

1 **Impact of thermal treatments on the bioaccessibility of phytoene and**
2 **phytofluene in relation to changes in the microstructure and size of**
3 **orange juice particles**

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22 **Declarations of interest:**

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HIGHLIGHTS

- Phytoene, phytofluene (PTF) and ζ -carotene behave similarly to thermal treatments.
- Phytoene and PTF have higher bioaccessibility than some carotenes and xanthophylls.
- TEM images show great disruptions of cell structures with the thermal treatments.
- Pasteurized juices have higher carotenoid bioaccessibility than frozen/thawed juices.

26 **ABSTRACT**

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The interest in the colourless carotenoids phytoene and phytofluene is expanding. In this study their bioaccessibility from thermally treated Pinalate orange juices, which contains high concentrations of these carotenoids, was evaluated in relation to microstructural and particle size changes. Other carotenoids were also considered for comparison. Fresh, pasteurized, and ultrafrozen juices thawed at room temperature (UF-RT), in microwave oven (UF-MW), and in the fridge (UF-FG) were investigated. Colourless carotenoids suffered less and higher degradation as a result of ultrafreezing and pasteurization, respectively, than xanthophylls. In the fresh juice the carotenoid with the highest bioaccessibility was phytoene (10%), followed by zeaxanthin (9%) and phytofluene (8%). Total carotenoid bioaccessibility followed the order: Pasteurized > UF-MW > UF-RT > UF-FG > Fresh. Thermal treatments decreased the particle sizes and ruptured the cell structures and hence increased the bioaccessibility. The best source of bioaccessible colourless carotenoids was UF-MW (0.93 mg/250 mL).

41 **KEYWORDS**

42 *In vitro* digestion; particle size distribution; pasteurization; transmission electron
43 microscopy; ζ -carotene.

44 **ABBREVIATIONS**

45 Conjugated double bonds (cdb)
46 Fresh juice (FRESH)
47 High performance liquid chromatography (HPLC)
48 Methanol (MeOH)
49 Methyl *tert*-butyl ether (MTBE)
50 One-way analysis of variance (ANOVA)

- 51 Particle size distribution (PSD)
- 52 Pasteurized juice (PAST)
- 53 Phytoene (PT)
- 54 Phytofluene (PTF)
- 55 Potassium hydroxide (KOH)
- 56 Strokes per minute (spm)
- 57 Transmission electron microscopy (TEM)
- 58 Ultrafrozen juice with thawing at room temperature (UF-RT)
- 59 Ultrafrozen juice with thawing in the microwave oven (UF-MW)
- 60 Ultrafrozen juice with thawing in the fridge (UF-FG)

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62 **1. Introduction**

63 Carotenoids are a wide family of isoprenoid compounds with beneficial health
64 properties (Kulczyński, Gramza-Michałowska, Kobus-Cisowska, & Kmiecik, 2017) and
65 only a few of the more than 700 natural carotenoids have been studied in depth (Britton,
66 Liaaen-Jensen, & Pfander, 2009). Phytoene (PT) and phytofluene (PTF) have been
67 largely ignored for a long time as compared to others but are currently being a focus of
68 major interest for the scientific community. In fact, some recent reviews have been
69 published focusing in the importance of these two carotenoids in the diet as well as in
70 their potential interest in the context of health promotion and cosmetics (Engelmann,
71 Clinton, & Erdman, 2011; Meléndez-Martínez, Mapelli-Brahm, & Stinco, 2018;
72 Meléndez-Martínez, Mapelli-Brahm, Benítez-González, & Stinco, 2015). PT and PTF
73 are present in a wide variety of fruits and vegetables and their juices, and hence are
74 among the predominant carotenoids in the diet (Biehler et al., 2012; Mapelli-Brahm,
75 Corte-Real, Meléndez-Martínez, & Bohn, 2017). It is well known that tomato and
76 tomato-based food products have a high concentration of both compounds (Khachik et

