

Full-Space Potential Gradient Driven Charge Migration inside BiFeO₃ Photocathode

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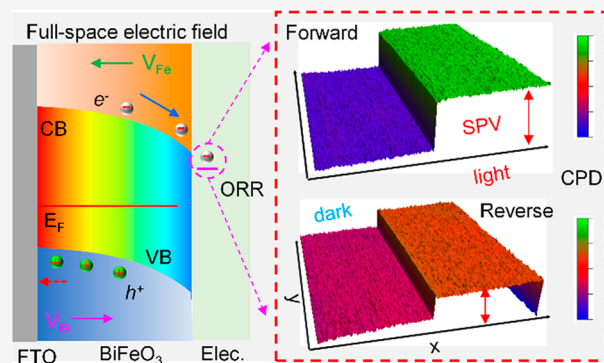
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ABSTRACT: Solar–fuel conversion depends on effective collection of photocarriers in the photoelectrode. In practice, however, considerable photocarriers are lost in the bulk through recombination due to the absence of a driving force. To overcome this, herein, a full-space electric field is induced in BiFeO₃ photocathodes by building a gradient homojunction through Fermi level engineering. As expected, the BiFeO₃ photocathodes with forward electric field show significantly enhanced performance: a state-of-the-art photocurrent of $-1.02 \text{ mA}\cdot\text{cm}^{-2}$ at 0.5 V vs RHE and H₂O₂ production of $380 \text{ mmol}\cdot(\text{L}\cdot\text{m}^2)^{-1}$ within 50 min. First-principles calculations and experimental analysis suggest that the Bi vacancies as shallow acceptors can significantly modulate the Fermi level of BiFeO₃. The resulting internal electric field serves as an additional driving force to promote charge collection. This work provides an approach to induce a full-space electric field in semiconductor films through gradient defects modulation, which can be broadly applied to other optoelectronic systems.



Photoelectrochemical (PEC) synthesis has received growing attention as a green approach to produce hydrogen peroxide (H₂O₂).^{1–4} This approach takes advantage of the charge-separation ability of semiconductor–liquid junctions, which provides excited photocarriers to produce H₂O₂ via dioxygen reduction or water oxidation.⁵ Despite great efforts, the practical application of PEC H₂O₂ production is still challenging due to severe photocarrier losses through recombination.^{6–8} For many emerging photoelectrode materials, photocarrier diffusion length is limited to 10–50 nm and only 10% of photocarriers have been collected.^{9,10} Generally, photoelectrodes are separated into a space charge region and a natural region, while recombination majorly (90%) occurs in the natural region due to the absence of a drive force.^{11,12} Therefore, promoting photocarrier collection in the natural region can provide measurable improvements toward realizing high-efficient PEC H₂O₂ production.

Constructing an internal electric field in the natural region as an additional driving force is critical to promote photocarriers collection.¹² So far, various strategies have been widely employed to build an electric field, such as surface tailoring,¹³ heterojunction,^{14,15} and facet junction.^{16,17} However, the influence of these electric fields is mostly spatially limited to the surface or depletion area, rarely modulating the recombination process in the field-free region.^{15,18,19} The depolarization field is a typical way to promote charge

separation in both bulk and surface, while it is confined to ferroelectrics.^{20,21} Thus, a universal approach to construct full-space electric field through the photoelectrode is the remedial approach to inhibit carrier recombination.

Gradient homojunction provides an alternative approach to creating a full-space electric field, which is often employed in multijunction photovoltaic devices, like CuInGaSe (CIGS) solar cells.²² In these devices semiconductors with equal band gaps but different Fermi levels are placed in contact.^{23,24} Compared to heterojunctions, homojunctions have the following advantages: (1) easier to prepare since only a single semiconductor material is required; (2) same material with less lattice mismatch is beneficial to mitigate interfacial defects. For instance, Berglund and co-workers developed a gradient self-doped CuBi₂O₄ photocathode with forward electric field to achieve a charge separation efficiency of 34%.^{25,26} Therefore, gradient homojunction depends on a fine modulation of the semiconductors' Fermi level, which is challenging to achieve

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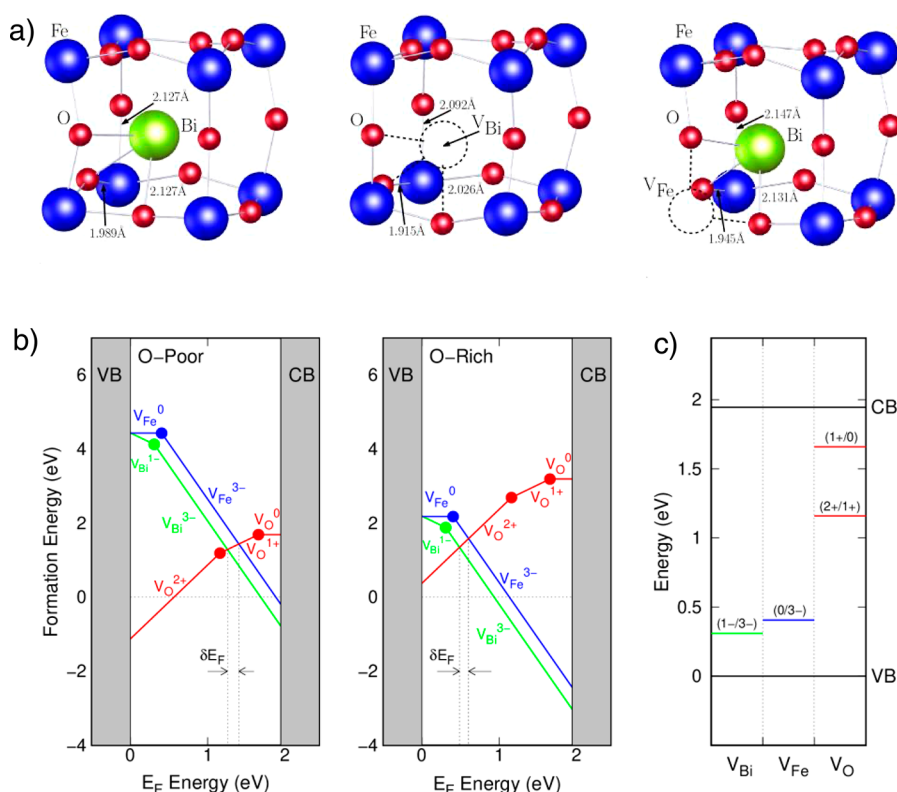


Figure 1. Local structure for pristine BFO, Fe vacancy, and Bi vacancy (a); formation energies for neutral and charged Bi, Fe, O vacancies (b); and defect transition levels (c).

64 without a deeper understanding of materials' electronic
65 structure.

