 1 | 66 | 66 | GLU | H | H | 1 | 6.796 | 0.001 |  | 1 |
| 66 | GLU | H | . |  | . . | . | . . | . | . |  | 26578 | 1 |  |
|  | 135 | . 1 | 1 | 66 | 66 | GLU | N | N | 15 | 117.722 | 0.001 | . | 1 |
| 66 | GLU | N |  | . | . . | . | . . | . | . | . . | 26578 | 1 |  |
|  | 136 | . 1 | 1 | 67 | 67 | TYR | H | H | 1 | 8.099 | 0.001 |  | 1 |
| 67 | TYR | H |  | . | . . | . | . . | . | . | . | 26578 | 1 |  |
|  | 137 | . 1 | 1 | 67 | 67 | TYR | N | N | 15 | 121.308 | 0.001 |  | 1 |
| 67 | TYR | N | . | . | . . | . | . . | . | . | . | 26578 | 1 |  |
|  | 138 | . 1 | 1 | 68 | 68 | LEU | H | H | 1 | 8.007 | 0.001 |  | 1 |
| 68 | LEU | H | . | . | - | . | . . | . | . | . . . | 26578 | 1 |  |
|  | 139 | 1 | 1 | 68 | 68 | LEU | N | N | 15 | 110.992 | 0.001 | . | 1 |
| 68 | LEU | N | . | . | . | . | . . | . | . | . | 26578 | 1 |  |
|  | 140 | . 1 | 1 | 69 | 69 | GLU | H | H | 1 | 6.747 | 0.001 |  | 1 |
| 69 | GLU | H | . | . | . . | . | . . | . | . | . . . | 26578 | 1 |  |
|  | 141 | 1 | 1 | 69 | 69 | GLU | N | N | 15 | 118.836 | 0.001 |  | 1 |
| 69 | GLU | N | . | . | - | . | . . | . | . | - | 26578 | 1 |  |
|  | 142 | . 1 | 1 | 70 | 70 | ASN | H | H | 1 | 6.638 | 0.001 |  | 1 |
| 70 | ASN | H | . | . | . | . | . . | . |  |  | 26578 | 1 |  |
|  | 143 | . 1 | 1 | 70 | 70 | ASN | HD21 | H | 1 | 6.904 | 0.001 |  | 1 |
| 70 | ASN | HD21 | . | . | - | . | . . | . | . | . . . | 26578 | 1 |  |
|  | 144 | . 1 | 1 | 70 | 70 | ASN | HD22 | H | 1 | 7.881 | 0.001 |  | 1 |
| 70 | ASN | HD22 | . | . | - | . | . . | . | . | . . . | 26578 | 1 |  |
|  | 145 | . 1 | 1 | 70 | 70 | ASN | N | N | 15 | 105.368 | 0.001 |  | 1 |
| 70 | ASN | N | . | . | . | . | . . | . | . | . | 26578 | 1 |  |
|  | 146 | . 1 | 1 | 70 | 70 | ASN | ND2 | N | 15 | 112.340 | 0.001 |  | 1 |
| 70 | ASN | ND2 | . | . | . . | . | . . | . | . | . . . | 26578 | 1 |  |


|  | 147 | . | 1 | 1 | 72 | 72 | LYS | H |  | H | 1 | 9.327 | 0.001 | . | 1 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 72 | LYS | H | H |  |  | . . | . | . | . |  |  |  | 26578 | 1 |  |
|  | 148 | . | 1 | 1 | 72 | 72 | LYS | N |  | N | 15 | 115.751 | 0.001 |  | 1 |
| 72 | LYS | N | N |  |  | . . | . | . | . | . |  | . . . | 26578 | 1 |  |
|  | 149 | . | 1 | 1 | 73 | 73 | LYS | H |  | H | 1 | 7.773 | 0.001 |  | 1 |
| 73 | LYS | H | H |  | . | . . | . | . | . | . | . | . . . | 26578 | 1 |  |
|  | 150 | . | 1 | 1 | 73 | 73 | LYS | N |  | N | 15 | 119.376 | 0.001 | . | 1 |
| 73 | LYS | N | N |  | . | . . | . | . | - | . |  | . . . | 26578 | 1 |  |
|  | 151 | . | 1 | 1 | 74 | 74 | TYR | H |  | H | 1 | 8.089 | 0.001 | . | 1 |
| 74 | TYR | H | H | . | . | . . | . | . | . | . | . | . . . | 26578 | 1 |  |
|  | 152 | . | 1 | 1 | 74 | 74 | TYR | N |  | N | 15 | 120.499 | 0.001 | . | 1 |
| 74 | TYR | N | N | . | . | . . | . | . | . | . | . | . . . | 26578 | 1 |  |
|  | 153 | . | 1 | 1 | 75 | 75 | ILE | H |  | H | 1 | 9.377 | 0.001 | . | 1 |
| 75 | ILE | H | H | . | . | . . | . | . | - | . | . | . . . | 26578 | 1 |  |
|  | 154 | . | 1 | 1 | 75 | 75 | ILE | N |  | N | 15 | 114.999 | 0.001 | . | 1 |
