

1 **Drying kinetics and effective water diffusivities in olive stone and olive-**  
2 **tree pruning**

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11

12 **Abstract**

13 Drying kinetics of olive stone and olive-tree pruning, two important biomasses from olive  
14 culture, was experimentally assessed at six different temperatures (from 343 K to 393 K)  
15 and four sample thicknesses (from 15 to 50 mm). Analysis of the drying curves revealed  
16 that Page’s model was suitable for predicting the drying characteristics of both solid  
17 biofuels. From this analysis, two new mathematical equations to describe the dependence  
18 of moisture ratio with temperature and drying time were also proposed. The values of  
19 effective water diffusivity, calculated at the falling rate period by using Fick’s second law  
20 of diffusion, increased when increasing drying temperature and sample thickness.  
21 Diffusivities for olive-tree pruning ( $3.41 \times 10^{-8} - 32.5 \times 10^{-8} \text{ m}^2/\text{s}$ ) were almost twice  
22 higher than those for olive stone ( $1.87 \times 10^{-8} - 16.4 \times 10^{-8} \text{ m}^2/\text{s}$ ).

23 **Keywords:** olive stone; olive-tree pruning; drying; mathematical modelling; effective  
24 diffusivity.

## 25 **1. Introduction**

26 Developed countries are constantly striving to achieve a stable source of renewable  
27 energy to reduce their dependence on fossil fuels and fight against global warming. In the  
28 European Union, biomass is expected to contribute to around half of the renewable energy  
29 share in 2020 according to projections [1]. Therefore, the use of all available biomasses  
30 for energy production in a sustainable way should be taken into account.

31 Over 11 million ha olive trees are cultivated worldwide, mainly in the Mediterranean  
32 countries [2]. The olive culture plays a vital role in the economic and social development  
33 of these countries. Spain produces about 33% of the world's olive oil [3]. More than 350  
34 million olives are grown all over Spain, most of them in the southern region of Andalusia  
35 where olive groves represent roughly  $1.55 \times 10^6$  ha cultivated land [4].

36 The industrial processing of the olive fruit as well the management of olive orchard  
37 generates lignocellulosic by-products such as olive stone and olive-tree pruning,  
38 respectively. In olive oil mills, decanters of two exits separate a liquid stream (olive oil)  
39 and a solid stream (pomace) [5]. Fragmented olive stones are then separated from olive  
40 pomace by pitting machines. Around 0.6 t olive stones are estimated to be produced in  
41 the olive mills per hectare olive grove [2], resulting in a worldwide production of  $6.6 \times$   
42  $10^6$  t/year. Olive stone (OP) can be used for different applications (bioethanol [6],  
43 activated carbon [7], furfural production [8], natural antioxidants [9] ...) however the vast  
44 majority of olive stone production is used for thermal energy generation due to its high  
45 heating value and density, low ash, nitrogen and sulphur content, and uniform particle  
46 size [10]. On the other hand, pruning operation is essential to enhance the productivity of  
47 olive grove because it helps to remove unproductive branches thus enhancing air  
48 circulation and light penetration through the foliage to prevent microbial diseases. Given  
49 that one hectare of olive grove can annually produce on average 3 t of pruning [2,11], it

50 can be estimated that the annual worldwide production of this biomass is around  $3.3 \times$   
51  $10^7$  t, which illustrates the great availability and potential of this biomass for energy  
52 production. However, in most cases olive-tree pruning (OP) is left on the land to be  
53 incinerated or ploughed into the soil with the disadvantages that this can result in: soil  
54 mineralization, air pollution and fire risks [11]. Biomass from olive-tree pruning can be  
55 used as fuel for heating systems in boilers [11,12], and its current price in Andalusia is  
56 30-40 €/t while that of olive stone is 80-100 €/t, so that several industries in Spain are  
57 currently producing pellets from this lignocellulose material.

58 Water content in biomass is a key factor when the material is used for combustion.  
59 High moisture percentage increases the cost of transport and pelletizing, reduces the  
60 combustion efficiency and causes water vapour condensation, which can reduce the  
61 lifetime of boilers. Besides, during storage and handling of biomass, high moisture levels  
62 could promote microbial activity that is harmful to human health [13]. The monitoring of  
63 moisture level of biomass is of major importance to obtain high-quality pellets, the  
64 required water content for the pellet press being lower than 10%. Other thermo-chemical  
65 processes such as pyrolysis and gasification of biomass are normally carried out at low  
66 water contents [14,15]. For OS, an analysis of 15 samples collected from different  
67 factories in Andalusia showed that its moisture percentages ranged between 10.2% and  
68 30.5%, with an average value of 22.3% [10], while some studies about OP reported  
69 moisture values in the range 30–35% [16]. Therefore, a drying operation should be  
70 applied to these biomasses before storage or other operations to reduce their moisture  
71 content to a suitable level.

72 Drying process occurs in several stages for high-moisture solid materials. During the  
73 initial stage (warming-up period) the rate of drying increases due to the temperature  
74 increase. In the second stage (constant-rate period) the surface of the solid is saturated

75 with free water and the rate of moisture removal is constant and maximum. In the final  
76 stage of drying (falling-rate period), the area of the saturated surface gradually decreases  
77 when the water movement within the solid can no longer supply enough water to wet the  
78 surface. Therefore, the instantaneous drying rate continually decreases in this stage. The  
79 rate-controlling factors in the falling-rate period are complex, being water diffusion the  
80 predominant phenomenon in this stage [17]. Effective water diffusivity ( $D_{\text{eff}}$ ) is an  
81 important drying parameter for biomass materials, useful for estimating drying times in  
82 the falling rate period and for designing and modelling the mass transfer during this period  
83 [18]. The  $D_{\text{eff}}$  value usually varies with the thickness of the material and the external  
84 drying conditions (gas temperature and rate). Several authors have applied Fick's second  
85 law of diffusion for  $D_{\text{eff}}$  values determination in by-products derived from olive grove.  
86 Thus, drying of wet olive pomaces has been researched at a wide range of temperatures  
87 (293–413 K) and sample thicknesses (7–63 mm) [19–21] while olive stone drying has  
88 only been studied at four temperatures (373, 423, 473 and 523 K) and three sample  
89 thicknesses (10, 20 and 30 mm) [22,23]. To the best of our knowledge, there are not works  
90 in the literature about  $D_{\text{eff}}$  determination in olive-tree pruning and the drying kinetics of  
91 OP has been not studied so far.

92 The two main objectives of the present study were: (a) to study the drying kinetics of  
93 both biomasses as a function of process time, air temperature and sample thickness and  
94 (b) to determine the effective water diffusivities during drying processes. These data  
95 would complete the available information on drying of biomasses derived from olive  
96 culture, which is of major importance for the design of dryers.

97

## 98 **2. Materials and methods**

### 99 *2.1. Raw materials*

100 The biomass samples used in this study were olive stone and olive-tree pruning. Olive  
101 stones, olive endocarps crushed into fragments, were collected from an olive oil mill  
102 (S.C.A. San Juan, Jaén province, Spain, UTM coordinates: 37°47′58.57″N,  
103 3°47′07.97″W) and air-dried at room temperature ( $293\pm 2$  K) in laboratory for 10 days.  
104 These olive stones came from olives of the variety 'Picual'. Olive-tree prunings were  
105 collected on-site after fruit-harvesting from a 'Picual' olive grove situated in Alhama de  
106 Granada (Granada province, Spain, UTM coordinates: 37°01′59.08″N, 3°56′10.80″W).  
107 This biomass consisted of thin branches (< 5 cm diameter) and leaves. Leaves were  
108 removed from the woody fraction. The branches were air-dried at room temperature  
109 ( $293\pm 2$  K) in laboratory for 10 days and then grounded using a blade mill (Retsch, mod.  
110 SM1, Germany). Particle size distributions of OS and OP were determined using a  
111 vibratory screen (Restch, Mod. Vibro, Germany).

112 Prior to drying experiments, OS and OP were submerged in a distilled water bath at  
113 293 K for 2 h and then submitted to gravity filtration for 15 min to obtain wet samples of  
114 olive stone (WOS) and wet samples of olive-tree pruning (WOP). Using 14 samples of  
115 each wet biomass, the moisture contents (wet basis) were found to be  $22.9\pm 1.0\%$  and  
116  $51.4\pm 3.3\%$  for WOS and WOP, respectively.

## 117 *2.2. Physical–chemical characterization of raw materials*

### 118 *2.2.1. Bulk density*

119 Bulk density of biomass particles represents the ratio between the mass of biomass and  
120 its volume including the contribution of the interparticulate void volume. This physical  
121 property was determined according to a method previously described [24]. A glass  
122 graduated cylinder, with a total volume of 10 mL and an inner diameter of 16 mm, was  
123 used. A funnel filled with biomass was allowed to flow freely into the cylinder at a height  
124 of 160 mm. Solid mass in the container was determined using an electronic balance

125 ( $\pm 0.001$  g accuracy). Filling and weighing were repeated five times to calculate five  
126 values for bulk density.

### 127 2.2.2. *Raw materials composition*

128 Moisture content of the samples was analysed by drying in an oven at 378 K for 24 h  
129 (TAPPI T 264 cm-07). Extractives in raw materials were determined gravimetrically  
130 using a two-step sequential extraction process to remove water and ethanol soluble  
131 materials [25]. Determinations of structural carbohydrates (cellulose and hemicelluloses),  
132 acid-insoluble lignin (AIL) and acid soluble lignin (ASL) in raw materials were carried  
133 out using a two-step acid hydrolysis previously described [26]. Ash was determined  
134 according to TAPPI Standard Method T 15 os-58. All the analyses were carried out in  
135 duplicate.