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77 al., 2002). However, another source rich in these colourless carotenoids has recently
78 been characterized, specifically the orange Pinalate (Lado et al., 2015; Rodrigo, Marcos,
79 Alférez, Mallent, & Zacarías, 2003). This orange is a spontaneous mutant derived from
80 the ordinary Navelate orange with a partial blockage at the ζ -carotene desaturation
81 which seems to be the cause of the accumulation of these lineal carotenes. The flavedo
82 of the Pinalate orange has one of the highest concentrations of PT in fruits described so
83 far (Lado et al., 2015) and also contains very high concentrations of PTF and ζ -carotene
84 (Rodrigo et al., 2003). However, it is important to consider that the beneficial actions of
85 carotenoids and other species do not only depend on how much we ingest, but also on
86 other factors, like for instance their bioavailability. Before their absorption, carotenoids
87 have to be incorporated into mixed micelles to become absorbable for the intestinal
88 enterocytes. It is widely accepted that the bioaccessibility of a given lipophilic
89 compound is the amount of such compound that is released from the food matrix and
90 incorporated into mixed micelles with respect to the initial amount present in the matrix.
91 *In vitro* methods have been widely used to determine the bioaccessibility of carotenoids
92 and to estimate their bioavailability (Estévez-Santiago, Olmedilla-Alonso, &
93 Fernández-Jalao, 2016; Ornelas-Paz, Failla, Yahia, & Gardea-Bejar, 2008; Rodríguez-
94 Roque, Rojas-Graü, Elez-Martínez, & Martín-Belloso, 2014). The bioaccessibility of
95 PT and PTF has not been as widely studied as that of other dietary carotenoids but there
96 are several articles that demonstrate their high bioaccessibility compared to that of other
97 carotenes and even to some xanthophylls (Jeffery, Turner, & King, 2012; Mapelli-
98 Brahm et al., 2017; Rodrigo, Cilla, Barberá, & Zacarías, 2015; Rodrigues,
99 Chitchumroonchokchai, Mariutti, Mercadante, & Failla, 2017). Thermal treatments are
100 widely used in the food industry for preservation (Ibarz, Pagán, & Garza, 1999) but
101 these processes can affect the bioaccessibility of carotenoids and lead to the degradation

102 of some of them (Aschoff et al., 2015). Thus, thermal treatments can produce changes in
103 food colour and be detrimental for the nutritional value (Palmero, Lemmens, Hendrickx,
104 & Van Loey, 2014). Within this context, the main goals of this study were two. On the
105 one hand, to understand how different thermal treatments of Pinalate orange juice affect
106 the stability and the bioaccessibility of the colourless carotenoids phytoene and
107 phytofluene and compare the results with those of other carotenoids present in the
108 orange juice. On the other hand, as the thermal treatments can disrupt the food matrix
109 and hence, enhance the release of the carotenoids and increase their bioaccessibility
110 (Hof van het, West, Weststrate, & Hautvast, 2000), the effects of the changes in particle
111 size distribution (PSD) and plastids integrity (assessed by transmission electron
112 microscopy (TEM)) were also evaluated.

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114 **2. Materials and methods**

115 *2.1. Standards and reagents*

116 HPLC solvents, i.e. methanol (MeOH) and methyl *tert*-butyl ether (MTBE), were
117 of HPLC grade and were acquired from Panreac (Barcelona, Spain). The necessary
118 reagents to obtain the images with the electronic transmission microscope were
119 purchased from Ted Pella, Inc. (Redding, USA). Porcine bile extract, pepsin (porcine,
120 367 units/mg solid, measured as TCA- soluble products using hemoglobin as substrate)
121 and pancreatin (porcine, 8x USP specifications of amylase, lipase and protease) were
122 acquired from Sigma-Aldrich (Bornem, Belgium). All reagents were of analytical grade
123 or higher.

124 *2.2. Samples*

125 The mutant Pinalate orange was chosen for this study due to its large accumulation
126 in colourless carotenoids. It has been found that the flavedo and pulp of this orange are

127 rich in PT, PTF and ζ -carotene (Lado et al., 2015; Rodrigo et al., 2003). However, to the
128 best of our knowledge, the juice has not been characterized. Mature Pinalate orange
129 fruits (*Citrus sinensis* L. Osbeck), were harvested at full mature stage (11.9 °Brix) in
130 January 2016, from adult trees grown at The Citrus Germplasm Bank at the *Instituto*
131 *Valenciano de Investigaciones Agrarias* (IVIA, Moncada, Valencia, Spain). All
132 analyses were made with the same primary juice, which was obtained by manually
133 squeezing representative orange replicates. Five aliquots (fresh samples) were stored in
134 the fridge under nitrogen until analysis (one day). Pasteurization was carried out by
135 immersion of 50 mL plastic tubes containing 20 mL of juice in a water bath at 120 °C.
136 After 30 seconds at 90 °C, the tubes were immediately cooled in an ice bath until the
137 samples reached 1.8 °C. Pasteurized juices were stored under nitrogen atmosphere in the
138 fridge until their analysis a few hours later. Ultrafreezing was carried out by direct
139 immersion of 50 mL plastic tubes containing 20 mL of juice in liquid nitrogen for two
140 minutes. The ultrafrozen samples were stored at - 80 °C until their analysis. These were
141 performed no more than three days later, after their thawing. The ultrafreezing and
142 thawing treatments evaluated were the same considered in a previous study (Stinco,
143 Fernández-Vázquez, Heredia, Meléndez-Martínez, & Vicario, 2013). On the day of the
144 analysis, five replicates of the ultrafrozen juices were thawed at room temperature
145 (avoiding light), another five were thawed in a fridge (4 °C) and the last five were
146 thawed in a microwave oven at 800 watts during 20 s, these latter conditions not leading
147 to increases of temperature. Thus, five samples were analysed, fresh juice (FRESH),
148 pasteurized juice (PAST) and ultrafrozen juices with thawing at room temperature (UF-
149 RT), in the microwave oven (UF-MW) and in the fridge (UF-FG). FRESH was taken as
150 reference sample in order to study the effect of the different thermal treatments on the

151 colour and particle size distribution of the juice and on the concentration and
152 bioaccessibility of the carotenoids.