| 75 | ILE | N | N | . | . | . . | . | . | . | . | . | . . . | 26578 | 1 |  |
|  | 155 | . | 1 | 1 | 77 | 77 | GLY | H |  | H | 1 | 9.140 | 0.001 | . | 1 |
| 77 | GLY | H | H | . | . | . . | . | . | . | . | . | . . . | 26578 | 1 |  |
|  | 156 | . | 1 | 1 | 77 | 77 | GLY | N |  | N | 15 | 112.120 | 0.001 | . | 1 |
| 77 | GLY | N | N | . | . | . . | . | . | . | . | . | . . . | 26578 | 1 |  |
|  | 157 | . | 1 | 1 | 78 | 78 | THR | H |  | H | 1 | 9.001 | 0.001 | . | 1 |
| 78 | THR | H | H | . | . | . . | . | . | . | . | . | . . . | 26578 | 1 |  |
|  | 158 | . | 1 | 1 | 78 | 78 | THR | N |  | N | 15 | 115.318 | 0.001 | . | 1 |
| 78 | THR | N | N | . | . | . . | . | . | - | . | . | . . . | 26578 | 1 |  |
|  | 159 | . | 1 | 1 | 79 | 79 | LYS | H |  | H | 1 | 8.233 | 0.001 | . | 1 |
| 79 | LYS | H | H | . | . | . . | . | . | . | . | . |  | 26578 | 1 |  |
|  | 160 | . | 1 | 1 | 79 | 79 | LYS | N |  | N | 15 | 123.039 | 0.001 | . | 1 |
| 79 | LYS | N | N | . | . | . . | . | . | - | . | . | . . . | 26578 | 1 |  |
|  | 161 | . | 1 | 1 | 80 | 80 | MET | H |  | H | 1 | 9.174 | 0.001 | . | 1 |
| 80 | MET | H | H | . | . | . . | . | . | . | . | . | . . . | 26578 | 1 |  |
|  | 162 | . | 1 | 1 | 80 | 80 | MET | N |  | N | 15 | 123.254 | 0.001 | . | 1 |
| 80 | MET | N | N | . | . | . . | . | . | . | . | . | . . . | 26578 | 1 |  |
|  | 163 | . | 1 | 1 | 81 | 81 | ILE | H |  | H | 1 | 8.440 | 0.001 | . | 1 |
| 81 | ILE | H | H | . | . | . . | . | . | . | . | . | . . . | 26578 | 1 |  |
|  | 164 | . | 1 | 1 | 81 | 81 | ILE | N |  | N | 15 | 135.513 | 0.001 | . | 1 |
| 81 | ILE | N | N | . | . | . . | . | . | . | . | . | . . . | 26578 | 1 |  |
|  | 165 | . | 1 | 1 | 82 | 82 | PHE | H |  | H | 1 | 9.031 | 0.001 | . | 1 |
| 82 | PHE | H | H | . | . | . . | . | . | . | . | . | . . . | 26578 | 1 |  |
|  | 166 | . | 1 | 1 | 82 | 82 | PHE | N |  | N | 15 | 126.565 | 0.001 | . | 1 |
| 82 | PHE | N | N |  | . | . . | . | . | . | . | . | . . . | 26578 | 1 |  |
|  | 167 | . | 1 | 1 | 83 | 83 | VAL | H |  | H | 1 | 8.003 | 0.001 | . | 1 |
| 83 | VAL | H | H | . | . | . . | . | . | . | . | . | . . . | 26578 | 1 |  |
|  | 168 | . | 1 | 1 | 83 | 83 | VAL | N |  | N | 15 | 127.001 | 0.001 | . | 1 |
| 83 | VAL | N | N | . | . | . . | . | . | . | . | . | . | 26578 | 1 |  |
|  | 169 | . | 1 | 1 | 85 | 85 | ILE | H |  | H | 1 | 7.870 | 0.001 | . | 1 |
| 85 | ILE | H | H |  | . | . . | . | . | . | . | . | . | 26578 | 1 |  |
|  | 170 |  | 1 | 1 | 85 | 85 | ILE | N |  | N | 15 | 120.180 | 0.001 | . | 1 |
| 85 | ILE | N | N |  |  | . . |  | . | . | . | . | . . . | 26578 | 1 |  |
|  | 171 |  | 1 | 1 | 86 | 86 | LYS | H |  | H | 1 | 8.232 | 0.001 |  | 1 |
| 86 | LYS | H | H | . |  | . . |  |  | . | . |  | . . . | 26578 | 1 |  |
|  | 172 |  | 1 | 1 | 86 | 86 | LYS | N |  | N | 15 | 127.669 | 0.001 | . | 1 |
| 86 | LYS | N | N | . | . | . . |  |  | . | . | . | . . . | 26578 | 1 |  |
|  | 173 |  | 1 | 1 | 87 | 87 | LYS | H |  | H | 1 | 8.188 | 0.001 | . | 1 |
| 87 | LYS | H | H | . | . | . . | . | . | . | . | . | . . . | 26578 | 1 |  |