### 136 2.2.3. *Elemental analysis*

137 Carbon, hydrogen, nitrogen and sulphur content in the raw materials were determined  
138 according to ASTM Standard method D5142-09 by using Thermo Finnigan Flash  
139 EA1112 CHNS-O Elemental Analyser. Oxygen content was calculated by subtraction of  
140 the CHNS content from the total content.

### 141 2.2.4. *Higher heating value (HHV)*

142 Higher heating value (HHV) of biomasses was measured by using an automatic Parr  
143 6400 calorimeter. According to the results from elemental analysis, the value of HHV  
144 was also calculated by Demirbas' equation [27]:

$$\text{HHV (MJ/kg)} = 0.335C + 1.423H - 0.154O - 0.145N \quad (1)$$

145 where C, H, O and N are the weight percentages of carbon, hydrogen, oxygen and  
146 nitrogen in the biomass, respectively. Results were expressed as MJ/kg in dry weight  
147 basis.

### 148 2.2.5. *Equilibrium moisture content (EMC)*

149 Olive stones and olive-pruning debris were exposed to three constant levels of relative  
150 humidity at 282 K (43.1%, 57.4% and 77.5%) and 303 K (43.2%, 51.4% and 73.1%).  
151 Relative humidity was maintained by using the static gravimetric method and different  
152 saturated salt solutions ( $K_2CO_3$ ,  $Mg(NO_3)_2$  and  $NaNO_3$ ), which can provide the  
153 respective humidity conditions [28]. Samples and salt solutions were maintained  
154 separately within a sealed container. The mass of each sample was initially weighed and  
155 then periodically removed, weighed and replaced in the container. Equilibrium was  
156 achieved when three consecutive weight measurements showed a difference of less than  
157 1 mg. When equilibrium was reached, the samples were dried in an oven ( $378\pm 1$  K for  
158 24 h) in order to obtain the dry matter content. All measurements were performed in  
159 duplicate.

### 160 *2.3. Drying experiments and mathematical modelling of drying curves*

161 The cabinet dryer (Selecta, Mod. 204, Barcelona, Spain) used for drying experiments  
162 had a height of 500 mm, a width of 400 mm and a depth of 450 mm. The dryer was  
163 composed of an electrical heater and a temperature controller. The relative humidity of  
164 the ambient air ranged from 50 to 70% during experiments. Drying experiments were  
165 performed by natural convection at 343, 353, 363, 373, 383 and 393 K for sample  
166 thicknesses of 25 and 50 mm. Furthermore, two additional assays were carried out at 393  
167 K with sample thicknesses of 15 and 40 mm.

168 The samples were dried in 100-mL beakers (85 mm height and 46 mm internal  
169 diameter). For all test, weight variation with time was recorded by removing the beakers  
170 from the dryer, weighing them on a digital balance ( $\pm 0.0001$  g accuracy) and immediately  
171 returning them to the dryer. Each process of weight measurement lasted about 10 s. When  
172 drying process finished, the samples were dried in an oven ( $378\pm 1$  K for 24 h) in order to  
173 obtain the dry matter content. All drying experiments were performed in duplicate. The

174 relative error was generally less than 7%. The standard deviation of the data is shown in  
 175 the Figures.

176 Drying curves represent the moisture ratio ( $M_R$ ) function versus drying time. The  
 177 dimensionless moisture ratio for thin layer drying can be expressed as (Eq. 2):

$$M_R = (M_t - M_e)/(M_0 - M_e) \quad (2)$$

178 where  $M_t$  is the moisture content at a given time (kg water/kg dry matter),  $M_0$  is the initial  
 179 moisture content and  $M_e$  is the equilibrium moisture content. Since the values of  $M_e$  are  
 180 small compared to  $M_t$  or  $M_0$ , the dimensionless moisture ratio could be expressed like,

$$M_R = M_t / M_0 \quad (3)$$

181 In our previous research [29], the experimental drying curves were fitted to six  
 182 equations widely used to describe the kinetics of the drying process, namely, Lewis [30],  
 183 Page [31], modified Page [32], Henderson and Pabis [33], logarithmic [34] and Midilli  
 184 [35] models. These mathematical models for drying curves are illustrated in Table 1.

185 *TABLE 1*

186 Non-linear regression techniques were used to obtain the different constants in each  
 187 selected model using the function ‘Solver’ in a Microsoft Excel spreadsheet. The  
 188 coefficients of determination ( $r^2$ ), reduced chi-squared ( $\chi_r^2$ ) and root mean square error  
 189 (RMSE) were calculated to evaluate the fitting of each model to experimental data. The  
 190 higher the values of  $r^2$  and the lower the values of  $\chi_r^2$  and RMSE, the better the goodness  
 191 of fit. These parameters can be calculated as,

$$r^2 = \frac{\sum_{i=1}^N (M_{R-pre,i} - \overline{M_{R-exp,i}})^2}{\sum_{i=1}^N (M_{R-exp,i} - \overline{M_{R-exp,i}})^2} \quad (4)$$

193



$$\chi_r^2 = \frac{\sum_{i=1}^N (M_{R-\text{exp},i} - M_{R-\text{pre},i})^2}{N-p} \quad (5)$$

194

$$RMSE = \left[ \frac{1}{N} \sum_{i=1}^N (M_{R-\text{pre},i} - M_{R-\text{exp},i})^2 \right]^{1/2} \quad (6)$$

195

196 where  $M_{R-\text{exp},i}$  is the experimental moisture ratio,  $M_{R-\text{pre},i}$  the predicted moisture ratio,  $N$   
 197 the number of data point and  $p$  is the number of constants in the regression model.

198 Tables S1, S2 and S3 illustrate, the fit values obtained for one particle size (50 mm)  
 199 and 3 temperatures (70, 100 and 120 °C). The two-parameter Page model and four-  
 200 parameter Midilli model presented the best fit results [29]. However, Page model (Eq. 7)  
 201 was selected for this work because of its better balance between accuracy and analytical  
 202 simplicity,

$$M_R = \exp(-k t^n) \quad (7)$$

203 where  $t$  is drying time in hours, and  $k$  and  $n$  are empirical constants of the model.

204

### 205 **3. Results and discussion**

#### 206 *3.1. Biomasses*

207 Olive stone and olive-tree pruning are two types of biomass with a high energetic  
 208 potential in the European countries of the Mediterranean basin. Some physical, chemical  
 209 and energetic properties of these materials are shown in Table 2. In relation to particle  
 210 size distribution, OS had mainly diameters between 2 and 5 mm (73.9%, wt), and only  
 211 1.46% of the total particles weight corresponded to grains with diameters below 1.4 mm.  
 212 For OP, 77.8% of the material had diameters between 1.030 and 1.400 mm after the  
 213 grinding process carried out in laboratory.

214 The bulk density is a parameter with great importance in material handling and storage  
215 [36], and it was considerably higher for OS samples ( $721.6 \pm 12.7 \text{ kg/m}^3$ ) than for OP  
216 samples ( $347.9 \pm 13.4 \text{ kg/m}^3$ ). The bulk density value for olive stone is in agreement with  
217 that ( $693.0 \pm 54.9 \text{ kg/m}^3$ ) reported by other authors [10].

218 The structural composition of OS and OP is shown in Table 2. Both biomasses were  
219 mainly composed of cellulose, hemicelluloses and lignin. The percentages of  
220 hemicelluloses and lignin were higher for OS, while cellulose was higher for OP. The rest  
221 corresponded to extractives, which include non-structural sugars, tannins and chlorophyll,  
222 and mineral salts. Elemental analysis results showed that OS and OP were mainly  
223 composed of carbon (49.7% and 44.4%, respectively) and oxygen (43.4% and 48.3%,  
224 respectively). Besides, negligible values of sulphur and low percentages of nitrogen  
225 (0.128% and 0.771%, respectively) were found, which is important from an  
226 environmental point of view because it contributes to lowering SO<sub>x</sub> and NO<sub>x</sub> emissions.

227 With regards to higher heating values (HHV) obtained from tests in the calorimeter,  
228 OS and OP samples reached values of 18.8 and 17.4 MJ/kg, respectively (Table 2). These  
229 data indicate that OS and OP have a great potential as solid biofuels in combustion  
230 processes, being the potential of OS higher due to its higher HHV and bulk density. The  
231 predicted HHV by Demirbas' equation (19.7 MJ/kg for OS and 16.7 MJ/kg for OP)  
232 agreed well with the experimental results.

233 The equilibrium moisture content (EMC) can be used as an indicator of hydrophobicity  
234 of a solid. EMC of olive stone and olive-tree pruning was measured at 283 and 303 K  
235 with relative humidity (RH) ranging from 43.1% to 77.5% (Table 2). For a given material,  
236 the EMC increased with RH and decreased with an increase in the temperature.  
237 Depending on the RH values and temperatures, the EMC ranged from 9.13% to 13.3%  
238 for OS and from 10.1% to 15.5% for OP. Data from Table 2 shows that OP is slightly

239 more hydrophilic than OS, probably due to the higher porosity of the olive wood. The  
240 differences in EMC values also could be related with cell wall compositions, especially  
241 with water-extractives (hydrophilic materials) and lignin (hydrophobic material)  
242 percentages. The EMC values were similar than those reported at 293 K for willow (EMC  
243 = 12.5% for RH = 75.0%) [37], and at 288 K for miscanthus stems (EMC = 15.0% for  
244 RH = 75.6%) [38].