153 *2.3. Transmission electron microscopy*

154 Fresh juice, processed juices and pulp were analysed by TEM. Regarding the juices,
155 1.5 mL of juice was centrifuged at 18000 g for 5 min at 4 °C in 2 mL vials (microfuge
156 22R, Beckman Coulter, Krefeld, Germany) and the aqueous phase was discarded. In the
157 case of the pulp, few juice vesicles were manually cut by using a blade and were
158 introduced in a 2 mL vial. Immediately, 1 mL of modified Karnovsky fixative (0.5%
159 glutaraldehyde, 2.5% formaldehyde) was added to each vial. The cells were fixing
160 during few hours in darkness and then centrifuged at 18000 g for 15 min at 4 °C. The
161 upper phase was discarded and the pellet was rinsed three times with 0.1 M sodium
162 cacodylate buffer (pH 7.4). Then, the post-fixation was performed with 1% osmium
163 tetroxide in the buffer for 1 h at 25 °C. The sample was washed for 20 min at 4 °C with
164 distilled water and then stained with 2% aqueous solution of uranyl acetate for 2 h at 25
165 °C. Dehydration was made in an acetone series (50, 70, 90, 100%). Subsequently, the
166 sample was embedded in Spurr resin, following a gradual procedure with different
167 ratios of acetone/spurr. Polymerization was carried out overnight at 70 °C. Ultrathin
168 sections of 70-100 nm were obtained by cutting semithin sections with an
169 ultramicrotome (Leica UC7, Wetzlar, Germany). These ultrathin sections were examined
170 with a Zeiss Libra 120 transmission electron microscope (Oberkochen, Germany)
171 equipped with a SSCCD digital camera.

172 *2.4. Epifluorescence microscopy*

173 The pulp and juice of the Pinalate orange were observed under a BX61 motorized
174 epifluorescence microscope (Olympus, Tokio, Japan) in order to detect phytofluene.
175 The pulp and juice of the parental Navelate were used as a reference. Fluorescence

176 microscopy allows for identifying specific fluorescent substances, named
177 fluorochromes, by observing their characteristic emission properties when illuminated
178 with radiation of appropriate wavelength. PTF, upon being excited with near-UV light
179 (300 - 400 nm), emits light at approximately 510 nm and therefore can be detected by
180 fluorescence microscopy (reviewed in Meléndez-Martínez et al., 2015). The
181 arrangement of optical components of this microscope allows the illumination from
182 above of the sample and therefore offers a high signal-to-noise ratio. The microscope
183 was equipped with a mercury vapour lamp as light source (X-Cite 120PC) and with
184 10×/0.4 and 20×/0.75 UPLANAPO objective lenses. The images were acquired with an
185 Olympus camera with a CellSens Dimension Software. An U-MNU2 filter (exciting
186 filter BP 360 - 370 nm, emission filter BA420, and dichromatic mirror DM 400 nm)
187 were applied. To obtain the images one drop of juice or one small piece of a cut vesicle
188 juice was spotted on a microscope slide glass (26 × 76 mm) and a microscope cover
189 glass (18 × 18 mm) was used to cover the sample. The use of a mounting medium was
190 not necessary.

191 *2.5. Particle Size Distribution*

192 Particles scatter light in all directions with an intensity pattern, which is dependent
193 on particle size. On the basis of this principle, the particle size distribution (PSD) of the
194 juices was determined by laser diffraction using a Mastersizer 3000E equipped with a
195 He-Ne laser (Malvern Instruments, Inc., Worcs, U.K.). The Mie model of light
196 scattering recommended by ISO13320-1 November 1999 was used. This model
197 assumes that particles are uniform spherical particles, which are illuminated by a plane
198 wave of infinite extent with a known wavelength (633 nm). The optical properties of the
199 sample and of the surrounding medium should be known in this model; the refractive
200 index and absorption rate of the cloud particles for orange juice is 1.73 and 0.1,

