Trehalose- and glucose-derived glycoamphiphiles: small-molecule and nanoparticle Toll-like Receptor

4 (TLR4) modulators

Julio Rodriguez Lavado, † Stefania E. Sestito, † Roberto Cighetti, † Eva M. Aguilar Moncayo, † Alja
Oblak, § Duško Lainšček, § José Luis Jiménez Blanco, † José Manuel García Fernández, ¶ Carmen
Ortiz Mellet, † Roman Jerala, § Valentina Calabrese, † Francesco Peri* †

[†] Department of Organic Chemistry, Faculty of Chemistry, University of Sevilla, E-41012 Sevilla, Spain.

[‡]Department of Biotechnology and Biosciences, University of Milano Bicocca; Piazza della Scienza, 2; 20126 Milano, Italy.

§ National Institute of Chemistry, Dept. of Biotechnology; Hajdrihova 19; SI-1001 Ljubljana and EN-FIST center of excellence, SI-1000, Ljubljana, Slovenia.

[¶]Instituto de Investigaciones Químicas (IIQ), CSIC - Univ. Sevilla, E-41092 Sevilla, Spain.

KEYWORDS: TLR4, MD-2, CD14, lipid A, LPS, trehalose, glucose, glycolipids, medicinal chemistry, drug development, gold nanoparticles, HEK cells, *in vivo* activity.

ABSTRACT. An increasing number of pathologies have been linked to Toll-like Receptor 4 (TLR4) activation and signaling. New hit and lead compounds that target the TLR4 activation

process are urgently needed. We report on the synthesis and biological properties of protonatable glycolipids based on glucose and trehalose scaffolds. Some of these compounds potently inhibit TLR4 activation and signaling in cells transfected with human and mouse variants of TLR4, MD-2 and CD14 receptors and are active in inhibiting TLR4-dependent cytokine production *in vivo*. Structure-activity relationship studies indicate that the presence of fatty ester chains in the molecule is a primary prerequisite for biological activity in this family of compounds and points to facial amphiphilicity as a preferred architecture for TLR4 antagonist design. Experimental data also support that single molecules, and not micellar aggregates, are the active species.

Because the cationic glycolipids here presented exhibit negligible toxicity at their active concentrations, they can be cosidered as new lead compounds for the development of drugs targeting TLR4 activation and signaling in infectious, inflammatory and autoimmune diseases. Interestingly, the biological activity of the best drug candidate was retained after adsorption at the surface of colloidal gold nanoparticles, broadening the options for clinical development.

INTRODUCTION

Toll-like receptors (TLRs) play a critical role in the recognition of conserved pathogen-associated molecular patterns (PAMPs) derived from various microbial pathogens including viruses, bacteria, protozoa and fungi, and in the subsequent initiation of innate immune response. Among TLRs, TLR4 selectively responds to bacterial endotoxin (E), composed by bacterial lipopolysaccharides (LPS) or part of it (lipooligosaccharides, LOS, lipid A). LPS is the main chemical component of the Gram negative bacteria outer membrane, and the lipid A, a negatively charged phosphorylated lipodisaccharide represents the LPS moiety that is responsible for TLR4 activation through specific molecular recognition processes (Figure 1).

TLR4 is also activated by endogenous molecules, generally known as danger-associated molecular patterns (DAMPs).4 Typical DAMPs acting as TLR4 agonists are released as a consequence of injury and inflammation. Most of the reported DAMPs, are proteins, which are very different from lipid A, and the molecular details of DAMP interaction with the TLR4 receptorial system are still unknown, although in some cases endotoxin contamination seems to be responsible for TLR4 activity of DAMPs. Chemical entities that block TLR4 activation by bacterial endotoxin (LPS) thus acting as antagonists, are hit compounds for developing drugs active against acute sepsis and septic shock derived from excessive and deregulated TLR4 activation and signaling.⁵ On the other hand, the inhibition of TLR4 stimulation by DAMPs could be used to contrast a wide range of inflammatory and autoimmune disorders associated to the release of inflammatory cytokines. In this context, TLR4 is an emerging molecular target related to an impressively broad spectrum of modern day disorders including autoimmune disorders, chronic inflammations, allergies, asthma, atherosclerosis, aortic aneurysm, CNS diseases such as neuropathic pain, Amyotrophic Lateral Sclerosis (ALS), Alzheimer's Disease (AD), and some types of cancer.⁶ As the majority of these pathologies still lack specific pharmacological treatment, small molecules active in inhibiting TLR4 activation have attracted increasing interest in a wide range of possible clinical settings.⁷

The molecular mechanism by which endotoxin activate TLR4 is a complex process⁸ and depends on LPS binding protein (LBP)⁹-catalyzed extraction and transfer of individual LPS molecules from aggregated LPS to the CD14 (cluster of differentiation 14) receptor,¹⁰ and then from CD14 to myeloid differentiation protein 2 (MD-2).^{11, 12} This process is followed by engagement and dimerization of TLR4 thus forming the cell surface complex (LPS:MD-2:TLR4)₂,¹³ which

initiates the intracellular signaling by recruiting specific adaptor proteins and activating downstream signaling pathways.

Several natural and synthetic small molecules are known to modulate TLR4 activation and subsequent intracellular signaling acting as agonists (activators) or antagonists (inhibitors).⁶ The majority of these molecules are lipid A variants and synthetic lipid A mimetics that reproduce the structural motif of the anionic disaccharide. Monophosphoryl lipid A (MPL) and some aminoalkyl glucosaminide phosphates (AGPs) are agonists in use as vaccine adjuvants, 14 while underacylated variants such as natural lipid IVa¹⁵ and synthetic Eritoran^{®16} are antagonists (Figure 1). In general Lipid A variants are anionic lipids, bearing one or two negatively charged phosphate groups and a hydrophobic domain (lipid chains). Although counterintuitive, several cationic lipids made of positively charged headgroups (usually tertiary or quaternary ammonium salts or polyamines) and a hydrophobic domain (alkyl chains or steroids) have been found to be active in modulating TLR4 activity, 17 acting either as agonists or antagonists of TLR4. Thus, some positively charged liposomes formed by cationic amphiphiles induce the expression of proinflammatory mediators. For instance, diC14-amidine (Figure 1) liposomes trigger the secretion of a cytokine pattern reminiscent of the TLR4-dependent LPS secretion pattern by activating both MyD88/NF-κB/JNK and TRAM/TRIF pathways. 18 Other cationic lipids activate cytokine production through NF-κB-independent, TRIF-dependent pathways, which requires the presence of CD14. 19, 20 Structural changes make cationic lipids switch from agonism to antagonism, as in the case of dioleovl trimethylammonium propane (DOTAP), that inhibits TLR4 signal by competing with LPS for interaction with LBP or/and CD14.²⁰ Complexes of the commercial cationic lipid formulation Lipofectamine® with LPS reduce its TLR4 activity. Interestingly LPS complexed with Lipofectamine® co-localizes with CD14 at the cell surface and inside cells, but

does not co-localizes with TLR4'MD-2 complex, suggesting that the mechanism of inhibition may result from the uncoupling of CD14 from TLR4'MD-2.²¹

Specific binding of amino glycolipids and aromatic ammonium salts to CD14 (compounds IAXO-101, -102, -103, Figure 1), was recently shown by our group.^{22, 23} These compounds are active in inhibiting LPS-stimulated TLR4-dependent cytokine production in cells and in animals.²⁴

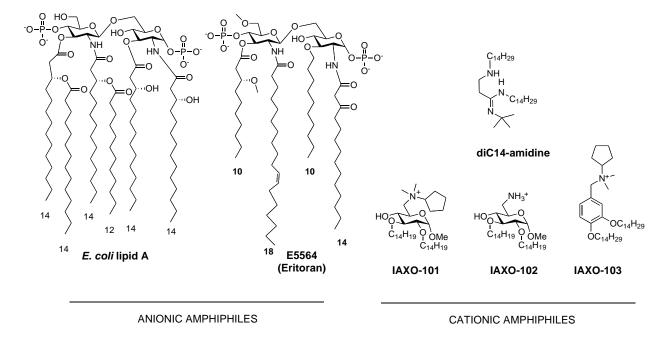


Figure 1. Anionic and cationic TLR4 modulators. From the left: lipid A from *E. coli*, the natural TLR4 agonist, synthetic anionic (the antagonist Eritoran[®]) and cationic amphiphiles (diC14-amidine, IAXO compounds).

Evaluation of transfer of LOS from the monomeric soluble form of CD14 (sCD14) to His₆-tagged CD14 or MD-2 by co-capture to a metal chelating resin clearly showed that the cationic lipids derived from D-glucose or benzylamine inhibit the transfer of LOS from sCD14 to CD14-His₆, but not the transfer of LOS from sCD14 to MD-2.²² Finally, saturation transfer difference (STD) NMR data demonstrated direct binding of the cationic lipids to CD14, through alkyl

chains mainly.²² Altogether, these data suggest that the lipid tails of cationic amphiphiles interact with the hydrophobic binding site of CD14²⁵ and compete with LPS or LOS chains. The carbohydrate scaffold in amino glycolipids probably acts by preventing random conformations and providing a favorable orientation of the lipid chains that is reminiscent of that found in lipid A. Most interestingly, through the interplay of regioselective functionalization methodologies and conformational bias, the installation of differentiated cationic and hydrophobic domains in carbohydrate platforms can be made compatible with molecular diversity-oriented strategies and structure-activity relationship (SAR) studies. As a proof of concept, we have now prepared a series of new cationic glycoamphiphiles using the monosaccharide methyl α -D-glucopyranoside and the disaccharide α , α '-trehalose as the sugar cores. Systematic modification of the cationic heads and the lipophilic tails and evaluation of their capacity to interfere with TLR4 activation and signaling *in vitro* and *in vivo* allowed the identification of a drug lead that has been further incorporated in gold nanoparticles to test the effect of multivalent presentation on its biological activity.

RESULTS AND DISCUSSION

Ligand-based rational design. It is known that the self-assembling capabilities and the ability of cationic amphiphiles to interact with CD14 as liposomes or micellar aggregates have a strong impact on their TLR4 modulatory activity.¹⁷ Yet, very little is still known on the molecular aspects underlying the mechanisms at play, mainly because structural data of cationic compounds bound to MD-2 or CD14 receptors are still lacking. Conducting SAR studies on series of homologous cationic amphiphiles and relating the biological activity to the aggregation properties is expected to provide new insights in this matter. Glucose-derived cationic glycolipids are particularly appealing for this purpose. First, the secondary hydroxyls of the

glucopyranose ring are well suited anchoring points to link lipophilic chains in a similar orientation as the fatty acid acyl chains in lipid A. Second, the incorporation of protonatable headgroups at the primary position imparts facial amphiphilicity to the molecule, a biomimetic feature that is associated to improved cell-membrane crossing abilities and proneness to form supramolecular complexes with complementary biomolecules by either electrostatic or hydrophobic interactions.²⁶ On these grounds, we have now synthesized amino glycolipids derived from methyl α -D-glucopyranoside (1-6) and from α,α' -trehalose (7-11, Figure 2). The latter can be formally considered as dimeric homologues of the glucose amphiphiles. We keep in mind that the confluence of two exoanomeric effects at the 1-1 interglycosidic linkage in α,α' trehalose strongly limits rotation about the glycosidic bonds.²⁷ preserving a rigid conformation that warrants facial anisotropy after differential functionalization at the primary and secondary positions, even in highly constrained constructs. ²⁸ Both the methyl α-D-glucopyranoside and the α,α '-trehalose scaffolds have previously demonstrated their efficiency in the design of TLR4 modulators with anionic amphiphilic structures.^{29,30} Structural modifications have been projected by varying the number, the nature and the length of the lipid chains and the number and disposition of amino groups, in order to evaluate how these structural elements influence the TLR4 activity.

Figure 2. Synthetic monosaccharide and disaccharide protonatable amphiphiles derived, respectively, from D -glucose (1-6) and α,α' -trehalose (7-11).

Synthesis of glucose-derived cationic glycolipids. The syntheses of the tri-*O*-alkylated 6-amino-6-deoxyglucoside derivatives **1** and **2** (Scheme 1) were successfully accomplished by reaction of the known methyl 6-azido-6-deoxy-α-D-glucopyranoside **12**³¹ with hexyl or tetradecyl bromide and sodium hydride (**13** and **14**) followed by reduction of the azido group by either catalytic hydrogenation or Staudinger reaction with triphenylphosphine and hydrolysis of the corresponding phosphazene intermediate.³² The target ether-type amino glycolipids **1** and **2** were isolated as the corresponding hydrochloride salts.

Scheme 1. Syntheses of monosaccharides 1-4.

Reagents and conditions: a) 1-bromohexane, NaH, DMF, overnight, 48% or 1-bromotetradecane, NaH, DMF, 55 °C, 52%; b) H₂, Pd/C, MeOH, 2 h, 87%; or PPh₃, THF; then, NH₄OH, 50 °C, overnight, 82%;c) 1 M LiAlH₄ in THF, AlCl₃, DCM, Et₂O, 83%; d) I₂, PPh₃, imidazole, toluene, 94%; e) HS(CH₂)₂NHBoc, Cs₂CO₃, DMF, 60 °C, 99%; f) 1:1 TFA-DCM, 80%; g) AlCl₃, DCM, Et₂O, 87%; h) I₂, PPh₃, imidazole, toluene, 91%; i) HS(CH₂)₂NHBoc, Cs₂CO₃, DMF, 60 °C, 95%; j) 1:1 TFA-CH₂Cl₂, quant.

The 2,3-di-O-hexyl and -O-tetradecyl glucose derivatives 3 and 4 were synthesized from the corresponding 4.6-O-(p-methoxybenzylidene) protected precursor 15 and 16³³ respectively. which at their turn were obtained by standard alkylation of methyl 4,6-O-(pmethoxybenzylidene)-α-D-glucopyranoside.³⁴ The regioselective opening of the acetal ring of 15 using lithium aluminium hydride (LiAlH₄) gave the 4-O-p-methoxybenzyl (PMB) ether 17. Compound 16 on his side was completely deprotected on C-4 and C-6 positions using aluminium trichloride, affording compound 20. Iodination of the C-6 hydroxyl groups of 17 and 20 using Garegg's conditions³⁵ afforded compounds 18 and 21, which were subjected to cesium carbonate-promoted nucleophilic displacement with t-butoxycarbonyl (Boc)-protected cysteamine (20 and 22) and final Boc removal in acidic conditions to give compounds 3 and 4. The synthetic routes to obtain compounds 5 and 6, having a tertiary and two primary amino groups in the cationic head (Scheme 2), are based on the thiourea-forming and the copper(I)catalyzed azide-alkyne cycloaddition (CuAAC) reactions, two "click"-type ligation strategies already proven very efficient to generate polycationic clusters.³⁶ The preparation of 5 started by hexanoylation of methyl 6-deoxy-6-iodo-α-D-glucopyranoside 23³⁵ (24; Scheme 2a) followed by nucleophilic displacement of the iodine by Boc-protected cysteamine (25) and carbamate hydrolysis, affording the cysteaminyl derivative 26 in 92% overall yield. Condensation of 26

with 2-[*N*,*N*-bis(2-(*N*-*tert*-butoxycarbonylamino)ethylamino]ethyl isothiocyanate³⁷ and final acid-promoted Boc deprotection provided **5**, which was isolated as the dihydrochloride salt. **Scheme 2**. Syntheses of monosaccharides **5** (2a) and **6** (2b).

Reagents and conditions (2a). a) Hexanoic anhydride, DMAP, DMF, Ar, rt, 4 h, 55%; b) *tert*-butyl *N*-(2-mercaptoethyl)carbamate, Cs₂CO₃, DMF, Ar, overnight, 85%; c) 1:1 TFA-DCM, quant.; d) 2-[*N*,*N*-bis(2-(*N*-tert-butoxyaminocarbonyl)ethylamino]ethyl isothiocyanate, Et₃N, DCM, Ar, overnight, 50%; e) 1:1 TFA-DCM, rt, 1 h, quant. **Reagents and conditions (2b)**. a)

hexanoic anhydride, DMAP, DMF, Ar, 4 h, 55% b) 3-bis[2-(*tert*-butoxycarbonylamino)ethyl]propargylamine, Si-BPA·Cu⁺, 9:1 H₂O/^tBuOH, 85 °C, 36 h. 78%; c) 1:1 TFA/H₂O, rt, 1 h, quant; d) 2-(*N-tert*-butoxyaminocarbonyl)ethyl isothiocyanate, Et₃N, DCM, overnight, 52%; e) 1:1 TFA/H₂O, rt, 1 h, quant.

The cationic amphiphile **6** was obtained in good yield from azide **12** following a divergent synthetic strategy in which the hydrophobic and cationic domains are sequentially installed onto the glucopyranoside scaffold. Acylation of **12** with hexanoic anhydride and *N,N*-dimethylaminopyridine (DMAP) in DMF afforded triester **28** (Scheme 2b) that was reacted with 3-bis[2-(*tert*-butoxycarbonylamino)ethyl]propargylamine³⁸ in the presence of silica based particles incorporating bis(pyridyl)amine (BPA) Cu(I) chelating agent³⁹ to give the triazol adduct **29** in 78% yield. The use of the solid-supported catalyst has proven advantageous even in multi-CuAAC ligation strategies, highly simplifying the purification step to a simple filtration process.⁴⁰ Acid hydrolysis of the Boc protecting groups in **29** provided the corresponding triamine **30**, which was next reacted with 2-(*N-tert*-butoxycarbonylamino)ethyl isothiocyanate⁴¹ to give bis(thiourea) **31**. Final hydrolysis of the Boc protecting groups, led to the target compound **6**.

Syntheses of trehalose-derived cationic glycolipids. The strategies implemented for the preparation of the α , α '-trehalose amino glycolipids 7-11 parallel those above commented for the corresponding ether- (1-4) or ester-type (5-7) methyl α -D-glucopranoside counterparts. Thus, compounds 7 and 8 were obtained in good overall yield in an efficient five-step synthesis starting from 6,6'-di-O-trityl- α , α '-trehalose 32⁴² after alkylation of the six secondary hydroxyl groups (33 and 34), trityl cleavage with p-toluenesulfonic acid in DCM-MeOH (35 and 36), Garegg's iodination of the primary hydroxyls (37 and 38), nucleophilic displacement of the iodines with

Boc-protected cysteamine (**39** and **40**) and hydrolysis of the carbamate groups (Scheme 3). The hexanoylated analogue **9** was similarly obtained from 6,6'-dideoxy-6,6'-diiodo- α,α '-trehalose **41**⁴³ by esterification of the secondary hydroxyls (**42**), incorporation of the Boc-protected cysteamine substituents at the primary positions (**43**) and final deprotection (Scheme 4).

Scheme 3. Syntheses of trehalose derivatives 7 and 8.

Reagents and conditions. a) Hexyl Bromide, NaH, DMF, overnight, 92% or 1-Bromotetradecane, NaH, DMF, 50 °C, 77%; b) p-Toluenesulfonic acid,1:1 DCM-MeOH, rt, 4 h, 48% for **35** and 47% for **36**); c) I₂, PPh₃, imidazole, toluene, 70 °C, 3 h, 94% for **37** and 96% for

38; d) HS(CH₂)₂NHBoc, Cs₂CO₃, DMF, 60 °C, overnight, 85% for **39** and 99% for **40**; e) 1:1 TFA/DCM, rt, 15 min, quant. for **7**, 98% for **8**.

Scheme 4. Synthesis of trehalose derivative **9**.

Reagents and conditions. a) HS(CH₂)₂NHBoc, Cs₂CO₃, DMF, 60 °C, 24 h, 58%; b) 1:1 TFA/DCM, rt, 15 min, quant.

The cationic trehalose amphiphiles **10** and **11** were prepared starting from the common diazide precursor **44**, readily accessed by nucleophilic displacement of the iodine in **42** with sodium azide, by CuAAC ligation using the silica-supported Si-BPA-Cu⁺ catalyst. Thus, sequential reaction of **44** with *N*-Boc-propargylamine (**45**), carbamate hydrolysis (**46**), thiourea-coupling with Boc-protected 2-aminoethyl isothiocyanate (**47**) and final Boc removal yielded the diaminoethylthioureido adduct **10**. Alternatively, the CuAAC coupling of **44** with 3-bis[2-(*tert*-butoxycarbonylamino)ethyl]propargylamine (**48**) followed by carbamate hydrolysis afforded compound **11** (Scheme 5).

Scheme 5. Syntheses of trehalose derivatives **10** and **11**.

Reagents and conditions. a) *N-tert*-butoxycarbonylpropargylamine, Si-BPA·Cu⁺, 9:1 H₂O/^tBuOH, 24 °C, 36 h, quant; b) 1:1 TFA/H₂O, rt, 1 h, 97%; c) *tert*-butyl-N-(2-isothiocyanoethyl)carbamate, Et₃N, DCM, overnight, 67%; d) 1:1 TFA/H₂O, rt, 1 h, quant; e) 3-bis[2-(*tert*-butoxycarbonylamino)ethyl]propargylamine, Si-BPA·Cu⁺, 9:1 H₂O/^tBuOH, reflux, 36 h, 91%; f) 1:1 TFA/H₂O, rt, 1 h, quant.

TLR4 modulation in HEK-BlueTM cells.

Cationic amphiphiles **1-11** were first screened for their capacity to interfere with TLR4 activation and signaling on HEK-BlueTM cells. HEK-BlueTM cells are stably transfected with TLR4, MD-2, and CD14 genes. In addition, these cells stably express an optimized alkaline phosphatase gene engineered to be secreted (sAP), placed under the control of a promoter inducible by several transcription factors such as NF-κB and AP-1.²⁹ This reporter gene allows monitoring the activation of TLR4 signal pathway by endotoxin. Compounds **5**, **9-11** were inactive in stimulating TLR4 signal when provided alone while inhibited in a dose-dependent way the LPS-stimulated TLR4 signal (Table 1). Compounds **1-4** and **6-8** resulted inactive or with very low activity both as agonists and antagonists.

Table 1. TLR4 antagonist activity of cationic glycolipids **5**, **9-11** on HEK-BlueTM cells, HEK293 hMD-2/hTLR4 and HEK293 mMD-2/mTLR4 stimulated with *E. coli* O55:B5 LPS (10 nM).