66 In this work, a full-space electric field was induced in BiFeO₃
67 (denoted as BFO) films by constructing a gradient
68 homojunction through Fermi level engineering. First-principles
69 calculations reveal that the Fermi level of BFO depends on the
70 density of Fe and Bi vacancies in the film, which are shallow
71 acceptors to define the p-type conductivity of BFO. These
72 results were validated by surface potential and ultraviolet
73 photoelectron spectroscopy (UPS) analysis. PEC perform-
74 ances of BFO films, including current–voltage (*J–V*) response
75 and H₂O₂ production activity, confirm enhanced performance
76 for the forward gradient doped photocathode. Incident and
77 absorbed photon-to-current efficiency (IPCE and APCE) and
78 surface photovoltage (SPV) analysis proved that the enhanced
79 performance is primarily attributed to the improved charge
80 collection efficiency. Thus, this work demonstrated the
81 successful inhibition of photocarrier recombination by gradient
82 homojunction which can be broadly applied to other
83 optoelectronic systems.

84 Insight into the structural and electronic properties of Bi and
85 Fe vacancy defects and their effect on the Fermi level energy of
86 BFO was obtained by means of density functional theory
87 (DFT) based first-principles calculations at the DFT+*U* level
88 of theory. Band structure and total density of states
89 calculations for the R3c phase structure of BFO (Figure
90 S1a,b) predict a band gap of 1.95 and 2.30 eV for Hubbard *U*
91 values of 3.0 and 4.5 eV, respectively, in line with previous
92 results and in good agreement with measured optical gaps of
93 2.10–2.86 eV.^{27–31} Point defects are calculated for *U* = 3.0 eV
94 level of theory using the supercell method (see Supporting
95 Information) for ease of comparison with previous DFT

results.^{27–31} Tables S1–S3 report computed chemical
96 potentials and formation energies for neutral defects. Neutral
97 and charged Fe and Bi vacancies distort the local geometry of
98 the R3c polar structure. For charged vacancies, V_{Fe}³⁻ and V_{Bi}³⁻,
99 bond lengths are slightly modified around the vacancy with
100 respect to the pristine structure (Figure 1a). The additional
101 negative charge of the defects reproduces the charge provided
102 by Fe and Bi to the crystal and therefore promote their
103 stability. The stability of Bi, Fe, and O vacancies is further
104 studied by computing formation energies as a function of the
105 Fermi energy (Figure 1b).³² For O-poor and O-rich
106 conditions, Fe and Bi vacancies both act as shallow acceptors,
107 with Bi vacancies more stable than Fe vacancies. Oxygen
108 vacancies act as donors, as shown by the defect transition levels
109 (Figure 1c). Our results are in good qualitative agreement with
110 previous first-principles results using different levels of
111 theory.^{27–31} 112

113 Notably, the position of the Fermi level is greatly affected by
114 the stability of O vacancies, suggesting that growth conditions
115 greatly modulate the Fermi level position. Different exper-
116 imental Bi:Fe ratios for BFO are modeled using our theoretical
117 results for isolated point defects. For the case of Bi:Fe ratio of
118 1:0.95 in growth solution (see below), Fe vacancies are
119 expected to be dominant in the solid. Hence, in this case, the
120 Fermi level is estimated as the intersection between the
121 formation energies of V_{Fe}³⁻ and V_O vacancies. In this case, the
122 Fermi level is set at 1.40 eV for O-poor conditions and at 0.61
123 eV for O-rich conditions, as shown in Figure 1b (δE_F).
124 Similarly, for the case of Bi:Fe ratio of 0.95:1, where Bi
125 vacancies are dominant, the Fermi level is set at 1.26 and 0.49
126 eV for O-poor and O-rich conditions, respectively. Therefore,
127 our results suggest that the Fermi level can be modulated

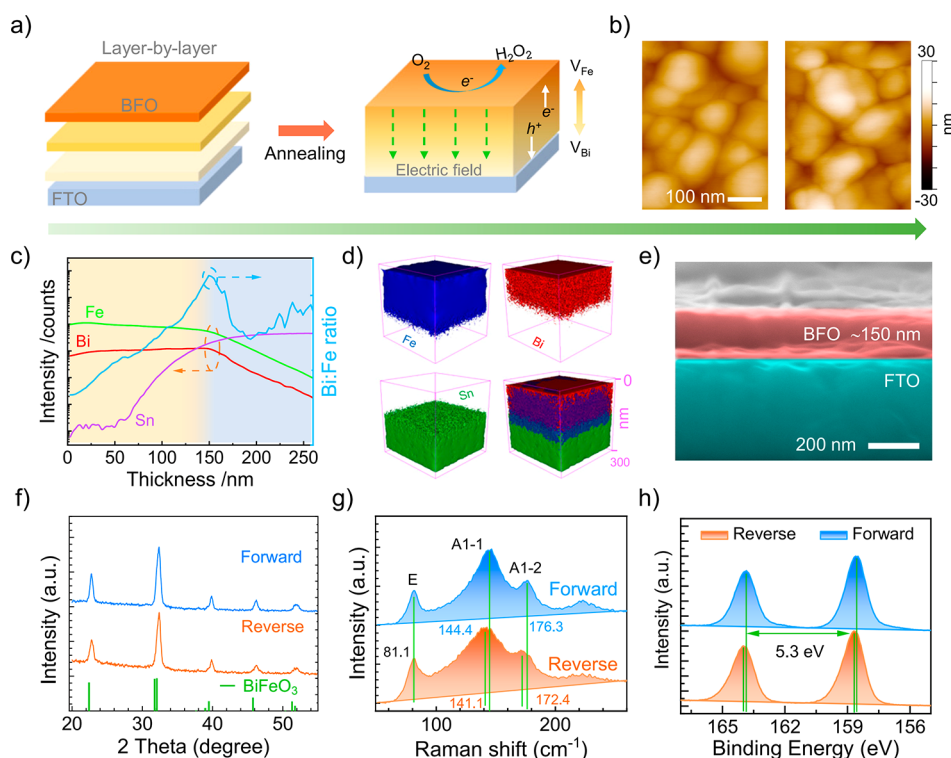


Figure 2. Schematic illustration of the fabrication procedure for gradient doped films (a); plane-view AFM images of the gradient doped films, forward (left) and reverse (right) (b); TOF-SIMS depth profile of the thin film composition (c); TOF-SIMS 3D element distribution of Fe, Bi, Sn and the overlay (d) in the gradient BFO film; the cross-section SEM image of BFO films (e); the XRD pattern (f), Raman spectra (g) and XPS core-level spectra of gradient BFO films (h).

128 (through Fe and Bi vacancies) in the ranges indicated by δE_F
 129 in Figure 1b. Our results are qualitatively similar for different
 130 values of the Hubbard U and hybrid functionals.

131 We can further estimate the position of the Fermi level as a
 132 function of O vacancy concentration by solving the electro-
 133 neutrality equation for charge balance.³³ We use arbitrary
 134 defect concentrations for Fe and Bi vacancy and the transition
 135 levels obtained from DFT as input parameters. Figure S1c
 136 shows our results for the extreme cases of 5% of Fe and Bi
 137 vacancies. In both cases, the Fermi level changes from p-type at
 138 O-rich to n-type under O-poor conditions, in agreement with
 139 the results of Figure 1b.