## 245 *TABLE 2*

### 246 *3.2. Mathematical modelling of drying curves*

247 The experimental drying curves of wet olive stone and olive-tree pruning (WOS and  
248 WOP, respectively) for different values of drying air temperature and sample thickness  
249 are shown in Fig. 1. As expected, the moisture content decreased continuously with  
250 drying time for all assays. Furthermore, an increase in temperature and a decrease in  
251 sample thickness resulted in reduced drying time of the two biomasses. The drying times  
252 to reach a moisture content of 10% (wet basis) working with 50 mm sample thickness at  
253 343, 373 and 393 K were 13.4, 6.2 and 2.8 h for WOS, and 22.0, 9.5 and 6.3 h for WOP,  
254 respectively. The shortest drying times were obtained at 393 K using a sample thickness  
255 of 15 mm: 1.0 h for WOS and 1.6 h for WOP.

256 For comparison with experimental data, calculated moisture contents from Page model  
257 (Eq. 7) are also shown in Fig. 1. The values of  $k$  and  $n$  coefficients of model are  
258 summarised in Table 3 along with values obtained for  $r^2$  (0.996–0.999),  $\chi_r^2$  ( $1.82 \times 10^{-5}$   
259  $-9.07 \times 10^{-4}$ ) and RMSE values (0.0037–0.0246), indicating a good fit.

## 261 *FIGURE 1*

## 263 *TABLE 3*

264

265 The previous results show the suitability of the Page model to describe the drying  
266 behaviour of wet olive stone and olive-tree pruning. However, this model does not  
267 reproduce the influence of drying temperature. The experimental data obtained for  
268 thickness sample of 50 mm were used to assess the dependence of  $k$  and  $n$  with drying  
269 temperature. The experimental  $n$  values had not a high variability with temperature (Table  
270 3) and it could be defined by means of a uniform exponent ( $n = 1.44 \pm 0.13$  for WOS, and  
271  $n = 1.81 \pm 0.06$  for WOP) as the average of all  $n$  values. On the other hand, the  $k$   
272 coefficients dependence with temperature can be described by an Arrhenius-type  
273 equation. Thus, the following mathematical formulas were proposed to evaluate the  
274 moisture ratio of the two biomasses at the drying times (h) and air temperatures (K)  
275 assayed:

276

$$\ln M_R = - \exp[8.954 - (4286.613/T)] t^{1.439} \quad (8)$$

277

278 for WOS, and

279

$$\ln M_R = - \exp[10.780 - (5260.929/T)] t^{1.810} \quad (9)$$

280

281 for WOP.

282

283 Figure 2 shows the comparison between experimental and predicted values of  $M_R$   
284 calculated using the new mathematical models (Eqs. 8 and 9), indicating good fits to the  
285 experimental data, especially for wet olive-tree pruning ( $r^2 = 0.992$ ,  $\chi_r^2 = 1.07 \times 10^{-3}$  and  
286 RMSE = 0.0324). Considering the whole range of assayed temperatures (343-393 K), the

287 value of  $r^2$  for olive stone is relatively low ( $r^2 = 0.982$ ), but it markedly improves ( $r^2 =$   
288  $0.994$ ,  $\chi_r^2 = 1.34 \times 10^{-3}$  and  $RMSE = 0.0361$ ) when excluding the data obtained at 373  
289 K (i.e. using the drying temperatures between 343 and 383 K).

290

291

## FIGURE 2

292

### 3.3. Drying rates ( $D_R$ )

294 In order to deepen the knowledge about the drying processes, the experimental drying  
295 rates ( $D_R$ ) for WOS and WOP were calculated by Eq. (10),

296

$$D_R = \frac{d(M_R)}{dt} \approx \frac{(M_R)_{t+\Delta t} - (M_R)_t}{\Delta t} \quad (10)$$

297

298 where  $t$  is the drying time and  $(M_R)_{t+\Delta t}$  and  $(M_R)_t$  stand for the moisture content at times  
299  $(t+\Delta t)$  and  $t$ , respectively.

300 Fig. 3 depicts the relation between  $D_R$  and time, and between  $D_R$  and moisture ratio,  
301 for experiments carried out in the range 343–393 K with 50 mm of sample thickness. The  
302 figure also shows the predicted values of drying rate calculated from the time derivative  
303 of Eq. (7) using the values of  $k$  and  $n$  tabulated in Table 3. It can be stated that the Page  
304 model gives an adequate estimation of drying rates for experiments performed between  
305 343 K and 383 K. However, errors were considerable at 393 K, especially for wet olive-  
306 tree pruning.

307 The analysis of drying rates allowed determining the number of stages and their  
308 characteristics during the drying of biomasses. Warming-up period was observed at the  
309 beginning of all drying processes, and it was longer for WOP than for WOS. For both

310 biomasses, this period increased when increasing drying temperature, which is in  
311 agreement with previous studies [22].

312 The constant drying period (CDP) was not observed for WOS under any drying  
313 condition. However, this stage was found working with WOP for some temperatures and  
314 samples thicknesses such as at 353 K and 50 mm thicknesses (Fig. 4). Experimental data  
315 indicated that, for a fixed temperature, the decrease of sample thickness led, on the one  
316 hand, to an increase of warming-up period and, on the other hand, to the reduction or even  
317 the disappearance of the constant drying period. This behaviour can be clearly observed  
318 in Fig. 4 for WOP thicknesses of 25 and 50 mm. The presence of constant drying periods  
319 for WOP could be explained taking into account that this biomass had smaller particle  
320 sizes than olive stone (Table 2) which could imply higher interface area per volume of  
321 packed bed. The increase in the sample thickness from 25 to 50 mm does not change the  
322 sample surface in contact with the external drying air ( $S_{CDA}$ ) but increases the volume of  
323 packed bed, so the internal moisture transfer could be sufficient to maintain  $S_{CDA}$   
324 saturated for a long time. Air temperature and sample thickness had a significant effect  
325 on the maximum drying rate ( $D_{R,max}$ ) for both WOS and WOP. Thus,  $D_{R,max}$  increased  
326 from  $0.0598 \text{ h}^{-1}$  to  $0.225 \text{ h}^{-1}$  for WOS, and from  $0.057 \text{ h}^{-1}$  to  $0.170 \text{ h}^{-1}$  for WOP when  
327 thickness was set to 50 mm and the process temperature was increased from 343 K to 393  
328 K (Fig. 3). For a sample thickness of 25 mm at the same range of temperatures (343–393  
329 K) the  $D_{R,max}$  values ranged from 0.151 to  $0.390 \text{ h}^{-1}$  and from 0.146 to  $0.323 \text{ h}^{-1}$  for WOS  
330 and WOP, respectively. The highest values of maximum drying rate were achieved at 393  
331 K with a sample thickness of 15 mm:  $0.739 \text{ h}^{-1}$  for WOS and  $0.776 \text{ h}^{-1}$  for WOP. The  
332 calculated value of  $D_{R,max}$  for wet olive stone ( $0.739 \text{ h}^{-1}$ ) is consistent with those ( $0.72$   
333 and  $1.08 \text{ h}^{-1}$ ) obtained by other authors [22] for the drying of olive stone with thickness  
334 of 20 mm at 373 and 423 K, respectively. With regard to OP, since its drying kinetics has

335 been not previously studied, no comparison with other authors' results could be  
336 performed.

337

338 *FIGURE 3*

339

340 *FIGURE 4*

341

#### 342 *3.4. Effective diffusivity*

343 During the falling-rate period, the drying rate begins to fall because the rate of moisture  
344 transfer from the interior of the packed bed towards external surface is lower than the rate  
345 of evaporation from that surface. In this stage, the drying process is controlled by the  
346 molecular transport of moisture which occurs according to a concentration gradient of  
347 water across the packed bed. Therefore, experimental results could be interpreted by using  
348 Fick's second law of diffusion (Eq. (11)),

349

$$\frac{dC}{dt} = D \frac{d^2C}{dx^2} \quad (11)$$

350

351 where  $C$  is the water concentration,  $D$  is the diffusion coefficient and  $x$  is the distance in  
352 the flow direction.

353 An analytical solution to the differential equation has been established for the one-  
354 dimensional mass transport in infinite slab geometry under the following limitations [39]:  
355 isothermal drying conditions, constant effective diffusivity, and negligible shrinkage and  
356 external resistance (Eq. (12)),

357

$$M_R = \frac{8}{\pi^2} \sum_{n=0}^{\infty} \frac{1}{(2n+1)^2} \exp\left(-\frac{(2n+1)^2 \pi^2 D_{\text{eff}} t}{4L^2}\right) \quad (12)$$

358

359 where  $D_{\text{eff}}$  is the effective diffusivity,  $n$  is the number of terms taken into consideration,  
 360  $L$  is the thickness of slab and  $t$  is the drying time.

361 For long drying periods, Eq. (12) can be simplified to the first term of the series,

362

$$M_R = \frac{8}{\pi^2} \exp\left(-\frac{\pi^2 D_{\text{eff}} t}{4L^2}\right) \quad (13)$$

363

364 Eq. (13) can be linearized in the following way,

365

$$\ln M_R = \ln \frac{8}{\pi^2} - \frac{\pi^2 D_{\text{eff}} t}{4L^2} \quad (14)$$

366

367 From the representation of the first member ( $\ln M_R$ ) against time ( $t$ ), the effective water  
 368 diffusivity ( $D_{\text{eff}}$ ) can be determined from the slope ( $-\pi^2 D_{\text{eff}} 4^{-1}L^{-2}$ ) (Fig. 5).