	IC ₅₀ (μM)		
Compound	HEK-Blue TM	HEK293	HEK293
		hMD-2·hTLR4	mMD-2·mTLR4
5	3.7±0.4	3.9±1.5	3.3±1.2
9	1.3 ± 0.1	1.4 ± 0.3	0.8 ± 0.2
10	5.0±1.0	0.6 ± 0.02	0.6 ± 0.03
11	0.6 ± 0.05	0.2±0.02	0.2±0.03

The lack of significant activity of all compounds bearing C_6 or C_{14} ether-linked lipophilic chains, namely compounds **1-4** and **7-8**, strongly suggests that the presence of ester-type linkages at the hydrophobic domain is a primary structural requirement to elicit LPS-antagonist behavior in cationic glycolipids. From the acylated sub-library all compounds are active with the exception of compound **6**, meaning that the cationic head also has an impact on the TLR4 antagonist

activity Among these, the trehalose-based disaccharides 9-11, bearing more compact cationic headgroups, showed higher potency as TLR4 antagonists than monosaccharide 5. The observed trend points to a positive relationship between well-ordered facial amphiphilicity and TLR4 antagonist activity of cationic glycolipids. In agreement with this, disaccharide 11 with six hexanoyl chains and six protonable amino groups oriented towards opposite faces in a rather compact arrangement (Figure 3) proved to be the most active TLR4 antagonist. Compounds 5, 9, 10, 11 were further tested for their toxicity by a standard MTT viability test and all resulted non toxic or with very low toxicity in the concentration range used to test their activity (Supp. Info.).

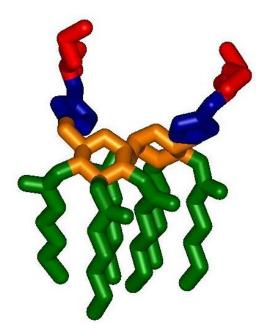


Figure 3. 3D molecular model of compound **11** (cationic headgroups in red, triazol linkers in blue, α,α '-trehalose scaffold in orange, hexanoyl chains in green) evidencing its compact facial amphiphilic character. Hydrogens have been omitted for the sake of clarity.

Activity on HEK293 cells transfected with human and murine MD-2⁻TLR4.

Biologically active cationic glycolipids 5, 9, 10 and 11 were further examined for their capacity to stimulate or to inhibit LPS-induced TLR4 activation and signaling in HEK293 cells

transfected with human or murine TLR4 and MD-2 and a dual luciferase reporter gene (Figure 4).

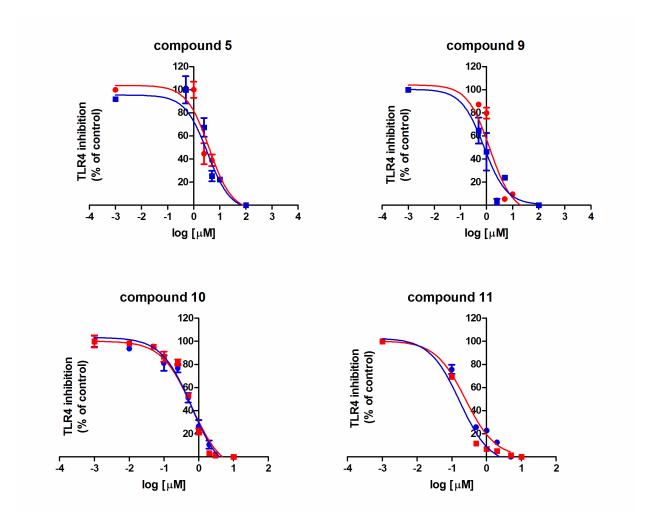


Figure 4. Dose-dependent inhibition of LPS-stimulated TLR4 activation by synthetic glycolipids. HEK293 cells transfected with human MD-2·TLR4 (red line) or murine MD-2/TLR4 (blue line), were treated with increasing concentrations of compounds and stimulated with LPS (5 nM). Normalized data are representative of three independent experiments.

In the absence of LPS, none of the cationic glycolipids stimulate TLR4 signaling (no agonist activity) in cells transfected with hMD-2'hTLR4 or mMD-2'mTLR4. Conversely, in the presence of LPS all compounds were able to inhibit human and murine MD-2'TLR4 activation in a

concentration-dependent manner. The antagonist potency (IC₅₀ values, Table 1) of **5**, **9**, **10** and **11** was similar in cells transfected with human and mouse receptors and also very similar to the activity found in HEK-BlueTM cells (Table 1). In cells transfected with hMD-2:hTLR4 or mMD-2:mTLR4, triacylated monosaccharide **5** and hexacylated disaccharide **9** were less active (IC₅₀ = $3.3-3.9 \mu M$ and $0.8-1.4 \mu M$, respectively) than disaccharides **10** and **11** (IC₅₀ = $0.6 \mu M$ and $0.2 \mu M$, respectively). The high potency of trehalose-derived glycolipids **9** and **11** in inhibiting both mouse and human MD-2:TLR4 signals is reminiscent of the activity of synthetic Eritoran[®] that has potent TLR4 antagonist activity in all species. ^{16,44} In contrast, natural lipid IVa is agonist on murine and antagonist on human TLR4. ⁴⁵

Experiments on murine macrophages.

The activity of compounds **5**, **9** and **11** on LPS-induced TLR4 signaling in bone marrow-derived murine macrophages (BMDM) was subsequently tested. Compounds **5** and **9** showed very low/no activity in activating TLR4 or in inhibiting LPS-stimulated TLR4 signal (Supp. Info.), while compound **11** (Figure 5) gave a concentration-dependent inhibition of IL-6 and TNF- α production at concentrations of 1 and 2 μ M (Figure 5), while at 0.1 and 0.5 μ M concentrations had no effect or a slightly potentiating effect.

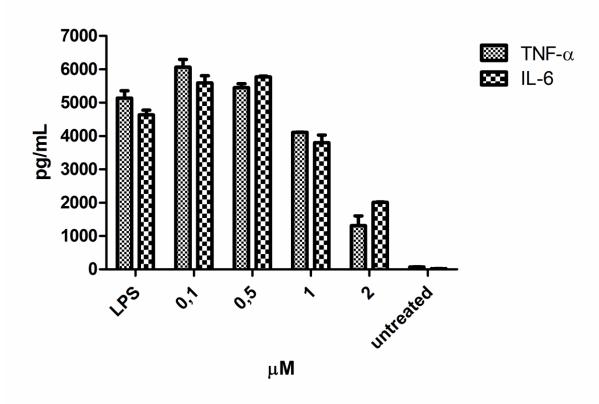


Figure 5. BMDM were treated with increasing concentrations (0-2 μM) of compound **11** in RPMI+FBS 10% in the presence of LPS, administered 1 hour after the treatment with synthetic compound. The ELISA assay, performed after overnight incubation, revealed a dose dependent decrease of LPS-induced IL-6 and TNF-α production. Cytokines productions in cells not treated with LPS are reported as negative controls.

Aggregation properties of cationic glycolipids

Cationic lipids can spontaneously assembly into liposomal structures. Some cationic liposomes induce the expression of pro-apoptotic and pro-inflammatory mediators through the activation of cellular pathways.¹⁷ However, it is still controversial if the initiation of inflammatory and apoptotic response is due to specific interaction with receptors at the cell surface or to the internalization of liposomes into cells through endocytosis and endocytosis-like mechanisms followed by interaction of charged compounds with downstream effectors. In the particular case

of cationic lipids modulating TLR4 activity, there is no information available on whether aggregated species or single molecules are the active species. To have an insight in this question, we have determined the critical micelle concentrations (CMC) of cationic glycolipids **5**, **9**, **10** and **11** using an established technique based on the polarity-induced change in the fluorescence spectra of pyrene when incorporated in micelle formed by synthetic compounds (Table 2). In all cases the CMC values of active compounds are higher than the corresponding IC₅₀ values as TLR4 antagonists, suggesting that the biologically active species are prevalently single molecules in solution.

Table 2. Critical micelle concentrations (CMC) in aqueous environment of TLR4 antagonists.

Compound	CMC (µM)
5	59.7 ± 7.4
9	97.7 ± 10.0
10	10.9 ± 2.1
11	350.5 ± 70.5

Synthesis and biological activity of gold nanoparticles coated with glycolipid 11

LPS is an amphiphilic molecule and it is mainly present in the form of micellar aggregates in a concentration range that is relevant for its biological activity. It has been recently proposed that the multiple presentation of LPS or other TLR4 ligands on nanoparticles could be a way to potentiate the agonist or antagonist action of chemicals by mimicking the 3D-structure of LPS aggregates. An Moreover, the possibility of *in vitro* and/or *in vivo* delivery based on NP is considered advantageous for clinical development, as it can maximize the effectiveness of drugs, minimize the invasiveness and toxic side effects and speed up the clinical development program.

To test the suitability of this approach in the case of cationic glycolipids, the preparation and biological evaluation of gold nanoparticles coated with the most active compound 11 has been undertaken. Colloidal gold nanoparticles were synthesized by a variation of the Brust-Schiffrin method⁴⁹ and coated by surface adsorption with the α,α'-trehalose derivative 11. The resulting cationic glycolipid-modified nanoparticles (11-NP) kept a small size with low-polydispersity. Most interestingly, they retained the biological activity without apparent increase in cytotoxicity. These results represent a proof of concept of the possibility of developing nanoparticulate systems based on cationic glycolipids as modulators of TLR4 signaling pathway, an approach previously demonstrated only for LPS itself.⁴⁷ Based on these data, the prepared 11-NP nanoparticles were engaged in an *in vivo* assay. The inhibitory activity of the nanoparticles coated with glycolipid 11 was tested on HEK293 cells. They exhibited strong LPS-inhibitory activity already at very low concentrations (Figure 6 A, B) on human as well as murine MD-2·TLR4 receptor complex.

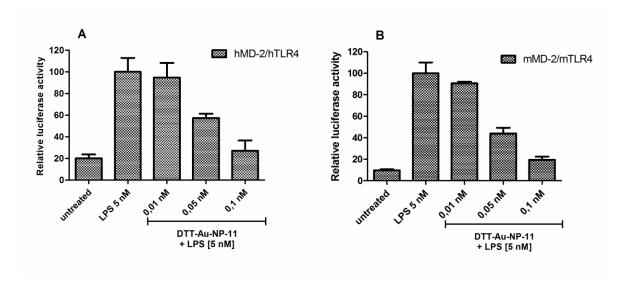


Figure 6. Dose-dependent TLR4 antagonism in HEK293 cells treated with DTT-Au-NP-11. HEK293 cells were transfected with NF-κB-dependent luciferase and constitutive Renilla luciferase reporter plasmids as well as with (A) human or (B) murine MD-2 and TLR4 plasmids.

The indicated amount of the DTT-Au-NP-11 was added to the cells, followed one hour later by stimulation with LPS. Luciferase activity was measured 16 hours later.

In vivo activity

Since the synthetized compounds exhibited strong inhibition of the LPS-induced MD-2·TLR4 activation *in vitro*, we next wished to determine their inhibitory potential *in vivo*. All four selected candidates (5, 9, 10 and 11) potently inhibited the LPS-induced immune activation in C57/Bl6 mice (Figure 7). The strongest inhibition was exhibited by compound 11, which totally abolished LPS-induced immune activation. Results in Figure 8 show that compounds 5, 9, 10 and 11 are strong MD-2·TLR4 inhibitors not only *in vitro* but also *in vivo*.

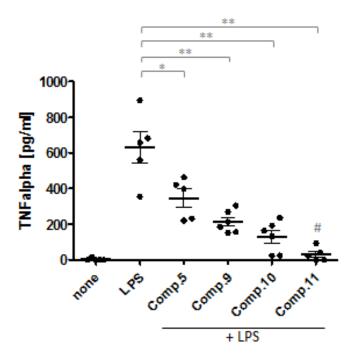


Figure 7. in vivo activity of cationic amphiphiles. C57/Bl6 mice were injected i.p. with the indicated compounds $(2 \cdot 10^{-7} \text{ mol/mouse})$, followed 1 hour later by i.p. injection of LPS $(1 \cdot 10^{-9} \text{ mol/mouse})$. Three hours later sera were collected and TNF-α concentration was determined by

ELISA. (data shown with mean and standard error, N=5-6) two-tailed t-test (* p<0,1; ** p<0,01 – compared to »LPS«) (# not significant – compared to »none«)

Conclusions and perspectives

New cationic amphiphiles 1-11 based on monosaccharide and disaccharide glycolipid scaffolds have been designed, synthesized and their capacity to modulate TLR4 activation and signaling evaluated. Glucose-based compound 5 and trehalose-based compounds 9-11 were active in inhibiting the LPS-triggered TLR4 activation and signaling in HEK cells with IC₅₀ values ranging from about 5 to 0.2 μM. The cell toxicity of these molecules is low, and the potency of TLR4 antagonism is in the same order of magnitude of the best synthetic TLR4 antagonists so far tested by us²⁹ and other groups.³⁰ The active molecules inhibited the TLR4 signal in HEK cells transfected with human and murine MD-2·TLR4 complexes with very similar potency, similarly to the very efficient TLR4 antagonist Eritoran[®] and differently from the natural TLR4 antagonist lipid IVa that has species-specific activity (antagonist on human and agonist on mice MD-2·TLR4). Compounds 5, 9, 10 and 11 significantly inhibited LPS-triggered IL-6 production in mice, with compound 11 showing the most evident effect. Since these compounds are active *in vitro* and *in vivo*, and show low toxicity, they represent good leads for the development of drugs targeting TLR4 signaling.

The biological evaluation of active compounds compared to inactive, structurally related monosaccharides (compounds 1-4, 6) and disaccharides (compounds 7 and 8), suggests some general structure-activity relationships in this type of compounds: i) the presence of acyl lipophilic chains at the hydrophobic domain seems to be a primary requisite since all compounds with ether bonds are inactive, ii) the higher *in vitro* and *in vivo* activity of compound 11 suggests that the trehalose scaffold favors the biological activity, probably by providing a well-ordered

facial amphiphilic character. Compound **11** adsorbed on gold nanoparticles (**11**-NP) is still active as TLR4 antagonist in cells, but the high toxicity of these functionalized nanoparticles could prevent their use *in vivo* as carriers for TLR4 antagonists.

Finally, the experimentally determined CMC values for cationic glycolipids **5**, **9-11** are one order of magnitude higher than the corresponding IC₅₀ as TLR4 antagonists, suggesting that they are active as single monomers in solution. This very likely means that specific molecular interaction with CD14 and MD-2 receptors regulate the TLR4 activity of these compounds. It will be important in the future to define more precisely the molecular determinants of the interaction with CD14 and MD-2 receptors to allow a structure-based rational design of cationic TLR4 modulators.

EXPERIMENTAL SECTION

General Synthetic Methods. Optical rotations were measured at 20 ± 2 °C in 1-dm tubes on a Jasco P-2000 polarimeter. Ultraviolet-visible (UV) spectra were recorded in 1-cm tubes on a Jasco V-630 spectrophotometer. Infrared (IR) spectra were recorded on a Jasco ATR MIRacleTM spectrophotometer. ¹H (and ¹³C NMR) spectra were recorded at 300 (75.5), 500 (125.7) MHz with Bruker 300 AMX, 500 AMX and 500 DRX. 1D TOCSY, 2D COSY, HMQC and HSQC experiments were used to assist on NMR assignments. Thin-layer chromatography (TLC) was carried out on aluminium sheets coated with *Silica Gel 60 F*₂₅₄ *Merck* with visualization by UV light and by charring with 10% H₂SO₄. With preparative purposes, column chromatography was carried out on Silica Gel 60 F₂₅₄ Merck. ESI mass spectra were recorded in the positive mode on Bruker Daltonics *esquire6000*TM ion-trap mass spectrometer. Typically, samples were dissolved in appropriate volumes of deionised water to give samples concentration of 50 mg/L. Aliquots were mixed with 25:25:1 deionised water-methanol-trifluoroacetic acid, generally in a ratio of

1:10, to give a total volume of 200 μ L. Samples were introduced by direct infusion, using a Cole-Parmer syringe at a flow rate of 2 μ L/min. Ions were scanned between 300 and 3000 Da with a scan speed of 13000 Da/s at unit resolution using resonance ejection at the multipole resonance of one-third of the radio frequency ($\Omega = 781.25 \text{ kHz}$). Calibration of the mass spectrometer was performed using ES tuning mix (Hewlett Packard). Recorded data were processed using Bruker Daltonics Esquire 5.0 software (Bruker). Elemental analyses were carried out at the Instituto de Investigaciones Químicas (Sevilla, Spain) using an elemental analyser *Leco CHNS-932* or *Leco TruSpec CHN*.

Methyl 6-azido-6-deoxy-α-D-glucopyranoside, 28 methyl 4,6-O-(4-methoxybenzylidene)-α-D-glucopyranoside, 34 methyl 6-deoxy-6-iodo-α-D-glucopyranoside, 35 2-[N,N-bis(2-(N-tert-butoxyaminocarbonyl)ethylamino]ethyl isothiocyanate, 37 6,6'-dideoxy-6,6'-diiodo-α,α'-trehalose 41, 43 6,6'-di-O-trityl-α,α'-trehalose 32, 42 N-(2-isothiocyanoethyl) tert-butylcarbamate, 41 and 3-bis[2-tert-butoxycarbonylamino)ethyl]propargylamine 38 were obtained according to described procedures. Purity of all final compounds was confirmed to be \ge 95% by 1 H NMR and combustion microanalysis.

Methyl 6-Azido-6-deoxy-2,3,4-tri-O-hexyl- α -D-glucopyranoside (13). To a solution of methyl 6-azido-6-deoxy- α -D-glucopyranoside (0.40 m, 1.82 mmol) in dry DMF (9 mL), NaH (0.65 g, 16.42 mmol) was added, under Ar atmosphere, at 0 °C. Then 1-bromohexane (2.3 mL, 16.42 mmol) was added dropwise and the reaction mixture was stirred overnight at rt. The solvent was evaporated and the residue diluted in DCM (10 mL) and washed with H₂O (2 x 10 mL). The organic layer was dried (MgSO₄), concentrated and purified by column chromatography (1:30 EtOAc-cyclohexane). Yield 48% (0.50 m, 1.06 mmol). $R_f = 0.34$ (1:18 EtOAc-cyclohexane); $[\alpha]_D = +90.0$ (c 1.0, DCM); IR: $v_{max} = 2099$, 1094 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): $\delta = 4.74$

(d, 1 H, $J_{1,2} = 3.5$ Hz, H-1), 3.79, 3.64, 3.58 (m, 6 H, OCH₂), 3.67 (m, 1 H, H-5), 3.64-3.24 (m, 2 H, H-3, H-6a), 3.39 (s, 3 H, OCH₃), 3.35 (m, 1 H, H-6b), 3.23 (dd, 1 H, $J_{2,3} = 9.6$ Hz, H-2), 3.12 (t, 1 H, $J_{3,4} = J_{4,5} = 9.3$ Hz, H-4), 1.55 (m, 6 H, CH₂), 1.27 (m, 18 H, CH₂), 0.86 (m, 9 H, CH₃); ¹³C NMR (75.5 MHz, CDCl₃): $\delta = 98.0$ (C-1), 81.3 (C-3), 80.7 (C-2), 78.8 (C-4), 73.7, 73.6, 71.8 (OCH₂), 70.2 (C-5), 55.2 (OCH₃), 51.5 (C-6), 31.7-22.6 (CH₂), 14.3 (CH₃); ESI MS: m/z = 965.8 [2M + Na]⁺, 494.6 [M + Na]⁺. Anal. Calcd for C₂₅H₄₉N₃O₅: C, 63.66; H, 10.47; N, 8.91. Found: C, 63.72; H, 10.97; N, 8.33.

Methyl 6-Amino-6-deoxy-2,3,4-tri-*O*-hexyl-α-D-glucopyranoside Hydrochloride (1). To a solution of **13** (0.15 g, 0.32 mmol) in degassed MeOH (12 mL), Pd/C (10%, 0.06 g) was added and the mixture was stirred under H₂ atmosphere (1 bar) at rt until complete consumption of the starting compound. The catalyst was filtered off, the solution concentrated, and the resulting residue purified by column chromatography (1:9 EtOAc-cyclohexane → 45:5:3 EtOAc-EtOH-H₂O) and freeze-dried from 0.1 N HCl solution. Yield 87% (0.12 g, 0.26 mmol); [α]_D = +83.0 (*c* 1.0, DCM); ¹H NMR (300 MHz, CD₃OD): δ = 4.82 (d, 1 H, $J_{1,2}$ = 3.4 Hz, H-1), 3.81, 3.64, 3.56 (m, 6 H, OCH₂), 3.65 (m, 1 H, H-5), 3.53 (t, 1 H, $J_{2,3}$ = $J_{3,4}$ = 9.3 Hz, H-3), 3.41 (s, 3 H, OCH₃), 3.25 (dd, 1 H, H-2), 3.11 (dd, 1 H, $J_{6a,6b}$ = 13.1 Hz, $J_{5,6a}$ = 2.7 Hz, H-6a), 3.01 (t, 1 H, $J_{4,5}$ = 9.3 Hz, H-4), 2.84 (dd, 1 H, $J_{5,6b}$ = 8.5 Hz, H-6b), 1.56 (m, 6 H, CH₂), 1.33 (m, 18 H, CH₂), 0.91 (m, 9 H, CH₃); ¹³C NMR (75.5 MHz, CD₃OD): δ = 99.1 (C-1), 82.6 (C-3), 81.8 (C-2), 81.1 (C-4), 74.5, 74.3, 72.2 (OCH₂), 70.7 (C-5), 55.8 (OCH₃), 42.9 (C-6), 33.0-24.2 (CH₂), 14.4 (CH₃); ESIMS: m/z = 891.7 [2 M + H]⁺, 446.5 [M + H]⁺. Anal. Calcd for C₂₅H₅₁NO₅·HCl: calcd. C, 62.28; H, 10.87; N, 2.91. found: C, 62.33; H, 10.69; N, 2.70.