140 The above mechanistic insights leveraged a path to build a
 141 BFO film with continuous homojunction through modulating Bi
 142 and Fe vacancies. As shown in Figure 2a, the gradient self-
 143 doped BFO films were fabricated using a three-step spin-
 144 coating and annealing process. Typically, for a forward gradient
 145 BFO film (defined as the V_{Fe} at the surface and at V_{Bi} the
 146 bottom), the lower Bi:Fe mole ratio (0.95:1, V_{Bi} dominated)
 147 precursor was used for the first two spin-coating cycles, then
 148 the moderate ratio (1:1) precursor for the following two cycles,
 149 and finally the higher ratio (1:0.95, V_{Fe} dominated) for the last
 150 two cycles. After that, the films were annealed at 600 °C for 1 h
 151 to improve the diffusion of Bi and Fe vacancies in films. The
 152 reversed gradient BFO film is fabricated using the same
 153 method but with a lower Bi:Fe ratio precursor introduced first.
 154 Plan-view AFM images of the resultant film revealed a
 155 homogeneous, compact layer with granular morphology and
 156 uniform grain size of around 50–200 nm. As shown in Figure
 157 2b, the forward and reverse doped films show identical
 158 morphology.

159 Gradient distribution of Bi and Fe through the thickness of
 160 the BFO film was investigated by time-of-flight secondary ion
 161 mass spectrometry (TOF-SIMS).³³ In the depth profile
 162 (Figure 2c) of the gradient BFO film, the Fe concentration
 163 monotonically decreased from the surface to 150 nm in depth,
 164 while the Bi concentration increased and then dramatically
 165 decreased. As a result, the Bi:Fe ratio continuously increased
 166 from the surface to 150 nm depth in the bulk. In Figure 2d, the
 167 3D element distribution images indicate layered structure and
 168 the uniform spatial distribution of BFO film horizontally on
 169 the FTO substrate. The cross-section scanning electron
 170 microscopy (SEM) image in Figure 2e further confirmed the
 171 layered structure of prepared films with BFO thickness of
 172 around 150 nm.

173 X-ray diffraction (XRD) patterns of BFO films, in Figure 2f,
 174 illustrates that all sharp and strong diffraction peaks are well-
 175 matched to those of BiFeO₃ (JCPDS 86-1518) with good
 176 crystallinity. Moreover, the well-matched diffraction peaks of
 177 homogeneous BFO films with varied V_{Bi} and V_{Fe} density in
 178 Figure S2 implies the excellent phase purity and defects
 179 tolerance of BFO films. Likewise, Raman spectra shown in
 180 Figure 2g are indicative of phase-pure BFO. Meanwhile, slight
 181 shifts can be found for the A1-1 and A1-2 signals, which are
 182 probably due to vacancies caused lattice distortion.³⁴ The
 183 X-ray photoelectron spectroscopy (XPS) in Figure 2h also
 184 displays a binding energy shift in gradient doped films. The Bi
 185 and Fe core-level spectra of the reverse BFO film show a higher
 186 binding energy shift compared with the forward BFO film
 187 (Figure 2h and Figure S3). Considering the forward film has
 188 higher V_{Fe} density at the surface than the reverse film, the shift
 189 could generally be attributed to the increase of equilibrium
 190 electron density and hence the binding energy.^{35,20} Thus, 190

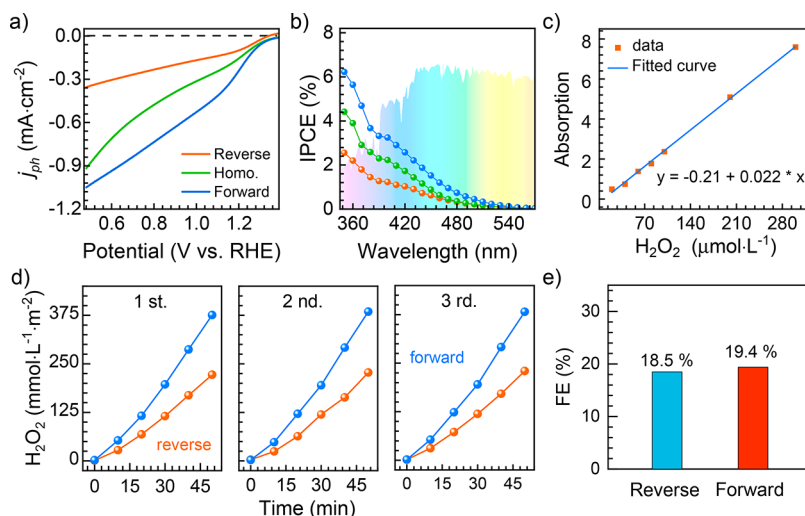


Figure 3. J - V responses of the reverse, homogeneous, and forward gradient BFO films under blue LED (430 nm, 60 mW·cm⁻²) illumination in 0.1 M KHCO₃ with 0.1 M Na₂S₂O₈ (pH 8.2) as an electron scavenger (a); IPCE spectra of BFO photocathodes at 0.58 V vs RHE (b); standard curve of H₂O₂ detection (c); PEC H₂O₂ production activities (d) and FE (e) for H₂O₂ production in KHCO₃ solution.