201 respectively, while distilled water has a refractive index of 1.33 (Corredig, Kerr, &
 202 Wicker, 2001). To measure the PSD, an aliquot of 2.5 mL of juice was introduced into
 203 the sampling unit of the Mastersizer. The sample was pumped through the optical cell
 204 using a stirrer rotating at 2000 rpm and diluted with approximately 100 mL of distilled
 205 water to achieve an obscuration level of between 10-20%. The analysis report provided
 206 the frequency distribution graph, the surface area (A_S), the standard percentiles (D_v (10),
 207 D_v (50) and D_v (90)), and the volume- and area- based mean diameters ($D_{[4,3]}$ and $D_{[3,2]}$,
 208 respectively). D_v (10), D_v (50) and D_v (90) are the sizes in microns at which 90, 50 and
 209 10% of the sample is smaller and 10, 50 and 90% is larger, respectively. $D_{[4,3]}$ and $D_{[3,2]}$
 210 are defined by the following equation:

$$D_{[4,3]} = \frac{\sum_i n_i d_i^4}{\sum_i n_i d_i^3} \quad (\text{Equation 1})$$

$$D_{[3,2]} = \frac{\sum_i n_i d_i^3}{\sum_i n_i d_i^2} \quad (\text{Equation 2})$$

where n_i is the number of particles of diameter d_i .

2.6. Colour measurement

The reflectance spectra were measured using a CAS 140 B spectroradiometer
 (Instrument Systems, Instrument Systems, Munich, Germany) in the visible region
 (380–770 nm) in 2-nm steps. The D65 standard illuminant, corresponding to the natural
 daylight, and 10 ° Observer were assumed (CIE 1978). The spectroradiometer was
 equipped with a Top 100 telescope optical probe, a Tamron zoom mod. SP 23 A
 (Tamron USA, Inc., Commack, NY, USA), and as external light source a white light
 150 W metal halide lamp Phillips MHN-TD Pro (12,900 lumen, 4200 K colour
 temperature). A 10 mm path length cuvette filled with distilled water against a white

223 background was used as a blank. The colour parameters of the uniform colour space
224 CIELAB were obtained directly from the reflectance spectra from the apparatus. The
225 CIELAB colour space is a coordinate cartometer system defined by three colorimetric
226 coordinates, i.e. L^* (lightness), a^* (ranging from green to red) and b^* (ranging from blue
227 to yellow). The hue angle (h_{ab} , the qualitative expression of colour) and the chroma
228 (C_{ab}^* , the quantitative expression of colourfulness) are defined from these coordinates.
229 The total colour differences (ΔE_{ab}^*) between the fresh juice sample (as a reference) and
230 the rest of the samples were calculated using the following formula:

$$\Delta E_{ab}^* = ((L_1^* - L_2^*)^2 + (a_1^* - a_2^*)^2 + (b_1^* - b_2^*)^2)^{1/2} \quad (\text{Equation 3})$$

234 In the preceding formula, subscript 1 refers to the fresh juice while the subscript 2
235 refers to the sample of interest in each case.

236 2.7. Total Soluble Solids (°Brix) and pH

237 The maturity stage of the samples was measured as °Brix with an ABBE
238 refractometer (WYA-S, Biotech) at 20 °C. The pH was measured with a pH-meter (GLP
239 21 pH-meter, Crison, Barcelona, Spain).

240 2.8. *In vitro* digestion

241 The *in vitro* gastro-intestinal digestion model was based on the protocol described
242 by Stinco et al. (2012) with some modifications. Three grams of juice and 5.4 mL of
243 aqueous saline solution (140 mM NaCl/5 mM KCl) were mixed in a 50-mL plastic tube.
244 The pH was adjusted to 2 by adding 0.1 M HCl. All samples were complemented with
245 distilled water up to a final volume of 12.4 mL. Then, 0.6 mL of pepsin solution (40
246 mg/mL in 0.1 M HCl, prepared the day of usage) was incorporated into the mixture. To
247 simulate the gastric digestion, the samples were incubated for 1 h at 37 °C in a Max

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248 Q5000 shaker (Labware, Madrid, Spain) with reciprocating motion at 100 strokes per
249 minute (spm). Afterward, samples were maintained in ice few minutes to inactivate the
250 enzymes. The pH was then increased to 6.9 by adding 1 M NaOH and the solution was
251 made up to a final volume of 15 mL with distilled water. Subsequently, 0.75 mL of a
252 bile extract and pancreatin solution (2 mg/mL pancreatin and 12 mg/mL bile in 0.1 M
253 NaHCO₃, prepared the day of usage) was added. Samples were incubated for 2 h at 37
254 °C in the shaker at 100 spm to mimicking the intestinal digestion. Once completed the
255 digestion, samples were centrifuged at 5000 g for 20 min at 4 °C. The supernatant was
256 filtered through a 0.22 µm nylon membrane (Millipore Iberica S.A., Madrid, Spain) into
257 a new 50 mL plastic tube. The mixed micelle fractions obtained, i.e. the bioaccessible
258 fraction, were flushed with nitrogen and stored at -80 °C until carotenoids extraction
259 (one day).