Methyl 6-Azido-6-deoxy-2,3,4-tri-*O*-tetradecyl-α-D-glucopyranoside (14). To a solution of methyl 6-azido-6-deoxy-α-D-glucopyranoside (0.28 g, 1.30 mmol) in dry DMF (4 mL), NaH

(0.55 mg, 13.60 mmol) was added, under Ar atmosphere, at 0 °C. Then 1-bromotetradecane (4.1 mL, 13.60 mmol) was added dropwise, and the reaction mixture was stirred overnight at 55 °C. The solvent was evaporated and the residue diluted in DCM (10 mL) and washed with H₂O (2 x 10 mL). The organic layer was dried (MgSO₄), concentrated and purified by column chromatography (cyclohexane \rightarrow 1:40 EtOAc-cyclohexane). Yield 42% (0.45 g, 0.26 mmol) . R_f = 0.25 (1:15 EtOAc-cyclohexane); [α]_D = +52.1 (c 1.0, DCM); IR: v_{max} = 2100, 1096 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ = 4.76 (d, 1 H, $J_{1,2}$ = 3.4 Hz, H-1), 3.79, 3.64, 3,58 (m, 6 H, OCH₂), 3.69 (m, 1 H, H-5), 3.66-3.44 (m, 2 H, H-3, H-6a), 3.39 (s, 3 H, OCH₃), 3.37 (dd, 1 H, $J_{6a,6b}$ = 13.3 Hz, $J_{5,6b}$ = 5.5 Hz, H-6b), 3.25 (dd, 1 H, $J_{2,3}$ = 9.4 Hz, H-2), 3.13 (t, 1 H, $J_{3,4}$ = $J_{4,5}$ = 9.4 Hz, H-4), 1.57 (m, 6 H, CH₂), 1.27 (m, 66 H, CH₂), 0.86 (m, 9 H, CH₃); ¹³C NMR (75.5 MHz, CDCl₃): δ = 98.0 (C-1), 81.3 (C-3), 80.7 (C-2), 78.8 (C-4), 73.7, 73.6, 71.8 (OCH₂), 70.2 (C-5), 55.2 (OCH₃), 51.5 (C-6), 31.9-22.7 (CH₂), 14.1 (CH₃); ESI MS: m/z = 830.8 [M + Na]⁺. Anal. Calcd for C₄₉H₉₇N₃O₅: C, 72.81; H, 12.10; N, 5.20. Found: C, 72.89; H, 11.87; N, 4.88.

Methyl 6-Amino-6-deoxy-2,3,4-tri-*O*-tetradecyl-α-D-glucopyranoside Hydrochloride (2). To a solution of 14 (0.15 g, 0.19 mmol) in THF (16.4 mL), TPP (0.10 g, 0.39 mmol) was added and the mixture was stirred at rt for 15 min. Then NH₄OH (1.6 mL) was added, the solution was stirred overnight at 50 °C and then concentrated. The resulting residue was purified by column chromatography (EtOAc→9:1 DCM-MeOH) and freeze-dried from 10:1 H₂O-HCl 0.1 N solution. Yield 82% (0.13 mg, 0.16 mmol); $[\alpha]_D = +43.2$ (*c* 1.0, DCM); IR: $\nu_{max} = 1092$ cm⁻¹; ¹H NMR (300 MHz, CDCl₃): $\delta = 4.79$ (d, 1 H, $J_{1,2} = 3.4$ Hz, H-1), 3.78, 3.60, 3.55 (m, 7 H, OCH₂, H-3), 3.73 (m, 1 H, H-5), 3.46 (s, 3 H, OCH₃), 3.27 (m, 1 H, H-6^a), 3.22 (dd, 1 H, $J_{1,2} = 9.5$ Hz, H-2), 2.95 (m, 2 H, H-4, H-6b), 1.55 (m, 6 H, CH₂), 1.25 (m, 66 H, CH₂), 0.87 (t, 9 H, ³ $J_{H,H} = 6.6$ Hz, CH₃); ¹³C NMR (75.5 MHz, CD₃OD): $\delta = 98.0$ (C-1), 81.0 (C-3), 80.4 (C-2), 79.9 (C-4), 77.2

(C-5), 73.7, 73.5, 71.8 (OCH₂), 67.1 (C-6), 55.9 (OCH₃), 31.9-22.7 (CH₂), 14.0 (CH₃); ESI MS: $m/z = 783.0 \text{ [M + H]}^+$. Anal. Calcd for C₄₉H₉₉NO₅·HCl: calcd. C, 71.88; H, 12.31; N, 1.71; found: C, 71.64; H, 12.26; N, 1.49.

Methyl 4,6-*O*-(4-Methoxybenzylidene)-2,3-di-*O*-hexyl-α-D-glucopyranoside (15). To a solution of methyl 4,6-O-(4-methoxybenzylidene)-α-D-glucopyranoside (0.80 g, 2.57 mmol) in DMF (8 mL), NaH (0.62 g, 15.42 mmol) was added. Then, 1-bromohexane (1.8 mL, 12.85 mmol) was added dropwise and the resulting mixture was stirred at 60 °C overnight. After cooling to rt, the reaction was quenched with MeOH (2 mL) and the solution was stirred for 20 min. Solvents were then evaporated and the residue was diluted with EtOAc (50 mL) and citric acid (satd aq soln, 40 mL). The layers were separated and the organic phase was washed with H₂O (3 x 40 mL), dried (MgSO₄), evaporated, and purified by column chromatography (1:9 EtOAc-cyclohexane). Yield 57% (0.71 g, 0.15 mmol); $R_f = 0.44$ (1:9 EtOAc-cyclohexane); $[\alpha]_D$ = +37.2 (c 1.0, DCM); 1 H NMR (300 MHz, CDCl₃): δ = 7.39, 6.87 (2 d, 4 H, A₂X₂, aromatics), 5.49 (s, 1 H, PhCH), 4.78 (d, 1 H, $J_{1,2} = 3.8$ Hz, H-1), 4.25 (dd, 1 H, $J_{6a,6b} = 9.6$ Hz, $J_{5,6a} = 4.5$ Hz, H-6a), 3.80 (s, 3 H, PhOCH₃), 3.76 (m, 1 H, H-5), 3.72 (t, 1 H, $J_{2,3} = J_{3,4} = 9.2$ Hz, H-3) 3.72-3.57 (m, 5 H, H-6b, OCH₂), 3.47 (t, 1 H, $J_{4,5} = 9.2$ Hz, H-4), 3.42 (s, 3 H, OCH₃), 3.34 (dd, 1 H, H-2), 1.66-1.49 (m, 4 H, CH₂), 1.41-1.18 (m, 12 H, CH₂), 0.88, 0.84 (2 t, 6 H, ${}^{3}J_{H,H} = 6.5$ Hz, CH₃); ¹³C NMR (75.5 MHz, CDCl₃): $\delta = 159.9-113.5$ (Ph), 101.2 (Ph*C*H), 99.1 (C-1), 81.9 (C-4), 80.4 (C-2), 78.2 (C-3), 73.4, 72.2 (OCH₂), 69.0 (C-6), 62.4 (C-5), 55.6 (OCH₃), 31.7-22.6 (CH₂), 14.0 (CH₃); ESI MS: m/z: 519.5 [M + K]⁺, 503.6 [M + Na]⁺. Anal. Calcd for C₂₇H₄₄O₇: C, 67.47; H, 9.23. Found: C, 67.54; H, 9.30.

Methyl 4,6-O-(4-Methoxybenzylidene)-2,3-di-O-tetradecyl- α -D-glucopyranoside (16). To a solution of methyl 4,6-O-(4-methoxybenzylidene)- α -D-glucopyranoside (0.80 g, 2.57 mmol) in

DMF (8 mL), NaH (60% suspension in mineral oil, 0.62 g, 15.42 mmol) was carefully added in small portions. Tetradecylbromide (3.8 mL, 12.85 mmol) was added dropwise and the resulting mixture was stirred at 60 °C overnight. After cooling to rt, the mixture was quenched with methanol (2 mL) then the solution was stirred for 20 min. Solvents were then evaporated and the residue was diluted with EtOAc (50 mL). Citric acid (satd aq soln, 40 mL) was added, the layers were separated, the organic layer was washed with water (3 x 40 mL), dried (Na₂SO₄) and evaporated. Flash column chromatography on silica gel of the residue (1:9 EtOAc-cyclohexane) afforded **16.** Yeld 74% (1.33 g, 1.89 mmol). $R_f = 0.65$ (1:9 EtOAc-cyclohexane); $[\alpha]_D = +23.3$ (c1.0 in CHCl₃); 1 H NMR (300 MHz, CDCl₃): δ = 7.40, 6.87 (2 d, 4 H, A₂X₂, aromatics), 5.49 (s, 1 H, PhCH), 4.78 (d, 1 H, $J_{1,2} = 3.8$ Hz, H-1), 4.25 (dd, 1 H, $J_{6a,6b} = 9.6$, $J_{5,6a} = 4.6$ Hz, H-6a), 3.80 (s, 3 H, OCH₃), 3.76 (m, 1 H, H-5), 3.75-3.59 (m, 6 H, H-3, H6b, OCH₂), 3.47 (t, 1 H, $J_{3,4} = J_{4,5}$ = 9.3 Hz, H-4), 3.42 (s, 3 H, OCH₃), 3.34 (dd, 1 H, $J_{2,3}$ = 9.3, H-2), 1.70-1.49 (m, 4 H, CH₂), 1.22 (bs, 44 H, CH₂), 0.87 (t, 6 H, ${}^{3}J_{H,H}$ = 5.8 Hz, CH₃). ${}^{13}C$ NMR (75.5 MHz, CDCl₃): δ = 159.1, 130.0, 127.3, 113.5 (Ph), 101.2 (PhCH), 99.1 (C-1), 81.9 (C-4), 80.4 (C-2), 78.2 (C-3), 73.4, 72.3 (OCH₂), 69.0 (C-6), 62.4 (C-5), 55.2 (OCH₃), 31.9-22.7 (CH₂), 14.1 (CH₃). ESI MS: m/z: 503.6 [M + Na]⁺, 519.5 [M + K]⁺.

Methyl 2,3-Di-*O*-hexyl-4-*O*-(*p*-methoxybenzyl)-α-D-glucopyranoside (17). To a solution of 15 (0.71 g, 1.48 mmol) in a mixture of Et₂O-DCM (2:1, 75 mL), under Ar atmosphere, 1 M LiAlH₄ in THF (7.4 mL, 7.40 mmol) and AlCl₃ (0.81 g, 6.06 mmol) in Et₂O (25 mL) were added dropwise, and the resulting mixture was refluxed for 4 h. After cooling to rt, EtOAc (250 mL) and H₂O (250 mL) were added. The organic layer was washed with brine (2 x 200 mL), dried (MgSO₄), evaporated and purified by column chromatography (1:2 EtOAc-cyclohexane). Yield 83% (0.60 g, 1.22 mmol); $R_f = 0.26$ (1:2 EtOAc-cyclohexane); $[\alpha]_D = +76.2$ (*c* 1.0, DCM); IR:

 v_{max} = 1076, 1035cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ =7.26, 6.87 (A₂X₂ system, 4 H, aromatics), 4.81 (d, 1 H, ${}^2J_{\text{Ha,Hb}}$ = 10.8 Hz, PhC*Ha*), 4.75 (d, 1 H, $J_{1,2}$ = 3.5 Hz, H-1), 4.57 (d, 1 H, PhC*Hb*), 3.86 (m, 1 H, OCH₂), 3.79 (s, 3 H, PhOCH₃), 3.77-3.64 (m, 3 H, H-6a, H-6b, OCH₂), 3.68 (m, 1 H, H-3), 3.67-3.53 (m, 3 H, OCH₂, H-5), 3.42 (t, 1 H, $J_{3,4}$ = $J_{4,5}$ = 9.5 Hz, H-4), 3.37 (s, 3 H, OCH₃), 3.26 (dd, 1 H, $J_{2,3}$ = 9.5 Hz, H-2), 1.75 (bs, 1 H, OH), 1.64-1.57 (m, 4 H, CH₂), 1.37-1.28 (m, 12 H, CH₂), 0.88 (t, 6 H, ${}^3J_{\text{H,H}}$ = 7.0 Hz, CH₃), 0.87 (t, 6 H, ${}^3J_{\text{H,H}}$ = 6.8 Hz, CH₃); ¹³C NMR (75.5 MHz, CDCl₃): δ =159.3-113.8 (Ph), 98.0 (C-1), 81.6 (C-3), 80.8 (C-2), 77.1 (C-4), 74.5 (PhCH₂), 73.7, 71.7 (OCH₂), 70.6 (C-5), 62.0 (C-6), 55.2, 55.0 (OCH₃), 31.8-22.6 (CH₂), 14.0 (CH₃); ESI MS: m/z: 505.6 [M + Na]⁺, 521.5 [M + K]⁺. Anal. Calcd for C₂₇H₄₆O₇: C, 67.19; H, 9.61; found: C, 66.92; H, 9.67.

Methyl 6-Deoxy-2,3-di-*O*-hexyl-6-iodo-4-*O*-(*p*-methoxybenzyl)-α-D-glucopyranoside (19). To a solution of 17 (0.60 g, 1.23 mmol) in toluene (25 mL) TPP (0.49 g, 1.85 mmol) and imidazole (0.25 g, 3.70 mmol) were added. Iodine (0.49 g, 1.73 mmol) was added in portions and the resulting solution was stirred at 70 °C for 5 h. After cooling at rt, NaHCO₃ sat. (25 mL) was added and the mixture was stirred for 5 min. Additional iodine was added and the mixture was stirred for 10 min. Then Na₂S₂O₃ aq. 10% was added to remove the iodine excess. The organic layer was separated, washed with H₂O (3 x 25 mL), dried (MgSO₄), filtered, concentrated and purified by column chromatography (1:4 EtOAc-cyclohexane). Yield 94% (0.69 g, 1.16 mmol) . R_f= 0.52 (1:4 EtOAc-cyclohexane); [α]_D = +84.3 (*c* 1.0, DCM); ¹H NMR (300 MHz, CDCl₃): δ = 7.26, 6.88 (A₂X₂ system, 4 H, , aromatics), 4.87 (d, 1 H, ²J_{Ha,Hb} = 10.6 Hz, PhC*Ha*), 4.77 (d, 1 H, J_{1,2} = 3.4 Hz, H-1), 4.62 (d, 1 H, PhC*Hb*), 3.87 (m, 1 H, CH₂), 3.80 (s, 3 H, PhOCH₃), 3.75-3.64 (m, 2 H, H-3, OCH₂), 3.63-3.50 (m, 2 H, OCH₂), 3.46 (dd, 1 H, J_{6a,6b} = 10.4 Hz, J_{5,6a} = 2.4 Hz, H-6a), 3.42 (s, 3 H, OCH₃), 3.37 (m, 1 H, H-5), 3.35-3.26 (m, 2

H, H-2, H-6b), 3.24 (t, 1 H, $J_{3,4} = J_{4,5} = 9.0$ Hz, H-4), 1.66-1.55 (m, 4 H, CH₂), 1.39-1.25 (m, 12 H, CH₂), 0.88 (t, 6 H, ${}^{3}J_{H,H} = 6.5$ Hz, CH₃), 0.87 (t, 6 H, ${}^{3}J_{H,H} = 6.8$ Hz, CH₃); ${}^{13}C$ NMR (75.5 MHz, CDCl₃): $\delta = 159.4$ -113.9 (Ph), 98.0 (C-1), 81.2 (C-3), 81.1 (C-2), 80.8 (C-4), 74.9 (PhCH₂), 73.7, 71.7 (OCH₂), 69.2 (C-5), 55.4, 55.2 (OCH₃), 31.7-22.6 (CH₂), 14.0 (CH₃), 8.1 (C-6); ESI MS: m/z: 631.3 [M + K]⁺, 615.4 [M + Na]⁺. Anal. Calcd for C₂₇H₄₅IO₆: C, 54.73; H, 7.65; found: C, 54.88; H, 7.71.

Methyl 6-(2-tert-Butoxycarbonylaminoethylthio)-2,3-di-O-hexyl-4-O-p-methoxybenzyl-α-D**glucopyranoside** (21). To a suspension of 19 (0.69 g, 1.16 mmol) and Cs₂CO₃ (0.53 g, 1.62 mmol) in dry DMF (10 mL), tert-butyl (2-mercaptoethyl)carbamate (0.27 mL, 1.62 mmol,) was added and the reaction mixture was stirred at 60 °C, under Ar atmosphere, for 24 hours. The mixture was concentrated, EtOAc (25 mL) and water (25 mL) were added then the organic layer was separated, washed with H₂O (3 x 25 mL), dried (MgSO₄), filtered, concentrated and the residue was purified by column chromatography (1:6→1:4 EtOAc-cyclohexane). Yield 99% (0.74 g, 1.15 mmol). $R_f = 0.17$ (1:6 EtOAc-cyclohexane); $[\alpha]_D = +64.2$ (c 1.0, DCM); IR: $\nu_{max} =$ 1714 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): $\delta = 7.24$, 6.87 (2 d, 4 H, ³ $J_{H,H} = 8.7$ Hz, aromatics), 4.98 (bs, 1 H, NH), 4.84 (d, 1 H, ${}^{2}J_{\text{Ha,Hb}} = 10.8$ Hz, PhCHa), 4.74 (d, 1 H, $J_{1,2} = 3.4$ Hz, H-1), 4.54 (d, 1 H, PhCHb), 3.85 (m, 1 H, OCH₂), 3.79 (s, 3 H, PhOCH₃), 3.74-3.52 (m, 3 H, OCH₂), 3.71 (m, 1 H, H-5), 3.64 (t, 1 H, $J_{2,3} = J_{3,4} = 9.3$ Hz, H-3), 3.40 (s, 3 H, OCH₃), 3.31-3.25 (m, 4 H, H-2, H-4, CH₂N), 2.84 (dd, 1 H, $J_{6a,6b} = 13.9$ Hz, $J_{5,6a} = 2.6$ Hz, H-6a), 2.71 (t, 2 H, ${}^{3}J_{H,H} = 6.4$ Hz, CH₂S), 2.57 (dd, 1 H, $J_{5.6b} = 7.5$ Hz, H6b), 1.65-1.55 (m, 4 H, CH₂), 1.44 (s, 9 H, CMe₃), 1.36-1.24 (m, 12 H, CH₂), 0.88, 0.87 (2 t, 6 H, ${}^{3}J_{H,H} = 6.6$ Hz, CH₃); ${}^{13}C$ NMR (75.5 MHz, CDCl₃): $\delta = 159.3$ (CO), 130.5-113.9 (Ph), 97.8 (C-1), 81.6 (C-3), 80.9 (C-2), 80.1 (C-4), 79.3 (CMe₃), 74.7 (CH₂Ph), 73.7, 71.7 (2 CH₂), 70.7 (C-5), 55.3, 55.0 (2 OCH₃), 39.7 (CH₂N), 33.6 (C-6), 33.5 (CH₂S), 31.7-29.6 (CH₂), 28.4 (CMe₃), 26.9-22.6 (CH₂), 14.0 (CH₃); ESI MS: m/z: 664 [M + Na]⁺, 680 [M + K]⁺. Anal. Calcd for C₃₄H₅₉NO₈S: C, 63.62; H, 9.26; N, 2.18; S, 5.00; found: C, 63.73; H, 9.21; N, 1.98; S, 4.86.

Methyl 6-(2-Aminoethylthio)-2,3-di-*O*-hexyl-α-D-glucopyranoside Hydrochloride (3). Treatment of 21 (0.35 g, 0.55 mmol) with 1:1 TFA-DCM (2 mL) and freeze-drying from 10:1 H₂O/0.1 N HCl solution afforded 3. Yield quant. (0.20 g, 0.54 mmol). R_f = 0.45 (45:5:3 EtOAc-EtOH-H₂O); [α]_D = +74.4 (*c* 1.0, MeOH); IR: v_{max} = 3404, 1109 cm⁻¹; ¹H NMR (300 MHz, CD₃OD): δ = 4.80 (d, 1 H, $J_{1,2}$ = 3.9 Hz, H-1), 3.74 (t, 1 H, ${}^3J_{H,H}$ = 6.8 Hz, OCH₂), 3.67-3.51 (m, 4 H, H-5, OCH₂), 3.42 (s, 3 H, OCH₃), 3.40 (t, 1 H, $J_{2,3}$ = $J_{3,4}$ = 9.3 Hz, H-3), 3.29 (t, 1 H, $J_{4,5}$ = 9.3 Hz, H-4), 3.23 (dd, 1 H, H-2), 3.16 (t, 2 H, ${}^3J_{H,H}$ = 6.8 Hz, CH₂N), 2.99 (dd, 1 H, $J_{6a,6b}$ = 14.4 Hz, $J_{5,6a}$ = 2.3 Hz, H-6a), 2.90 (m, 2 H, CH₂S), 2.72 (dd, 1H, $J_{5,6a}$ = 8.0 Hz, H6b), 1.60-1.52 (m, 4 H, CH₂), 1.42-1.31 (bs, 12 H, CH₂), 0.90, 0.89 (2 t, 6 H, ${}^3J_{H,H}$ = 6.6 Hz, CH₃); ¹³C NMR (75.5 MHz, CD₃OD): δ = 99.1 (C-1), 82.6 (C-3), 81.6 (C-2), 74.6 (OCH₂), 74.0 (C-4), 73.4 (C-5), 72.2 (OCH₂), 55.5 (OCH₃), 40.0 (CH₂N), 34.2 (C-6), 33.0-31.1 (CH₂), 31.1 (CH₂S), 26.9, 26.8, 23.7 (CH₂), 14.4 (CH₃); ESI MS: m/z: 422.5 [M - CI]⁺. Anal. Calcd for C₂₁H₄₃NO₅S·HCl: calcd. C, 55.06; H, 9.68; N, 3.06; S, 7.00; found: C, 54.87; H, 9.45; N, 2.79; S, 6.78.

Methyl 2,3-di-O-Tetradecyl- α -D-glucopyranoside (18). Methyl 4,6-O-(4-methoxybenzylidene)-2,3-di-O-tetradecyl- α -D-glucopyranoside (0.70 g, 0.98 mmol) was dissolved in a mixture of Et₂O-DCM (2:1, 15 mL), under Ar atmosphere. AlCl₃ (0.81 g, 6.06 mmol) in Et₂O (15 mL) were added dropwise and the resulting mixture was refluxed for 4 h. After cooling to rt, EtOAc (150 mL) and H₂O (150 mL) were added and the layers separated. The organic layer was washed with brine (3 × 100 mL), dried (Mg₂SO₄) and evaporated. Column chromatography of the residue (1:1 EtOAc-cyclohexane) afforded 18. Yield 87% (0.50 mg, 0.85)

mmol); $R_f = 0.47$ (1:1 EtOAc-cyclohexane); $[\alpha]_D = +24.7$ (c 1.0, DCM); IR: ν_{max} 3362, 2953, 1468 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): $\delta = 4.78$ (d, 1 H, $J_{1,2} = 3.5$ Hz, H-1), 3.94-3.43 (m, 9 H, H-3, H,4 H-5, H-6a, H-6b, 2 OCH₂), 3.41 (s, 3 H, OCH₃), 3.26 (dd, 1 H, $J_{2,3} = 9.2$ Hz, H-2), 1.69-1.48 (m, 4 H, CH₂), 1.25 (bs, 44 H, CH₂), 0.87 (t, 6 H, ${}^3J_{H,H} = 6.9$ Hz, CH₃); ¹³C NMR (75.5 MHz, CDCl₃): $\delta = 98.1$ (C-1), 81.0, 80.6 (C-2, C-3), 73.6 (C-4), 71.3, 70.7 (OCH₂), 70.5 (C-5), 62.5 (C-6), 55.2 (OCH₃), 31.9-22.6 (CH₂), 14.1 (CH₃); ESI MS: m/z: 625.6 [M + K]⁺, 609.8 [M + Na]⁺. Anal. Calcd for C₃₅H₇₀O₆: C, 71.62; H, 12.02; found: C, 71.38; H, 11.76.