369

370

### *FIGURE 5*

371

372 Values of  $D_{\text{eff}}$  for each experiment along with the coefficients of determination  $r^2$  are  
 373 shown in Table 4. This table illustrates that, under the same experimental conditions,  
 374 effective water diffusivities through WOS (between  $1.87 \times 10^{-8} \text{ m}^2/\text{s}$  and  $16.4 \times 10^{-8} \text{ m}^2/\text{s}$ )  
 375 were lower than though WOP (between  $3.41 \times 10^{-8} \text{ m}^2/\text{s}$  and  $32.5 \times 10^{-8} \text{ m}^2/\text{s}$ ). What is  
 376 more, diffusivities for wet olive-tree pruning were roughly twice than those measured  
 377 with wet olive stone (Fig. 6). It can be seen that the  $D_{\text{eff}}$  values obtained in the present



378 study are in agreement with those obtained for the drying of olive cake ( $6.80 \times 10^{-8} \text{ m}^2/\text{s}$   
379 to  $21.5 \times 10^{-8} \text{ m}^2/\text{s}$ ) at temperatures between 323 and 353 K [20] or coconut ( $5.99 \times 10^{-8}$   
380  $\text{m}^2/\text{s}$  to  $26.6 \times 10^{-8} \text{ m}^2/\text{s}$ ) in the range 333–393 K [40]. Similar  $D_{\text{eff}}$  values of  $1.64 \times 10^{-8}$   
381  $\text{m}^2/\text{s}$  and  $3.13 \times 10^{-8} \text{ m}^2/\text{s}$  for the drying process of olive stone at 373 K and 423 K,  
382 respectively, have been previously reported [22]. By contrast, there is not available data  
383 in literature for WOP. Data from Table 4 also indicates that  $D_{\text{eff}}$  values increased notably  
384 with air temperature. Thus, when WOS with 25 mm thickness were dried, the  $D_{\text{eff}}$  values  
385 continuously increased from  $1.87 \times 10^{-8} \text{ m}^2/\text{s}$  to  $6.34 \cdot 10^{-8} \text{ m}^2/\text{s}$  when air temperature rose  
386 from 343 K to 393 K. The same trend was observed when sample thickness was 50 mm  
387 and also in the experiments with WOP. On the other hand, results showed a general trend  
388 of increasing effective diffusivity with decreasing sample thickness. Thus, for WOP at  
389 393 K and sample thicknesses of 15 mm, 25 mm, 40 mm and 50 mm, the effective water  
390 diffusivities were  $12.0 \times 10^{-8} \text{ m}^2/\text{s}$ ,  $13.0 \times 10^{-8} \text{ m}^2/\text{s}$ ,  $30.2 \times 10^{-8} \text{ m}^2/\text{s}$  and  $32.5 \times 10^{-8}$   
391  $\text{m}^2/\text{s}$ , respectively, while for WOS under the same experimental conditions (393 K  
392 temperature, and 15 mm, 25 mm, 40 mm and 50 mm thickness) the effective water  
393 diffusivities were  $7.22 \times 10^{-8} \text{ m}^2/\text{s}$ ,  $6.34 \times 10^{-8} \text{ m}^2/\text{s}$ ,  $15.4 \times 10^{-8} \text{ m}^2/\text{s}$  and  $16.4 \times 10^{-8}$   
394  $\text{m}^2/\text{s}$ , respectively. This behaviour has been reported by other authors working with olive  
395 stone [22].

396

397

*TABLE 4*

398

399

*FIGURE 6*

400

401 In diffusion of solids, the temperature dependence of the effective diffusion coefficient  
402 ( $D_{\text{eff}}$ ) of water through the solid can be described by an empirical equation (Eq. (15)),  
403 which has the typical form describing an activated process [41],

$$D_{\text{eff}} = \alpha \exp\left(-\frac{\beta}{T}\right) \quad (15)$$

404

405 where  $\alpha$  ( $\text{m}^2/\text{s}$ ) and  $\beta$  (K) are empirical constants and  $T$  is the absolute temperature (K).

406 Therefore, the values of  $\alpha$  and  $\beta$  were calculated by plotting the natural logarithm of  
407  $D_{\text{eff}}$  versus the reciprocal of the absolute temperature, as presented in Fig. 7, where the  
408 coefficients of determination values are also shown.

409

410

#### *FIGURE 7*

411

#### 412 **4. Conclusions**

413 This study confirms that olive stone exhibits better energetic characteristics for  
414 combustion process than olive-tree pruning, because of its higher gross calorific value  
415 (18.8 vs 17.4 MJ/kg) and bulk density (721.6 vs 347.9  $\text{kg}/\text{m}^3$ ), and its lower ash percentage  
416 (0.69 vs 2.7%). By contrast, its current price (80-100 €/t) is more than twice higher than  
417 that of olive-tree pruning (30-40 €/t), which can be a hindrance for industrial applications.

418 With regards to the drying process of wet olive stone, no constant rate period was  
419 observed, and the water loss was mainly accomplished during the falling rate period. By  
420 contrast, wet olive-tree pruning showed constant drying period under some experimental  
421 conditions, which could be related to the smaller particle size obtained for this biomass  
422 after grinding. Temperature increase and sample thickness decrease resulted in a  
423 reduction of the drying time for both biomasses, the highest drying rates being achieved

424 at the maximum temperature (393 K) and sample thickness (15 mm) assayed ( $0.739 \text{ h}^{-1}$   
425 for wet olive stone and  $0.776 \text{ h}^{-1}$  for wet olive-tree pruning). Page Model was successfully  
426 applied for drying kinetics prediction and two new mathematical equations were proposed  
427 to describe the dependence of moisture ratio with drying time and temperature.

428 The values of effective water diffusivity, calculated at the falling rate period, increased  
429 when the drying temperature and the sample thickness rose as well. Diffusion coefficient  
430 values for olive-tree pruning (between  $3.4 \times 10^{-8} \text{ m}^2/\text{s}$  and  $32.5 \times 10^{-8} \text{ m}^2/\text{s}$ ), which have  
431 been for first time determined in this work, were almost twice as high as those for olive  
432 stone (between  $1.87 \times 10^{-8} \text{ m}^2/\text{s}$  and  $16.4 \times 10^{-8} \text{ m}^2/\text{s}$ ). These differences can lead to  
433 modifications in the design of rotary dryers for these biomasses.

434

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441

#### 442 **References**

- 443 [1] C. Europeaia, REPORT FROM THE COMMISSION TO THE COUNCIL AND  
444 THE EUROPEAN PARLIAMENT on sustainability requirements for the use of  
445 solid and gaseous biomass sources in electricity, heating and cooling, 2010.  
446 doi:SEC(2010) 65; SEC(2010) 66.
- 447 [2] M. Cuevas, S. Sánchez, J.F. García, Thermochemical and biochemical  
448 conversion of olive stones, in: C.N. Foster (Ed.), Agric. Wastes Charact. Types

- 449 Manag., Nova Science Publishers, 2015: pp. 61–86.
- 450 [3] C. Galán, H. García-Mozo, L. Vázquez, L. Ruiz, C. Díaz De La Guardia, E.  
451 Domínguez-Vilches, Modeling olive crop yield in Andalusia, Spain, *Agron. J.*  
452 100 (2008) 98–104. doi:10.2134/agronj2006.0345.
- 453 [4] J. Gómez, J. Infante-Amate, M. de Molina, T. Vanwalleghem, E. Taguas, I.  
454 Lorite, Olive cultivation, its impact on soil erosion and its progression into yield  
455 impacts in Southern Spain in the past as a key to a future of increasing climate  
456 uncertainty, *Agriculture*. 4 (2014) 170–198. doi:10.3390/agriculture4020170.
- 457 [5] A. Roig, M.L. Cayuela, M.A. Sánchez-Monedero, An overview on olive mill  
458 wastes and their valorisation methods, *Waste Manag.* 26 (2006) 960–969.  
459 doi:10.1016/j.wasman.2005.07.024.
- 460 [6] M. Cuevas, S. Sánchez, J.F. García, J. Baeza, C. Parra, J. Freer, Enhanced  
461 ethanol production by simultaneous saccharification and fermentation of  
462 pretreated olive stones, *Renew. Energy*. 74 (2015) 839–847.  
463 doi:10.1016/j.renene.2014.09.004.
- 464 [7] T.M. Alslaibi, I. Abustan, M.A. Ahmad, A.A. Foul, Application of response  
465 surface methodology (RSM) for optimization of Cu<sup>2+</sup>, Cd<sup>2+</sup>, Ni<sup>2+</sup>, Pb<sup>2+</sup>, Fe<sup>2+</sup>,  
466 and Zn<sup>2+</sup> removal from aqueous solution using microwaved olive stone activated  
467 carbon, *J. Chem. Technol. Biotechnol.* 88 (2013) 2141–2151.  
468 doi:10.1002/jctb.4073.
- 469 [8] D. Montané, J. Salvadó, C. Torras, X. Farriol, High-temperature dilute-acid  
470 hydrolysis of olive stones for furfural production, *Biomass and Bioenergy*. 22  
471 (2002) 295–304. doi:10.1016/S0961-9534(02)00007-7.
- 472 [9] G. Rodríguez, A. Lama, M. Trujillo, J.L. Espartero, J. Fernández-Bolaños,  
473 Isolation of a powerful antioxidant from *Olea europaea* fruit-mill waste: 3,4-