|  | 174 |  | 1 | 1 | 87 | 87 | LYS | N |  | N | 15 | 120.154 | 0.001 | . | 1 |
| 87 | LYS | N | N | . |  | . . |  | . | . | . | . |  | 26578 | 1 |  |
|  | 175 |  | 1 | 1 | 88 | 88 | LYS | H |  | H | 1 | 8.734 | 0.001 | . | 1 |
| 88 | LYS | H | H | . |  | . . |  | . | - | . | . |  | 26578 | 1 |  |
|  | 176 |  | 1 | 1 | 88 | 88 | LYS | N |  | N | 15 | 129.999 | 0.001 | . | 1 |
| 88 | LYS | N | N | . |  | . . | . | . | . | . |  |  | 26578 | 1 |  |
|  | 177 | . | 1 | 1 | 89 | 89 | GLU | H |  | H | 1 | 9.281 | 0.001 | . | 1 |
| 89 | GLU | H | H | . |  | - | . | . | - | . | . |  | 26578 | 1 |  |
|  | 178 | . | 1 | 1 | 89 | 89 | GLU | N |  | N | 15 | 117.939 | 0.001 | . | 1 |
| 89 | GLU | N | N | . | . | - | . | . | . | . | . |  | 26578 | 1 |  |
|  | 179 | . | 1 | 1 | 90 | 90 | GLU | H |  | H | 1 | 6.175 | 0.001 | . | 1 |
| 90 | GLU | H | H | . | . | - | . | . | - | . | . | . . | 26578 | 1 |  |
|  | 180 | - | 1 | 1 | 90 | 90 | GLU | N |  | N | 15 | 116.411 | 0.001 | . | 1 |
| 90 | GLU | N | N | . | . | - | . | . | - | . |  | . . | 26578 | 1 |  |
|  | 181 | . | 1 | 1 | 91 | 91 | ARG | H |  | H | 1 | 7.136 | 0.001 | . | 1 |
| 91 | ARG | H | H | . | . | - | . | . | . | . | . | . ${ }^{\text {a }}$ | 26578 | 1 |  |
|  | 182 |  | 1 | 1 | 91 | 91 | ARG | N |  | N | 15 | 117.033 | 0.001 | . | 1 |
| 91 | ARG | N | N | . | . | - | . | . | . | . |  | . . | 26578 | 1 |  |
|  | 183 | - | 1 | 1 | 92 | 92 | ALA | H |  | H | 1 | 8.350 | 0.001 | . | 1 |
| 92 | ALA | H | H | . | . | . . | . | . | - | . |  | . . | 26578 | 1 |  |
|  | 184 |  | 1 | 1 | 92 | 92 | ALA | N |  | N | 15 | 119.991 | 0.001 | . | 1 |
| 92 | ALA | N | N | . | . | - | . | . | . | . |  | . . | 26578 | 1 |  |
|  | 185 |  | 1 | 1 | 93 | 93 | ASP | H |  | H | 1 | 8.034 | 0.001 | . | 1 |
| 93 | ASP | H | H |  | . | . . | . |  | - | . | . | . | 26578 | 1 |  |
|  | 186 |  | 1 | 1 | 93 | 93 | ASP | N |  | N | 15 | 122.088 | 0.001 |  | 1 |
| 93 | ASP | N | N |  |  | . | . |  | . | . |  |  | 26578 | 1 |  |
|  | 187 | . | 1 | 1 | 94 | 94 | LEU | H |  | H | 1 | 7.783 | 0.001 |  | 1 |
| 94 | LEU | H | H | . | . | . . | . | . | - | . | . | . . . | 26578 | 1 |  |



## Dataset

3. Moreno-Beltrán, B et al. (2015). NMR Assignment of the Y48pCMF variant of Homo sapiens cytochrome $c$ in its reduced state. BMRB and PDB entries (in preparation).






|  |
| :---: |
|  |
|  |
|  |
|  |
|  |



| 1354 | 0.896 | 0.002 | QG2 | 102 | THR |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 1355 | 2.705 | 0.000 | HB2 | 62 | ASP |
| 1356 | 2.626 | 0.000 | HB3 | 62 | ASP |
| 1357 | 21.577 | 0.000 | CG2 | 3 | VAL |
| 1358 | 21.697 | 0.000 | CG1 | 3 | VAL |
| 1359 | 0.988 | 0.000 | QG1 | 3 | VAL |
| 1360 | 0.954 | 0.000 | QG2 | 3 | VAL |
| 1361 | 36.428 | 0.000 | CG | 89 | GLU |
| 1363 | 37.068 | 0.000 | CG | 61 | GLU |
| 1364 | 2.376 | 0.003 | HG2 | 61 | GLU |
| 1365 | 2.306 | 0.001 | HG3 | 61 | GLU |
| 1366 | 2.320 | 0.001 | HG2 | 89 | GLU |
| 1367 | 2.253 | 0.001 | HG3 | 89 | GLU |
| 1368 | 35.918 | 0.000 | CG | 4 | GLU |
| 1369 | 2.053 | 0.001 | HB2 | 4 | GLU |
| 1370 | 2.021 | 0.002 | HB3 | 4 | GLU |