260 2.9. Carotenoids analysis

261 Six hundred microliters of diethyl ether were added to 0.5 g of juice in a 2-mL
262 plastic tube. The mixture was vortexed and ultrasonicated (Ultrasons, JP Selecta,
263 Barcelona, Spain) for 2 min. To promote phase separation, the sample was centrifuged
264 (Microfuge 22R, Beckman Coulter, Madrid, Spain) at 18000 g for 5 minutes at 4 °C.
265 Then, the upper phase containing the carotenoids was transferred to another 2 mL
266 plastic tube. The matrix was re-extracted twice with 600 µL of diethyl ether and the
267 organic phases were combined together. The pooled ether phase was concentrated to
268 dryness in a rotary evaporator at 30 °C (Eppendorf Concentrator Plus, Hamburg,
269 Germany). The extract was saponified by adding 600 µL dichloromethane and 600 µL
270 methanolic potassium hydroxide (KOH) (30%, w/v) and the mixture was maintained 30
271 min with mechanical shake (Gyromini Nutating 3-D Mixer, Labnet, Madrid, Spain) in
272 the dark under nitrogen. The saponification time and KOH concentration were chosen

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273 after carrying out some preliminary tests. Saponification tests included KOH at 30 or
274 40% in MeOH and saponification times from 30 min to overnight. After saponification,
275 the organic phase was washed with distilled water until neutral pH of the waste water.
276 Finally, the organic phase was concentrated to dryness by rotary evaporation at 30 °C.
277 The extracts were kept at -80 °C under a nitrogen atmosphere until the HPLC analysis.

278 The extraction of carotenoids from digesta was carried out similarly with the
279 following minor changes. Ten millilitres of diethyl ether and ten millilitres of NaCl
280 were added to the entire mixed micelle fraction. The mixture was vortexed for 1 min
281 and centrifuged for 5 min at 3280 g. The upper layer was transferred to a 15 mL plastic
282 tube. The extraction was repeated twice more by adding 5 mL of ethyl ether at each
283 step. Saponification of the dry extract was carried out in the same manner but with 2 mL
284 of KOH (30% MeOH) and 2 mL of dichloromethane.

285 *2.10. HPLC-DAD analysis*

286 The extracts were analysed by reverse-phase HPLC (Agilent 1260 system,
287 Waldbronn, Germany) with UV/VIS diode array detector. A C₃₀ YMC column (3 µm,
288 150 cm × 4.6 mm) and a C₃₀ YMC pre-column (2.7 µm, 50 mm × 4.6 mm)
289 (Wilmington, NC, USA) were used. The chromatographic method was similar to that
290 published by Stinco et al. (2012) with minor modifications. Thus, the mobile phase
291 consisting of MeOH (A), MTBE (B) and Milli-Q quality water (C). The linear gradient
292 elution was: 0 min, 90% A + 5% B + 5% C; 5 min, 95% A + 5% B; 10 min, 89% A +
293 11% B; 16 min, 75% A + 25% B; 20 min, 40% A + 60% B; 22.5 min, 15% A + 85% B;
294 25 min, 90% A + 5% B + 5% C. The run time was 27 min during which the flow rate
295 was 1 mL/min and the column was kept at 30 °C. The detector was set at 286 nm for the
296 detection of phytoene, at 350 nm for phytofluene, at 410 nm for ζ-carotene, and at 450
297 nm for the rest of the carotenoids. Prior to the injection the samples were dissolved in

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298 ethyl acetate. Juice extracts were dissolved in 100 μL and 7 μL were injected while the
299 digesta extracts were dissolved in 30 μL and 20 μL were injected. The identification of
300 the carotenoids was made by comparison of their spectroscopic and chromatographic
301 features with those of the collection of standards of the Food Colour and Quality Lab..
302 External calibration was used for quantification.

303 *2.11. Calculations and statistical analysis*

304 All analyses were carried out in triplicate except the *in vitro* digestions and the
305 colour analyses that were performed in quintuplicate. The results shown in the text and
306 the tables were expressed as mean values \pm standard deviations. In the graphics, the
307 error bars represent \pm standard deviation. Data processing was performed using the IBM
308 SPSS Statistics 20[®] software (SPSS Inc., 2012). Shapiro-Wilk and Levene tests ($P \leq$
309 0.05) were used to verify the normality and homoscedasticity, respectively. One-way
310 analysis of variance (ANOVA) followed by the Bonferroni *post hoc* test was performed
311 in order to detect significant differences among means. When the results did not allow
312 carrying out parametric studies, the comparisons were carried out following the *post hoc*
313 non-parametric T2-Tamhane test. Differences were considered statistically significant at
314 P value ≤ 0.05 (2-sided).

315 Relative bioaccessibility was calculated as the percentage of carotenoid content
316 that remained in the micellar aqueous fraction after centrifugation and filtration in
317 relation to the respective initial content in the original non-digested matrix. To estimate
318 the bioaccessible content the relative bioaccessibility and the initial concentration in the
319 matrix were taken into account. In this sense, it was regarded as the amount of
320 carotenoid that can be potentially absorbed per ration of food.