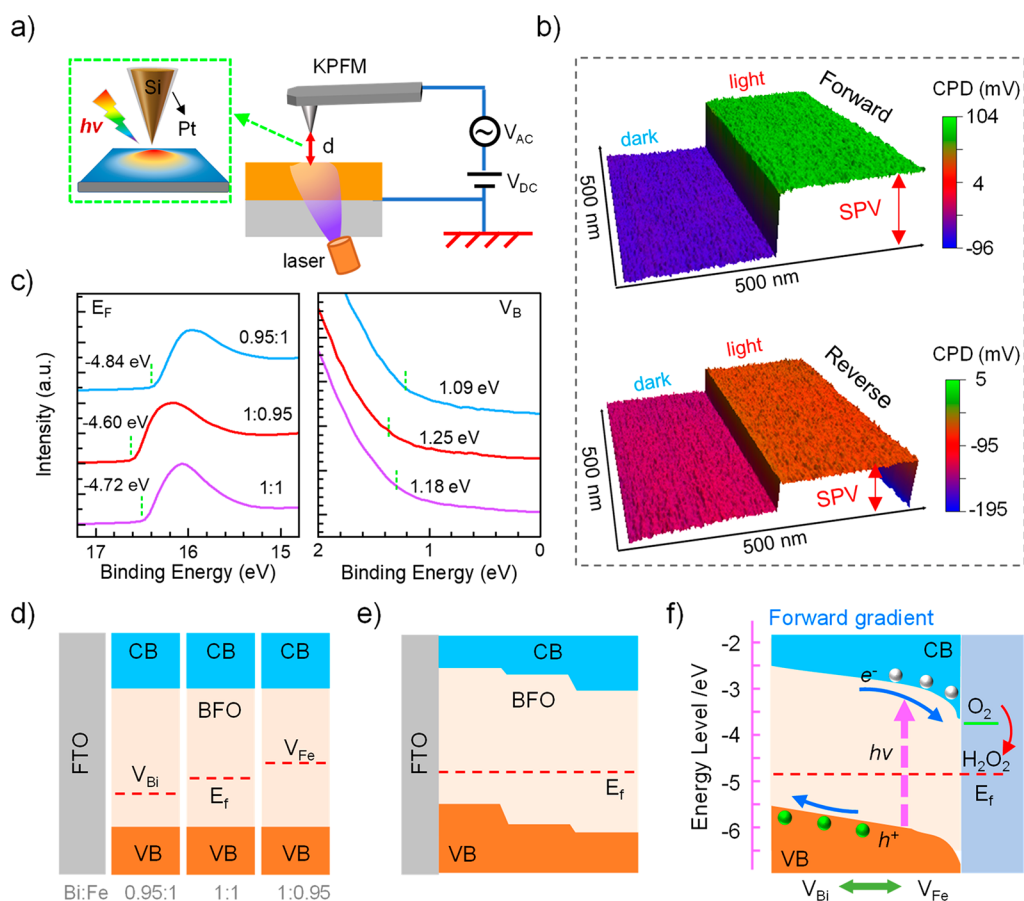


Figure 4. Schematic illustration of the SPV setup (a); CPD maps of the forward and reverse gradient BFO films under blue dark or LED light illumination (b); UPS Fermi level and valence band edge of BFO films (c); schematic band alignments of BFO films before contact (d) and after contact (e), and the formed full-space electric field (f).

191 taking the above results together, we claim successful
 192 fabrication of the gradient BFO films. As the emphasis of
 193 this work is to investigate the effect of gradient doping on the
 194 optoelectronic properties of BFO films, we elected to carefully
 195 evaluate and compare the forward and reversed gradient doped

films to uncover the specific origins of these observations 196
 through a suite of characterization methods. 197

In assessing the BFO films as photocathodes, the perform- 198
 ance was evaluated using standard three-electrodes system 199
 under blue LED (430 nm) illumination at 60 mW·cm⁻² in 0.1 200
 M KHCO₃ solution. An electron scavenger (0.1 M, Na₂S₂O₈) 201

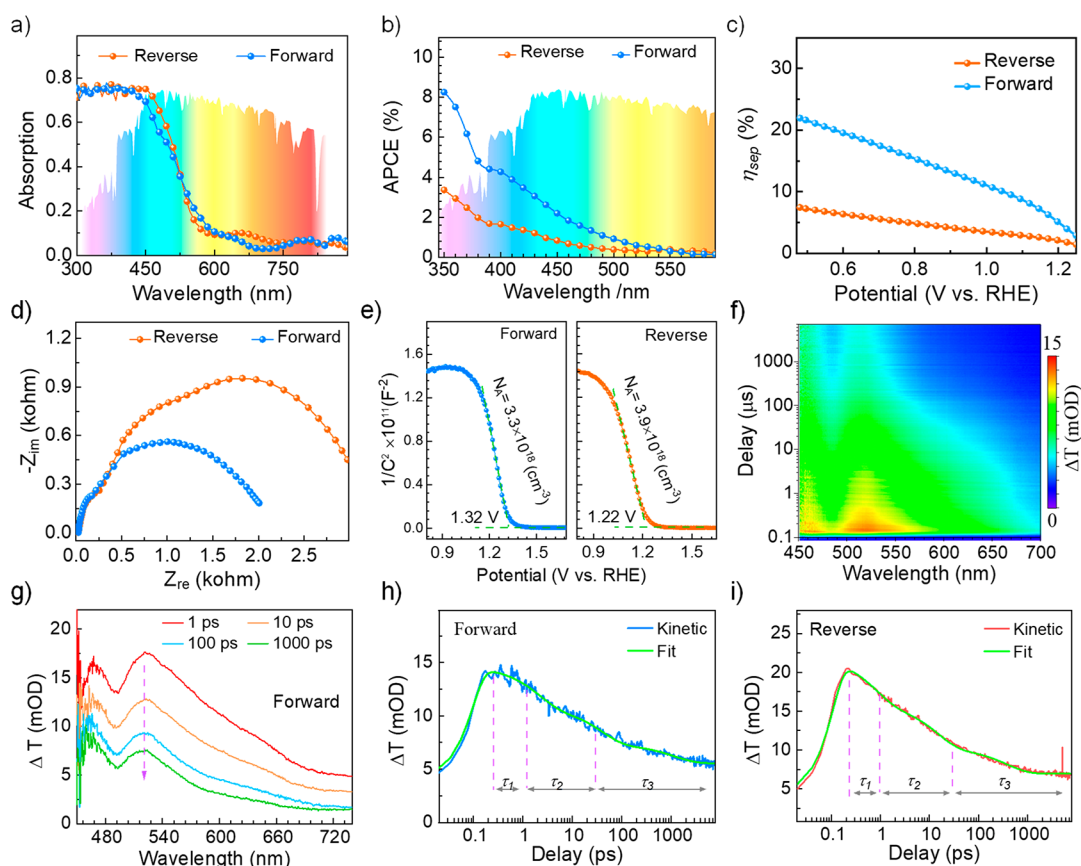


Figure 5. UV-vis absorption spectra (a), calculated APCE spectra (b), and the charge separation efficiency (c) of prepared BFO photocathodes; EIS plots of BFO films measured at 0.68 V vs RHE under blue LED illumination (d); Mott-Schottky plots of BFO films at 2 kHz (e); contour maps of transient transmission spectrum of forward gradient BFO film in the range of -100 fs to 7.4 ns (f); representative transmission spectra of forward gradient BFO film at various pump-probe delays (g); transient transmission decay and related fitting curves of forward (h) and reverse (i) doped BFO films in the range of -100 fs to 7.4 ns. Excitation wavelength: 400 nm. Power: $1.5 \mu\text{J}$ per pulse. Beam diameter: 0.3 mm.