- 474 Dihydroxyphenylglycol, *LWT - Food Sci. Technol.* 42 (2009) 483–490.  
475 doi:10.1016/j.lwt.2008.08.015.
- 476 [10] J. Mata-Sánchez, J.A. Pérez-Jiménez, M.J. Díaz-Villanueva, A. Serrano, N.  
477 Núñez-Sánchez, F.J. López-Giménez, Development of olive stone quality system  
478 based on biofuel energetic parameters study, *Renew. Energy.* 66 (2014) 251–256.  
479 doi:10.1016/j.renene.2013.12.009.
- 480 [11] J.F. García, S. Sánchez, J. García, Ethanol from biomass: Application to the  
481 olive-pruning debris, in: D.A. Carasillo (Ed.), *Liq. Fuels Types, Prop. Prod.*,  
482 Nova Science Publishers, 2012: pp. 239–254.
- 483 [12] F.J. Lopez, S. Pinzi, J.J. Ruiz, A. Lopez, M.P. Dorado, Economic viability of the  
484 use of olive tree pruning as fuel for heating systems in public institutions in  
485 South Spain, *Fuel.* 89 (2010) 1386–1391. doi:10.1016/j.fuel.2009.11.003.
- 486 [13] T. Gebreegziabher, A.O. Oyedun, C.W. Hui, Optimum biomass drying for  
487 combustion - A modeling approach, *Energy.* 53 (2013) 67–73.  
488 doi:10.1016/j.energy.2013.03.004.
- 489 [14] R. Moreno, G. Antolín, A. Reyes, P. Alvarez, Drying characteristics of forest  
490 biomass particles of *Pinus radiata*, *Biosyst. Eng.* 88 (2004) 105–115.  
491 doi:10.1016/j.biosystemseng.2004.02.005.
- 492 [15] Y. Huangfu, H. Li, X. Chen, C. Xue, C. Chen, G. Liu, Effects of moisture content  
493 in fuel on thermal performance and emission of biomass semi-gasified cookstove,  
494 *Energy Sustain. Dev.* 21 (2014) 60–65. doi:10.1016/j.esd.2014.05.007.
- 495 [16] R. Spinelli, G. Picchi, Industrial harvesting of olive tree pruning residue for  
496 energy biomass, *Bioresour. Technol.* 101 (2010) 730–735.  
497 doi:10.1016/j.biortech.2009.08.039.
- 498 [17] R. Perry, D. Green, J. Maloney, *Perry's chemical engineers' handbook*, 1997.

- 499           doi:10.10360071422943.
- 500   [18]   J. Srikiatden, J.S. Roberts, Moisture transfer in solid food materials: A review of  
501           mechanisms, models, and measurements, *Int. J. Food Prop.* 10 (2007) 739–777.  
502           doi:10.1080/10942910601161672.
- 503   [19]   I. Montero, T. Miranda, J.I. Arranz, C.V. Rojas, Thin layer drying kinetics of by-  
504           products from olive oil processing, *Int. J. Mol. Sci.* 12 (2011) 7885–7897.  
505           doi:10.3390/ijms12117885.
- 506   [20]   S. Meziane, Drying kinetics of olive pomace in a fluidized bed dryer, *Energy*  
507           *Convers. Manag.* 52 (2011) 1644–1649. doi:10.1016/j.enconman.2010.10.027.
- 508   [21]   A. Ruiz Celma, S. Rojas, F. Lopez-Rodríguez, Mathematical modelling of thin-  
509           layer infrared drying of wet olive husk, *Chem. Eng. Process. Process Intensif.* 47  
510           (2008) 1810–1818. doi:10.1016/j.cep.2007.10.003.
- 511   [22]   F.J. Gómez-de la Cruz, P.J. Casanova-Peláez, J.M. Palomar-Carnicero, F. Cruz-  
512           Peragón, Drying kinetics of olive stone: A valuable source of biomass obtained in  
513           the olive oil extraction, *Energy.* 75 (2014) 146–152.  
514           doi:10.1016/j.energy.2014.06.085.
- 515   [23]   F.J. Gómez-De La Cruz, J.M. Palomar-Carnicero, P.J. Casanova-Peláez, F. Cruz-  
516           Peragón, Experimental determination of effective moisture diffusivity during the  
517           drying of clean olive stone: Dependence of temperature, moisture content and  
518           sample thickness, *Fuel Process. Technol.* 137 (2015) 320–326.  
519           doi:10.1016/j.fuproc.2015.03.018.
- 520   [24]   P.S. Lam, S. Sokhansanj, X.T. Bi, C.J. Lim, S.H. Larsson, Drying characteristics  
521           and equilibrium moisture content of steam-treated Douglas fir (*Pseudotsuga*  
522           *menziesii* L.), *Bioresour. Technol.* 116 (2012) 396–402.  
523           doi:10.1016/j.biortech.2012.03.093.

- 524 [25] A. A. Sluiter, R. Ruiz, C. Scarlata, J. Sluiter, D. Templeton, Determination of  
525 Extractives in Biomass: Laboratory Analytical Procedure (LAP); Issue Date  
526 7/17/2005 - 42619.pdf, Tech. Rep. NREL/TP-510-42619. (2008) 1–9.  
527 doi:NREL/TP-510-42621.
- 528 [26] M. Cuevas, J.F. García, S. Sánchez, Enhanced enzymatic hydrolysis of pretreated  
529 almond-tree prunings for sugar production, *Carbohydr. Polym.* 99 (2014) 791–  
530 799. doi:10.1016/j.carbpol.2013.08.089.
- 531 [27] A. Demirbas, Calculation of higher heating values of fatty acids, *Energy Sources,*  
532 *Part A Recover. Util. Environ. Eff.* 38 (2016) 2693–2697.  
533 doi:10.1080/15567036.2015.1115924.
- 534 [28] L. Greenspan, Humidity fixed points of binary saturated aqueous solutions, *J.*  
535 *Res. Natl. Bur. Stand. Sect. A Phys. Chem.* 81A (1977) 89.  
536 doi:10.6028/jres.081A.011.
- 537 [29] L. Hernández, Estudio del secado de hueso de aceituna y poda de olivar para  
538 mejorar su uso como biocombustibles en calderas de biomasa, University of Jaén,  
539 2015.
- 540 [30] W.K. Lewis, The Rate of Drying of Solid Materials, *Ind. Eng. Chem.* 13 (1921)  
541 427–432. doi:10.1021/ie50137a021.
- 542 [31] G.E. Page, Factors influencing the maximum rates of air drying shelled corn in  
543 thin layers, Purdue University, 1949.
- 544 [32] D.G. Overhults, G.M. White, H.E. Hamilton, I.J. Ross, Drying Soybeans With  
545 Heated Air, *Trans. ASAE.* 16 (1973) 112–113. doi:10.13031/2013.37459.
- 546 [33] S.M. Henderson, S. Pabis, Grain drying theory II: Temperature effects on drying  
547 coefficients, *J. Agric. Eng. Res.* (1961) 169–174.
- 548 [34] I.T. Togrul, D. Pehlivan, Mathematical modelling of solar drying of apricots in

- 549 thin layers, *J. Food Eng.* 55 (2002) 209–216. doi:10.1016/S0260-8774(02)00065-  
550 1.
- 551 [35] A. Midilli, H. Kucuk, Z. Yapar, A new model for single layer drying, *Dry.*  
552 *Technol.* 20 (2002) 1503–1513. doi:10.1081/drt-120005864.
- 553 [36] Z. Liu, B. Mi, Z. Jiang, B. Fei, Z. Cai, X. Liu, Improved bulk density of bamboo  
554 pellets as biomass for energy production, *Renew. Energy.* 86 (2016) 1–7.  
555 doi:10.1016/j.renene.2015.08.011.
- 556 [37] B. Krupinska, I. Strømmen, Z. Pakowski, T.M. Eikevik, Modeling of sorption  
557 isotherms of various kinds of wood at different temperature conditions, *Dry.*  
558 *Technol.* 25 (2007) 1459–1466. doi:10.1080/07373930701537062.
- 559 [38] J.J. Fitzpatrick, C. O’Sullivan, H. Boylan, O. Cribben, D. Costello, K. Cronin,  
560 Moisture sorption isotherm study of Sitka spruce, larch, willow and miscanthus  
561 chips and stems, *Biosyst. Eng.* 115 (2013) 474–481.  
562 doi:10.1016/j.biosystemseng.2013.05.004.
- 563 [39] J. Crank, *The mathematics of diffusion*, Clarendon Press, Oxford, England, 1975.
- 564 [40] T. Madhiyanon, A. Phila, S. Soponronnarit, Models of fluidized bed drying for  
565 thin-layer chopped coconut, *Appl. Therm. Eng.* 29 (2009) 2849–2854.  
566 doi:10.1016/j.applthermaleng.2009.02.003.
- 567 [41] C.O. Bennett, J.E. Myers, *Molecular diffusion and diffusivity*, in: *Momentum,*  
568 *Heat, and Mass Transfer*, McGraw-Hill (2<sup>a</sup> Ed.), New York, 1974.



**Table 1**

Mathematical models of drying curves.