| 1371 | 2.231 | 0.001 | HG2 | 4 | GLU |
| 1372 | 2.208 | 0.002 | HG3 | 4 | GLU |
| 1373 | 36.943 | 0.000 | CG | 90 | GLU |
| 1374 | 1.999 | 0.002 | HB2 | 90 | GLU |
| 1375 | 1.919 | 0.001 | HB3 | 90 | GLU |
| 1376 | 35.887 | 0.000 | CG | 66 | GLU |
| 1377 | 1.874 | 0.002 | HG2 | 66 | GLU |
| 1378 | 1.853 | 0.002 | HG3 | 66 | GLU |
| 1379 | 1.590 | 0.001 | HB2 | 66 | GLU |
| 1380 | 1.453 | 0.002 | HB3 | 66 | GLU |
| 1381 | 26.036 | 0.041 | CG1 | 81 | ILE |
| 1382 | 16.961 | 0.000 | CG2 | 81 | ILE |
| 1383 | 10.665 | 0.000 | CD1 | 81 | ILE |
| 1384 | 1.118 | 0.001 | HG13 | 81 | ILE |
| 1385 | 1.348 | 0.005 | HG12 | 81 | ILE |
| 1386 | 0.667 | 0.001 | QG2 | 81 | ILE |
| 1387 | 0.618 | 0.001 | QD1 | 81 | ILE |
| 1388 | 14.288 | 0.000 | CD1 | 95 | ILE |
| 1389 | 17.883 | 0.000 | CG2 | 95 | ILE |
| 1390 | 31.889 | 0.000 | CG1 | 95 | ILE |
| 1391 | 1.884 | 0.001 | HG12 | 95 | Ile |
| 1392 | 0.956 | 0.001 | HG13 | 95 | ILE |
| 1393 | 1.115 | 0.001 | QG2 | 95 | ILE |
| 1394 | 0.882 | 0.001 | QD1 | 95 | ILE |
| 1395 | 26.889 | 0.000 | CG1 | 85 | ILE |
| 1396 | 18.367 | 0.000 | CG2 | 85 | ILE |
| 1397 | 14.053 | 0.000 | CD1 | 85 | ILE |
| 1398 | 1.174 | 0.001 | HG13 | 85 | ILE |
| 1399 | 1.616 | 0.001 | HG12 | 85 | ILE |
| 1400 | 0.933 | 0.002 | QG2 | 85 | ILE |
| 1401 | 0.892 | 0.001 | QD1 | 85 | ILE |
| 1402 | 28.101 | 0.000 | CG1 | 9 | ILE |
| 1403 | 18.364 | 0.000 | CG2 | 9 | ILE |
| 1404 | 14.027 | 0.000 | CD1 | 9 | ILE |
| 1405 | 1.152 | 0.004 | HG13 | 9 | ILE |
| 1406 | 1.810 | 0.002 | HG12 | 9 | ILE |
| 1407 | 1.064 | 0.002 | QG2 | 9 | ILE |
| 1408 | 1.003 | 0.001 | QD1 | 9 | ILE |
| 1409 | 2.969 | 0.000 | HB2 | 26 | HIS |
| 1410 | 2.859 | 0.000 | HB3 | 26 | HIS |
| 1411 | 2.822 | 0.001 | HB2 | 33 | HIS |
| 1412 | 2.770 | 0.002 | HB3 | 33 | HIS |
| 1413 | 24.548 | 0.000 | CG | 99 | LYS |
| 1414 | 29.179 | 0.000 | CD | 99 | LYS |
| 1415 | 41.722 | 0.000 | CE | 99 | LYS |
| 1416 | 0.712 | 0.001 | HG2 | 99 | LYS |
| 1417 | 0.246 | 0.003 | HG3 | 99 | LYS |
| 1418 | 1.180 | 0.002 | HD3 | 99 | LYS |
| 1419 | 1.233 | 0.001 | HD2 | 99 | LYS |
| 1420 | 2.631 | 0.002 | QE | 99 | LYS |
| 1421 | 41.608 | 0.000 | CE | 39 | LYS |
| 1422 | 25.303 | 0.000 | CG | 39 | LYS |
| 1423 | 29.200 | 0.000 | CD | 39 | LYS |
| 1424 | 1.670 | 0.000 | HB2 | 39 | LYS |
| 1425 | 1.475 | 0.000 | HB3 | 39 | LYS |
| 1426 | 1.503 | 0.000 | QD | 39 | LYS |
| 1427 | 1.322 | 0.001 | QG | 39 | LYS |
| 1428 | 2.799 | 0.002 | QE | 39 | LYS |
| 1429 | 26.072 | 0.000 | CG | 38 | ARG |
| 1430 | 45.037 | 0.000 | CD | 38 | ARG |
| 1431 | 3.169 | 0.000 | HD2 | 38 | ARG |
| 1432 | 3.080 | 0.000 | HD3 | 38 | ARG |
| 1433 | 2.143 | 0.000 | HG2 | 38 | ARG |
| 1434 | 1.833 | 0.000 | HG3 | 38 | ARG |
| 1435 | 41.566 | 0.000 | CE | 22 | LYS |
| 1436 | 28.870 | 0.000 | CD | 22 | LYS |
| 1437 | 24.010 | 0.000 | CG | 22 | LYS |


| 1438 | 2.769 | 0.002 | QE | 22 | LYS |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 1439 | 1.339 | 0.000 | QD | 22 | LYS |
| 1440 | 0.851 | 0.001 | HG2 | 22 | LYS |
| 1441 | 0.686 | 0.001 | HG3 | 22 | LYS |