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322 **3. Results and discussion**

323 *3.1. Transmission electron microscopy (TEM)*

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2 324 TEM images of the pulp showed that mature Pinalate orange did not contain
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5 325 chloroplasts and was rich in chromoplasts containing round vesicles, achlorophyllous
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7 326 membranes, plastoglobuli, and some starch grains. Thus, in Figure 2-A two intact
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9 327 chromoplasts can be easily seen. Most of the chromoplasts were rich in round vesicles
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11 328 which were larger than the plastoglobuli, in agreement with the TEM results of Lado et
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13 329 al. (2015). These vesicles seem to contribute to the large accumulation of linear
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15 330 carotenes in this mutant (Lado et al., 2015). In the fresh juice, vesicles, plastoglobuli
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17 331 and starch grains fairly intact were observed, although in some cases outside the
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19 332 chromoplasts which were broken during juicing. In general, it can be observed a gradual
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21 333 degradation of the cell and plastid structures in the following order: PULP, FRESH, UF-
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23 334 FG, UF-RT, UF-MW and PAST. Thus, while fresh juice still contained a large number
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25 335 of intact plastid internal structures, in the PAST the great disruption of cell material was
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27 336 quite evident and were almost devoid of intact suborganellar structures. In agreement
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29 337 with our results, Gupta et al. (2011) found intact cellular components in a freshly
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31 338 extracted tomato juice while cellular components were indistinguishable in a
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33 339 pasteurized juice (100 °C, 10 min, 0.1 MPa). In the ultrafrozen juices, the ice crystals
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35 340 could be the cause of the rupture of the cell structures (Leong & Oey, 2012).

341 *3.2. Epifluorescence microscopy*

342 The pulp and juice were observed under an epifluorescence microscope in order to
343 evaluate the distribution of PTF in these matrices. Both, pulp and juice of the Pinalate
344 orange showed fluorescence, being more easily detected and concentrated in the pulp
345 (Figures 1-A and 1-B). In the juice the fluorescence was more scattered, indicating a
346 greater dispersion of phytofluene, which could have an impact in its release during
347 digestion. These images were compared to those of the parental Navelate orange which

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348 only contain a small amount of PTF (Stinco, Escudero-Gilete, Heredia, Vicario, &
349 Meléndez-Martínez, 2016) in comparison with that of Pinalate (Table 1). Fluorescence
350 was hardly observed in the pulp of Navelate orange (Figures 1-C and 1-D). From these
351 results, it can be inferred that epifluorescence microscopy could be an efficient, rapid
352 and reliable technique to detect PTF in orange juices.

353 *3.3. Particle Size Distribution (PSD)*

354 Regarding the PSD, all parameters analysed followed the decreasing order:
355 FRESH > UF-FG > UF-RT > UF-MW > PAST, being the differences between FRESH
356 and PAST in all cases statistically significant (Table 2). The volume- and area- based
357 mean diameters ($D_{[4,3]}$ and $D_{[3,2]}$, respectively) were 14.4% and 12.5% respectively
358 lower in PAST than in FRESH (Table 2). The cloud size distributions were bimodal in
359 all juices and, as can also be seen in Figure 3, FRESH was the juice with the highest
360 percentage of particles with a bigger size and with the lowest percentage of particles
361 with a lower size followed by UF-FG, UF-RT, UF-MW and PAST. That is, PAST was
362 the juice with the lowest particle size. PSD results of FRESH were similar to those
363 found by Stinco et al. (2012). Other researches have shown conflicting results on the
364 change in the particle size due to the pasteurization of fresh orange juices (Leizerson &
365 Shimoni, 2005; Stinco et al., 2012). To the best of our knowledge, the effect of freezing
366 and thawing on the particle size of orange juices is not well described in the literature.
367 However, it should be pointed out that, in many articles, juices which are regarded as
368 fresh juices were in fact frozen at some point. This is an important aspect to take into
369 account, as it has been proven that this process can have important effects in the PSD
370 (Table 2) and therefore in the colour and bioaccessibility of the carotenoids (sections 3.4
371 and 3.7).

372 *3.4. Colour measurement*

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373 Colour is an attribute of paramount importance for the consumers and in Pinalate
374 juice it could be strongly influenced by its peculiar profile of carotenoids. The main
375 coloured carotenoid in Pinalate juice, i.e. ζ -carotene (Table 3), has a b^* and C^*_{ab} values
376 much lower than those of the main carotenoids in common orange juices, i.e.
377 violaxanthin and β -cryptoxanthin (Meléndez-Martínez, Britton, Vicario, & Heredia,
378 2007; Stinco et al., 2016). This may be one of the main reasons why fresh Pinalate juice
379 has a significantly lower b^* and C^*_{ab} values, i.e. 30 and 32 respectively, compared to
380 those average values found in common orange juices, i.e. approximately between 59
381 and 62 and between 50 and 61, respectively (Fernández-Vázquez, Stinco, Meléndez-
382 Martínez, Heredia, & Vicario, 2011; Stinco et al., 2016). As a result, Pinalate juice has a
383 light-yellowish colour, which correlates well with that of ζ -carotene (Meléndez-
384 Martínez et al., 2007), while common orange juices are more orange. On the other hand,
385 the high content of colourless carotenoids PT and PTF and the low content of
386 xanthophylls could contribute to the pale colour of the Pinalate juice. The colour
387 attributes a^* and b^* were positive in all samples.