Methyl 2,3-di-O-Tetradecyl-6-deoxy-6-iodo-α-D-glucopyranoside (20). To a solution of 18 (0.49 g, 0.84 mmol) in toluene (17 mL) triphenylphosphine (0.33 g, 1.26 mmol) and imidazole (0.17 g, 2.52 mmol) were added. Iodine (0.33 g, 1.17 mmol) was added in portions and the resulting solution was stirred at 70 °C for 3 h. After cooling at rt, NaHCO₃ sat. (20 mL) was added and the mixture was stirred for 5 min. Additional iodine was added up to turn brown the organic phase and the mixture was stirred for 10 min. Then, aqueous 10% Na₂S₂O₃ was added to remove the iodine excess. The organic layer was separated, washed with H₂O (3 x 20 mL), dried (MgSO₄), filtered, concentrated, and purified by column chromatography (1:9 EtOAccyclohexane). Yield 91% (0.03 g, 0.76 mmol; $R_f = 0.30$ (1:9 EtOAc-cyclohexane); $[\alpha]_D = +44.1$ (c 1.0, DCM); IR : v_{max} = 1041, 722 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ = 4.81 (d, 1 H, $J_{1.2}$ = 3.6 Hz, H-1), 3.91 (m, 1 H, OCH₂), 3.66-3.49 (m, 5 H, H-6a, OCH₂, H-3), 3.47 (s, 3 H, OCH₃), 3.45 (ddd, 1 H, $J_{4,5} = 9.2$ Hz, $J_{5,6a} = 6.8$ Hz, $J_{5,6b} = 2.2$ Hz, H-5), 3.31 (m, 1 H, H-4) 3.29 (dd, 1 H, $J_{2,3}$ = 10.0 Hz, H-2), 3.26 (dd, 1 H, $J_{6a,6b}$ = 11.2 Hz, H-6b), 2.41 (d, 1 H, $J_{4,OH}$ = 2.3 Hz, OH-4), 1.61-1.54 (m, 4 H, CH₂), 1.25 (bs, 44 H, CH₂), 0.88 (t, 6 H, ${}^{3}J_{H,H} = 6.9$ Hz, CH₃); ${}^{13}C$ NMR (75.5) MHz, CDCl₃): $\delta = 98.2$ (C-1), 80.7, 80.5 (C-2, C-3), 73.8 (C-4), 73.6, 71.2 (OCH₂), 69.7 (C-5),

55.5 (OCH₃), 31.9-22.7 (CH₂), 14.1 (CH₃), 7.2 (C-6); ESI MS: m/z: 735.6 [M + K]⁺, 719.7 [M + Na]⁺. Anal. Calcd for $C_{35}H_{69}IO_5$: C, 60.33; H, 9.98. Found: C, 59.89; H, 9.72.

Methyl 6-(2-tert-Butoxycarbonylaminoethylthio)-2,3-di-O-tetradecyl-α-D-glucopyranoside (22). To a suspension of 20 (0.53 g, 0.77 mmol) and Cs_2CO_3 (0.35 g, 1.07 mmol) in DMF (7 mL), tert-butyl (2-mercaptoethyl)carbamate (0.18 mL, 1.07 mmol) was added and the reaction mixture was stirred at 60 °C, under Ar atmosphere for 24 h. The mixture was concentrated then EtOAc (20 mL) and H₂O (20 mL) were added and the organic layer was separated, washed with H₂O (3 x 20 mL), dried (MgSO₄), filtered, and concentrated. The residue was purified by column chromatography (1:3 EtOAc-cyclohexane) affording 22. Yield 95% (0.54 g,0.73 mmol); $R_f =$ 0.40 (1:3 EtOAc-cyclohexane); $[\alpha]_D = +50.4$ (c 1.0, DCM); IR: $\nu_{max} = 3631$, 1698cm^{-1} ; ^1H NMR (300 MHz, CDCl₃): $\delta = 4.99$ (bs, 1 H, NH), 4.77 (d, 1 H, $J_{1,2} = 3.6$ Hz, H-1), 3.90 (m, 1 H, OCH₂), 3.72 (ddd, 1 H, $J_{4,5} = 9.3$ Hz, $J_{5,6a} = 7.5$ Hz, $J_{5,6b} = 2.6$ Hz, H-5), 3.64-3.46 (m, 3 H, OCH_2), 3.48 (t, 1 H, $J_{2,3} = J_{3,4} = 9.3$ Hz, H-3), 3.43 (s, 3 H, OCH_3), 3.36 (t, 1 H, H-4), 3.31 (m, 2 H, CH₂N), 3.27 (dd, 1 H, H-2), 2.98 (dd, 1 H, $J_{6a,6b} = 14.1$ Hz, H-6a), 2.71 (m, 3 H, H6b, CH₂S), 2.63 (bs, 1 H, OH-4), 1.65-1.50 (m, 4 H, CH₂), 1.43 (s, 9 H, CMe₃), 1.24 (bs, 44 H, CH₂), 0.87 (t, 6 H, ${}^{3}J_{H,H} = 7.0$ Hz, CH₃); ${}^{13}C$ NMR (75.5 MHz, CDCl₃): $\delta = 156$ (CO), 97.9 (C-1), 80.9 (C-3), 80.6 (C-2), 79.4 (CMe₃), 73.5 (OCH₂), 72.4 (C-4), 71.2 (OCH₂), 71.0 (C-5), 55.2 (OCH₃), 39.6 (CH₂N), 33.5 (C-6), 33.4 (CH₂S), 31.8-29.3 (CH₂), 28.4 (CMe₃), 26.1-22.6 (CH₂), 14.0 (CH₃); ESI MS: m/z: 784.7 [M + K]⁺, 768.8 [M + Na]⁺. Anal. Calcd for C₄₂H₈₃NO₇S: C, 67.60; H, 11.21; N, 1.88; S, 4.30. Found: C, 67.45; H, 10.90; N, 1.62; S, 4.24.

Methyl 6-(2-Aminoethylthio)-2,3-di-*O*-tetradecyl-α-D-glucopyranoside Hydrochloride (4). Treatment of **22** (0.25 g, 0.33 mmol) with 1:1 TFA-DCM (2 mL) and freeze-drying from 10:1 H₂O/0.1 N HCl solution afforded **4**. Yield quant (0.22 g, , 0.33 mmol). Column chromatography

of the residue (4:1 EtOAc-cyclohexane→EtOAc→45:5:3 EtOAc-EtOH-H₂O). $R_f = 0.25$ (45:5:3 EtOAc-EtOH-H₂O); $[\alpha]_D = +41.3$ (c 0.9, 9:1 DCM-MeOH); $IR: \nu_{max} = 3406$, 722 cm⁻¹; ${}^{1}H$ NMR (300 MHz, CDCl₃): $\delta = 8.11$ (bs, 2 H, NH₂HCl), 4.78 (d, 1 H, $J_{1,2} = 3.2$ Hz, H-1), 3.87 (m, 1 H, OCH₂), 3.74 (ddd, 1 H, $J_{4,5} = 9.4$ Hz, $J_{5,6b} = 6.7$ Hz, $J_{5,6a} = 2.7$ Hz, H-5), 3.66-3.54 (m, 3 H, OCH₂), 3.49 (t, 1 H, $J_{2,3} = J_{3,4} = 8.9$ Hz, H-3), 3.42 (s, 3 H, OCH₃), 3.38 (t, 1 H, H-4), 3.27 (dd, 1 H, H-2), 3.18 (t, 2 H, ${}^{3}J_{H,H} = 6.2$ Hz, CH₂N), 2.97 (dd, 1 H, $J_{6a,6b} = 14.2$ Hz, H-6a), 2.90 (m, 2 H, CH₂S), 2.74 (dd, 1H, H6b), 1.60-1.54 (m, 4 H, CH₂), 1.25 (bs, 44 H, CH₂), 0.87 (t, 6 H, ${}^{3}J_{H,H} = 6.9.0$ Hz, CH₃); ${}^{13}C$ NMR (75.5 MHz, CDCl₃): $\delta = 97.9$ (C-1), 80.9 (C-3), 80.5 (C-2), 73.7 (OCH₂), 72.1 (C-4), 71.3 (OCH₂), 70.7 (C-5), 55.2 (OCH₃), 38.7 (CH₂N), 33.2 (C-6), 31.9 (CH₂), 30.3 (CH₂S), 30.3-22.7 (CH₂), 14.1 (CH₃); ESI MS: m/z: 646.7 [M - HCl]⁺. Anal. Calcd for C₃₇H₇₅NO₅S·HCl·2H₂O: C, 61.85; H, 11.22; N, 1.95; S, 4.46. Found: C, 61.79; H, 11.03; N, 2.01; S, 4.43.

Methyl 6-(2-tert-Butoxycarbonylaminoethylthio)-2,3,4-tri-*O*-hexanoyl-α-D-glucopyranoside (24). To a solution of 23 (0.49 g, 0.83 mmol) in dry DMF (7.5 mL), Cs₂CO₃ (0.38 g, 0.16 mmol) and *tert*-butyl *N*-(2-mercaptoethyl)carbamate (196 μL, 1.16 mmol) were added, under Ar atmosphere, and the reaction mixture was stirred at 60 °C for 48 h. The reaction mixture was concentrated and the crude product was dissolved in DCM (10 mL) and washed with H₂O (2 x 15 mL). The organic phase was dried (MgSO₄), filtered, concentrated, and the residue was purified by flash column chromatography (1:4 EtOAc-cyclohexane). Yield 85% (0.45 mg, 0.70 mmol). R_f = 0.38 (1:3 EtOAc-cyclohexane); [α]_D = +62.1 (*c* 1.0, DCM); IR: ν_{max} = 2958, 1747, 1701cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ = 5.46 (t, 1 H, $J_{2,3}$ = $J_{3,4}$ = 10 Hz, H-3), 4.95 (t, 1 H, $J_{4,5}$ = 10.0 Hz, H-4), 3.41 (s, 3 H, OCH₃), 4.90 (d, 1 H, $J_{1,2}$ = 3.7 Hz, H-1), 4.83 (dd, 1 H, H-2), 3.90 (dt, 1 H, $J_{5,6b}$ = 8.0 Hz, H-5), 3.27 (q, 2 H, $J_{H,H}$ = 6.0 Hz, C*H*₂N), 2.8-2.5 (m, 4 H, C*H*₂S, H-

 6^{a} , H-6b), 2.52 (m, 2 H, H-6a, H-6b), 2.31 (m, 6 H, H-2_{Hex},), 1.54 (m, 6 H, H-3_{Hex}), 1.41 (s, 9 H, CMe₃), 1.25 (m, 12 H, H-4_{Hex}, H-5_{Hex}), 0.86 (t, 9 H, H-6_{Hex}); ¹³C NMR (75.5 MHz, CDCl₃): δ = 172.9-172.0 (CO ester), 155.7 (CO carbamate) 96.7 (C-1), 71.5 (C-4), 70.9 (C-2), 69.3 (C-3), 69.9 (C-5), 55.1 (OMe), 39.58 (*C*H₂N), 33.7 (*C*H₂S), 33.0 (C-6), 34.0 (C-2_{Hex}), 24.5 (C-3_{Hex}), 28.2 (*CMe*₃), 22.1 (C-5_{Hex}), 13.6 (C-6_{Hex}); ESI MS: m/z = 670.4 [M + Na]⁺. Anal. Calcd for C₃₂H₅₇NO₁₀S: C, 59.32; H, 8.87; N, 2.16; S, 4.95 found: C, 59.45; H, 8.93; N, 2.31; S, 4.71.

Methyl 6-(2-Aminoethylthio)-2,3,4-tri-*O*-hexanoyl-α-D-glucopyranoside Hydrochloride (25). Treatment of 24 (0.29 g, 0.44 mmol) with 1:1 TFA-DCM (6 mL) and freeze-drying from 10:1 H₂O/0.1 N HCl solution afforded 25. Yield quant. (0.27 mg, 0.44 mmol); [α]_D =+72.0 (c 0.7, EtOAc); IR: v_{max} = 2962, 1749 cm⁻¹; ¹H NMR (300 MHz, CD₃OD): δ = 8.05 (bs, 3 H, N*H*₃), 5.28 (t, 1 H $J_{2,3}$ = 9.7 Hz, $J_{3,4}$ = 9.7 Hz, H-3), 4.94 (t, 1 H, $J_{4,5}$ = 9.7 Hz, H-4), 4.88 (m, 1 H, $J_{1,2}$ = 3.8 Hz, H-1), 4.82 (dd, 1 H, H-2), 3.8 (m, 1 H, H-5), 3.34 (s, 1 H, OCH₃), 2.93 (bs, 2 H, C*H*₂N), 2.78 (m, 2 H, C*H*₂S), 2.73 (dd, 1 H, $J_{6a,6b}$ = 14.0 Hz, $J_{5,6a}$ = 2.7, H-6^a), 2.62 (dd, 1 H, $J_{5,6b}$ = 7.7 Hz, H-6b), 2.26 (m, 6 H, H-2_{Hex}), 1.44 (m, 6 H, H-3_{Hex}), 1.21 (m, 12 H, H-4_{Hex}, H-5_{Hex}), 0.82 (t, 9 H, H-6_{Hex}); ¹³C NMR (75.5 MHz, CD₃OD): δ = 172-171 (CO), 95.98 (C-1), 71.2 (C-4), 70.6 (C-2), 69.4 (C-3), 69.1 (C-5), 55.1 (OMe), 39.1 (CH₂N), 32.5 (C-6), 30.5 (CH₂S), 33.1 (C-2_{Hex}), 31.1 (C-4_{Hex}), 24 (C-3_{Hex}), 22.1(C-2_{Hex}), 14.4 (C-6_{Hex}); ESIMS: m/z = 548.3 [M + H]⁺. Anal. Calcd for C₂₇H₅₀ClNO₈S·HCl: C, 55.51; H, 8.63; N, 2.40; S, 5.49; found: C, 55.37; H, 8.62; N, 2.26; S, 5.27.

Methyl 6-(2-(N'-(2-(N,N-di-(2-(N-tert-Butoxycarbonylamino)ethyl)amino)ethyl)thioureido)-ethylthio)-2,3,4-tri-O-hexanoyl- α -D-glucopyranoside (27). To a solution of 25 (0.12 g, 0.20 mmol) and Et₃N (56 μL, 0.4 mmol) in DCM (6 mL), 2-[N,N-bis(2-(N-tert-butoxyaminocarbonyl)ethylamino]ethyl isothiocyanate (0.09 g, 0.24 mmol) was added and the

reaction mixture was stirred, under Ar atmosphere, at rt for 48 h. The reaction mixture was washed with aqueous diluted HCl (2 x 20 mL), dried (MgSO₄), filtered and concentrated. The residue was purified by column chromatography (1:1→ 3:1 EtOAc-cyclohexane). Yield 50% (0.08 g, 0.10 mmol). $R_f = 0.2 \text{ (1:1 EtOAc-cyclohexane)}$; $[\alpha]_D = +71.6 \text{ (c 1.0, DCM)}$; IR: $\nu_{max} =$ 2959, 1748, 1685 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): $\delta = 7.46$, 6.96 (2 bs, 2 H, NHCS), 5.46 (t, 1 H, $J_{2,3} = J_{3,4} = 9.8$ Hz, H-3), 4.96 (t, 1 H, $J_{4,5} = 9.8$ Hz, H-4), 4.90 (d, 1 H, $J_{1,2} = 3.5$ Hz, H-1), 4.88 (bs, 2 H, NHBoc), 4.84 (dd, 1 H, H-2), 3.93 (bd, 1 H, H-5), 3.77 (q, 2 H, NHCH₂CH₂N), 3.52 (bs, 2 H, SCH₂CH₂N), 3.42 (s, 3 H, OCH₃), 3.11 (q, 4 H, NCH₂CH₂NHBoc), 2.82 (t, 2 H, $J_{H,H} = 7.0 \text{ Hz}, \text{ NHCH}_2\text{C}H_2\text{N}), 2.77-2.56 \text{ (m, 4 H, H-6a, H-6b, SC}H_2), 2.50 \text{ (bs, 4 H, H-6a, H-6b, SC}H_2)$ NCH_2CH_2NHBoc), 2.37-2.1 (m, 6 H, H-2_{Hex}), 1.57 (m, 6 H, $J_{H,H}$ = 7 Hz, H-3_{Hex}), 1.43 (s, 9 H, CMe₃), 1.27 (m, 12 H, H-4_{Hex}, H-5_{Hex}), 0.87 (t, 9 H, H-6_{Hex}); 13 C NMR (75.5 MHz, CDCl₃): $\delta =$ 182.1 (CS), 172.0-171.0 (CO ester), 155.2 (CO carbamate), 96.6 (C-1), 70.9 (C-4), 70.1 (C-2), 69.9 (C-5), 69.6 (C-3), 55.8 (NCH₂CH₂NHBoc), 55.5 (OMe), 54.0 (SCH₂), 44.1 (NHCH₂CH₂N), 42.4 (SCH₂CH₂N), 39.4 (NCH₂CH₂NHBoc), 33.1 (C-6), 32.6 (NHCH₂CH₂N), 34.1 (C-2_{Hex}), 31.2 (C-5_{Hex}), 28.4 (CMe₃) 24.5 (C-3_{Hex}), 22.1 (C-5_{Hex}), 13.9 (C-6_{Hex}); ESI MS: m/z = 958.6 [M + Na]⁺ Anal. Calcd for C₄₄H₈₁N₅O₁₂S₂: C, 56.44; H, 8.72; N, 7.48; S, 6.85; found: C, 56.61; H, 8.89; N, 7.21; S, 6.60.

Methyl 6-(2-(N'-(2-(N,N-Bis-(2-aminoethyl)amino)ethyl)thioureido)ethylthio)-2,3,4-tri-*O*-hexanoyl-α-D-glucopyranoside Trihydrochloride (5). Treatment of 27 (0.15 g, 0.16 mmol) with 1:1 TFA-DCM (2 mL) and freeze-drying from 10:1 H₂O/0.1 N HCl solution afforded 5. Yield quant. (0.13 g, 0.16 mmol). [α]_D = + 46.8 (c 0.85, MeOH); IR: v_{max} = 2958, 1747, 1675 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 7.98 (bs, 4 H, NHCS, NH₂HCl), 7.60 (bs, 1 H, NHCS), 5.47 (t, $J_{3,4}$ = $J_{2,3}$ = 9.5 Hz, 1 H, H-3), 4.99 (t, 1 H, $J_{4,5}$ = 9.5 Hz, H-4), 4.91 (dd, 1 H, $J_{1,2}$ = 4 Hz,

H-1), 4.85 (dd, 1 H, H-2), 3.96 (td, 1 H, $J_{5,6b} = 7.3$ Hz, $J_{5,6a} = 2.4$ Hz, H-5), 3.73 (bs, 2 H, SCH₂CH₂NHCS), 3.68 (bs, 2 H, NHCSCH₂CH₂N), 3.41 (s, 3 H, OCH₃), 3.11 (bs, 4 H, CH₂NH₃Cl), 2.84 (bs, 6 H, CH₂CH₂NH₃Cl, SCH₂CH₂NHCS), 2.75 (m, 3 H, H-6a, NHCSCH₂CH₂N), 2.66 (dd, 1 H, $J_{6a,6b} = 13.8$ Hz, H-6b), 2.32-2.19 (m, 6 H, H-2_{Hex}), 1.55 (m, 6 H, H-3_{Hex}), 1.27 (m, 12 H, H-4_{Hex}, H-5_{Hex}), 0.87 (m, 9 H, H-6_{Hex}); ¹³C NMR (75 MHz, CDCl₃): 173.1, 173.0, 172.6 (CO), 96.6 (C-1), 71.2 (C-4), 70.8 (C-2), 69.6 (C-3), 69.1 (C-5), 55.4 (OCH₃), 53.4 (NHCSCH₂CH₂N), 52.0 (CH₂CH₂NH₂HCl), 43.1 (SCH₂CH₂NHCS), 41.7 (NHCSCH₂CH₂N), 37.6 (CH₂NH₃Cl), 34.1, 34.0 (C-2_{Hex}), 33.0 (C-6), 32.7 (SCH₂CH₂NHCS), 31.2, 31.1 (C-4_{Hex}), 24.5, 24.4 (C-3_{Hex}), 22.2 (C-5_{Hex}), 13.8 (C-6_{Hex}); ESI MS: m/z = 736.4 [M]⁺. Anal. Calcd for C₃4H₆₅N₅O₈S₂·3HCl: C, 48.30; H, 8.11; N, 8.28; S, 7.58; found: C, 48.24; H, 8.39; N, 8.15; S, 7.41.