| 1442 | 24.348 | 0.000 | CG | 86 | LYS |
| 1444 | 41.883 | 0.000 | CE | 86 | LYS |
| 1445 | 2.831 | 0.000 | QE | 86 | LYS |
| 1446 | 29.648 | 0.000 | CD | 86 | LYS |
| 1447 | 25.694 | 0.000 | CG | 98 | LEU |
| 1448 | 22.455 | 0.000 | CD2 | 98 | LEU |
| 1449 | 25.508 | 0.000 | CD1 | 98 | LEU |
| 1450 | 2.106 | 0.001 | HG | 98 | LEU |
| 1452 | 33.202 | 0.000 | CG | 16 | GLN |
| 1453 | 2.604 | 0.000 | HG2 | 16 | GLN |
| 1454 | 2.444 | 0.000 | HG3 | 16 | GLN |
| 1455 | 26.849 | 0.000 | CG | 64 | LEU |
| 1456 | 24.353 | 0.000 | CD1 | 64 | LEU |
| 1457 | 24.438 | 0.000 | CD2 | 64 | LEU |
| 1458 | 0.637 | 0.004 | QQD | 64 | LEU |
| 1459 | 0.373 | 0.006 | HG | 64 | LEU |
| 1460 | 24.726 | 0.000 | CG | 32 | LEU |
| 1461 | 23.361 | 0.000 | CD1 | 32 | LEU |
| 1462 | 20.335 | 0.000 | CD2 | 32 | LEU |
| 1463 | 0.379 | 0.000 | HG | 32 | LEU |
| 1464 | -0.761 | 0.002 | QD2 | 32 | LEU |
| 1465 | -0.869 | 0.004 | QD1 | 32 | LEU |
| 1466 | 3.005 | 0.000 | HB2 | 10 | PHE |
| 1467 | 2.890 | 0.000 | HB3 | 10 | PHE |
| 1468 | 25.344 | 0.000 | CG | 87 | LYS |
| 1469 | 29.020 | 0.000 | CD | 87 | LYS |
| 1470 | 41.848 | 0.000 | CE | 87 | LYS |
| 1471 | 1.539 | 0.002 | HG2 | 87 | LYS |
| 1472 | 1.391 | 0.002 | HG3 | 87 | LYS |
| 1473 | 1.641 | 0.001 | QD | 87 | LYS |
| 1474 | 2.913 | 0.001 | QE | 87 | LYS |
| 1475 | 1.600 | 0.001 | QD | 53 | LYS |
| 1476 | 1.444 | 0.001 | QG | 53 | LYS |
| 1477 | 2.839 | 0.001 | QE | 53 | LYS |
| 1478 | 41.739 | 0.000 | CE | 5 | LYS |
| 1479 | 26.024 | 0.000 | CG | 5 | LYS |
| 1480 | 28.801 | 0.000 | CD | 5 | LYS |
| 1482 | 1.458 | 0.005 | HG2 | 5 | LYS |
| 1483 | 1.304 | 0.000 | HG3 | 5 | LYS |
| 1484 | 2.883 | 0.002 | QE | 5 | LYS |
| 1485 | 28.716 | 0.000 | CG | 94 | LEU |
| 1486 | 27.558 | 0.000 | CD1 | 94 | LEU |
| 1487 | 24.198 | 0.000 | CD2 | 94 | LEU |
| 1488 | 1.415 | 0.002 | HG | 94 | LEU |
| 1489 | 26.428 | 0.000 | CD | 13 | LYS |
| 1490 | 29.458 | 0.000 | CG | 13 | LYS |
| 1491 | 42.303 | 0.000 | CE | 13 | LYS |
| 1492 | 2.399 | 0.000 | HB2 | 13 | LYS |
| 1493 | 2.221 | 0.000 | HB3 | 13 | LYS |
| 1494 | 1.671 | 0.001 | QD | 13 | LYS |
| 1495 | 1.908 | 0.001 | HG2 | 13 | LYS |
| 1496 | 1.785 | 0.001 | HG3 | 13 | LYS |
| 1497 | 3.103 | 0.001 | HE2 | 13 | LYS |
| 1498 | 3.069 | 0.001 | HE3 | 13 | LYS |
| 1499 | 23.584 | 0.000 | CG2 | 63 | THR |
| 1500 | 1.276 | 0.002 | QG2 | 63 | THR |
| 1501 | 25.795 | 0.000 | CG | 68 | LEU |
| 1502 | 0.967 | 0.001 | HG | 68 | LEU |
| 1503 | 21.682 | 0.000 | CD1 | 68 | LEU |
| 1504 | 21.583 | 0.000 | CD2 | 68 | LEU |
| 1505 | 0.215 | 0.004 | QQD | 68 | LEU |
| 1506 | 24.281 | 0.000 | CG | 25 | LYS |
| 1507 | 28.514 | 0.000 | CD | 25 | LYS |
| 1508 | 41.714 | 0.000 | CE | 25 | LYS |
| 1509 | 2.797 | 0.000 | QE | 25 | LYS |
| 1510 | 1.451 | 0.000 | QD | 25 | LYS |
| 1511 | 1.216 | 0.000 | QG | 25 | LYS |
| 1512 | 31.430 | 0.000 | CG | 12 | MET |
| 1513 | 2.136 | 0.003 | HB3 | 12 | MET |
| 1514 | 2.250 | 0.002 | HB2 | 12 | MET |
| 1515 | 2.731 | 0.002 | HG2 | 12 | MET |
| 1516 | 2.575 | 0.003 | HG3 | 12 | MET |
| 1517 | 33.733 | 0.000 | CG | 42 | GLN |