Model name	Equation	References
Lewis	$M_R = \exp(-kt)$	[30]
Page	$M_R = \exp(-kt^n)$	[31]
Modified Page	$M_R = \exp(-(kt)^n)$	[32]
Henderson and Pabis	$M_R = a \exp(-kt)$	[33]
Logarithmic	$M_R = a \exp(-kt) + c$	[34]
Midilli et al.	$M_R = a \exp(-kt^n) + bt$	[35]

**Table 2**

Physicochemical properties of dry olive stone (OS) and olive-tree pruning (OP)

Property		OS	OP
Particle size (mm), %	5.00 > x > 2.00	73.905	0.086
	2.00 > x > 1.80	16.330	0.125
	1.80 > x > 1.40	8.313	7.448
	1.40 > x > 1.03	1.105	77.798
	1.03 > x > 0.730	0.245	12.673
	0.730 > x > 0.510	0.074	1.432
	0.510 > x > 0.360	0.012	0.280
	0.360 > x > 0.210	0.016	0.157
	0.210 > x	0.000	0.000
Bulk density, kg/m <sup>3</sup>		721.6±12.7	347.9±13.4
Composition	Extractives, %	6.01±0.06	19.5±0.1
	Cellulose, %	26.1±0.3	29.4±0.5
	Hemicelluloses, %	26.3±0.5	20.3±0.2
	AIL, %	31.3±0.8	20.2±0.7
	ASL, %	2.12±0.22	0.41±0.07
	Ash, %	0.691±0.031	2.67±0.18
	Elemental analysis	C, %	49.7±0.1
H, %		6.83±0.09	6.62±0.14
N, %		0.128±0.002	0.771±0.042
S, %		n.d.	n.d.
O, %		43.4±0.0	48.3±0.2
High heating value	Calorimetric pump, MJ/kg	18.8±0.2	17.4±0.1
	Demirbas equation, MJ/kg	19.7±0.1	16.7±0.2
EMC (283 K), % d.b.	RH = 43.1%	10.6	11.4
	RH = 57.4%	11.5	11.9
	RH = 77.5%	13.3	15.5
EMC (303 K), % d.b.	RH = 43.2%	9.13	10.1
	RH = 51.4%	10.3	11.7
	RH = 73.1%	11.5	12.3

AIL: Acid Insoluble Lignin; ASL: Acid Soluble Lignin. n.d.: not detected; EMC: Equilibrium Moisture Content; RH: Relative Humidity.

**Table 3**

Statistics for the Page model. Influence of temperature and sample thickness.

Biomass	<i>ST</i> (mm)	<i>T</i> (K)	Coefficient <i>k</i> (h <sup>-1</sup> )	Coefficient <i>n</i>	<i>r</i> <sup>2</sup>	$\chi_r^2$	RMSE	
WOS	15	393	1.01	1.84	1.00	$7.52 \times 10^{-5}$	0.00733	
		25	343	0.0474	1.67	0.998	$2.75 \times 10^{-4}$	0.0156
	25	353	0.207	1.16	0.999	$1.55 \times 10^{-4}$	0.0110	
		363	0.234	1.24	0.999	$1.58 \times 10^{-4}$	0.0112	
		373	0.217	1.50	1.00	$5.94 \times 10^{-5}$	0.00680	
		383	0.277	1.51	0.999	$2.67 \times 10^{-4}$	0.0138	
		393	0.503	1.40	1.00	$1.82 \times 10^{-5}$	0.00370	
		40	393	0.349	1.40	1.00	$2.22 \times 10^{-5}$	0.00408
		50	343	0.0295	1.35	1.00	$4.99 \times 10^{-5}$	0.00673
	50	353	0.0538	1.29	1.00	$4.68 \times 10^{-5}$	0.00647	
		363	0.0491	1.44	1.00	$3.36 \times 10^{-5}$	0.00536	
		373	0.0698	1.45	0.999	$7.83 \times 10^{-5}$	0.00819	
		383	0.0693	1.67	0.999	$1.89 \times 10^{-4}$	0.0124	
		393	0.223	1.43	1.00	$2.98 \times 10^{-5}$	0.00502	
		WOP	15	393	0.781	2.36	0.997	$9.07 \times 10^{-4}$
25				343	0.0153	2.19	0.999	$1.57 \times 10^{-4}$
25	353		0.0376	2.01	0.998	$5.04 \times 10^{-4}$	0.0207	
	363		0.0811	1.95	0.999	$1.87 \times 10^{-4}$	0.0121	
	373		0.122	1.86	0.997	$5.56 \times 10^{-4}$	0.0199	
	383		0.112	2.19	0.999	$2.51 \times 10^{-4}$	0.0134	
	393		0.257	1.93	0.999	$2.68 \times 10^{-4}$	0.0134	
	40		393	0.137	1.73	0.997	$6.26 \times 10^{-4}$	0.0217
	50		343	0.00953	1.77	1.00	$4.91 \times 10^{-5}$	0.00648
50	353		0.0163	1.76	0.998	$2.59 \times 10^{-4}$	0.0152	
	363		0.0303	1.76	1.00	$1.82 \times 10^{-5}$	0.00398	
	373		0.0368	1.82	0.998	$2.75 \times 10^{-4}$	0.0154	
	383		0.0461	1.91	0.999	$3.78 \times 10^{-4}$	0.0176	
	393		0.0744	1.86	0.996	$5.18 \times 10^{-4}$	0.0212	

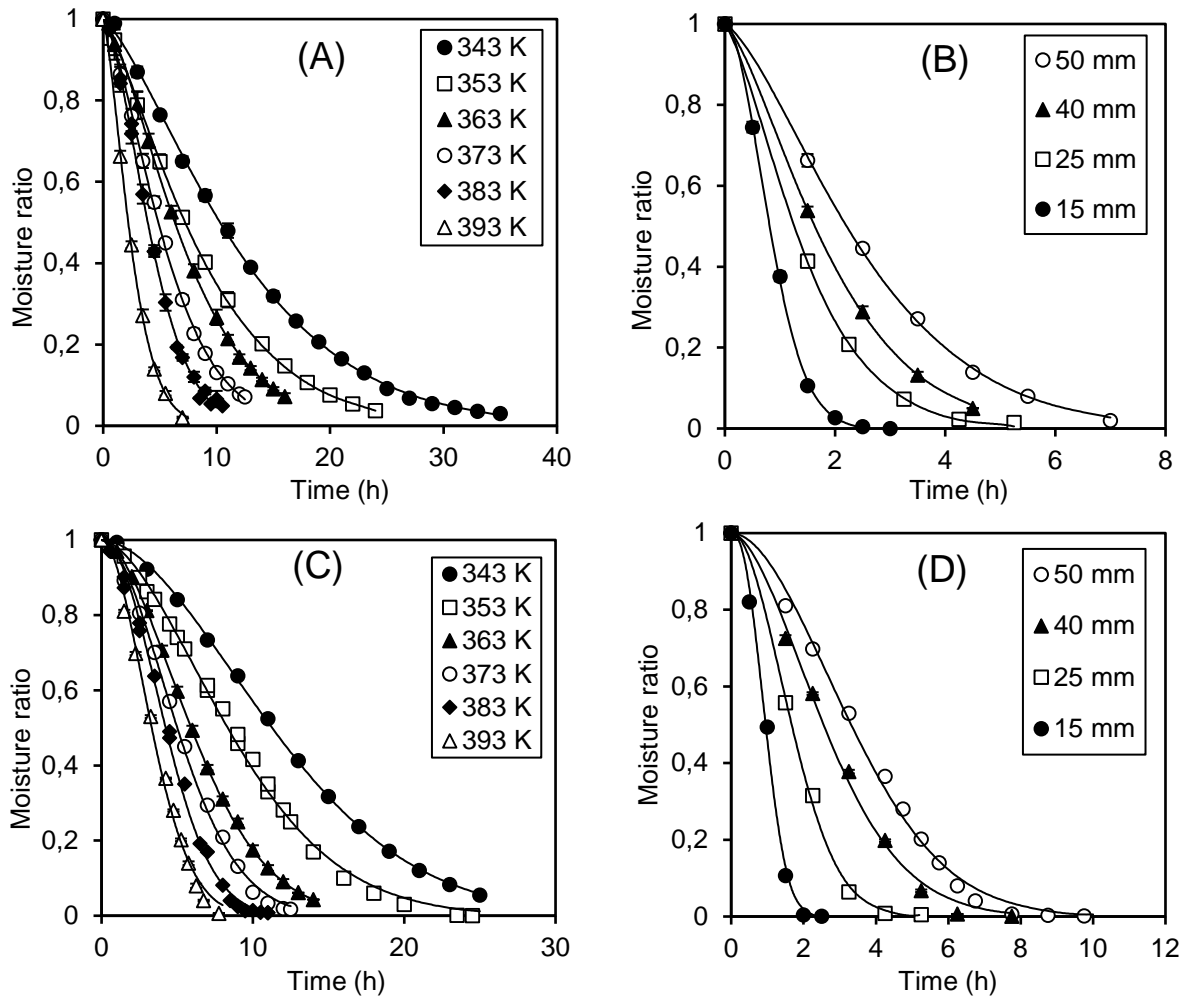
*ST*: Sample thickness; WOS: Wet olive stone; WOP: Wet olive-tree pruning.  $\chi_r^2$ : Reduced chi-square; RMSE: Root mean square error.