was added to the electrolyte to study the intrinsic properties of photocathodes without the influence of surface activity. Figure 3a shows the J - V responses of homogeneous, forward, and reverse gradient doped BFO films. The forward BFO film exhibits the best PEC performance: a 16% increase in the photocurrent to $-1.02 \text{ mA}\cdot\text{cm}^{-2}$ at 0.5 V vs RHE and a 26% improvement in the fill factor when compared with the homogeneous BFO film ($-0.88 \text{ mA}\cdot\text{cm}^{-2}$ at 0.5 V vs RHE). The reverse gradient doped photocathode shows a negative influence with photocurrent of only 40% of the homogeneous BFO film. To further support defects modulated PEC performance, three homogeneous BFO films with Bi:Fe mole ratios of 1:0.95, 1:1, and 0.95:1 were prepared and evaluated (Figure S4). The photocurrent significantly enhanced about three times from $-0.29 \text{ mA}\cdot\text{cm}^{-2}$ to $-0.88 \text{ mA}\cdot\text{cm}^{-2}$ at 0.5 V vs RHE as the Bi:Fe ratio varied from 1:0.95 to 0.95:1. The onset potential also displays a positive shift from 1.25 to 1.38 V vs RHE (taken at $-10 \mu\text{A}\cdot\text{cm}^{-2}$). Details of J - V response were summarized in Table S4. Incident photon-to-current conversion efficiency (IPCE) further confirmed the enhanced PEC performance through forward gradient doping. As shown in Figure 3b, forward gradient doping barely changes the onset wavelength of BFO films (550 nm) but improves the IPCE value by 43%. PEC performance was further evaluated by testing the H_2O_2 production activity in a homemade two-compartment cell as

illustrated in Figure S5. The BFO photocathode and counter electrode were separated by a Nafion membrane. The electrolyte was 0.1 M KHCO_3 solution and purged with O_2 (99.9%) gas for 30 min before and during the measurement. Figure 3c shows the standard curve of H_2O_2 detection with excellent linear correlation. Figure 3d displays the production of H_2O_2 as a function of time under AM 1.5 irradiation at an applied bias of 0.6 V vs RHE. The corresponding I - t curve is available in Figure S6. Constant with J - V responses, the forward gradient doped BFO photocathode exhibited better performance with an average H_2O_2 concentration of around $380 \text{ mmol}\cdot(\text{L}\cdot\text{m}^2)^{-1}$ within 50 min, about 1.7 times higher than the reverse gradient doped BFO ($90.6 \text{ mmol}\cdot(\text{L}\cdot\text{m}^2)^{-1}$). Additionally, recyclability tests revealed that the system could be used for stable H_2O_2 generation for three cycles without significant decay (Figure 3d). Finally, the faradic efficiency (FE) was calculated based on the equation described in the Supporting Information. As shown in Figure 3e, the similar FE with an average value of 19% indicates that gradient doping rarely influences the reaction selectivity. However, the low FE of BFO films indicates that a highly efficient cocatalyst is a critical need, which may be investigated in the following work.^{1,36,37} In summary, the forward gradient doped BFO films show the best PEC performance both in J - V response and H_2O_2 production activity.

253 The constructed full-space electric field corresponds to the
 254 enhanced PEC performance, as identified by Fermi level
 255 analysis, including surface potential and ultraviolet photo-
 256 electron spectroscopy (UPS).¹³ Figure 4a displays the
 257 schematic illustration of SPV setup, which is based on Kelvin
 258 probe force microscopy (KPFM) and light illumination
 259 system.^{38–40} For data collection, the film was illuminated
 260 from the back side with an LED light source and through
 261 optical fiber. Figure S7a illustrated the origin of the contact
 262 potential difference (CPD). Under illumination, photogen-
 263 erated minority carriers will shift the Fermi level upward at the
 264 interface and the resulting average potential difference
 265 (Δ CPD) in the dark and under illumination is defined as
 266 the surface photovoltage (SPV). Figure 4b and Figure S7b
 267 show the CPD maps and plots through BFO films. In the dark,
 268 the reverse gradient BFO has significantly lower CPD
 269 (approximately -126 mV) than the forward gradient BFO
 270 (approximately -85 mV), indicating that reverse BFO has a
 271 larger work function than the forward BFO at the surface.
 272 Under illumination, both films show clear enhanced CPD, and
 273 the SPV of forward BFO is nearly three times as large as that of
 274 reverse BFO. Since the SPV is mainly determined by the
 275 density of photogenerated minority carriers (electrons in p-
 276 type BFO), such an enhanced SPV indicates that the forward
 277 gradient is beneficial for photocarriers' separation. UPS results
 278 further confirmed the V_{Bi} and V_{Fe} adjusted Fermi level and
 279 valence band (VB) edge position of BFO films. As shown in
 280 Figure 4c, the work function of BFO films significantly
 281 increased from 4.60 to 4.84 eV, and the distance between the
 282 Fermi level and VB edge decreased from 1.25 to 1.09 eV, as the
 283 V_{Bi} concentration increased. These results matched well with
 284 first-principles analysis and confirmed the V_{Bi} and V_{Fe} as
 285 acceptors to modulate the position of the Fermi level.
 286 Altogether, our results illustrate how the gradient V_{Bi} and
 287 V_{Fe} and the resulting gradient of Fermi level lead to an internal
 288 electric field. Before contact, BFO films with different
 289 vacancies density possess different Fermi levels, as shown in
 290 Figure 4d. After contact, the Fermi level will equilibrate by
 291 redistributing the free carriers between the different regions so
 292 that the CB and VB bend at each interface, as illustrated by
 293 Figure 4e. A forward gradient BFO photocathode will contain
 294 a full-space electric field shown in Figure 4f, which would
 295 promote photocarrier separation by field-enhanced diffusion
 296 (drift) of the photogenerated electrons toward the BFO/
 297 electrolyte interface and drift of photogenerated holes toward
 298 the back contact.
 299 The full-space electric field through the film will have an
 300 influence on the carrier dynamics of semiconductors. To verify
 301 it, other effects should be excluded through suitable character-
 302 izations. First, light-harvesting capacities of gradient doped
 303 films have been measured and compared. As shown in Figure
 304 5a, the direction of gradient doping barely influences the light-
 305 harvesting capacity of BFO films. Additionally, absorbed
 306 photon-to-current conversion efficiency (APCE, Figure 5b)
 307 calculated from IPCE and related UV–vis spectrum shows a
 308 similar improvement trend as that of IPCE (Figure 3b),
 309 indicating that the enhanced PEC performance with forward
 310 gradient doping is mainly attributed to improved charge carrier
 311 separation and transport. This enhancement was further
 312 confirmed by the calculated charge separation efficiency. As
 313 shown in Figure 5c, the forward gradient doped photocathode
 314 shows a significant improvement of η_{sep} , 3 times larger than
 315 that of the reversed one.

Electrochemical impedance spectroscopy (EIS) plots in
 Figure 5d indicate that the forward gradient doped BFO film
 shows better charge transport capacity than the reverse BFO
 film, as proved by the significantly reduced charge transport
 resistance. Moreover, Mott–Schottky analysis (Figure 5e)
 indicates that the flat-band potential (E_{fb}) of BFO films was
 significantly modified by the gradient self-doping process. As
 summarized in Table S1, the E_{fb} of forward gradient BFO film
 is a round 1.32 V vs RHE, 100 mV larger than that of reversed
 gradient film (1.22 V vs RHE). Such a positive shift of the E_{fb}
 is a direct evidence of the Fermi level shifting toward the VB
 edge. The similar carrier density in the films indicates that
 forward or reversed gradient does not affect the overall carrier
 density in the film.