Methyl 6-(4-(2,2-Bis-tert-butoxycarbonylamino)ethylaminomethyl)-1*H*-1,2,3-triazol-1-yl-6-deoxy)-2,3,4-tri-*O*-hexanoyl-α-D-glucopyranoside (29). To a solution of 28 (0.20 g, 0.39 mmol) and 3-bis[2-tert-butoxycarbonylamino)ethyl]propargylamine (0.29 g, 0.85 mmol) in H₂O- $^{\text{t}}$ BuOH 9:1 (15 mL), the Cu-supported catalyst Si-BPA·Cu⁺ (0.02g) was added and the reaction mixture was refluxed for 36 h at 85 °C. The catalyst was filtered and the solvent was removed. The residue was purified by column chromatography (1:1→2:1 EtOAc-cyclohexane). Yield 78% (0.25 g, 0.30 mmol). R_f = 0.61 (9:1 DCM-MeOH); [α]_D = +50.5 (*c* 1.0, DCM); IR: v_{max} = 2957, 2359, 1748, 1703, 734 cm⁻¹; $^{\text{1}}$ H NMR (300 MHz, CDCl₃): δ = 7.59 (s, 1 H, =CH), 5.49 (t, 1 H, $J_{2,3} = J_{3,4} = 10.3$ Hz, H-3), 4.89 (bs, 2 H, N*H*Boc), 4.87 (d, 1 H, $J_{1,2} = 3.5$ Hz, H-1), 4.85 (t, 1 H, $J_{4,5} = 10.3$ Hz, H-4), 4.81 (dd, 1 H, H-2), 4.53 (dd, 1 H, $J_{6a,6b} = 14.0$ Hz, $J_{5,6a} = 2.6$ Hz, H-6a), 4.29 (dd, 1 H, $J_{5,6b} = 9.0$ Hz, H-6b), 4.17 (ddd, 1 H, H-5), 3.80 (m, 2 H, CH₂-triazole), 3.18 (bs, 4 H, CH₂NHBoc), 3.07 (s, 3 H, OCH₃), 2.55 (t, 4 H, CH₂CH₂NHBoc), 2.27 (m, 6 H, H-2_{Hex}), 1.58

(s, 9 H, CMe₃), 1.44 (m, 6 H, H-3_{Hex}), 1.30 (m, 12 H, H-4_{Hex},H-5_{Hex}), 0.89 (m, 9 H, H-6_{Hex}); ¹³C NMR (75.5 MHz, CDCl₃): δ = 173.9, 173.5, 173,1 (CO ester), 156.2 (CO carbamate), 144.1 (C-4 triazole), 124.3 (C-5 triazole), 91.4 (C-1), 78.1 (*C*Me₃), 70.6 (C-2), 69.8 (C-4), 69.3 (C-3), 68.0 (C-5), 53.1 (*C*H₂CH₂NHBoc), 50.6 (C-6), 48.3 (CH₂ triazole), 38.4 (*C*H₂NHBoc), 34.1 (C-2_{Hex}), 31.7 (C-4_{Hex}), 28.5 (*CMe*₃), 24.6 (C-3_{Hex}), 22.6 (C-5_{Hex}), 13.8 (C-6_{Hex}); ESI MS: m/z = 877.5 [M + Na]⁺. Anal. Calcd for C₄₂H₇₄N₆O₁₂: calcd. C, 59.00; H, 8.72; N, 9.83; found: C, 59.09; H, 8.77; N, 9.64.

Methyl 6-Deoxy-6-(4-(2,2-diaminoethylaminomethyl)-1H-1,2,3-triazol-1-yl)-2,3,4-tri-Ohexanoyl-α-D-glucopyranoside Dihydrochloride (30). Treatment of 29 (0.42 g, 0.49 mmol) with 1:1 TFA-DCM (5 mL) and freeze-drying from 10:1 H₂O/0.1 N HCl solution afforded 30. Yield quant. (0.35 g, 0.49 mmol). $[\alpha]_D = +38.1$ (c 1.0, MeOH); IR: $\nu_{max} = 2957$, 1748, 1675 cm⁻¹; ¹H NMR (300 MHz, CD₃OD): $\delta = 8.07$ (s, 1 H, =CH), 5.42 (t, 1 H, $J_{2,3} = J_{3,4} = 9.5$ Hz, H-3), 4.93 (d, 1 H, $J_{1,2}$ = 3.5 Hz, H-1), 4.82 (dd, 1 H, H-2), 4.76 (t, 1 H, $J_{4,5}$ = 9.5 Hz, H-4), 4.64 (m, 2 H, H-6a, H-6b), 4.27 (ddd, 1 H, $J_{5,6a} = 3.7$ Hz, $J_{5,6b} = 6.0$ Hz, H-5), 3.92 (s, 2 H, CH₂ triazole), 3.26 (s, 3 H, OMe), 3.16 (t, 4 H, J_{HH} = 6.4 Hz, $CH_2CH_2NH_2$), 2.82 (t, 4 H, $CH_2CH_2NH_2$), 2.48-2.15 (m, 6 H, H-2_{Hex}), 1.58 (m, 6 H, H-3_{Hex}), 1.33 (m, 12 H, H-4_{Hex}, H-5_{Hex}), 0.92 (m, 9 H, H- 6_{Hex}); ¹³C NMR (75.5 MHz, CD₃OD): $\delta = 174.2\text{-}173.9$ (CO), 143.8 (C-4 triazole), 126.9 (C-5 triazole), 98.6 (C-1), 71.8 (C-2), 70.7 (C-3), 70.3 (C-4), 68.9 (C-5), 56.0 (OCH₃), 52.0 (CH₂CH₂NH₂), 51.6 (C-6), 47.4 (CH₂ triazole), 38.2 (CH₂CH₂NH₂), 34.9, 34.8, 34.7 (C-2_{Hex}), 32.4, 32.2 (C-4_{Hex}), 25.6, 25.5 (C-3_{Hex}), 23.4 (C-5_{Hex}), 14.2 (C-6_{Hex}); ESI MS: m/z = 831.3 [M + TFA + Cl + Cu]⁺; 717.3 [M + Cu + Cl]⁺; Anal. Calcd for $C_{32}H_{59}N_6O_8 \cdot 2HCl$: calcd. C, 52.81; H, 8.31; N, 11.55; found: C, 52.69; H, 8.1; N, 11.72.

Dendritic Boc-protected Diminoethyl-bis(thiourea) Glucopyranoside Derivative 31. To a solution of **30** (0.20 g, 0.27 mmol) and Et₃N (115 μL, 0.82 mmol) in DCM (12 mL), tert-butyl N-(2-isothiocyanoethyl)carbamate (0.17 mg, 0.82 mmol) was added and the mixture was stirred overnight at rt. The reaction mixture was washed with aqueous diluted HCl (3 x 10 mL) and the organic phase was dried (MgSO₄), filtered, and concentrated. The residue was purified by column chromatography (3:1 EtOAc-cyclohexane→20:1 DCM-MeOH). Yield 52% (0.15 g, 0.14 mmol). $R_f = 0.44$ (9:1 DCM-MeOH); $[\alpha]_D = +31.7$ (c 1.0, DCM); IR: $v_{max} = 2959$, 1750, 1698, 736 cm⁻¹; UV (DCM): 248 nm (ε_{mM} 47.8). ¹H NMR (300 MHz, CDCl₃): $\delta = 7.62$ (s, 1 H, =CH), 7.18, 6.94 (bs, 4 H, NHCS), 5.49 (t, 1 H, $J_{2,3} = J_{3,4} = 9.5$ Hz, H-3), 5.38 (bs, 2 H, NHBoc), 4.82 (d, 1 H, $J_{1,2}$ = 3.5 Hz, H-1), 4.80 (t, 1 H, $J_{4,5}$ = 9.5 Hz, H-4), 4.80 (dd, 1 H, H-2), 4.53 (dd, 1 H, $J_{6a,6b} = 14.5 \text{ Hz}, J_{5,6a} = 2.6 \text{ Hz}, \text{ H-6a}, 4.37 \text{ (dd, 1 H, } J_{5,6b} = 8.0 \text{ Hz}, \text{ H-6b}), 4.17 \text{ (ddd, 1 H, H-5)},$ 3.79 (s, 2 H, CH₂ triazole), 3.62 (bs, 4 H, CH₂CH₂NHBoc), 3.54 (bs, 4 H, NCH₂CH₂NHCS), 3.31 (m, 4 H, CH₂NHBoc), 3.12 (s, 1 H, OCH₃), 2.69 (bs, 4 H, NCH₂CH₂NHCS), 2.42-2.13 (m, 6 H, H-2_{Hex}), 1.55 (m, 6 H, H-3_{Hex}), 1.42 (s, 18 H, CMe₃), 1.26 (m, 12 H, H-4_{Hex}, H-5_{Hex}), 0.87 (t, 9 H, H-6_{Hex}); 13 C NMR (75.5 MHz, CDCl₃): $\delta = 182.7$ (CS), 173.0-172.5 (CO ester), 155.5 (CO carbamate), 144.5 (C-4 triazole), 124.4 (C-5 triazole), 96.7 (C-1), 79.8 (CMe₃), 70.2 (C-2), 69.6 (C-4), 69.2 (C-3), 67.8 (C-5), 55.4 (OMe), 52.4 (NCH₂CH₂NHCS), 50.7 (C-6), 48.0 (CH₂ triazole), 44.6 (CH₂CH₂NHBoc), 42.1 (NCH₂CH₂NHCS), 40.1 (CH₂NHBoc), 34.3 (C-2_{Hex}), 30.8 $(C-4_{Hex})$, 28.1 (CMe_3) , 24.2 $(C-3_{Hex})$, 22.6 $(C-5_{Hex})$, 14.1 $(C-6_{Hex})$; ESI MS: m/z = 1081.5 [M + Na⁺. Anal. Calcd for $C_{48}H_{86}N_{10}O_{12}S_2$: calcd. C, 54.42; H, 8.18; N, 13.22; S, 6.05; found: C, 54.37; H, 7.98; N, 13.28; S, 5.85

Dendritic Diminoethyl-bis(thiourea) Glucopyranoside Dihydrochloride Derivative (6). Treatment of **3**1 (0.12 g, 0.12 mmol) with 1:1 TFA-DCM (2 mL) and freeze-drying from 10:1

H₂O/0.1 N HCl solution afforded **6**. Yield 91%(0.10 g, 0.11 mmol). [α]_D = +47.7 (c 1.0, MeOH); IR: v_{max} = 2955, 1748, 1676 cm⁻¹; UV (MeOH): 244 nm (ε_{mM} 29.1); ¹H NMR (300 MHz, CD₃OD): δ = 8.43 (s, 1 H, =CH), 5.43 (t, 1 H, $J_{2,3}$ = $J_{3,4}$ = 9.7 Hz, H-3), 4.97 (d, 1 H, $J_{1,2}$ = 3.5 Hz, H-1), 4.90 (dd, 1 H, H-2), 4.82 (t, 1 H, $J_{4,5}$ = 9.7 Hz, H-4), 4.77 (m, 2 H, CH₂ triazole), 4.70 (m, 2 H, H-6a, H-6b), 4.27 (ddd, 1 H, $J_{5,6a}$ = 3.7 Hz, $J_{5,6b}$ = 5.6 Hz, H-5), 4.06 (bs, 4 H, C H_2 CH₂NH₂), 3.87 (t, 4 H, $^3J_{H,H}$ = 5.8 Hz, NCH₂C H_2 NHCS), 3.52 (t, 4 H, $^3J_{H,H}$ = 5.8 Hz, NCH₂C H_2 NHCS), 3.52 (t, 4 H, $^3J_{H,H}$ = 5.8 Hz, C H_2 NH₂), 3.25 (s, 3 H, OCH₃), 3.21 (t, 4 H, $J_{H,H}$ = 6.0 Hz, NC H_2 CH₂NHCS), 2.50-2.15 (m, 6 H, H-2_{Hex}), 1.56 (m, 6 H, H-3_{Hex}), 1.31 (m, 12 H, H-4_{Hex}, H-5_{Hex}), 0.91 (m, 9 H, H-6_{Hex}); 13 C NMR (75.5 MHz, CD₃OD): δ = 184.1 (CS), 174.3-173.9 (CO), 137.2 (C-4 triazole), 130.1 (C-5 triazole), 98.2 (C-1), 71.8 (C-2), 71.2 (C-3), 70.5 (C-4), 68.7 (C-5), 56.2 (OMe), 54.4 (CH₂NH₂), 51.4 (C-6), 48.4 (CH₂ triazole), 42.2 (CH₂CH₂NH₂), 40.4 (NCH₂CH₂NHCS), 40.0 (CH₂NH₂), 34.4 (C-2_{Hex}), 32.0 (C-4_{Hex}), 25.2 (C-3_{Hex}), 22.8 (C-5_{Hex}), 14.0 (C-6_{Hex}). ESI MS: m/z = 859.5 [M + Na]⁺; 921.4 [M + Cu]⁺. Anal. Calcd for C₃₈H₇₂Cl₂N₁₀O₈S₂: calcd. C, 48.97; H, 7.79; N, 15.03; S, 6.88 found: C, 48.71; H, 77.40; N, 15.23; S, 6.65.

2,3,4,2',3',4'-Hexa-*O***-hexyl-6,6'-di-***O***-trityl-** α , α '-**trehalose (33).** To a solution of **32** (1.00 g, 1.21 mmol) in dry DMF (11 mL), NaH (871 mg, 21.78 mmol) was added and the mixture was stirred at 0 °C for 10 min. 1-Bromohexane (3.06 mL, 21.78 mmol) was added dropwise under Ar atmosphere and the mixture was stirred overnight at rt. The reaction was quenched with MeOH (5 mL) and stirred for 10 min. The solvents were removed and the resulting residue was suspended in DCM (50 mL), washed with H₂O (3 x 15 mL) and the organic layer was separated, dried (MgSO₄), filtered, concentrated, and purified by column chromatography (1:8 \rightarrow 1:6 EtOAc-cyclohexane). Yield 92% (1.50 g, 1.11 mmol). R_f= 0.74 (1:5 EtOAc-cyclohexane); [α]_D = +70.3 (*c* 1.0, DCM); IR: ν _{max} = 2923, 2855 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ = 7.53-7.23

(m, 30 H, Ph), 5.34 (d, 2 H, $J_{1,2}$ = 3.7 Hz, H-1), 4.03 (bd, 2 H, $J_{4,5}$ = 9.3 Hz, H-5), 3.80 (m, 2 H, OCH₂), 3.78-3.41 (m, 8 H, OCH₂), 3.57 (t, 2 H, $J_{2,3}$ = $J_{3,4}$ =9.3 Hz, H-3), 3.51-3.41 (m, 4 H, H-4, H-6a), 3.39 (dd, 2 H, H-2), 3.24 (m, 2 H, OCH₂), 3.13 (dd, 2 H, $J_{6a,6b}$ = 10.0 Hz, $J_{5,6b}$ = 3.3 Hz, H-6b), 1.65-1.56 (m, 12 H, CH₂), 1.38-1.04.89 (m, 36 H, CH₂), 0.93-0.82 (m, 18 H, CH₃); ¹³C NMR (75.5 MHz, CDCl₃): δ = 144.0, 128.8, 127.6, 126.8 (Ph), 93.6 (C-1), 86.1 (Ph₃C), 81.5 (C-3), 80.6 (C-2), 78.1 (C-4), 73.7, 73.0, 71.3 (OCH₂), 70.0 (C-5), 62.0 (C-6), 31.8-22.5 (CH₂), 14.1 (CH₃); ESI MS: m/z = 1353.8 [M + Na]⁺. Anal. Calcd for C₈₆H₁₂₂O₁₁: C, 77.55; H, 9.23. Found: C, 77.67; H, 9.31.

2,3,4,2',3',4-Hexa-O-tetradecyl-6,6'-di-O-trityl- α , α '-trehalose (34). To a solution of 32 (1.00 g, 1.21 mmol) in dry DMF (11 mL), NaH (0.87 g, 21.78 mmol) was added and the mixture was stirred at 0 °C for 10 min. 1-Bromotetradecane (6.68 mL, 21.78 mmol) was added dropwise, under Ar atmosphere, and the mixture was stirred overnight at 60 °C. The reaction was quenched with MeOH (5 mL) and stirred for 10 min. Solvents were removed and the resulting residue was suspended in DCM (50 mL). The suspension was washed with H₂O (3 x 15 mL) and the organic layer was dried (MgSO₄), filtered, concentrated and purified by column chromatography (1:50 \rightarrow 1:30 EtOAc-cyclohexane). Yield 77% (1.50 g, 0.93 mmol). R_f = 0.77 (1:15 EtOAccyclohexane); $[\alpha]_D = +50.2$ (c 1.0, DCM); ¹H NMR (300 MHz, CDCl₃): $\delta = 7.52-7.22$ (m, 30 H, Ph), 5.33 (d, 2 H, $J_{1,2}$ = 3.6 Hz, H-1), 4.03 (bd, 2 H, $J_{4,5}$ = 9.4 Hz, H-5), 3.79 (m, 2 H, OCH₂), 3.71-3.52 (m, 8 H, OCH₂), 3.55(t, 2 H, $J_{2,3} = J_{3,4} = 9.4$ Hz, H-3), 3.44 (t, 2 H, H-4), 3.46-3.35 (m, 6 H, OCH₂, H-6a, H-2), 3.22 (m, 2 H, OCH₂), 3.12 (dd, 2 H, $J_{6a,6b} = 10.0$ Hz, $J_{5,6b} = 3.1$ Hz, H-6b), 1.76-156 (m, 12 H, CH₂), 1.28 (m, 132 H, CH₂), 0.91 (t, 18 H, ${}^{3}J_{H,H} = 6.3$ Hz, CH₃); ${}^{13}C$ NMR (75.5 MHz, CDCl₃): $\delta = 144.1$, 128.8, 127.6, 126.8 (Ph), 93.7 (C-1), 86.2 (Ph₃C), 81.4 (C-3), 80.6 (C-2), 78.2 (C-4), 73.7, 73.0, 71.3 (OCH₂), 70.5(C-5), 62.1 (C-6), 31.9-22.7 (CH₂), 14.1

(CH₃); ESI MS: $m/z = 2027.4 \text{ [M + Na]}^+$. Anal. Calcd for C₁₃₄H₂₁₈O₁₁: C, 80.26; H, 10.96. Found: C, 80.35; H, 11.05.

2,3,4,2',3',4-Hexa-*O***-hexyl-α,α'-trehalose (35).** To a solution of **33** (0.68 g, 0.52 mmol) in 1:1 DCM-MeOH (25 mL), p-toluenesulphonic acid monohydrate (0.08 g, 0.42 mmol) was added and the solution was stirred at rt for 4 h. The mixture was diluted with DCM, washed with saturated aqueous NaHCO₃, dried (MgSO₄), filtered and concentrated. The resulting residue was purified by column chromatography (1:8 \rightarrow 1:2 EtOAc-cyclohexane). Yield 48% (0.21 g, 0.24 mmol); $R_f = 0.25$ (1:2 EtOAc-cyclohexane); $[\alpha]_D = +103.0$ (c 1.0, DCM); 1 H NMR (300 MHz, CDCl₃): $\delta = 5.06$ (d, 2 H, $J_{1,2} = 3.7$ Hz, H-1), 3.90 (dt, 2 H, $J_{4,5} = 9.2$ Hz, $J_{5,6a} = J_{5,6b} = 2.9$ Hz, H-5), 3.79 (m, 4 H, OCH₂), 3.73-3.63 (m, 8 H, H-6a, H-6b, OCH₂), 3.60-3.43 (m, 4 H, OCH₂), 3.57 (t, 2 H, $J_{2,3} = J_{3,4} = 9.2$ Hz, H-3), 3.23 (t, 1 H, H-4), 3.18 (dd, 1 H, H-2), 2.03 (bs, 2 H, OH), 1.60-1.47 (m, 12 H, CH₂), 1.37-1.27 (m, 36 H, CH₂), 0.87 (m, 18 H, CH₃); 13 C NMR (75.5 MHz, CDCl₃): $\delta = 93.6$ (C-1), 81.0 (C-3), 80.4 (C-2), 77.9 (C-4), 73.6, 73.3, 71.5 (OCH₂), 71.4 (C-5), 61.8 (C-6), 31.8-22.6 (CH₂), 14.0 (CH₃); ESI MS: m/z = 869.7 [M + Na]⁺, 885.7 [M + K]⁺. Anal. Calcd for C₄₈H₉₄O₁₁: C, 68.05; H, 11.18. Found: C, 67.89; H, 11.04.

2,3,4,2',3',4-Hexa-*O*-tetradecyl- α , α '-trehalose (36). To a solution of 34 (1.49 g, 0.74 mmol) in 1:1 DCM-MeOH (36 mL), *p*-toluenesulphonic acid monohydrate (0.11 g, 0.50 mmol) was added and the solution was stirred at rt for 4 h. The mixture was diluted with DCM, washed with saturated aqueous NaHCO₃, dried (MgSO₄), filtered and concentrated. Purification by column chromatography (1:9 \rightarrow 1:5 EtOAc-cyclohexane) of the residue afforded 37. Yield 47% (0.53 g, 0.34 mmol); R_f = 0.20 (1:5 EtOAc-cyclohexane); [α]_D = +58.4 (*c* 1.0, DCM); ¹H NMR (300 MHz, CDCl₃): δ = 5.06 (d, 2 H, J_{1,2} = 3.7 Hz, H-1), 3.90 (dt, 2 H, J_{4,5} = 9.2 Hz, J_{5,6a} = J_{5,6b} = 2.6 Hz, H-5), 3.84-3.64 (m, 4 H, OCH₂), 3.73-3.63 (m, 12 H, H-6a, H-6b, OCH₂), 3.58 (t, 2 H, J_{2,3}=

 $J_{3,4}$ = 9.2 Hz, H-3), 3.24 (t, 2 H, H-4), 3.19 (dd, 2 H, H-2), 1.86 (bs, 2 H, OH), 1.61-1.49 (m, 12 H, CH₂), 1.26 (bs, 132 H, CH₂), 0.88 (t, 18 H, ${}^{3}J_{H,H}$ = 6.9 Hz,CH₃); ${}^{13}C$ NMR (75.5 MHz, CDCl₃): δ = 93.7 (C-1), 81.1 (C-3), 80.5 (C-2), 78.0 (C-4), 73.6, 73.3, 71.6 (OCH₂), 71.1 (C-5), 61.9 (C-6), 31.9- 22.7 (CH₂), 14.1 (CH₃); ESI MS: m/z = 1543.2 [M + Na]⁺. Anal. Calcd for C₉₆H₁₉₀O₁₁: C, 75.83; H, 12.59. Found: C, 75.70; H, 12.41.

6,6'-Dideoxy-2,3,4,2',3',4-hexa-O-hexyl-6,6'-diiodo- α , α '-trehalose (37). To a solution of 35 (0.10 g, 0.12 mmol) in toluene (5 mL), triphenylphosphine (0.11 g, 0.43 mmol) and imidazole (0.05 g, 0.81 mmol) were added and the mixture was stirred at rt until complete dissolution. Iodine (0.11 g, 0.40 mmol) was added in portions and the solution was vigorously stirred at 70 °C for 5 h. Satd. aqueous NaHCO₃ solution (10 mL) was added and the mixture was stirred for 5 min. Additional iodine was then added until the aqueous solution turned to a slightly brown color, then aqueous 10% Na₂S₂O₃ was added until complete decoloration of both organic and aqueous layer. The organic layer was then separated, dried (MgSO₄), filtered, concentrated and purified by column chromatography (1:15 EtOAc-cyclohexane). Yield 94% (0.12 g, 0.12 mmol); $R_f = 0.75$ (1:8 EtOAc-cyclohexane); $[\alpha]_D = +50.1$ (c 1.0, DCM); ¹H NMR (300 MHz, CDCl₃): δ = 5.17 (d, 2 H, $J_{1,2}$ = 3.3 Hz, H-1), 3.83 (m, 4 H, OCH₂), 3.72-3.47 (m, 12 H, H-5, H-3, OCH₂), 3.39 (m, 4 H, H-6a, H-6b), 3.24 (dd, 1 H, Hz, $J_{2,3} = 9.1$ Hz, H-2), 3.05 (t, 1 H, $J_{3,4} = J_{3,4} = 9.1$ Hz, H-4), 1.62-1.49 (m, 12 H, CH₂), 1.30 (m, 36 H, CH₂), 0.89 (m, 18 H, CH₃); ¹³C NMR (75.5 MHz, CDCl₃): $\delta = 92.9$ (C-1), 81.9 (C-4), 80.8 (C-3), 80.2 (C-2), 73.5, 73.4, 71.9 (OCH₂), 69.1 (C-5), 31.8-22.6 (CH₂), 14.0 (CH₃), 8.7 (C-6); ESI MS: $m/z = 1089.6 \text{ [M + Na]}^+$. Anal. Calcd for C₄₈H₉₂I₂O₉: C, 54.03; H, 8.69. Found: C, 53.88; H, 8.77.