**Table 4**

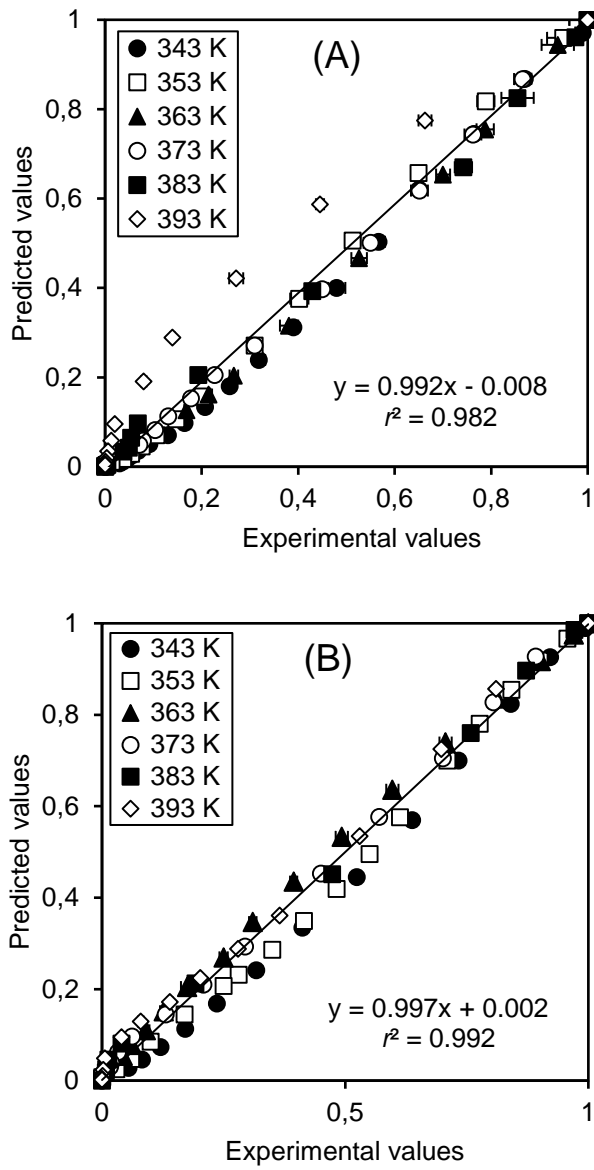
Effective moisture diffusivity ( $D_{\text{eff}}$ ) for each experimental condition.

$ST$ (mm)	$T$ (K)	Wet olive stone		Wet olive-tree pruning	
		$D_{\text{eff}} \times 10^8$ (m <sup>2</sup> /s)	$r^2$	$D_{\text{eff}} \times 10^8$ (m <sup>2</sup> /s)	$r^2$
15	393	7.22	0.996	12.0	0.978
25	343	1.87	0.999	3.41	0.998
	353	2.24	0.993	4.51	0.991
	363	3.14	0.992	6.09	0.995
	373	3.96	0.992	7.60	0.995
	383	5.18	0.994	10.4	0.999
	393	6.34	0.994	13.0	0.994
40	393	15.4	0.991	30.2	0.984
50	343	3.46	0.996	4.95	0.996
	353	4.59	0.998	7.59	0.996
	363	6.07	0.999	9.96	0.999
	373	7.66	0.999	15.7	0.991
	383	12.1	0.992	22.3	0.980
	393	16.4	0.997	32.5	0.991

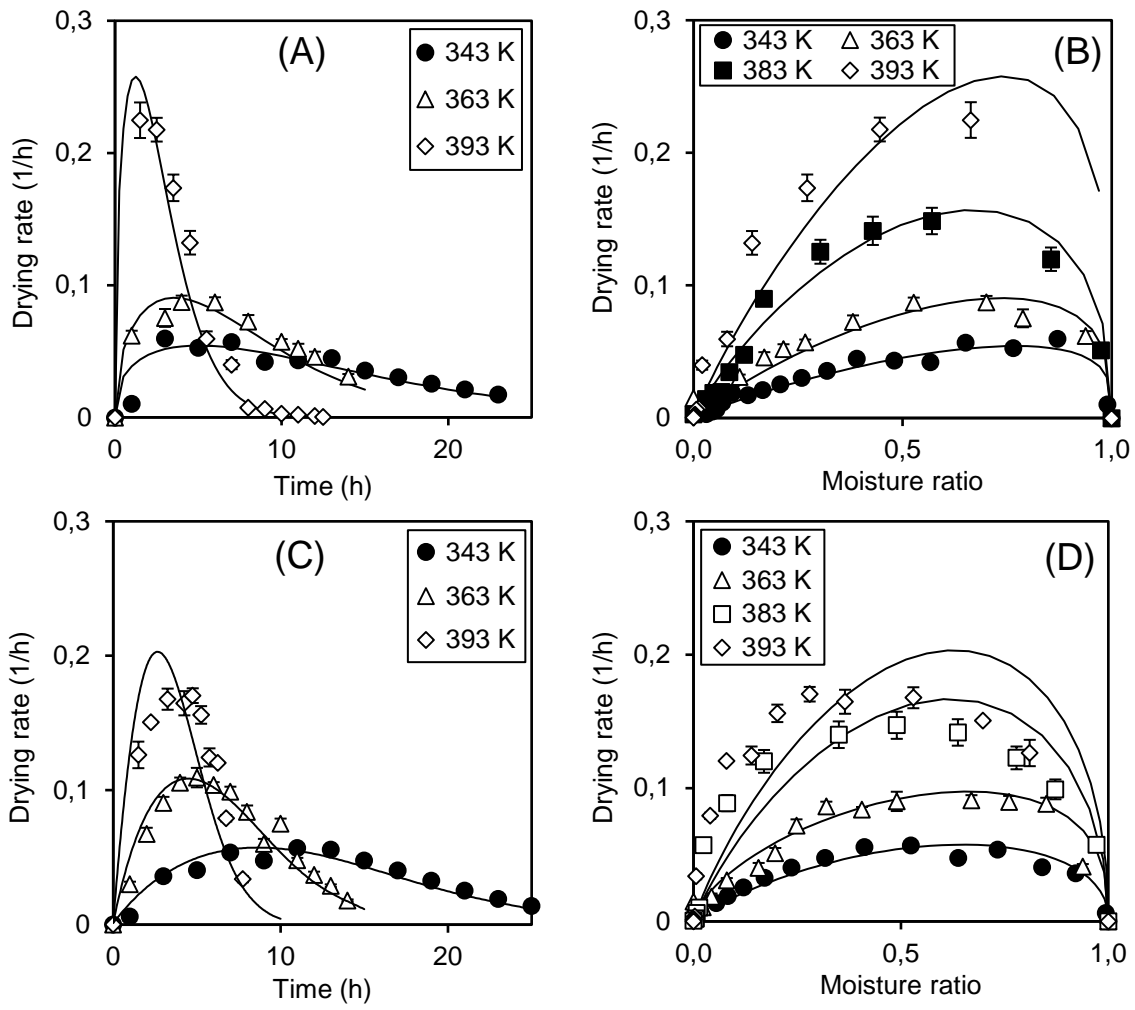
$ST$ : Sample thickness.



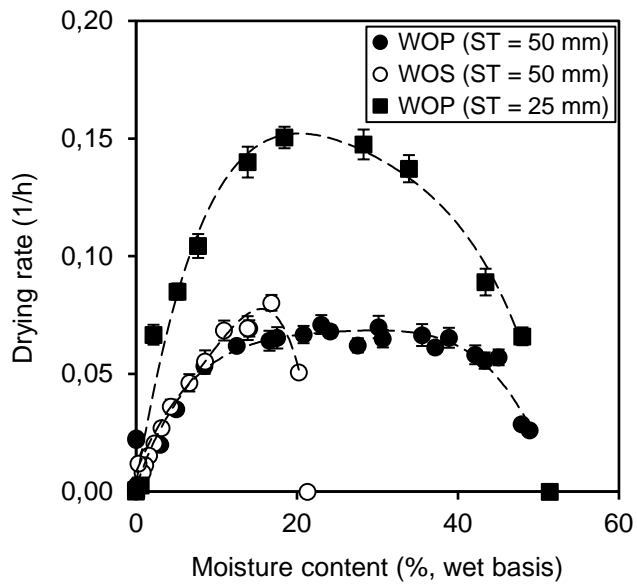
**Fig. 1.** Drying curves of WOS (A, B) and WOP (C, D) at different temperatures (A, C, sample thickness: 50 mm) and sample thicknesses (B, D, temperature: 393 K). Points: experimental data. Lines: predicted data by Page model.



**Fig. 2.** Experimental and predicted moisture ratio values according to Eqs. (8) and (9) for WOS (A) and WOP (B), respectively. Sample thickness: 50 mm.