Figure 5f and Figure S8a show the transmission contour
 maps of BFO film that contain two main peaks located at 460
 nm (2.65 eV) and 520 nm (2.38 eV) which decrease over time.
 These peaks are attributed to the ligand-to-metal charge
 transition between O-to-Bi (2.27 eV) and O-to-Fe (1.82 eV).¹⁵
 Figure 5g and Figure S8b are the transmission spectra
 extracted from the contour maps of BFO films in the visible
 region at different delay times. The ratio changes in BFO films
 rarely influence the sharpness of spectra. It should be noted
 that the sharp signals from thin film semiconductors are related
 to a modification of the optical properties (the complex index
 of refraction) of the semiconductor as a result of the
 perturbation of the charge carrier band population by the
 pump pulse according to eq 1,^{32,41–44}

$$\Delta n(N, P, \lambda) = - \left(\frac{e^2 \lambda^2}{8\pi^2 c^2 \epsilon_0 n} \right) \left(\frac{\Delta N}{m_e} + \frac{\Delta P}{m_h} \right) \quad (1)$$

where Δn is the change of complex refractive index; ΔN and
 ΔP are the photogenerated electrons and holes density; λ is the
 wavelength of incident light; c is the speed of photons in
 vacuum; m_e and m_h are the effective mass of electrons and
 holes.

The carrier lifetimes of BFO films were extracted from
 reflection decay, as shown in Figure 5h,i. The kinetic traces at
 520 nm with 400 nm excitation were well fitted via eq 2:

$$\Delta A(t) = k_1 e^{-t/\tau_1} + k_2 e^{-t/\tau_2} + k_3 e^{-t/\tau_3} \quad (2)$$

where τ_1 , τ_2 , and τ_3 are time constants and k_1 , k_2 , and k_3 are
 weightings (amplitudes). The first time constant (τ_1) is related
 to hot carrier cooling through carrier–carrier scattering and
 momentum relaxation, which is commonly less than 1 ps. The
 second time constant (τ_2) is associated with energy relaxation
 involving carrier-phonon scattering. The third time constant is
 attributed to the trapping and recombination of charge
 carriers.^{45–47} The final carrier lifetime was calculated by
 summing these three time constants with weights. Table S5
 shows the fitting parameters of BFO films. The total carrier
 lifetime of forward BFO film (201 ps) is almost 1.3 times as
 large as that of reversed film (153 ps), which further indicates
 that the forward gradient doping is an efficient approach to
 prolong the carrier lifetime and inhibit carrier recombination.
 Thus, we conclude that the full-space electric field through the
 BFO films improved PEC performance by enhanced carrier
 dynamics, including improved charge separation, transport,
 and prolonged lifetime.

In summary, a BFO-based photocathode with improved
 photocarriers collection was achieved by constructing a full-
 space electric field through the film. First-principles calcu-

lations revealed that Fe vacancies act as acceptor impurities in BFO, which can modulate the Fermi level. TOF-SIMS depth profile and XPS analysis showed the gradient distribution of Fe vacancies within BFO films and illustrated that its concentration can be modified by varying the Fe ratio in the precursor. The J - V response and H_2O_2 production tests show clearly enhanced PEC performance by forward gradient doping, which improved the photocurrent by 2 times, the onset potential by 130 mV, and H_2O_2 production by 69%, compared to the reverse gradient doping approach. SPV and UPS analyses show the modified Fermi level and reveal the construction of gradient homojunction and full-space electric field. EIS, Mott-Schottky, and transient spectroscopy analyses indicate the forward full-space electric field as additional driving force that significantly promoted photocarriers collection and prolonged carriers' lifetime. These results demonstrate that deeper understanding of electric structures of materials and constructing full-space electric field through gradient self-doping are a viable approach to promote photocarriers collection, which deserves great attention and wide applications.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acseenergylett.2c01750>.

Detailed experimental section, maximum photocurrent, XPS spectra, SEM images, EDX map, J - V response, optical photos, Tauc plot, and band energy diagram (PDF)

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Notes

The authors declare no competing financial interest.

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REFERENCES

- (1) Gao, R.; Pan, L.; Li, Z.; Shi, C.; Yao, Y.; Zhang, X.; Zou, J. J. Engineering Facets and Oxygen Vacancies over Hematite Single Crystal for Intensified Electrocatalytic H_2O_2 Production. *Adv. Funct. Mater.* **2020**, *30* (24), 1910539.
- (2) Liu, T.; Pan, Z.; Vequizo, J. J. M.; Kato, K.; Wu, B.; Yamakata, A.; Katayama, K.; Chen, B.; Chu, C.; Domen, K. Overall Photosynthesis of H_2O_2 by An Inorganic Semiconductor. *Nat. Commun.* **2022**, *13* (1), 1034.
- (3) Liu, J.; Zou, Y.; Jin, B.; Zhang, K.; Park, J. H. Hydrogen Peroxide Production from Solar Water Oxidation. *ACS Energy Lett.* **2019**, *4* (12), 3018–3027.
- (4) Cheng, W.-H.; de la Calle, A.; Atwater, H. A.; Stechel, E. B.; Xiang, C. Hydrogen from Sunlight and Water: A Side-by-Side Comparison between Photoelectrochemical and Solar Thermochemical Water-Splitting. *ACS Energy Lett.* **2021**, *6* (9), 3096–3113.
- (5) Li, C.; He, J.; Xiao, Y.; Li, Y.; Delaunay, J.-J. Earth-Abundant Cu-Based Metal Oxide Photocathodes for Photoelectrochemical Water Splitting. *Energy Environ. Sci.* **2020**, *13* (10), 3269–3306.
- (6) Piekner, Y.; Ellis, D. S.; Schleuning, M.; Grave, D. A.; Schnell, P.; Dotan, H.; Abdi, F.; Rothschild, A. External Quantum Efficiency Spectra of $BiVO_4$ Thin Film Photoanodes under Bias Illumination. *J. Electrochem. Soc.* **2022**, *169*, 046513.
- (7) Piekner, Y.; Ellis, D. S.; Grave, D. A.; Tsyganok, A.; Rothschild, A. Wasted Photons: Photogeneration Yield and Charge Carrier Collection Efficiency of Hematite Photoanodes for Photoelectrochemical Water Splitting. *Energy Environ. Sci.* **2021**, *14*, 4584–4598.
- (8) Zhang, K.; Lu, Y.; Zou, Q.; Jin, J.; Cho, Y.; Wang, Y.; Zhang, Y.; Park, J. H. Tuning Selectivity of Photoelectrochemical Water Oxidation via Facet-Engineered Interfacial Energetics. *ACS Energy Lett.* **2021**, *6* (11), 4071–4078.
- (9) Segev, G.; Jiang, C.-M.; Cooper, J. K.; Eichhorn, J.; Toma, F. M.; Sharp, I. D. Quantification of the Loss Mechanisms in Emerging Water Splitting Photoanodes through Empirical Extraction of the Spatial Charge Collection Efficiency. *Energy Environ. Sci.* **2018**, *11* (4), 904–913.
- (10) Gottesman, R.; Peracchi, I.; Gerke, J. L.; Irani, R.; Abdi, F. F.; van de Krol, R. Shining a Hot Light on Emerging Photoabsorber Materials: The Power of Rapid Radiative Heating in Developing Oxide Thin-Film Photoelectrodes. *ACS Energy Lett.* **2022**, *7* (1), 514–522.
- (11) Jiang, X.; Cheng, X.; Zhang, Z.; Chen, T.; Tao, K.; Han, W. Computation-Assisted Performance Optimization for Photoelectrochemical Photoelectrodes. *Appl. Phys. Lett.* **2022**, *120* (6), 063901.
- (12) Zhang, S.; Wang, J.; Wen, S.; Jiang, M.; Xiao, H.; Ding, X.; Wang, N.; Li, M.; Zu, X.; Li, S.; Yam, C.; Huang, B.; Qiao, L. Approaching Charge Separation Efficiency to Unity without Charge Recombination. *Phys. Rev. Lett.* **2021**, *126* (17), 176401.