6,6'-Dideoxy-2,3,4,2',3',4-hexa-*O***-tetradecyl-6,6'-diiodo-α,α'-trehalose (38).** To a solution of **36** (0.44 g, 0.29 mmol) in toluene (13 mL), triphenylphosphine (0.27 g, 1.02 mmol) and

imidazole (0.07 g, 1.89 mmol) were added and the mixture was stirred at rt until complete dissolution. Iodine (0.26 g, 0.93 mmol) was added in portions and the solution was vigorously stirred at 70 °C for 5 h. Satd. NaHCO₃ solution (20 mL) was added and the mixture was stirred for 5 min. Additional iodine was then added until the aqueous solution got slightly brown, and then an aqueous 10% Na₂S₂O₃ solution was added until complete decoloration of both organic and aqueous layer. The organic layer was then separated, dried (MgSO₄), filtered, concentrated, and purified by column chromatography (1:25 EtOAc-cyclohexane). Yield 96% (0.52 g, 0.28 mmol); $R_f = 0.69$ (1:20 EtOAc-cyclohexane); $[\alpha]_D = +47.2$ (c 1.0, DCM); ¹H NMR (300 MHz, CDCl₃): $\delta = 5.17$ (d, 2 H, $J_{1,2} = 3.5$ Hz, H-1), 3.83 (m, 4 H, OCH₂), 3.70-3.48 (m, 12 H, H-5, H-3, OCH₂), 3.42 (dd, 2 H, $J_{6a,6b}$ = 10.7 Hz, $J_{5,6a}$ = 2.9 Hz, H-6a), 3.36 (dd, 2 H, $J_{5,6a}$ = 5.2 Hz, H-6a) 6b), 3.24 (dd, 1 H, Hz, $J_{2,3} = 9.4$ Hz, H-2), 3.05 (t, 1 H, $J_{3,4} = J_{3,4} = 9.4$ Hz, H-4), 1.59-1.51 (m, 12 H, CH₂), 1.26 (bs, 132 H, CH₂), 0.88 (t, 18 H, ${}^{3}J_{H,H} = 6.9$ Hz, CH₃); ${}^{13}C$ NMR (75.5 MHz, CDCl₃): $\delta = 92.9$ (C-1), 81.9 (C-4), 80.8 (C-3), 80.3 (C-2), 73.6, 73.5, 71.9 (OCH₂), 69.1 (C-5), 31.9- 22.7 (CH₂), 14.1 (CH₃), 8.7 (C-6); ESI MS: $m/z = 1763.0 \text{ [M + Na]}^+$. Anal. Calcd for C₉₆H₁₈₈I₂O₉: C, 66.25; H, 10.89. Found: C, 66.09; H, 10.74.

6,6'-Di-(2-tert-butoxycarbonylaminoethylthio)-2,3,4,2',3',4'-hexa-O-hexyl-α,α'-trehalose

(39). To a solution of 37 (0.11 g, 0.10 mmol) in dry DMF (12 mL), Cs₂CO₃ (0.09 g, 0.29 mmol) and *tert*-butyl *N*-(2-mercaptoethyl)carbamate (49µL, 0.29 mmol) were added, under Ar atmosphere, and the reaction mixture was stirred at 60 °C for 24 h. The reaction mixture was concentrated and the crude product was dissolved in DCM (20 mL) and washed with H₂O (2 x 30 mL). The organic phase was dried (MgSO₄), filtered, concentrated and purified by column chromatography (1:8 \rightarrow 1:6 EtOAc-cyclohexane). Yield 85 %(0.10 g, 0.08 mmol). R_f = 0.26 (1:5 EtOAc-cyclohexane); [α]_D = +86.4 (c 1.0, DCM); IR: ν _{max} = 3350, 2928, 2855, 1710 cm⁻¹, ¹H

NMR (300 MHz, CDCl₃): $\delta = 5.12$ (d, 2 H, $J_{1,2} = 3.3$ Hz, H-1), 4.99 (bs, 2 H, NHBoc), 4.01 (ddd, 2 H, $J_{4,5} = 9.2$ Hz, $J_{5,6b} = 6.3$ Hz, $J_{5,6a} = 2.7$ Hz, H-5), 3.81 (m, 6 H, H-3, OCH₂), 3.70-3.46 (m, 8 H, OCH₂), 3.58 (t, 2 H, $J_{2,3} = J_{3,4} = 9.2$ Hz, H-3), 3.29 (q, 2 H, ${}^{3}J_{H,H} = {}^{3}J_{H,NH} = 6.0$ Hz, CH₂N), 3.23 (dd, 1 H, H-2), 3.16 (t, 1 H, H-4), 2.82 (dd, 1 H, $J_{6a,6b} = 13.5$ Hz, H-6a), 2.72 (dd, 1 H, H-6b), 2.69 (t, 2 H, CH₂S), 1.60-1.47 (m, 12 H, CH₂), 1.43 (s, 18 H, C(CH₃)₃), 1.34-1.28 (m, 36 H, CH₂), 0.89 (m, 18 H, CH₃); 13 C NMR (75.5 MHz, CDCl₃): $\delta = 155.7$ (CO), 92.3 (C-1), 81.0 (C-3), 80.5 (C-2), 80.4 (C-4), 79.2 (CMe₃), 73.4, 73.2, 71.7 (OCH₂), 71.2 (C-5), 39.8 (CH₂S), 33.8 (C-6), 33.7 (CH₂N), 31.8- 30.2 (CH₂), 28.4 (CMe₃), 25.9, 25.8, 22.6 (CH₂), 14.0 (CH₃); ESI MS: m/z = 1187.9 [M + Na]⁺, 1203.8 [M + K]⁺. Anal. Calcd for C₆₂H₁₂₀N₂O₁₃S₂: C, 63.88; H, 10.38; N, 2.40 Found: C, 63.69; H, 10.21; N, 5.19.

6,6'-Di-(2-tert-butoxycarbonylaminoethylthio)-2,3,4,2',3',4'-hexa-O-tetradecyl-\alpha,\alpha'-

trehalose (40). To a solution of 38 (0.28 g, 0.13 mmol) in dry DMF (15 mL), Cs₂CO₃ (0.12 g, 0.36 mmol) and *tert*-butyl *N*-(2-mercaptoethyl)carbamate (61μL, 0.36 mmol) were added, under Ar atmosphere, and the reaction mixture was stirred at 60 °C for 24 h. The solvent was evaporated and the crude product was dissolved in DCM (20 mL) and washed with H₂O (2 x 30 mL). The organic phase was separated, dried (MgSO₄), filtered, concentrated and purified by column chromatography (1:8 EtOAc-cyclohexane). Yield 99% (0.24 g, 0.13 mmol). R_f = 0.53 (1:5 EtOAc-cyclohexane); [α]_D = +49.3 (*c* 1.0, DCM); ¹H NMR (300 MHz, CDCl₃): δ = 5.12 (d, 2 H, $J_{1,2}$ = 3.4 Hz, H-1), 4.99 (bs, 2 H, NHBoc), 4.01 (ddd, 2 H, $J_{4,5}$ = 9.6 Hz, $J_{5,6a}$ = 2.6 Hz, $J_{5,6b}$ = 6.1 Hz, H-5), 3.81 (m, 4 H, OCH₂), 3.69-3.46 (m, 8 H, OCH₂), 3.58 (t, 2 H, $J_{2,3}$ = $J_{3,4}$ = 9.6 Hz, H-3), 3.30 (bq, 2 H, $^3J_{H,H}$ = $^3J_{H,NH}$ = 6.0 Hz, CH₂N), 3.23 (dd, 1 H, H-2), 3.17 (t, 1 H, H-4), 2.83 (dd, 1 H, $J_{6a,6b}$ = 13.7 Hz, H-6a), 2.72 (dd, 1 H, H-6b), 2.69 (t, 2 H, CH₂S), 1.60-1.51 (m, 12 H, CH₂), 1.44 (s, 18 H, CMe₃), 1.26 (bs, 132 H, CH₂), 0.88 (t, 18 H, $^3J_{H,H}$ = 6.9 Hz, CH₃); 13 C NMR

 $(75.5 \text{ MHz}, \text{CDCl}_3)$: $\delta = 155.7 \text{ (CO)}, 92.3 \text{ (C-1)}, 81.0 \text{ (C-3)}, 80.5 \text{ (C-2)}, 80.4 \text{ (C-4)}, 79.3 \text{ (CMe}_3), 73.5, 73.2, 71.6 \text{ (OCH}_2), 71.2 \text{ (C-5)}, 39.8 \text{ (CH}_2\text{S)}, 33.8 \text{ (C-6)}, 33.7 \text{ (CH}_2\text{N)}, 31.9, 30.7, 30.5, 30.3, (CH}_2), 29.7 \text{ (CMe}_3), 29.4-22.7 \text{ (CH}_2), 14.1 \text{ (CH}_3); ESI MS: <math>m/z = 1861.5 \text{ [M + Na]}^+$. Anal. Calcd for $C_{110}H_{216}N_2O_{13}S_2$: C, 71.84; H, 11.84; N, 1.52; S, 3.49. Found: C, 71.90; H, 11.72; N, 1.44; S, 3.38.

6,6'-Di-(2-aminoethylthio)-2,3,4,2',3',4'-hexa-*O***-hexyl-**α,α'**-trehalose Dihydrochloride** (7). Treatment of **39** (0.10 g, 0.09 mmol) with 1:1 TFA-DCM (2 mL) and freeze-drying from 10:1 H₂O/0.1 N HCl solution afforded **7**. Yield quant. (0.09 g, 0.09 mmol); R_f = 0.72 (10:1:1CH₃CN-H₂O-NH₄OH); [α]_D = +90.3 (*c* 1.0, DCM); IR: v_{max} = 3300, 2928, 2859, 1099 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ = 8.13 (bs, 6 H, NH₃⁺), 5.20 (d, 2 H, $J_{1,2}$ = 3.2 Hz, H-1), 3.98 (dt, 2 H, $J_{4,5}$ = 9.5 Hz, $J_{5,6a}$ = $J_{5,6b}$ = 4.3 Hz, H-5), 3.79 (m, 4 H, OCH₂), 3.69-3.43 (m, 8 H, OCH₂), 3.56 (t, 2 H, $J_{2,3}$ = $J_{3,4}$ = 9.5 Hz, H-3), 3.25 (dd, 2 H, H-2), 3.18 (bs, 4 H, CH₂N), 3.11 (t, 2H, H-4), 2.97 (bd, 2 H, $J_{6a,6b}$ = 14.0 Hz, H-6a), 2.82 (m, 6 H, CH₂S, H-6b), 1.59-1.50 (m, 12 H, CH₂), 1.28 (m, 36 H, CH₂), 0.88 (m, 18 H, CH₃); ¹³C NMR (75.5 MHz, CDCl₃): δ = 91.9 (C-1), 80.9 (C-3), 80.2 (C-2), 80.0 (C-4), 73.5, 73.4, 71.8 (OCH₂), 71.0 (C-5), 39.4 (CH₂N), 34.4 (C-6), 31.8 (CH₂S), 31.7,30.6, 30.4, 30.2, 25.9, 25.8, 22.6, 22.5 (CH₂), 14.0 (CH₃); ESI MS: m/z = 965.9 [M – 2CI]⁺. Anal. Calcd for C₅₂H₁₀₂N₂O₉S₂·2HCl: C, 60.14; H, 10.29; N, 2.70; S, 6.17. Found: C, 59.86 H, 10.02; N, 2.41; S, 5.88.

6,6'-Di-(2-aminoethylthio)-2,3,4,2',3',4'-hexa-*O***-tetradecyl-α,α'-trehalose Dihydrochloride (8).** Treatment of **40** (0.09 mg, 0.05 mmol) with 1:1 TFA-DCM (1 mL) and freeze-drying from 10:1 H₂O/0.1 N HCl solution afforded **8**. Yield quant (0.09 g, 0.05 mmol); R_f = 0.21 (EtOAc); $[\alpha]_D = +61.2$ (*c* 1.0, DCM); ¹H NMR (300 MHz, CDCl₃): $\delta = 6.39$ (bs, 6 H, NH₃⁺), 5.21 (d, 2 H, $J_{1,2} = 3.5$ Hz, H-1), 3.97 (dt, 2 H, $J_{4,5} = 9.6$ Hz, $J_{5,6a} = J_{5,6b} = 4.7$ Hz, H-5), 3.79 (m, 4 H, OCH₂),

3.69-3.43 (m, 8 H, OCH₂), 3.56 (t, 2 H, $J_{2,3} = J_{3,4} = 9.6$ Hz, H-3), 3.25 (dd, 2 H, H-2), 3.18 (t, 4 H, ${}^{3}J_{H,H} = 6.3$ Hz,CH₂N), 3.11 (t, 2H, H-4), 3.07-2.94 (m, 4 H, H-6a, H-6b), 2.82 (m, 4 H, CH₂S), 1.58-1.49 (m, 12 H, CH₂), 1.26 (bs, 132 H, CH₂), 0.88 (m, 18 H, ${}^{3}J_{H,H} = 7.0$ Hz, CH₃); 13 C NMR (75.5 MHz, CDCl₃): $\delta = 91.9$ (C-1), 80.9 (C-3), 80.2 (C-2), 80.0 (C-4), 73.5, 73.4, 71.8 (OCH₂), 71.0 (C-5), 39.5 (CH₂N), 34.6 (C-6), 31.9(CH₂S), 31.9-22.7 (CH₂), 14.1 (CH₃); ESI MS: m/z = 1639.4 [M - 2 Cl]⁺. Anal. Calcd for C₁₀₀H₂₀₀N₂O₉S₂·2HCl: C, 70.17; H, 11.90; N, 1.64; S, 3.75 Found: C, 69.82; H, 11.77; N, 1.39; S, 3.41.

6,6'-Dideoxy-2,3,4,2',3',4'-hexa-O-hexanoyl-6,6'-diiodo- α , α '-trehalose (42). To a solution of 6,6'-dideoxy-6,6'-diiodo-trehalose41, (5.78 g, 10.30 mmol) and DMAP (6.37 g, 52.20 mmol) in dry DMF (80 mL), hexanoic anhydride (16 mL, 69.60 mmol) was added dropwise, under Ar atmosphere at 0° C and the reaction mixture was stirred at rt for 6 h. Then MeOH (60 mL) was added and the mixture was stirred ad rt for 2 h. The solution was poured into H₂O-ice mixture (100 mL) and extracted with DCM (50 mL). The organic phase was then washed with 2N H₂SO₄ (2 x 50 mL), H₂O (2 x 50 mL) and saturated aqueous NaHCO₃ (2 x 50 mL), dried (MgSO₄), filtered, concentrated and purified by column chromatography (1:12 EtOAc-petroleum ether). Yield 76% (1.20 g, 7.82 mmol) $R_f = 0.35$ (1:6 EtOAc-petroleum ether); ¹H NMR (300 MHz, CDCl₃): $\delta = 5.44$ (t, 1 H, $J_{2,3} = J_{3,4} = 9.5$ Hz, H-3), 5.35 (d, 1 H, $J_{1,2} = 3.7$ Hz, H-1), 5.1 (dd, 1 H, H-2), 4.84 (t, 1 H, $J_{4.5} = 9.5$ Hz, H-4), 3.82 (ddd, 1 H, $J_{5.6a} = 2.5$ Hz, H-5), 3.11 (dd, 1 H, $J_{6a.6b} =$ 11.0 Hz, H-6a), 2.97 (dd, 1 H, H-6b), 2.22 (m, 6 H, H-2_{Hex}), 1.58 (m, 6 H, H-3_{Hex}), 1.17 (m, 12 H, H-4_{Hex}, H-5_{Hex}), 0.84 (m, H-6_{Hex}). 13 C NMR (75.5 MHz, CDCl₃): $\delta = 172.6$, 172.5, 172.4 (CO), 91.5 (C-1), 72.0 (C-4), 70.0 (C-5), 69.3 (C-3), 69.2 (C-2), 34 (C-2_{Hex}), 31.2 (C-4_{Hex}), 24.4 $(C-3_{Hex})$, 22.2 $(C-5_{Hex})$, 14.1 $(C-6_{Hex})$, 2.6 $(C-6_{Hex})$. ESI MS: $m/z = 1173.4 \text{ [M + Na]}^+$. Anal. Calcd for C₄₇H₇₈I₂O₁₅: calcd. C, 49.65; H, 6.92; found: C, 50.12; H, 7.01.

2,3,4,2',3',4'-Hexa-O-hexanoyl-6,6'-bis(2-tert-butoxycarbonylaminoethylthio)-α,α'-

trehalose (43). To a solution of 42 (0.19 g, 0.17 mmol) in dry DMF (1.5 mL), Cs₂CO₃ (0.15 g, 0.47 mmol) and tert-butyl N-(2-mercaptoethyl)carbamate (80 µL, 0.47 mmol) were added, under Ar atmosphere, and the reaction mixture was stirred at 60° C for 24 h. The reaction mixture was concentrated and the crude product was dissolved in DCM (10 mL) and washed with water (2 x 20 mL). The organic phase was dried (MgSO₄), filtered and concentrated. The residue was purified by column chromatography (1:3 EtOAc-cyclohexane). Yield 58% (0.12 mg, 0.10 mmol). $R_f = 0.52$ (1:2 EtOAc-cyclohexane); $[\alpha]_D = +84.5$ (c 1.0, DCM); IR: $v_{max} = 2959$, 1749, 1709 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): $\delta = 5.43$ (t, 1 H, $J_{3,2} = J_{3,4} = 9.5$ Hz, H-3), 5.25 (d, 1 H, $J_{1,2}$ = 4.1 Hz, H-1), 5.00 (dd, 1 H, H-2), 4.94 (t, 1 H, H-4), 3.88 (m, 1 H, H-5), 3.2 (s, 2 H, CH_2N), 2.61 (m, 2 H, CH_2S), 2.52 (m, 2 H, H-6^a, H-6b), 2.23 (m, 6 H, H-2_{Hex}), 1.50 (m, 6 H, H- 3_{Hex}), 1.41 (s, 9 H, CMe₃), 1.23 (m, 12 H, H-4_{Hex}, H-5_{Hex}), 0.83 (t, 9 H, H-6_{Hex}); 13 C NMR (75.5) MHz, CDCl₃): $\delta = 172.6$, 172.5, 172.4 (C-1_{Hex}), 155.7 (CO carbamate), 91.4 (C-1), 71.2 (C-4), 71.1 (C-5), 69.6 (C-2, C-3), 39.7 (CH₂N), 34.1, 34.0 (C-2_{Hex}), 33.8 (CH₂S), 24.4 (C-3_{Hex}), 28.4 (CMe_3) , 22.3 $(C-4_{Hex}, C-5_{Hex})$, 13.8 $(C-6_{Hex})$; ESI MS: $m/z=1271.8 [M + Na]^+$. Anal. Calcd for C₆₂H₁₀₈N₂O₁₉S₂: C, 59.59; H, 8.71; N, 2.24; S, 5.13; found: C, 59.67; H, 8.69; N, 2.32; S, 4.89.

2,3,4,2',3',4'-Hexa-O-hexanoyl-6,6'-bis(2-tert-aminoethylthio)-α,α'-trehalose

dihydrochloride (9). Treatment of **43** (0.07 g, 0.04 mmol) with 1:1 TFA-DCM (2 mL) and freeze-drying from 10:1 H₂O/0.1 N HCl solution afforded **9**. Yield quant (0.07 g, 0.04 mmol). [α]_D = + 66.7 (c 0.9, EtOAc); IR: ν _{max} = 2957, 1740, 1686 cm⁻¹. ¹H NMR (300 MHz, DMSO-d₆): δ = 5.35 (t, J_{2,3} = 9.7 Hz, 1 H, H-3), 5.29 (d, 1 H, J_{1,2} = 4.0 Hz, H-1), 5.08 (m, 2 H, H-2, H-3), 3.95 (m, 1 H, H-5), 2.95 (bs, 2 H, CH₂N), 2.69 (t, 2 H, J_{H,H} = 7.0 Hz, CH₂S), 2.67 (bd, 2 H, H-6^a, H-6b), 2.27 (m, 6 H, H-2_{Hex}), 1.49 (m, 6 H, H-3_{Hex}), 1.24 (m, 12 H, H-4_{Hex}, H-5_{Hex}), 0.85 (t,

9 H, H-6_{Hex}); 13 C NMR (75.5 MHz, DMSO-d₆): $\delta = 171.8$, 171.7 (CO), 90.9 (C-1), 70.9 (C-5), 70.0, 69.2, (C-2, C-4), 69.4 (C-3), 39.2 (CH₂N), 31.5 (C-6), 33.9 (C-2_{Hex}), 30.3 (CH₂S), 23.9 (C- 3_{Hex}), 30.6 (C-4_{Hex}), 21.8 (C-5_{Hex}), 13.3 (C-6_{Hex}); ESI MS: $m/z = 1049.6 \text{ [M]}^+$. Anal. Calcd for C₅₂H₉₄Cl₂N₂O₁₅S₂: C, 55.65; H, 8.44; N, 2.50; S, 5.71; found: C, 55.29; H, 8.18; N, 2.14; S, 5.33 6,6'-Diazido-6,6'-dideoxy-2,3,4,2',3',4'-hexa-O-hexanoyl- α , α '-trehalose (45). To a solution of 42 (1.58 g, 1.37 mmol) in dry DMF (8 mL), NaN3 (0.25 g, 3.60 mmol) was added. The reaction mixture was stirred overnight at 40 °C, under Ar atmosphere. The mixture was poured into H₂O-ice mixture (20 mL), and the product was extracted with DCM (4 x 20 mL). The organic phase was dried (MgSO₄), filtered and concentrated. The residue was purified by column chromatography (1:12 EtOAc-cyclohexane) to afford 45. Yield 72% (0.91 g, 0.98 mmol). $R_f =$ 0.53 (1:6 EtOAc-cyclohexane). $[\alpha]_D = +108.7$ (c 1.0, DCM); IR: $\nu_{max} = 2958$, 2104, 1750, 735 cm⁻¹; ¹H NMR (300 MHz, CD₃OD): $\delta = 5.52$ (t, 2 H, $J_{2,3} = J_{3,4} = 9.8$ Hz, H-3), 5.41 (d, 2 H, $J_{1,2} = J_{1,2} = J_{1,2}$ = 4.0 Hz, H-1), 5.13 (dd, 2 H, H-2), 5.09 (t, 2 H, $J_{4.5}$ = 9.8 Hz, H-4), 4.08 (ddd, 2 H, $J_{5.6a}$ = 7.0 Hz, $J_{5,6b} = 2.6$ Hz, H-5), 3.44 (dd, 2 H, $J_{6a,6b} = 13.0$ Hz, H-6a), 2.97 (dd, 2 H, H-6b), 2.33 (m, 12) H, H-2_{Hex}), 1.58 (m, 12 H, H-3_{Hex}), 1.33 (m, 24 H, H-4_{Hex},H-5_{Hex}), 0.91 (m, 18 H, H-6_{Hex}); ¹³C NMR (75.5 MHz, CD₃OD): $\delta = 175.2-174.2$ (CO), 93.7 (C-1), 71.5 (C-3), 71.3 (C-2), 71.6 (C-1) 5), 71 (C-4), 52.2 (C-6), 34.9 (C-2_{Hex}), 33.2 (C-4_{Hex}), 25.8 (C-3_{Hex}), 23.6 (C-5_{Hex}), 14.8 (C-6_{Hex}); ESI MS: $m/z = 1003.5 \text{ [M + Na]}^+$. Anal. Calcd for $C_{25}H_{43}N_8O_{15}$: calcd. C, 58.76; H, 8.22; N, 8.57; found: C, 58.84; H, 8.32; N, 8.60.