388 The colour of FRESH was affected by the thermal treatments. Considering the
389 significant decrease in the lightness value of PAST in comparison to that of FRESH
390 (Table 3) it can be concluded that pasteurization caused a darkening of the juice.
391 Regarding C^*_{ab} and h_{ab} parameters, no statistically significant differences were found
392 with any of the treatments. These results are in agreement with those of a study with
393 ultrafrozen juice of Valencia late oranges thawed under conditions analogous to those of
394 this study (Stinco, Fernández-Vázquez, Heredia, Meléndez-Martínez, & Vicario, 2013).
395 However, contradictory results have been found in the literature regarding the changes
396 in the colour parameters of thermally-treated orange juices (Ortiz et al., 2017; Stinco et
397 al., 2016; Stinco et al., 2013; Wibowo et al., 2015). These differences may be due to

398 some extent to differences in the carotenoid levels of the oranges (Wibowo et al., 2015).
399 However, it should be noted that changes in the concentration of ascorbic acid can also
400 influence the colour of orange juices (Meléndez-Martínez, Vicario, & Heredia, 2009;
401 Roig, Bello, Rivera, & Kennedy, 1999) and that orange juices browning could be due to
402 the oxidation of phenols to quinones (Eissa, Fadelb, Ibrahim, Hassan, & Elrashid,
403 2006). ΔE^* is a good parameter for understanding how observers perceive colour
404 differences and it is considered that, ΔE^* over 2.8 CIELAB units can be perceived by
405 even inexperienced observers (Melgosa, Pérez, Yebra, Huertas, & Hita, 2001).
406 Differences above this threshold were found in UF-FG, UF-MW, and PAST (Table 3).
407 The differences found in the ultrafrozen juices are very similar to those found in other
408 studies (Cinquanta, Albanese, Cuccurullo, & Di Matteo, 2010; Stinco et al., 2013).
409 Among the thermal treatments, pasteurization generated the greatest colour difference
410 (ΔE^*) with respect to FRESH while the smallest difference occurred in the UF-RT
411 (Table 3). This could be related to the fact that UF-RT was the juice with the lowest
412 degradation of carotenoids and PAST the one that suffered the greatest degradation
413 (Supplementary Table 1).

414 *3.5. Carotenoid profile in the fresh orange juice*

415 Violaxanthin, antheraxanthin (cyclic epoxy-carotenoids), zeaxanthin (cyclic
416 dihydroxycarotenoid), β -carotene (cyclic carotene) and three linear carotenes, i.e. ζ -
417 carotene, PT and PTF, were detected. That is, three xanthophylls and four carotenes. β -
418 Carotene is one of most studied carotenoids due to its high provitamin A activity and
419 zeaxanthin has also been extensively studied for its role in the human macula (Johnson
420 et al., 2000). On the other hand, although PT, PTF and ζ -carotene are found in human
421 serum they are very poorly studied carotenoids. The most abundant carotenoid was ζ -
422 carotene followed by PTF and PT (Table 1). Other researchers have found a higher

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423 concentration of PT with respect to that of ζ -carotene in the pulp of Pinalate (Lado et
424 al., 2015). This difference could be due to the possible difference in the degree of
425 release between the carotenoids when juicing, and this could, in turn, be due to the
426 difference in the subcellular structures where each carotenoid accumulates (Rodrigo,
427 Cilla, Reyes, & Zacarías, 2015). Regarding the isomers, 4 of ζ -carotene and PTF, 2 of
428 PT, and only the 9-*cis* isomer of violaxanthin and antheraxanthin were detected. In
429 addition, four compounds were detected but could not be identified. As their retention
430 times were similar to those of the common carotenoid esters, different saponification
431 conditions were tested in order to be certain whether they were free carotenoids or not.
432 The areas of their respective peaks were not altered with the different saponification
433 times and KOH concentrations so it was concluded that they were free carotenoids.
434 Taking into account that they could not be identified and that they were not one of the
435 six carotenoids that represent more than 95% of the total carotenoids in human blood
436 (Maiani et al., 2009), they were not included in the present study. However, their
437 chromatographic and spectroscopic characteristics are summarized in Supplementary
438 Table 2. The carotenoid profile found in the Pinalate juice is very different from that of
439 other coloured common citrus fruits. In the latter, xanthophylls represent up to 80% of
440 the total carotenoid content and linear carotenes account for no more than 20% (Rodrigo
441 et al., 2003) while, in the Pinalate juice, ζ -carotene, PT and PTF together represented
442 nearly 98% of the total carotenoids and only 1.2% of the carotenoids were xanthophylls
443 (Table 1). Thus, while in juices of common varieties of orange the concentration of PT
444 and PTF is approximately 1 and 0.35 mg/L respectively (Stinco et al., 2016), in Pinalate
445 juice the concentration was 14 and 19 mg/L, respectively. Although Cara Cara is also
446 considered an orange rich in these carotenoids, with reported values of 12 and 3 mg/L of