- 498 (13) Zhang, Z.; Lindley, S. A.; Guevarra, D.; Kan, K.; Shinde, A.;
499 Gregoire, J. M.; Han, W.; Xie, E.; Haber, J. A.; Cooper, J. K. Fermi
500 Level Engineering of Passivation and Electron Transport Materials for
501 p-Type CuBi_2O_4 Employing a High-Throughput Methodology. *Adv.*
502 *Funct. Mater.* **2020**, *30* (24), 2000948.
- 503 (14) Zhang, Z.; Lindley, S. A.; Dhall, R.; Bustillo, K.; Han, W.; Xie,
504 E.; Cooper, J. K. Beneficial CuO Phase Segregation in the Ternary p-
505 Type Oxide Photocathode CuBi_2O_4 . *ACS Appl. Energy Mater.* **2019**, *2*
506 (6), 4111–4117.
- 507 (15) Zhang, Z.; Tan, B.; Ma, W.; Liu, B.; Sun, M.; Cooper, J. K.;
508 Han, W. BiFeO_3 Photocathodes for Efficient H_2O_2 Production via
509 Charge Carrier Dynamics Engineering. *Mater. Horiz.* **2022**, *9*, 1999–
510 2006.
- 511 (16) Tan, B.; Liu, B.; Sun, M.; Li, Y.; Cao, Z.; Zhang, Z. Enhanced
512 Charge Collection and Surface Activity of a CuBi_2O_4 Photocathode
513 via Crystal Facet Engineering. *J. Mater. Chem. A* **2022**, *10* (17),
514 9427–9434.
- 515 (17) Li, R.; Zhang, F.; Wang, D.; Yang, J.; Li, M.; Zhu, J.; Zhou, X.;
516 Han, H.; Li, C. Spatial Separation of Photogenerated Electrons and
517 Holes Among $\{010\}$ and $\{110\}$ Crystal Facets of BiVO_4 . *Nat.*
518 *Commun.* **2013**, *4*, 1432.
- 519 (18) Sun, M.; Liu, B.; Han, W.; Zhang, Z.; Xie, M. CuBi_2O_4
520 Photocathode with Integrated Electric Field for Enhanced H_2O_2
521 Production. *Appl. Catal., B* **2022**, *304*, 120980.
- 522 (19) Liu, B.; Sun, M.; Tan, B.; Zhang, Z.; Han, W. Enhanced
523 Photocarrier Collection in Bismuth Vanadate Photoanode through
524 Modulating the Inner Potential Distribution. *Adv. Opt. Mater.* **2022**,
525 *10*, 2200046.
- 526 (20) Tian, Y.; Wang, A.; Wei, Y.; Pei, M.; Cao, R.; Gu, Z.; Yuan, Q.;
527 Hu, Y.; Wang, J.; Liu, K.; Shang, D.; Niu, J.; An, X.; Long, R.; Zhang,
528 J. Large-Area Printing of Ferroelectric Surface and Super-Domain for
529 Solar Water Splitting. *Adv. Funct. Mater.* **2022**, *32*, 2111180.
- 530 (21) Tian, Y.; Wei, L.; Zhang, Q.; Huang, H.; Zhang, Y.; Zhou, H.;
531 Ma, F.; Gu, L.; Meng, S.; Chen, L. Q.; Nan, C. W.; Zhang, J. Water
532 Printing of Ferroelectric Polarization. *Nat. Commun.* **2018**, *9* (1),
533 3809.
- 534 (22) Choi, P.-P.; Cojocaru-Mirédin, O.; Wuerz, R. Compositional
535 Gradients and Impurity Distributions in CuInSe_2 Thin-Film Solar
536 Cells Studied by Atom Probe Tomography. *Surf. Interface Anal.* **2012**,
537 *44* (11–12), 1386–1388.
- 538 (23) Zhang, H.; Li, D.; Byun, W. J.; Wang, X.; Shin, T. J.; Jeong, H.
539 Y.; Han, H.; Li, C.; Lee, J. S. Gradient Tantalum-Doped Hematite
540 Homojunction Photoanode Improves Both Photocurrents and Turn-
541 on Voltage for Solar Water Splitting. *Nat. Commun.* **2020**, *11* (1),
542 4622.
- 543 (24) Mirzaei, A.; Eddah, M.; Roualdes, S.; Ma, D.; Chaker, M.
544 Multiple-Homojunction Gradient Nitrogen Doped TiO_2 for Photo-
545 catalytic Degradation of Sulfamethoxazole, Degradation Mechanism,
546 and Toxicity Assessment. *Chem. Eng. J.* **2021**, *422*, 130507.
- 547 (25) Wang, F.; Septina, W.; Chemseddine, A.; Abdi, F. F.; Friedrich,
548 D.; Bogdanoff, P.; van de Krol, R.; Tilley, S. D.; Berglund, S. P.
549 Gradient Self-Doped CuBi_2O_4 with Highly Improved Charge
550 Separation Efficiency. *J. Am. Chem. Soc.* **2017**, *139*, 15094–15103.
- 551 (26) Xiao, Y.; Feng, C.; Fu, J.; Wang, F.; Li, C.; Kunzelmann, V. F.;
552 Jiang, C.-M.; Nakabayashi, M.; Shibata, N.; Sharp, I. D.; Domen, K.;
553 Li, Y. Band Structure Engineering and Defect Control of Ta_3N_5 for
554 Efficient Photoelectrochemical Water Oxidation. *Nat. Catal.* **2020**, *3*
555 (11), 932–940.
- 556 (27) Paudel, T. R.; Jaswal, S. S.; Tsymbal, E. Y. Intrinsic Defects in
557 Multiferroic BiFeO_3 and Their Effect on Magnetism. *Phys. Rev. B*
558 **2012**, *85*, 104409.
- 559 (28) Shimada, T.; Matsui, T.; Xu, T.; Arisue, K.; Zhang, Y.; Wang, J.;
560 Kitamura, T. Multiferroic nature of intrinsic point defects in BiFeO_3 ;
561 A hybrid Hartree-Fock density functional study. *Phys. Rev. B* **2016**, *93*
562 (17), 174107.
- 563 (29) McDonnell, K. A.; Wadnerkar, N.; English, N. J.; Rahman, M.;
564 Dowling, D. Photo-active and Optical Properties of Bismuth Ferrite
565 (BiFeO_3): An Experimental and Theoretical Study. *Chem. Phys. Lett.*