6,6'-Di-(4-tert-butoxycarbonylaminomethyl-1H-1,2,3-triazol-1-yl)-6,6'-dideoxy-

2,3,4,2',3',4'-hexa-*O***-hexanoyl-α,α'-trehalose (46).** To a solution of **44** (0.25 g, 0.25 mmol) and **45** (0.08 g, 0.56 mmol) in H₂O-^tBuOH 9:1 (15 mL) the Cu-supported catalyst Si-BPA·Cu⁺ (0.02 g) was added and the reaction mixture was refluxed for 36 h at 85 °C. The catalyst was

filtered and the solvent was concentrated. The residue was purified by column chromatography (1:1 \rightarrow 2:1 EtOAc-cyclohexane). Yield quant. (0.33 g, 0.25 mmol). R_f =0.73 (2:1 EtOAc-cyclohexane). [α]_D = +56.7 (c 1.0, DCM); IR: v_{max} = 2957, 1752, 1714, 735 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ = 7.52 (s, 2 H, =CH), 5.45 (t, 2 H, $J_{2,3} = J_{3,4} = 9.8$ Hz, H-3), 5.30 (bs, 2 H, NHBoc), 4.97 (dd, 2 H, $J_{1,2}$ = 4.0 Hz, H-2), 4.87 (t, 2 H, $J_{4,5}$ = 9.8 Hz, H-4), 4.83 (d, 2 H, H-1), 4.48 (bd, 2 H, $J_{6a,6b}$ = 13.7 Hz, H-6a), 4.37 (d, 4 H, $J_{H,H}$ = 6.0 Hz, CH₂ triazole), 4.12 (dd, 2 H, $J_{5,6b}$ = 8.8 Hz, H-6b), 4.07 (m, 2 H, H-5), 2.25 (m, 12 H, H-2_{Hex}), 1.55 (m, 12 H, H-3_{Hex}), 1.46 (s, 18 H, CMe₃), 1.29 (m, 24 H, H-4_{Hex}, H-5_{Hex}), 0.89 (m, 18 H, H-6_{Hex}); ¹³C NMR (75.5 MHz, CD₃OD): δ = 173.5-173.2 (CO ester), 155.5 (CO carbamate), 146.0 (C-4 triazole), 122.3 (C-5 triazole), 91.7 (C-1), 79.5 (CMe₃), 69.9 (C-3, C-5), 69.8 (C-4), 69.3 (C-2), 50.7 (C-6), 36.4 (CH₂-triazole), 34.3 (C-2_{Hex}), 31.9 (C-4_{Hex}), 28.7 (CMe₃) 24.6 (C-3_{Hex}), 22.6 (C-5_{Hex}), 14.1 (C-6_{Hex}); ESI MS: m/z = 1313.4 [M + Na]⁺. Anal. Calcd for C₆₄H₁₀₆N₈O₁₉: calcd. C, 59.52; H, 8.27; N, 8.68; found: C, 59.61; H, 8.33; N, 8.84.

6,6'-Di-(4-Aminomethyl-1*H***-1,2,3-triazol-1-yl)-6,6'-dideoxy-2,3,4,2',3',4'-hexa-***O***-hexanoyl)** -α,α'-trehalose Dihydrochloride (47). Treatment of 46 (0.30 g, 0.23 mmol) with 1:1 TFA-DCM (4 mL) and freeze-drying from 10:1 H₂O/0.1 N HCl solution afforded 47. Yield 97% (0.26 g, 0.22 mmol). [α]_D = +48.9 (c 1.0, MeOH); IR: v_{max} = 2956, 1751, 1464, 1026 cm⁻¹; ¹H NMR (300 MHz, CD₃OD): δ = 8.05 (s, 2 H, CH triazole), 5.49 (t, 2 H, $J_{3,4}$ = $J_{2,3}$ = 9.8 Hz, H-3), 5.06 (dd, 2 H, $J_{1,2}$ = 4.0 Hz, H-2), 4.96 (d, 2 H, $J_{1,2}$ = 3.6 Hz, H-1), 4.95 (t, 2 H, $J_{4,5}$ = 9.8 Hz, H-4), 4.63 (dd, 2 H, $J_{5,6a}$ = 2.6 Hz, $J_{5,6b}$ = 14.6 Hz, H-6), 4.53 (dd, 2 H, $J_{5,6b}$ = 7.9 Hz, H-6b), 4.27 (bs, 4 H, CH_2 NH₂), 4.24 (m, 1 H, H-5), 2.32 (m, 12 H, H-2_{Hex}), 1.59 (m, 12 H, H-3_{Hex}), 1.33 (m, 24 H, H-4_{Hex},H-5_{Hex}), 0.93 (m, 18 H, H-6_{Hex}); ¹³C NMR (75.5 MHz, CD₃OD): δ = 174.5-173.6 (CO), 141.5 (C-4 triazole) 127.1 (C-5 triazole), 92.7 (C-1), 71.2 (C-2), 70.6 (C-3), 70.4 (C-4),

70.3 (C-5), 51.5 (C-6), 35.2 (CH_2NH_2), 34.8 ($C-2_{Hex}$), 32.1 ($C-4_{Hex}$), 25.8 ($C-3_{Hex}$), 22.6 ($C-5_{Hex}$), 14.2 ($C-6_{Hex}$); ESI MS: m/z = 1091.4 [M]⁺. Anal. Calcd for $C_{54}H_{92}N_8O_{15}$: calcd. C, 55.71; H, 7.96; N, 9.62; found: C, 55.48; H, 7.77; N, 9.97.

6,6'-Di-[4-(2-N'-(2-(N-tert-butoxycarbonyl)aminoethyl)thioureido))methyl-1H-1,2,3-

triazolyl- $[2,3,4,2',3',4'-hexa-O-hexanoyl)-6,6'-dideoxy-\alpha,\alpha'-trehalose (48).$ To a solution of 47 (0.14 g, 0.12 mmol) and Et₃N (37 μ L, 0.36 mmol) in DCM (10 mL), tert-butyl N-(2isothiocyanoethyl) carbamate (0.07 g, 0.36 mmol) was added and the reaction mixture was stirred overnight at rt. The reaction mixture was washed with aqueous diluted HCl (3 x 10 mL) and the organic phase was dried (MgSO₄), filtered, and concentrated. The residue was purified by column chromatography (2:1 \rightarrow 3:1 EtOAc-cyclohexane). Yield 67% (0.12 g, 0.08 mmol). R_f = 0.40 (3:1 EtOAc-cyclohexane). $[\alpha]_D$ = +59.8 (*c* 1.0, DCM); UV (DCM): 249 nm (ϵ_{mM} 53.9); IR: $v_{\text{max}} = 2959$, 1752, 1701, cm⁻¹; ¹H NMR (300 MHz, CDCl₃): $\delta = 7.80$ (s, 2 H, CH triazole), 7.07 (bs, 4 H, NHCS), 5.39 (t, 2 H, $J_{3,4} = J_{2,3} = 9.7$ Hz, H-3), 5.29 (bs, 2 H, NHBoc), 4.98-4.76 (m, 4 H, H-1, H-4, CH₂ triazole), 4.62 (dd, 2 H, H-2), 4.47 (d, 2 H, $J_{6a.6b}$ = 14.0 Hz, H-6^a), 4.28 $(dd, 2 H, J_{5.6b} = 8.0 Hz, H-6b), 3.97 (bt, 2 H, H-5), 3.60 (bs, 4 H, CH₂CH₂NHBoc), 3.31 (m, 4 H,$ $^{3}J_{H,H} = 5.4 \text{ Hz}, \text{ C}H_{2}\text{NHBoc}), 2.22 \text{ (m, 12 H, H-2}_{Hex}), 1.53 \text{ (m, 12 H, H-3}_{Hex}), 1.39 \text{ (s, 18 H, H-2}_{Hex})$ CMe₃), 1.27 (m, 24 H, H-4 $_{Hex}$, H-5 $_{Hex}$), 0.86 (m, 18 H, H-6 $_{Hex}$); 13 C NMR (75.5 MHz, CDCl₃): δ = 173.2-172.5 (CO ester), 156.2 (CO carbamate), 141.5 (C-4 triazole), 124.3 (C-5 triazole), 96.7 (C-1), 79.7 (CMe₃), 69.6-68.7 (C-2, C-3, C-4, C-5), 50.3 (C-6), 40.0 (CH₂CH₂NHBoc), 39.5 (CH₂NHBoc), 33.9 (C-2_{Hex}), 31.2 (C-4_{Hex}), 28.3 (CMe₃), 24.3 (C-3_{Hex}), 22.2 (C-5_{Hex}), 13.1 (C- 6_{Hex}). ESI MS: $m/z = 1517.4 \text{ [M + Na]}^+$. Anal. Calcd for $C_{70}H_{118}N_{12}O_{19}S_2$: calcd. C, 56.20; H, 7.95; N, 11.24; O, 20.32; S, 4.29; found: C, 55.95; H, 7.84; N, 11.04; S, 11.04

[(6,6'-Di-4-(2-N'-(2-aminoehtyl)thioureido))methyl-1*H*-1,2,3-triazol-1-yl)-6,6'-dideoxy-2,3,4,2',3',4'-hexa-O-hexanoyl]- α,α' -trehalose Dihydrochloride (10). Treatment of 48 (0.11) g, 0.07 mmol) with TFA-DCM (2 mL) and freeze-drying from 0.1 N HCl solution afforded 10. Yield quant (0.10 g, 0.07 mmol). $[\alpha]_D = +38.2$ (c 1.0, MeOH); UV (MeOH): 243 nm (ϵ_{mM} 16.9); IR: $v_{max} = 2956$, 2862, 1752, 1675 721 cm⁻¹; ¹H NMR (300 MHz, CD₃OD): $\delta = 7.92$ (s, 2 H, CH triazole), 5.44 (t, 2 H, $J_{2,3} = J_{3,4} = 9.7$ Hz, H-3), 5.08 (dd, 2 H, $J_{1,2} = 4.2$ Hz, H-2), 4.95 (t, 2 H, H-4), 4.82 (m, 4 H, H-4, H-1, CH₂ triazole), 4.62 (dd, 2 H, $J_{6a,6b} = 14.0$ Hz, $J_{5,6a} = 2.5$ Hz, H-6^a), $4.48 \text{ (dd, 2 H, } J_{5,6b} = 8.2 \text{ Hz, H-6b)}, 4.19 \text{ (ddd, 2 H, H-5)}, 3.91 \text{ (m, 4 H, C} H_2\text{CH}_2\text{NH}_2\text{)}, 3.21 \text{ (dt, 4 H, C} H_2\text{CH}_2\text{CH}_2\text{NH}_2\text{)}, 3.21 \text{ (dt, 4 H, C} H_2\text{CH}_2\text{CH}_2\text{NH}_2\text{)}, 3.21 \text{ (dt, 4 H, C$ H, $J_{H,H} = 5.8$ Hz, CH_2NH_2), 2.31 (m, 12 H, H-2_{Hex}), 1.59 (m, 12 H, H-3_{Hex}), 1.32 (m, 24 H, H- 4_{Hex} , H-5_{Hex}), 0.93 (m, 18 H, H-6_{Hex}); ¹³C NMR (75.5 MHz, CD₃OD): $\delta = 182.6$ (CS), 176.0-174.5 (CO), 146.1 (C-4 triazole), 125.6 (C-5 triazole), 92.4 (C-1), 71.4 (C-4), 70.6 (C-3), 70.4 (C-5), 70.3 (C-2), 51.5 (C-6), 42.6 (CH₂CH₂NH₂), 40.9 (CH₂NH₂), 40.2 (CH₂-triazole), 35-34.8 $(C-2_{Hex})$, 32.5-32.4 $(C-4_{Hex})$, 25.7-25.4 $(C-3_{Hex})$, 23.2 $(C-5_{Hex})$, 14.3 $(C-6_{Hex})$; ESI MS: m/z = $1357 [M + Cu]^+$; $711 [M + Cu]^{2+}$ Anal. Calcd for $C_{60}H_{104}N_{12}O_{15}S_2$: calcd. C, 52.66; H, 7.66; N, 12.28; S, 4.69; found: C, 52.71; H, 7.50; N, 12.55, 4.69.

6,6'[-4-(2,2-Bis-*tert*-butoxycarbonylamino)ethylaminomethyl)-1*H*-1,2,3-triazol-1-yl)-6-deoxy-2,3-di-*O*-hexanoyl]- α , α '-trehalose (49). To a solution of 44 (0.31 g, 0.31 mmol) and 29 (0.24 g, 0.73 mmol) in H₂O-^tBuOH 9:1(15 mL), the Cu-supported catalyst Si-BPA·Cu⁺ (0.02 g) was added and the reaction mixture was refluxed for 36 h at 85 °C. The catalyst was filtered and the solvent was removed. The residue was purified by column chromatography (2:1 \rightarrow 3:1 EtOAc-cyclohexane). Yield 91% (0.54 g, 0.29 mmol). R_f = 0.36 (3:1 EtOAc-cyclohexane). [α]_D = +57.0 (c 1.0, DCM); IR: ν _{max} = 2958, 2359, 2341, 1751, 1700 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ = 7.57 (s, 2 H, =C*H*), 5.47 (t, 2 H, J_{2,3} = J_{3,4} = 9.8 Hz, H-3), 5.20 (bs, 4 H, NHBoc),

4.93 (dd, 2 H, $J_{1,2}$ = 3.8 Hz, H-2), 4.86 (t, 2 H, $J_{4,5}$ = 10.0 Hz, H-4), 4.77 (d, 2 H, H-1), 4.46 (bd, 2 H, $J_{6a,6b}$ = 14.0 Hz, H-6^a), 4.25 (dd, 2 H, $J_{5,6b}$ = 8.7 Hz, H-6b), 4.12 (m, 2 H, H-5), 3.83 (bs, 4 H, CH₂ triazole), 3.19 (bd, 8 H, ${}^{3}J_{H,H}$ = 6.0 Hz, $CH_{2}NHBoc$), 2.55 (t, 8 H, $J_{H,H}$ = 6.0 Hz, $CH_{2}CH_{2}NHBoc$), 2.27 (m, 12 H, H-2_{Hex}), 1.58 (s, 36 H, CMe₃), 1.44 (m, 12 H, H-3_{Hex}), 1.30 (m, 24 H, H-4_{Hex}, H-5_{Hex}), 0.89 (m, 18 H, H-6_{Hex}); ${}^{13}C$ NMR (75.5 MHz, CDCl₃): δ = 174.2-173.0 (CO ester), 156.2 (CO carbamate), 143.9 (C-4 triazole), 124.3 (C-5 triazole), 91.4 (C-1), 79.1 (CMe_{3}), 69.5 (C-4), 69.3 (C-3), 69.2 (C-5), 68.8 (C-2), 53.1 ($CH_{2}CH_{2}NHBoc$), 50.6 (C-6), 48.0 (CH₂-triazole), 38.4 ($CH_{2}NHBoc$), 34.1 (C-2_{Hex}), 31.7 (C-4_{Hex}), 28.5 (CMe_{3}), 24.6 (C-3_{Hex}), 22.6 (C-5_{Hex}), 13.8 (C-6_{Hex}); ESI MS: m/z = 1686.6 [M + Na]⁺ Anal. Calcd for $C_{82}H_{142}N_{12}O_{23}$: calcd. C, 59.18; H, 8.60; N, 10.10; found: C, 59.20; H, 8.51; N, 10.15.

6,6'[4-(2,2-Diaminoethylaminomethyl)-1*H***-1,2,3-triazol-1-yl)-6,6'-dideoxy-2,3,4,2',3',4'-tri- O-hexanoyl]-α,α'-trehalose Tetrahydrochloride (11)**. Treatment of **49** (0.56 g, 0.34 mmol) with TFA-DCM 1:1 (6 mL) and freeze-drying from 0.1 N HCl solution afforded **11**. Yield quant. (0.48 g, 0.33 mmol). [α]_D = +22.1 (*c* 1.0, MeOH); IR: v_{max} = 2956, 2356, 1753, 1676, 721 cm⁻¹; ¹H NMR (300 MHz, CD₃OD): δ = 8.33 (s, 2 H, CH triazole), 5.51 (t, 2 H, $J_{3,4}$ = $J_{2,3}$ = 9.7 Hz, H-3), 5.08 (dd, 2 H, $J_{1,2}$ = 3.9 Hz, H-2), 5.01 (t, 2 H, $J_{4,5}$ = 9.7 Hz, H-4), 4.95 (d, 2 H, H-1), 4.65 (dd, 2 H, $J_{5,6a}$ = 3.0 Hz, $J_{6a,6b}$ = 14.5 Hz, H-6a), 4.56 (dd, 2 H, $J_{5,6b}$ = 8.1 Hz, H-6b), 4.28 (bs, 4 H, CH₂ triazole), 4.18 (ddd, 2 H, H-5), 3.41 (bt, 8 H, $^3J_{H,H}$ = $^3J_{H,NH}$ = 6.0 Hz, CH₂NH₂), 3.22 (bt, 8 H, CH₂CH₂NH₂), 2.31 (m, 12 H, H-2_{Hex}), 1.59 (m, 12 H, H-3_{Hex}), 1.33 (m, 24 H, H-4_{Hex}, H-5_{Hex}), 0.93 (m, 18 H, H-6_{Hex}); 13 C NMR (75.5 MHz, CD₃OD): δ = 174.2-173.2 (CO), 141.1 (C-4 triazole), 128.8 (C-5 triazole), 91.9 (C-1), 71.0 (C-4), 70.6 (C-3), 70.5 (C-5), 70.4 (C-2), 51.8 (C-6), 51.6 (CH₂CH₂NH₂), 47.9 (CH₂-triazole), 37.2 (CH₂NH₂), 35.1-34.8 (C-2_{Hex}), 32.5-32.4 (C-4_{Hex}), 25.6-25.5 (C-3_{Hex}), 23.4 (C-5_{Hex}), 14.3 (C-6_{Hex}); ESI MS: m/z = 1263.5 [M]⁺; 632.0 [M]²⁺.

Anal. Calcd for $C_{62}H_{114}N_{12}O_{15}$: calcd. C, 52.83; H, 8.15; N, 11.93; found: C, 52.87; H, 8.04; N, 11.74.

Determination of CMC via Pyrene Fluorescence Measurements. In order to assess the amphiphilicity, the critical micelle concentrations (CMC) of all derivatives have been determined using an established fluorescence technique based on pyrene. 46 This extremely hydrophobic dye is preferentially incorporated in the interior of micelles. The onset of micelle formation can be observed in a shift of the fluorescence excitation spectra of the samples at an emission wavelength of 372 nm. In the concentration range of aqueous micellar solutions, a shift of the excitation band in the 335 nm region toward higher wavelengths confirms the incorporation of pyrene in the hydrophobic interior of micelles. The ratio of the fluorescence intensities at 339 and 335 nm was used to quantify the shift of the broad excitation band. The critical micelle concentrations were determined from the crossover point in the low concentration range. Fluorescence spectra were recorded with an F-2500 Hitachi spectrofluorophotometer and conventional 1-cm quartz cuvettes at 37 ± 0.1 °C, using 2.5 mm excitation and emission slits. Synthesis of Dodecanethiol Coated Gold Nanoparticles (DDT-Au NPs). A solution of tetrachloroaurate acid in milli-Q water (25 mL, 0.03 M) was mixed with a solution of tetraoctylammonium bromide in toluene (80 mL, 0.05 M). The two phases mixture was vigorously stirred until all the tetrachloroaurate was transferred into the organic layer, and the aqueous layer was discarded. To the solution was added dropwise a NaBH₄ aqueous solution (25 mL, 0.35 M) for 1 minute, then the mixture was stirred for 1 h. The biphasic system was washed with 0.01 M HCl (1 x 25 mL), 0.01 M NaOH (1 x 25 mL) and H₂O milli-Q (3 x 25 mL). Aqueous layers were discarded and the organic phase was stirred overnight at rt. Dodecanethiol (10 mL, 42 mmol) was added and the mixture was refluxed for 3 h. The system was cooled to rt

and spin-dried at 2000 rpm for 10 min. The supernatant was recovered, and MeOH was added to reach 1:1 mixture to precipitate the NPs and eliminate the excess of dodecanethiol, and the system was spin-dried for 10 min at 2000 rpm. Then supernatant was discarded and the precipitate was suspended in 1 mL of CHCl₃.

The final concentration of DDT-Au NPs was determined by UV spectrometry. A small aliquot of NPs was 1000-fold diluted and absorbance was measured using a $\varepsilon = 8.63 \cdot 10^6 \, \text{M}^{-1} \text{cm}^{-1}$.