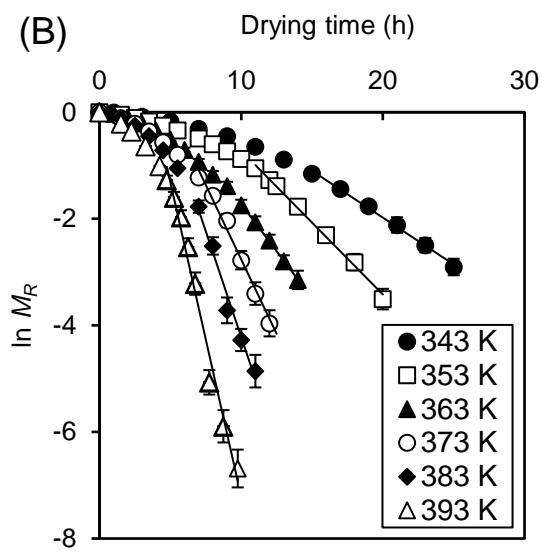
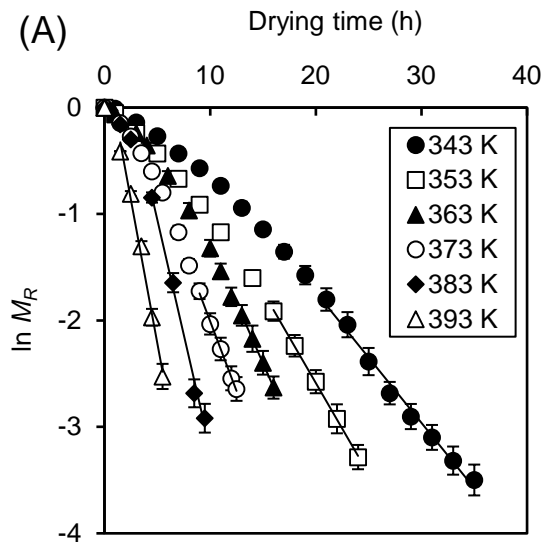


**Fig. 3.** Drying rate of OS (A, B) and OP (C, D) vs drying time and moisture ratio at different temperatures (sample thickness: 50 mm). Lines: Page model.



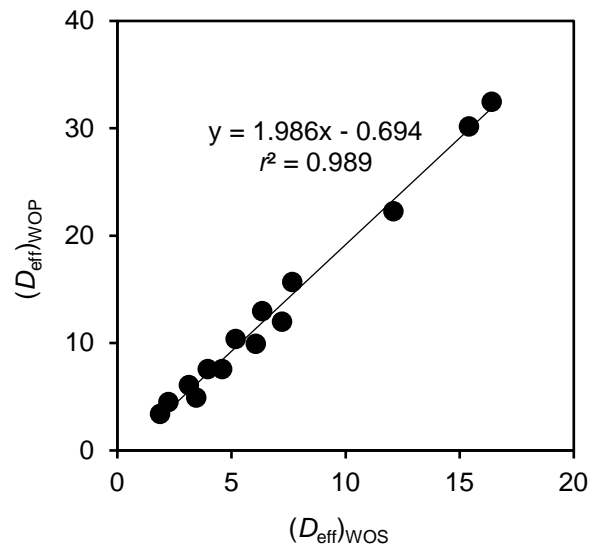
**Fig. 4.** Profile of drying rate versus moisture content at 353 K for WOS and WOP.



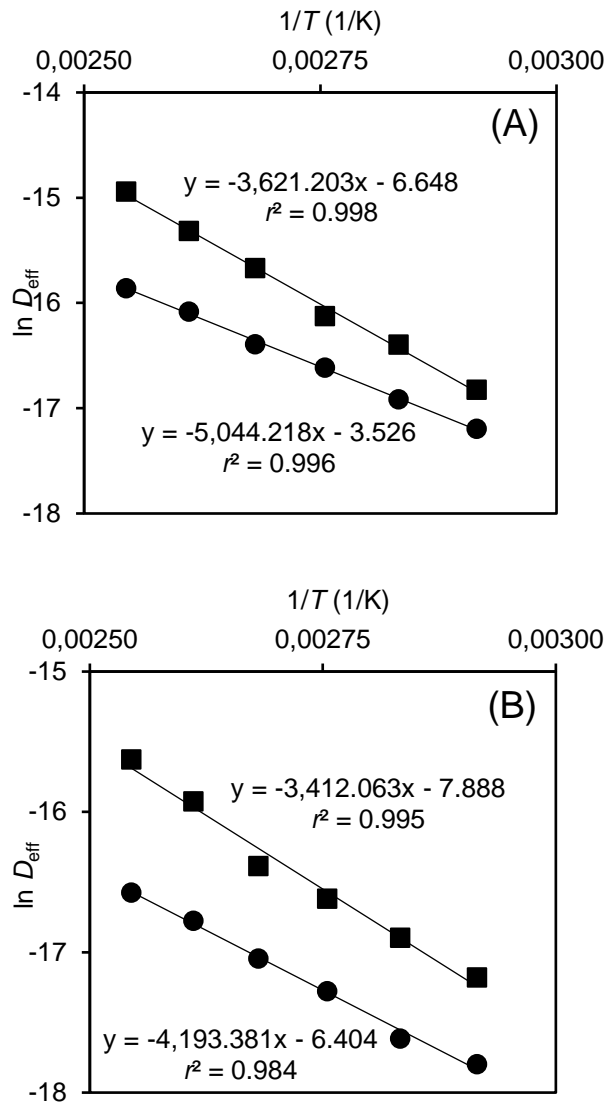


**Fig. 5.** Plot of  $\ln M_R$  vs drying time, and fit of data for the falling rate period (lines). (A)

WOS, (B) WOP. Sample thickness: 50 mm.



**Fig. 6.** Comparison between effective water diffusivities obtained with WOP and WOS.



**Fig 7.** Relationship between effective diffusivity and temperature for the drying process of WOS (A) and WOP (B). Sample thickness: 25 mm (solid circles) and 50 mm (solid squares).

**Table S1**

Fit values of the mathematical models for 50 mm particle size and 70 °C temperature

Biomass	Model name	Model constants	$r^2$	$\chi_r^2$	RMSE
WOS	Lewis	$k = 0.0768$	0.992	$2.29 \times 10^{-3}$	0.0467
	Page	$k = 0.0295, n = 1.35$	1.00	$4.99 \times 10^{-5}$	0.00673
	Modified Page	$k = 0.0740, n = 1.35$	1.00	$4.99 \times 10^{-5}$	0.00673
	Henderson and Pabis	$k = 0.0830, a = 1.08$	0.984	$1.47 \times 10^{-3}$	0.0366
	Logarithmic	$k = , a = , c =$	0.990	$1.55 \times 10^{-3}$	0.0366
	Midilli et al.	$k = , n = , a = , b =$	1.00	$9.72 \times 10^{-5}$	0.00892
WOP	Lewis	$k = 0.0713$	0.963	$9.07 \times 10^{-3}$	0.0918
	Page	$k = 0.00953, n = 1.77$	1.00	$4.91 \times 10^{-5}$	0.00648
	Modified Page	$k = 0.0722, n = 1.77$	1.00	$4.91 \times 10^{-5}$	0.00648
	Henderson and Pabis	$k = 0.0806, a = 1.12$	0.953	$6.79 \times 10^{-3}$	0.0763
	Logarithmic	$k = , a = , c =$	0.953	$7.41 \times 10^{-3}$	0.0763
	Midilli et al.	$k = 0.00905, n = 1.79, a = 0.995, b = 0.00$	1.00	$5.30 \times 10^{-5}$	0.00615

WOS: Wet olive stone; WOP: Wet olive-tree pruning;  $\chi_r^2$ : Reduced chi-square; RMSE: Root mean square error.

**Table S2**

Fit values of the mathematical models for 50 mm particle size and 100 °C temperature

Biomass	Model name	$r^2$	$\chi_r^2$	RMSE
WOS	Lewis	0.981	$3.92 \times 10^{-3}$	0.0603
	Page	1.00	$7.83 \times 10^{-5}$	0.00819
	Modified Page	1.00	$7.83 \times 10^{-5}$	0.00819
	Henderson and Pabis	0.977	$3.01 \times 10^{-3}$	0.0508
	Logarithmic	0.977	$3.28 \times 10^{-3}$	0.0508
	Midilli et al.	1.00	$8.51 \times 10^{-5}$	0.00779
WOP	Lewis	0.957	$1.04 \times 10^{-2}$	0.0982
	Page	0.998	$2.75 \times 10^{-4}$	0.0154
	Modified Page	0.998	$2.75 \times 10^{-4}$	0.0154
	Henderson and Pabis	0.947	$8.41 \times 10^{-3}$	0.0849
	Logarithmic	0.947	$9.17 \times 10^{-3}$	0.0849
	Midilli et al.	0.998	$8.74 \times 10^{-5}$	0.00790

WOS: Wet olive stone; WOP: Wet olive-tree pruning;  $\chi_r^2$ : Reduced chi-square; RMSE: Root mean square error.

**Table S3**

Fit values of the mathematical models for 50 mm particle size and 120 °C temperature

Biomass	Model name	$r^2$	$\chi_r^2$	RMSE
WOS	Lewis	0.988	$1.80 \times 10^{-3}$	0.0407
	Page	1.00	$2.98 \times 10^{-5}$	0.00502
	Modified Page	1.00	$2.98 \times 10^{-5}$	0.00502
	Henderson and Pabis	0.987	$1.76 \times 10^{-3}$	0.0385
	Logarithmic	0.987	$1.93 \times 10^{-3}$	0.0385
	Midilli et al.	1.00	$3.60 \times 10^{-5}$	0.00499
WOP	Lewis	0.948	$9.89 \times 10^{-3}$	0.0956
	Page	0.996	$6.12 \times 10^{-4}$	0.0228
	Modified Page	0.996	$6.12 \times 10^{-4}$	0.0228
	Henderson and Pabis	0.937	$9.01 \times 10^{-3}$	0.0873
	Logarithmic	0.937	$9.91 \times 10^{-3}$	0.0873
	Midilli et al.	0.996	$6.62 \times 10^{-4}$	0.0214

WOS: Wet olive stone; WOP: Wet olive-tree pruning;  $\chi_r^2$ : Reduced chi-square; RMSE: Root mean square error.