566 **2013**, *572*, 78–84.
- (30) Geneste, G.; Paillard, C.; Dkhil, B. Polarons, Vacancies, 567
Vacancy Associations, and Defect States in Multiferroic BiFeO_3 . *Phys.* 568
Rev. B **2019**, *99* (2), 024104. 569
- (31) Lee, J.; Nahm, H.-H.; Kim, Y.-H. Diversity of Hole-Trap 570
Centers Due to Small Polarons and Bipolarons in Ca-doped BiFeO_3 : 571
Origin of Electrochromism. *Phys. Rev. B* **2020**, *101* (1), 014110. 572
- (32) Zhang, Z.; Liu, B.; Quinteros, F.; Zhai, X.; Wang, Q.; Han, W.; 573
Xie, E.; Reyes-Lillo, S. E.; Cooper, J. K. Understanding the Role of 574
Oxygen and Hydrogen Defects in Modulating the Optoelectronic 575
Properties of P-Type Metal Oxide Semiconductors. *Chem. Mater.* 576
2021, *33* (19), 7829–7838. 577
- (33) Yang, J. H.; Yin, W. J.; Park, J. S.; Ma, J.; Wei, S. H. Review on 578
First-Principles Study of Defect Properties of CdTe as a Solar Cell 579
Absorber. *Semicond. Sci. Technol.* **2016**, *31* (8), 083002. 580
- (34) Jeon, N.; Rout, D.; Kim, I. W.; Kang, S.-J. L. Enhanced 581
Multiferroic Properties of Single-Phase BiFeO_3 Bulk Ceramics by Ho 582
Doping. *Appl. Phys. Lett.* **2011**, *98* (7), 072901. 583
- (35) Yang, T.; Wei, J.; Guo, Y.; Lv, Z.; Xu, Z.; Cheng, Z. 584
Manipulation of Oxygen Vacancy for High Photovoltaic Output in 585
Bismuth Ferrite Films. *ACS Appl. Mater. Interfaces* **2019**, *11* (26), 586
23372–23381. 587
- (36) Zheng, L.; Su, H.; Zhang, J.; Walekar, L. S.; Vafaei 588
Molamahmood, H.; Zhou, B.; Long, M.; Hu, Y. H. Highly Selective 589
Photocatalytic Production of H_2O_2 on Sulfur and Nitrogen co-doped 590
Graphene Quantum Dots Tuned TiO_2 . *Appl. Catal., B* **2018**, *239*, 591
475–484. 592
- (37) Yuan, Q.; Zhao, J.; Mok, D. H.; Zheng, Z.; Ye, Y.; Liang, C.; 593
Zhou, L.; Back, S.; Jiang, K. Electrochemical Hydrogen Peroxide 594
Synthesis from Selective Oxygen Reduction over Metal Selenide 595
Catalysts. *Nano Lett.* **2022**, *22* (3), 1257–1264. 596
- (38) Jiang, Y.; Chen, H. Y.; Li, J. Y.; Liao, J. F.; Zhang, H. H.; Wang, 597
X. D.; Kuang, D. B. Z-Scheme 2D/2D Heterojunction of $\text{CsPbBr}_3/\text{Bi}_2\text{WO}_6$ 598
for Improved Photocatalytic CO_2 Reduction. *Adv. Funct.* 599
Mater. **2020**, *30* (50), 2004293. 600
- (39) Ben, H.; Liu, Y.; Liu, X.; Ling, C.; Liang, C.; Zhang, L. 601
Diffusion-Controlled Z-Scheme-Steered Charge Separation across 602
PDI/ BiOI Heterointerface for Ultraviolet, Visible, and Infrared Light- 603
Driven Photocatalysis. *Adv. Funct. Mater.* **2021**, *31* (37), 2102315. 604
- (40) Woo, G.; Lee, D. H.; Heo, Y.; Kim, E.; On, S.; Kim, T.; Yoo, H. 605
Energy-Band Engineering by Remote Doping of Self-Assembled 606
Monolayers Leads to High-Performance IGZO/p-Si Heterostructure 607
Photodetectors. *Adv. Mater.* **2022**, *34* (6), 2107364. 608
- (41) Xu, Z.; Hou, B.; Zhao, F.; Cai, Z.; Shi, H.; Liu, Y.; Hill, C. L.; 609
Musaev, D. G.; Mecklenburg, M.; Cronin, S. B.; Lian, T. Nanoscale 610
 TiO_2 Protection Layer Enhances the Built-In Field and Charge 611
Separation Performance of GaP Photoelectrodes. *Nano Lett.* **2021**, *21* 612
(19), 8017–8024. 613
- (42) Sabbah, A. J.; Riffe, D. M. Femtosecond Pump-probe 614
Reflectivity Study of Silicon Carrier Dynamics. *Phys. Rev. B* **2002**, 615
66, 165217. 616
- (43) Prabhu, S. S.; Vengurlekar, A. S. Dynamics of the Pump-probe 617
Reflectivity Spectra in GaAs and GaN. *J. Appl. Phys.* **2004**, *95* (12), 618
7803–7812. 619
- (44) Cooper, J. K.; Reyes-Lillo, S. E.; Hess, L. H.; Jiang, C.-M.; 620
Neaton, J. B.; Sharp, I. D. Physical Origins of the Transient 621
Absorption Spectra and Dynamics in Thin-Film Semiconductors: The 622
Case of BiVO_4 . *J. Phys. Chem. C* **2018**, *122* (36), 20642–20652. 623
- (45) Shenje, L.; Larson, S.; Zhao, Y.; Ullrich, S. Composition Effects 624
on Ultrafast Optical Properties of Cu_xO_y Thin Films: A Transient 625
Absorption Study. *J. Phys. Chem. C* **2020**, *124* (45), 24908–24918. 626
- (46) Fischer, M. C.; Wilson, J. W.; Robles, F. E.; Warren, W. S. 627
Invited Review Article: Pump-probe microscopy. *Rev. Sci. Instrum.* 628
2016, *87* (3), 031101. 629
- (47) Born, B.; Krupa, J. D. A.; Geoffroy-Gagnon, S.; Hristovski, I. R.; 630
Collier, C. M.; Holzman, J. F. Ultrafast Charge-Carrier Dynamics of 631
Copper Oxide Nanocrystals. *ACS Photonics* **2016**, *3* (12), 2475–2481. 632