Coating of DDT-Au_NPs with Compound 11. To a solution of DDT-Au NP (120 μ L) in CHCl₃ (30 mL) was added a solution of compound 11 in MeOH (1 mL, 2 mM). The mixture was concentrated and milli-Q water (300 μ L) was added before 1 min sonication. The dark red solution was spin-dried in a ultrafiltration device with a polyethersulfone membrane (Corning® Spin-X® UF) for 5 min at 6000 rpm. The precipitate was recovered by addition of milli-Q water (500 μ L). Final concentration was determined measuring the absorbance at 450 nm using an ϵ = 3.07· 10⁷ M⁻¹cm⁻¹.

Biological Assays

Reagents and Cell Cultures. Expression plasmid for mouse MD-2 was a gift from Dr. Y. Nagai (University of Tokyo, Japan). Expression plasmid for mouse TLR4 was purchased from InvivoGen (CA, USA). Expression plasmids containing sequences of human TLR4 and MD-2 as well as the pELAM-1 firefly luciferase plasmid were a gift from Dr. C. Kirschning (Technical University of Munich, Germany). The Renilla luciferase phRL-TK plasmid was purchased from Promega (WI, USA).

The human embryonic kidney (HEK) 293 cells were provided by Dr. J. Chow (Eisai Research Institute, Andover, USA). HEK293 cells were grown in DMEM supplemented with 10 % FBS. Compounds were dissolved in 100% DMSO to provide 4 mM stock solutions; further working

dilutions were prepared immediately before stimulation with cell medium (DMEM supplemented with 10 % FBS).

Cell Activation Assay – NF-κB-Luciferase Reporter Assay. HEK 293 cells were seeded in 96-well Costar plates (Corning, NY, USA) at 1.6·104 cells/well and incubated overnight in a humidified atmosphere (5 % CO₂) at 37 °C. The next day, when cells were 40-60 % confluent, they were co-transfected with MD-2 (10 ng), NF-κB-dependent luciferase (70 ng) and constitutive Renilla (15 ng) reporter plasmids and TLR4 plasmid (1 ng) using PEI (7.5 molar polyethylenimine pH 7.5, Polysciences) transfection reagent. Cells were stimulated 4 hours after transfection with the synthetic compounds, then1 h later with LPS (5 nM) that was extensively vortexed immediately prior to stimulation. Cells were lysed after 16 hours of stimulation in 1x reporter assay lysis buffer (Promega, USA) and analyzed for reporter gene activities using a dual-luciferase reporter assay system. Relative luciferase activity (RLA) was calculated by normalizing each sample's firefly luciferase activity for constitutive Renilla activity measured within the same sample. When plotting the data the value of the wild type MD-2·TLR4 sample stimulated with LPS was normalized to 100 and other values were adjusted accordingly.

HEK-BlueTM Assay. HEK-Blue-TLR4 cells (InvivoGen) were cultured according to manufacturer's instructions. Briefly, cells were cultured in DMEM high glucose medium supplemented with 10% fetal bovine serum (FBS), 2 mM glutamine, 1x Normocin (InvivoGen), 1x HEK-Blue Selection (InvivoGen). Cells were detached by the use of a cell scraper and the cell concentration was estimated by using Trypan Blue (Sigma Aldrich). The cells were diluted in DMEM high glucose medium supplemented as described before and seeded in multiwell plate at a density of 2x10⁴ cells/well in 200 μL. After overnight incubation (37°C, 5% CO₂, 95% humidity), supernatant was removed, cell monolayers were washed with warm PBS without Ca²⁺

and Mg^{2+} and treated with increasing concentrations of compounds dissolved in DMSO-ethanol (1:1). After 30 minutes, the cells were stimulated with 10 nM LPS from *E. coli* O55:B5 (Sigma Aldrich) and incubated overnight at 37°C, 5% CO_2 and 95% humidity. As a control, the cells were treated with or without LPS (10 nM) alone. Then, the supernatants were collected and 50 μ L of each sample was added to 100 μ L PBS, pH 8, 0.84 mM para-Nitrophenylphosphate (pNPP) for a final concentration of 0.8 mM pNPP. Plates were incubated for 2-4 h in the dark at rt and then the plate reading was assessed by using a spectrophotometer at 405 nm (LT 4000, Labtech). The results were normalized with positive control (LPS alone) and expressed as the mean of percentage \pm SD of at least three independent experiments.

MTT Cell Viability Assay. HEK-Blue cells were seeded in 100 μ L DMEM without Phenol Red at a density of $2x10^4$ cells per well. After overnight incubation, 10 μ L compounds were added and the plates were incubated overnight at 37 °C, 5% CO₂, 95% humidity. DMSO and PBS were included as control. Then 10 μ l of MTT solution (5 mg/mL in PBS) were added to each well. After 3 h incubation (37 °C, 5% CO₂, 95% humidity), HCl 0.1 N in isopropanol was added (100 μ l/well) to dissolve formazan crystals. Formazan concentration in the wells was determined measuring the absorbance at 570 nm (LT 4000, Labtech). The results were normalized with untreated control (PBS) and expressed as the mean of percentage \pm SD of three independent experiments.

In vivo Endotoxin Inhibition. C57BL/6J mice (11-13 weeks old) were randomly assigned into groups and injected intraperitoneally with vehicle control (5% DMSO in PBS) (groups none and LPS only) or the inhibitory compound ($2x10^{-7}$ mol compound/mouse for compounds **5-11**, all in 5% DMSO solution). One hour later the mice were injected intraperitoneally with vehicle control (PBS) (group none) or with LPS from *E. coli* 055:B5 ($1x10^{-9}$ mol / mouse $\approx 10 \mu g$ LPS / mouse).

Three hours later the blood was collected. Serum was tested with the mouse TNF-α ELISA kit

("ReadySetGo", eBioscience) to determine the levels of mouse TNF-a. The experiment was

performed according to the manufacturer's instructions.

Supporting Information. Activity of compounds 5, 9, 10 on bone marrow-derived murine

macrophages (BMDM); MTT cell toxicity tests for synthetic molecules. This material is

available free of charge via the Internet at http://pubs.acs.org.

AUTHOR INFORMATION

Corresponding Author

E-mail: francesco.peri@unimib.it

ACKNOWLEDGMENTS

This study was financially supported by the Italian Ministry of University and Research (MIUR),

PRIN 2010-11, project: "Italian network for the development of multivalent nanosystems," the

Spanish Ministerio de Economía y Competitividad (contract numbers SAF2013-44021-R and

CTQ2010-15848), the Junta de Andalucía (contract number FQM 1467), COST actions CM1102

"MultiGlycoNano" and BM 1003 "Microbial cell surface determinants of virulence as targets for

new therapeutics for Cystic Fibrosis"; the Slovenian Research Agency grant P4-176 and J1-

4170, and the European Regional Development Funds (FEDER and FSE). Technical assistance

from the research support services of the University of Seville (CITIUS) is acknowledged.

ABBREVIATIONS

BMDM, bone marrow derived machophages; CNS, central nervous system; DC, dendritic cell;

DCC, dicyclohexylcarbodiimide; DMEM, Dulbecco's modified Eagle's medium; DMAP, 4-

dimethylaminopyridine; DMSO, dimethysulfoxide; HEK, human embryonic kidney; PTSA, p-

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toluenesulfonic acid; MTT, 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide test; NF-κB, nuclear factor kappa-light-chain-enhancer of activated B cells; TPP, triphenylphosphine; TRAM, TRIF-related adaptor molecule; TRIF, TIR-domain-containing adapter-inducing interferon-β.

REFERENCES

- 1. Akira, S.; Takeda, K. Toll-like receptor signalling. *Nat. Rev. Immunol.* **2004,** *4*, 499-511.
- 2. Poltorak, A.; He, X.; Smirnova, I.; Liu, M.; Van Huffel, C.; Du, X.; Birdwell, D.; Alejos, E.; Silva, M.; Galanos, C.; Freudenberg, M.; Ricciardi-Castagnoli, P.; Layton, B.; Beutler, B. Defective LPS signaling in C3H/HeJ and C57BL/10ScCr mice: mutations in TLR4 gene. *Science* **1998**, *282*, 2085-2088.
- 3. Beutler, B. TLR4 as the mammalian endotoxin sensor. *Curr. Top. Microbiol. Immunol.* **2002,** *270*, 109-120.
- 4. Lucas, K.; Maes, M. Role of the Toll like receptor (TLR) radical cycle in chronic inflammation: possible treatments targeting the TLR4 pathway. *Mol. Neurobiol.* **2013**, *48* (*1*), 190-204.
- 5. Cribbs, S.; Martin, G. Expanding the global epidemiology of sepsis. *Crit. Care Med.* **2007,** *35*, 2646-2648.
- 6. Peri, F.; Calabrese, V. Toll-like Receptor 4 (TLR4) modulation by synthetic and natural compounds: an update. *J. Med. Chem.* **2014,** *57*, 3612-3622.
- 7. Peri, F.; Piazza, M. Therapeutic targeting of innate immunity with Toll-like receptor 4 (TLR4) antagonists. *Biotechnol. Adv.* **2012**, *30*, 251-260.
- 8. Jerala, R. Structural biology of the LPS recognition. *Int. J. Med. Microbiol.* **2007,** 297, 353-363.

- 9. Schumann, R.; Leong, S.; Flaggs, G.; Gray, P.; Wright, S.; Mathison, J.; Tobias, P.; Ulevitch, R. Structure and function of lipopolysaccharide binding protein. *Science* **1990**, *249*, 1429-1431.
- 10. Wright, S.; Ramos, R.; Tobias, P.; Ulevitch, R.; Mathison, J. CD14, a receptor for complexes of lipopolysaccharide (LPS) and LPS binding protein. *Science* **1990**, *249*, 1431-1433.
- 11. Shimazu, R.; Akashi, S.; Ogata, H.; Nagai, Y.; Fukudome, K.; Miyake, K.; Kimoto, M. MD-2, a molecule that confers lipopolysaccharide responsiveness on Toll-like receptor 4. *J. Exp. Med.* **1999**, *189*, 1777-1782.
- 12. Gioannini, T.; Teghanemt, A.; Zhang, D.; Coussens, N.; Dockstader, W.; Ramaswamy, S.; Weiss, J. Isolation of an endotoxin-MD-2 complex that produces Toll-like receptor 4-dependent cell activation at picomolar concentrations. *Proc. Natl. Acad. Sci. U.S.A* **2004**, *101*, 4186-4191.
- 13. Park, B.; Song, D.; Kim, H.; Choi, B.; Lee, H.; Lee, J. The structural basis of lipopolysaccharide recognition by the TLR4-MD-2 complex. *Nature* **2009**, *458*, 1191-1195.
- 14. Baldridge, J.; McGowan, P.; Evans, J.; Cluff, C.; Mossman, S.; Johnson, D.; Persing, D. Taking a Toll on human disease: Toll-like receptor 4 agonists as vaccine adjuvants and monotherapeutic agents. *Expert. Opin. Biol. Ther.* **2004**, *4*, 1129-1138.
- 15. Golenbock, D. T.; Hampton, R. Y.; Qureshi, N.; Takayama, K.; Raetz, C. R. Lipid A-like molecules that antagonize the effects of endotoxins on human monocytes. *J. Biol. Chem.* **1991,** *266*, 19490-19498.
- 16. Kim, H.; Park, B.; Kim, J.; Kim, S.; Lee, J.; Oh, S.; Enkhbayar, P.; Matsushima, N.; Lee, H.; Yoo, O.; Lee, J. Crystal structure of the TLR4-MD-2 complex with bound endotoxin antagonist Eritoran. *Cell* **2007**, *130*, 906-917.

- 17. Lonez, C.; Vandenbranden, M.; Ruysschaert, J. M. Cationic lipids activate intracellular signaling pathways. *Adv. Drug. Deliv. Rev.* **2012,** *64*, 1749-1758.
- 18. Tanaka, T.; Legat, A.; Adam, E.; Steuve, J.; Gatot, J. S.; Vandenbranden, M.; Ulianov, L.; Lonez, C.; Ruysschaert, J. M.; Muraille, E.; Tuynder, M.; Goldman, M.; Jacquet, A. DiC14-amidine cationic liposomes stimulate myeloid dendritic cells through Toll-like receptor 4. *Eur. J. Immunol.* **2008**, *38*, 1351-1357.
- 19. Yan, W.; Chen, W.; Huang, L. Mechanism of adjuvant activity of cationic liposome: phosphorylation of a MAP kinase, ERK and induction of chemokines. *Mol. Immunol.* **2007**, *44*, 3672-3681.
- 20. Vangasseri, D. P.; Cui, Z.; Chen, W.; Hokey, D. A.; Falo, L. D.; Huang, L. Immunostimulation of dendritic cells by cationic liposomes. *Mol. Membr. Biol.* **2006**, *23*, 385-395.
- 21. Leon-Ponte, M.; Kirchhof, M. G.; Sun, T.; Stephens, T.; Singh, B.; Sandhu, S.; Madrenas, J. Polycationic lipids inhibit the pro-inflammatory response to LPS. *Immunol. Lett.* **2005**, *96*, 73-83.
- 22. Piazza, M.; Yu, L.; Teghanemt, A.; Gioannini, T.; Weiss, J.; Peri, F. Evidence of a specific interaction between new synthetic antisepsis agents and CD14. *Biochemistry* **2009**, *48*, 12337-12344.
- 23. Piazza, M.; Calabrese, V.; Baruffa, C.; Gioannini, T.; Weiss, J.; Peri, F. The cationic amphiphile 3,4-bis(tetradecyloxy)benzylamine inhibits LPS signaling by competing with endotoxin for CD14 binding. *Biochem. Pharmacol.* **2010**, *80*, 2050-2056.

- 24. Piazza, M.; Rossini, C.; Della Fiorentina, S.; Pozzi, C.; Comelli, F.; Bettoni, I.; Fusi, P.; Costa, B.; Peri, F. Glycolipids and benzylammonium lipids as novel antisepsis agents: synthesis and biological characterization. *J. Med. Chem.* **2009**, *52*, 1209-1213.
- 25. Kim, J.; Lee, C.; Jin, M.; Lee, C.; Paik, S.; Lee, H.; Lee, J. Crystal structure of CD14 and its implications for lipopolysaccharide signaling. *J. Biol. Chem.* **2005**, *280*, 11347-11351.
- 26. Ortiz Mellet, C.; Benito, J. M.; García Fernández, J. M. Preorganized, macromolecular, gene-delivery systems. *Chemistry* **2010**, *16*, 6728-6742.
- 27. Tvaroska, I.; Bleha, T. Anomeric and Exo-Anomeric Effects in Carbohydrate Chemistry. *Adv. Carbohydr. Chem. Biochem.* **1989**, *47*, 45-123.
- 28. Rodríguez-Lucena, D.; Benito, J. M.; Alvarez, E.; Jaime, C.; Perez-Miron, J.; Ortiz Mellet, C.; García Fernández, J. M. Synthesis, structure, and inclusion capabilities of trehalose-based cyclodextrin analogues (cyclotrehalans). *J. Org. Chem.* **2008**, *73*, 2967-2979.
- 29. Cighetti, R.; Ciaramelli, C.; Sestito, S. E.; Zanoni, I.; Kubik, Ł.; Ardá-Freire, A.; Calabrese, V.; Granucci, F.; Jerala, R.; Martín-Santamaría, S.; Jiménez-Barbero, J.; Peri, F. Modulation of CD14 and TLR4·MD-2 activities by a synthetic lipid A mimetic. *ChemBioChem* **2014**, *15*, 250-258.
- 30. Artner, D.; Oblak, A.; Ittig, S.; Garate, J. A.; Horvat, S.; Arrieumerlou, C.; Hofinger, A.; Oostenbrink, C.; Jerala, R.; Kosma, P.; Zamyatina, A. Conformationally constrained lipid A mimetics for exploration of structural basis of TLR4/MD-2 activation by lipopolysaccharide. *ACS Chem. Biol.* **2013**, *8*, 2423-2432.
- 31. Barath, M.; Petrusova, M.; Bystricky, S.; Křen, V.; Petruš, L. Synthesis of a precursor of a lipid A mimic. *Chem. Pap.* **2003**, 57, 125-130.

- 32. Gobolobov, Y. G.; Zhmurova, I. N.; Kasukhin, L. F. Sixty years of Staudinger reaction. *Tetrahedron* **1981,** *37*, 437-472.
- 33. Peri, F.; Marinzi, C.; Barath, M.; Granucci, F.; Urbano, M.; Nicotra, F. Synthesis and biological evaluation of novel lipid A antagonists. *Bioorg. Med. Chem.* **2006**, *14*, 190-199.
- 34. Chen, J.; Dormán, G.; Prestwich, G.; Glenn, D. Asymmetric total synthesis of D-myoinositol 1,2,4,5-tetrakisphosphateand its P-2-(*O*-aminopropyl) derivative. *J. Org. Chem.* **1996,** *61*, 393-397.
- 35. Garegg, P. J.; Samuelsson, B. Novel reagent system for converting a hydroxy-group into an iodo-group in carbohydrates with inversion of configuration. Part 2. *J. Chem. Soc., Perkin Trans. I* **1980**, 2866-2868.
- 36. Méndez-Ardoy, A.; Guilloteau, N.; Di Giorgio, C.; Vierling, P.; Santoyo-González, F.; Ortiz Mellet, C.; García Fernández, J. M. β-Cyclodextrin-based polycationic amphiphilic "click" clusters: effect of structural modifications in their DNA complexing and delivery properties. *J. Org. Chem.* **2011**, *76*, 5882-5894.
- 37. Díaz-Moscoso, A.; Le Gourriérec, L.; Gómez-García, M.; Benito, J. M.; Balbuena, P.; Ortega-Caballero, F.; Guilloteau, N.; Di Giorgio, C.; Vierling, P.; Defaye, J.; Ortiz Mellet, C.; García Fernández, J. M. Polycationic amphiphilic cyclodextrins for gene delivery: synthesis and effect of structural modifications on plasmid DNA complex stability, cytotoxicity, and gene expression. *Chemistry* **2009**, *15*, 12871-12888.
- 38. Méndez-Ardoy, A.; Gómez-García, M.; Ortiz Mellet, C.; Sevillano, N.; Girón, M. D.; Salto, R.; Santoyo-González, F.; García Fernández, J. M. Preorganized macromolecular gene delivery systems: amphiphilic beta-cyclodextrin "click clusters". *Org. Biomol. Chem.* **2009,** *7*, 2681-2684.

- 39. Megia-Fernandez, A.; Ortega-Muñoz, M.; Lopez-Jaramillo, J.; Hernandez-Mateo, F.; Santoyo-Gonzalez, F. Non-magnetic and magnetic supported copper(I) chelating adsorbents as efficient heterogeneous catalysts and copper scavengers for click chemistry. *Adv. Synth. Catal.* **2010,** *352*, 3306-3320.
- 40. Gallego-Yerga, L.; González-Álvarez, M. J.; Mayordomo, N.; Santoyo-González, F.; Benito, J. M.; Ortiz Mellet, C.; Mendicuti, F.; García Fernández, J. M. Dynamic self-assembly of polycationic clusters based on cyclodextrins for pH-sensitive DNA nanocondensation and delivery by component design. *Chemistry* **2014**, *20*, 6622-6627.
- 41. Kneeland, D. M.; Ariga, K.; Lynch, V. M.; Huang, C. Y.; Anslyn, E. V. Bis(alkylguanidinium) receptors for phosphodiesters: effect of counterions, solvent mixtures, and cavity flexibility on complexation. *J. Am. Chem. Soc.* **1993**, *115*, 10042-10055.
- 42. Liav, A.; Goren, M. B. An improved synthesis of (2,3,4-tri-O-acetyl-a-D-glucopyranosyl)uronic acid (2,3,4-tri-O-acetyi-a-D-glucopyranosid)uronic acid. *Carbohydr*. *Res.* **1980,** *84*, 171-174.
- 43. García Fernández, J. M.; Ortiz Mellet, C.; Jiménez Blanco, J. L.; Fuentes Mota, J.; Gadelle, A.; Coste-Sarguet, A.; Defaye, J. Isothiocyanates and cyclic thiocarbamates of α,α' -trehalose, sucrose, and cyclomaltooligosaccharides. *Carbohydr. Res.* **1995**, *268*, 57-61.
- 44. Shirey, K. A.; Lai, W.; Scott, A. J.; Lipsky, M.; Mistry, P.; Pletneva, L. M.; Karp, C. L.; McAlees, J.; Gioannini, T. L.; Weiss, J.; Chen, W. H.; Ernst, R. K.; Rossignol, D. P.; Gusovsky, F.; Blanco, J. C.; Vogel, S. N. The TLR4 antagonist Eritoran protects mice from lethal influenza infection. *Nature* **2013**, *497*, 498–502

- 45. Ohto, U.; Fukase, K.; Miyake, K.; Shimizu, T. Structural basis of species-specific endotoxin sensing by innate immune receptor TLR4/MD-2. *Proc. Natl. Acad. Sci. U.S.A.* **2012,** *109*, 7421-7426.
- 46. Hofmann, A. M.; Wurm, F.; Frey, H. Rapid access to polyfunctional lipids with complex architecture via oxyanionic ring-opening polymerization. *Macromolecules*, **2011**, *44*, 4648-4657.
- 47. Piazza, M.; Colombo, M.; Zanoni, I.; Granucci, F.; Tortora, P.; Weiss, J.; Gioannini, T.; Prosperi, D.; Peri, F. Uniform lipopolysaccharide (LPS)-loaded magnetic nanoparticles for the investigation of LPS-TLR4 Signaling. *Angew. Chem., Int. Ed.* **2011,** *50*, 622-626.
- 48. Barr, T. A.; Krembuszewski, M.; Gupta, M.; Gray, D.; Mareque-Rivas, J. C. Quantum dots decorated with pathogen associated molecular patterns as fluorescent synthetic pathogen models. *Mol. Biosyst.* **2010**, *6*, 1572-1575.
- 49. Pellegrino, T.; Manna, L.; Kudera, S.; Liedl, T.; Koktysh, D.; Rogach, A. L.; Keller, S.; Rädler, J.; Natile, G.; Parak, W. Hydrophobic nanocrystals coated with an amphiphilic polymer Shell: a general route to water soluble nanocrystals. *Nano Lett.* **2004**, *4*, 703-707.
- 50. Klar, T. A.; Dulkeith, E.; Feldmann, J. *Time-resolved fluorescence measurements of fluorophores close to metal nanoparticles*. Springer: Berlin, 2004; Vol. 8, pp 249-273.

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