| 1518 | 2.145 | 0.000 | HG3 | 42 | GLN |
| 1519 | 2.206 | 0.000 | HG2 | 42 | GLN |
| 1520 | 32.089 | 0.000 | CG | 65 | MET |
| 1521 | 2.766 | 0.000 | HG2 | 65 | MET |
| 1522 | 2.542 | 0.001 | HG3 | 65 | MET |
| 1523 | 43.817 | 0.005 | CD | 91 | ARG |



| 1613 | 2.682 | 0.002 | QB | 70 | ASN |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 1614 | 2.316 | 0.001 | HG2 | 90 | GLU |
| 1615 | 2.253 | 0.001 | HG3 | 90 | GLU |
| 1616 | 0.672 | 0.000 | QG2 | 78 | THR |
| 1617 | 4.167 | 0.001 | HB | 78 | THR |
| 1618 | 3.721 | 0.000 | HA2 | 77 | GLY |
| 1619 | 3.852 | 0.001 | HA3 | 77 | GLY |
| 1620 | 1.001 | 0.001 | QQD | 94 | LEU |
| 1621 | 1.019 | 0.001 | QD1 | 98 | LEU |
| 1622 | 0.544 | 0.001 | QD2 | 98 | LEU |
| 1623 | 2.169 | 0.002 | QG | 104 | GLU |
| 1624 | 3.266 | 0.001 | HD2 | 91 | ARG |
| 1625 | 3.179 | 0.001 | HD3 | 91 | ARG |
| 1626 | 1.706 | 0.001 | HG2 | 91 | ARG |
| 1627 | 1.180 | 0.001 | HG3 | 91 | ARG |
| 1628 | 58.867 | 0.000 | CA | 58 | ILE |
| 1629 | 38.504 | 0.000 | CB | 58 | ILE |
| 1630 | 177.495 | 0.000 | C | 58 | ILE |
| 1631 | 4.082 | 0.000 | HA | 58 | ILE |
| 1632 | 1.491 | 0.000 | HB | 58 | ILE |
| 1633 | 12.387 | 0.000 | CD1 | 58 | ILE |
| 1634 | 17.084 | 0.000 | CG2 | 58 | ILE |
| 1635 | 27.350 | 0.000 | CG1 | 58 | ILE |
| 1636 | 1.351 | 0.001 | HG12 | 58 | ILE |
| 1637 | 0.834 | 0.003 | HG13 | 58 | ILE |
| 1638 | 0.613 | 0.002 | QG2 | 58 | ILE |
| 1639 | 0.616 | 0.001 | QD1 | 58 | ILE |
| 1644 | 3.860 | 0.000 | HB2 | 15 | SER |
| 1645 | 3.621 | 0.000 | HB3 | 15 | SER |
| 1646 | 3.768 | 0.006 | HA | 15 | SER |
| 1647 | 1.353 | 0.001 | HG3 | 8 | LYS |
| 1648 | 1.531 | 0.001 | HG2 | 8 | LYS |
| 1649 | 1.591 | 0.002 | QD | 8 | LYS |
| 1650 | 2.160 | 0.002 | HA | 7 | LYS |
| 1651 | 41.814 | 0.000 | CE | 7 | LYS |
| 1652 | 24.023 | 0.000 | CG | 7 | LYS |
| 1653 | 29.272 | 0.000 | CD | 7 | LYS |
| 1655 | 1.670 | 0.002 | HB2 | 7 | LYS |
| 1656 | 1.324 | 0.005 | HB3 | 7 | LYS |
| 1657 | 0.986 | 0.002 | QD | 7 | LYS |
| 1658 | 1.536 | 0.001 | QG | 7 | LYS |
| 1660 | 24.155 | 0.000 | CG | 27 | LYS |
| 1661 | 29.035 | 0.000 | CD | 27 | LYS |
| 1662 | 41.341 | 0.000 | CE | 27 | LYS |
| 1663 | 1.671 | 0.000 | HB2 | 27 | LYS |
| 1664 | 0.964 | 0.000 | HB3 | 27 | LYS |
| 1665 | 1.235 | 0.000 | HD2 | 27 | LYS |
| 1666 | 1.011 | 0.000 | HD3 | 27 | LYS |
| 1667 | 0.954 | 0.000 | QG | 27 | LYS |
| 1668 | 2.386 | 0.000 | QE | 27 | LYS |
| 1669 | 1.563 | 0.000 | HB2 | 25 | LYS |
| 1670 | 1.379 | 0.000 | HB3 | 25 | LYS |
| 1671 | 1.813 | 0.001 | HB2 | 53 | LYS |
| 1672 | 1.742 | 0.002 | HB3 | 53 | LYS |
| 1673 | 4.032 | 0.002 | HA | 55 | LYS |
| 1674 | 24.386 | 0.000 | CG | 55 | LYS |
| 1675 | 28.097 | 0.000 | CD | 55 | LYS |
| 1676 | 1.858 | 0.001 | HB2 | 55 | LYS |
| 1677 | 1.646 | 0.001 | HB3 | 55 | LYS |
| 1678 | 1.516 | 0.000 | QG | 55 | LYS |
| 1679 | 1.199 | 0.003 | QG | 73 | LYS |
| 1680 | 3.562 | 0.001 | HA | 88 | LYS |
| 1681 | 24.802 | 0.000 | CG | 88 | LYS |
| 1682 | 29.271 | 0.000 | CD | 88 | LYS |
| 1683 | 41.650 | 0.000 | CE | 88 | LYS |
| 1684 | 1.781 | 0.001 | HB2 | 88 | LYS |
| 1685 | 1.699 | 0.001 | HB3 | 88 | LYS |
| 1686 | 1.321 | 0.001 | QG | 88 | LYS |
| 1687 | 2.857 | 0.000 | QE | 88 | LYS |
| 1688 | 1.538 | 0.001 | QD | 88 | LYS |
| 1689 | 1.789 | 0.000 | HB2 | 87 | LYS |
| 1690 | 1.674 | 0.001 | HB3 | 87 | LYS |
| 1691 | 1.261 | 0.000 | HG2 | 100 | LYS |
| 1692 | 1.119 | 0.001 | HG3 | 100 | LYS |
| 1693 | 1.606 | 0.002 | QD | 100 | LYS |
| 1747 | 54.344 | 0.000 | CA | 31 | ASN |
| 1748 | 4.023 | 0.002 | HA | 31 | ASN |
| 1749 | 39.785 | 0.000 | CB | 31 | ASN |
| 1750 | 2.013 | 0.004 | HB2 | 31 | ASN |
| 1751 | 1.919 | 0.002 | HB3 | 31 | ASN |
| 1752 | 3.752 | 0.000 | HA3 | 1 | GLY |
| 1753 | 42.567 | 0.000 | CA | 1 | GLY |
| 1754 | 3.341 | 0.000 | HA2 | 1 | GLY |



| 1844 | 3.993 | 0.001 | HA | 51 | ALA |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 1845 | 6.957 | 0.000 | HE21 | 16 | GLN |
| 1846 | 7.559 | 0.000 | HE22 | 16 | GLN |
| 1849 | 10.179 | 0.000 | HE1 | 59 | TRP |
| 1850 | 6.404 | 0.002 | HD2 1 | 103 | ASN |
| 1851 | 7.807 | 0.001 | HD22 | 103 | ASN |
| 1852 | 9.464 | 0.000 | HD1 | 18 | HIS |
| 1853 | 7.660 | 0.002 | HD22 | 70 | ASN |
| 1854 | 6.570 | 0.002 | HD21 | 70 | ASN |
| 1855 | 6.491 | 0.000 | HD2 | 97 | TYR |
| 1856 | 7.040 | 0.000 | HD1 | 97 | TYR |
| 1857 | 5.457 | 0.000 | HE2 | 97 | TYR |
| 1858 | 6.599 | 0.000 | HE1 | 97 | TYR |
| 1859 | 6.856 | 0.000 | HD2 | 10 | PHE |
| 1860 | 6.027 | 0.000 | HE1 | 10 | PHE |
| 1861 | 6.928 | 0.000 | HE2 | 10 | PHE |
| 1862 | 6.093 | 0.000 | HZ | 10 | PHE |
| 1863 | 0.010 | 0.004 | HD2 | 18 | HIS |
| 1864 | 6.773 | 0.000 | QE | 36 | PHE |
| 1865 | 7.305 | 0.000 | QD | 36 | PHE |
| 1866 | 6.744 | 0.000 | QD | 46 | TYR |
| 1867 | 6.889 | 0.000 | HD1 | 59 | TRP |
| 1868 | 7.472 | 0.005 | HE3 | 59 | TRP |
| 1869 | 6.532 | 0.004 | HZ3 | 59 | TRP |
| 1870 | 6.536 | 0.000 | QE | 67 | TYR |
| 1871 | 7.091 | 0.000 | QD | 67 | TYR |
| 1872 | 6.507 | 0.000 | QE | 74 | TYR |
| 1873 | 7.105 | 0.000 | QD | 74 | TYR |
| 1874 | 6.580 | 0.000 | QD | 82 | PHE |
| 1876 | 7.276 | 0.000 | QE | 82 | PHE |
| 1877 | 7.106 | 0.000 | HZ | 82 | PHE |
| 1878 | 111.391 | 0.000 | NE2 | 16 | GLN |
| 1879 | 111.664 | 0.005 | ND2 | 70 | ASN |
| 1880 | 111.964 | 0.000 | ND2 | 54 | ASN |
| 1881 | 7.423 | 0.001 | HD22 | 54 | ASN |
| 1882 | 6.854 | 0.000 | HD21 | 54 | ASN |
| 1883 | 112.579 | 0.000 | NE2 | 42 | GLN |
| 1884 | 7.470 | 0.000 | HE22 | 42 | GLN |
| 1885 | 6.783 | 0.000 | HE21 | 42 | GLN |
| 1886 | 114.098 | 0.000 | ND2 | 103 | ASN |
| 1887 | 115.086 | 0.005 | ND2 | 31 | ASN |
| 1888 | 7.325 | 0.000 | HD21 | 31 | ASN |
| 1889 | 7.924 | 0.002 | HD22 | 31 | ASN |
| 1890 | 127.918 | 0.000 | NE1 | 59 | TRP |
| 1891 | 6.230 | 0.000 | HE | 38 | ARG |
| 1892 | -3.750 | 0.000 | HG3 | 80 | MET |
| 1893 | -2.050 | 0.000 | HG2 | 80 | MET |
| 1894 | 3.342 | 0.000 | QM5 | 105 | HEM |
| 1895 | 9.530 | 0.008 | HBM | 105 | HEM |
| 1896 | 2.461 | 0.001 | QT4 | 105 | HEM |
| 1897 | 6.272 | 0.000 | HT4A | 105 | HEM |
| 1898 | 3.735 | 0.000 | QM3 | 105 | HEM |
| 1899 | 9.167 | 0.000 | HAM | 105 | HEM |
| 1900 | 1.380 | 0.000 | QT2 | 105 | HEM |
| 1901 | 5.109 | 0.000 | HT2A | 105 | HEM |
| 1902 | 3.403 | 0.000 | QM1 | 105 | HEM |
| 1903 | 8.884 | 0.000 | HDM | 105 | HEM |
| 1904 | 2.057 | 0.000 | QM8 | 105 | HEM |
| 1905 | 4.062 | 0.000 | HA71 | 105 | HEM |
| 1906 | 3.311 | 0.000 | HA 72 | 105 | HEM |
| 1907 | 999.000 | 0.000 | QA7 | 105 | HEM |
| 1908 | 2.346 | 0.000 | HB73 | 105 | HEM |
| 1909 | 3.001 | 0.000 | HB74 | 105 | HEM |
| 1910 | 999.000 | 0.000 | QB7 | 105 | HEM |
| 1911 | 9.571 | 0.000 | HGM | 105 | HEM |
| 1912 | 4.389 | 0.000 | QA6 | 105 | HEM |
| 1913 | 999.000 | 0.000 | HA62 | 105 | HEM |
| 1914 | 999.000 | 0.000 | HA61 | 105 | HEM |
| 1915 | 3.680 | 0.000 | HB63 | 105 | HEM |
| 1916 | 2.580 | 0.000 | HB64 | 105 | HEM |
| 1917 | 999.000 | 0.000 | QB6 | 105 | HEM |
| 1918 | -3.429 | 0.000 | QE | 80 | MET |
| 1919 | 5.578 | 0.003 | HH2 | 59 | TRP |
| 1920 | 6.956 | 0.002 | HZ2 | 59 | TRP |
| 1921 | 0.381 | 0.000 | HE1 | 18 | HIS |
| 3000 | 999.000 | 0.000 | HD2 | 48 | CMF |
| 3001 | 999.000 | 0.000 | HD1 | 48 | CMF |
| 3002 | 7.206 | 0.000 | QD | 48 | CMF |
| 3003 | 7.165 | 0.000 | QE | 48 | CMF |
| 3004 | 999.000 | 0.000 | HB2 | 48 | CMF |
| 3005 | 999.000 | 0.000 | HB3 | 48 | CMF |
| 3006 | 6.948 | 0.000 | HD1 | 26 | HIS |
| 3008 | 6.730 | 0.003 | HD2 | 26 | HIS |


| 3009 | 999.000 | 0.000 | QHH | 48 | CMF |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 3010 | 175.638 | 0.000 | C | 40 | THR |
| 3011 | 178.294 | 0.000 | C | 44 | PRO |
| 3012 | 175.235 | 0.000 | C | 49 | THR |
| 3013 | 178.634 | 0.000 | C | 51 | ALA |
| 3014 | 176.255 | 0.000 | C | 52 | ASN |
| 3099 | 999.000 | 0.000 | H | 48 | CMF |
| 3100 | 6.206 | 0.000 | HE | 91 | ARG |
| 3101 | 6.579 | 0.000 | H | 47 | SER |
| 3102 | 4.151 | 0.000 | HA | 47 | SER |
| 3103 | 3.486 | 0.001 | QB | 47 | SER |
| 5000 | 3.590 | 0.000 | HA | 30 | PRO |
| 5001 | 1.230 | 0.000 | QB | 30 | PRO |
| 5002 | 0.733 | 0.000 | QG | 30 | PRO |
| 5003 | 2.540 | 0.001 | HD2 | 30 | PRO |
| 5004 | 1.516 | 0.000 | HD3 | 30 | PRO |
| 5011 | 6.704 | 0.000 | QE | 46 | TYR |
| 5020 | 10.198 | 0.000 | H | 31 | ASN |
| 5021 | 60.269 | 0.000 | CA | 30 | PRO |
| 5022 | 31.020 | 0.000 | CB | 30 | PRO |
| 5023 | 26.377 | 0.000 | CG | 30 | PRO |
| 5024 | 48.500 | 0.000 | CD | 30 | PRO |
| 5100 | 7.285 | 0.000 | HE1 | 33 | HIS |
| 7000 | 0.752 | 0.005 | HG12 | 57 | ILE |
| 7001 | 0.806 | 0.000 | HG13 | 57 | ILE |
| 7002 | 7.076 | 0.000 | HD1 | 10 | PHE |
| 7004 | 112.024 | 0.000 | N | 45 | GLY |
| 7005 | 8.831 | 0.000 | H | 45 | GLY |
| 8000 | 4.356 | 0.000 | HA | 44 | PRO |
| 8001 | 1.870 | 0.000 | HB2 | 44 | PRO |
| 8002 | 2.147 | 0.000 | HB3 | 44 | PRO |
| 8003 | 1.934 | 0.000 | HG2 | 44 | PRO |
| 8004 | 2.064 | 0.000 | HG3 | 44 | PRO |
| 8005 | 3.658 | 0.000 | HD2 | 44 | PRO |
| 8006 | 3.949 | 0.000 | HD3 | 44 | PRO |
| 8007 | 63.456 | 0.000 | CA | 44 | PRO |
| 8008 | 31.514 | 0.000 | CB | 44 | PRO |
| 8009 | 50.659 | 0.000 | CD | 44 | PRO |
| 8010 | 27.383 | 0.000 | CG | 44 | PRO |
| 8021 | 46.052 | 0.000 | CA | 56 | GLY |
| 8022 | 3.748 | 0.000 | HA2 | 56 | GLY |
| 8023 | 3.637 | 0.000 | HA3 | 56 | GLY |