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447 PT and PTF respectively (Stinco et al., 2016), the concentration is lower than that of
448 Pinalate.

449 *3.6. Carotenoid degradation during thermal treatments*

450 The degradation of carotenoids is an important aspect for the food industry since
451 it can affect the colour and the nutritional value of foods. Important percentages of
452 degradation of carotenoids were observed across the different treatments
453 (Supplementary Table 1). PT and PTF were the carotenoids which suffered the highest
454 degradation during pasteurization, with approximately 67% of degradation. On the other
455 hand, in all ultrafrozen juices, PT and PTF were more stable than xanthophylls
456 (Supplementary Table 1). These results seem to indicate that the colourless carotenoids
457 are less prone to degradation due to the combination of oxygen, enzymes and acid, but
458 less heat stable than the xanthophylls. This difference in the behaviour of PT and PTF
459 compared to the xanthophylls may be due to some extent to structural and physico-
460 chemical differences, for instance their absence of terminal rings, differences in
461 polarity, or the smaller number of conjugated double bonds, which is known to have a
462 remarkable impact in the electron affinity and ionization energy of carotenoids
463 (Martínez, Stinco, & Meléndez-Martínez, 2014). Likewise, they could be due, at least in
464 part, to differences in their accumulation within the cell, like for instance the
465 substructures they may deposit or in the aggregates that might be formed. However, it
466 must be taken into account that orange juice is a complex matrix and that, to be able to
467 obtain more specific conclusions about the degradation of PT and PTF as compared to
468 other orange juice carotenoids, it seems more reasonable to carry out oxidation studies
469 with carotenoid standards under identical conditions. Violaxanthin was the carotenoid
470 which suffered the highest losses with the three types of ultrafreezing and thawing, with
471 an average degradation of 76.5% among the different thawing conditions. This is

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472 consistent with the fact that 5,6-epoxycarotenoids suffer a re-arrangement to their
473 respective 5,8-furanoxide in acidic media (Meléndez-Martínez, Vicario, & Heredia,
474 2007).

475 Regarding total carotenoid concentration, FRESH was the juice with the highest
476 content, i.e. 7.7 mg/100 g, while PAST was the juice with the lowest concentration, i.e.
477 2.6 mg/100g (Table 1), ca. 3-fold lower compared to FRESH. Significant differences
478 were observed in total carotenoid content between fresh and ultrafrozen juices, although
479 previous work did not find differences (Stinco et al., 2013). This differential behaviour
480 may be related to the differences in carotenoid composition between standard sweet
481 orange juices and Pinalate juice. The percentages of carotenoid degradation in the juices
482 followed the order: UF-RT (32%), UF-FG (41%), UF-MW (42%) and PAST (67%)
483 (Supplementary Table 1). Interesting, this increasing order agreed well with that of the
484 degree of microstructural changes (section 3.2). In this sense, it seems reasonable to
485 hypothesize that the greater the degradation of the cell structures, the more exposed the
486 carotenoids are to the acid environment, oxygen, and enzymes, and so, the greater
487 degradation they suffer. The greater degradation of carotenoids and plastids found in
488 PAST could be due to the combination of the aforementioned factors and the heat, as it
489 is well known that the oxidative degradation of carotenoids is stimulated by heat
490 (reviewed in Rodríguez-Amaya & Kimura, 2004; Rodríguez-Amaya, 1999). Among the
491 ultrafrozen juices, UF-FG suffered the highest loss of carotenoids, a result that is in
492 good agreement with the fact that a slow defrosting causes greater carotenoid losses
493 than a rapid thawing (reviewed in Rodríguez-Amaya & Kimura, 2004).

494 *3.7. Carotenoid bioaccessibility*

495 In all the juices, PT had a bioaccessibility higher than that of PTF and the
496 bioaccessibility of both was higher than that of ζ -carotene. This order among the linear

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497 carotenes might be due to the difference in the number of conjugated double bonds
498 (cdb). In this regard, it could be argued that the higher number of cdb of ζ -carotene (7
499 cdb) in comparison with those of PTF (5 cdb) and PT (3 cdb) makes this carotene more
500 rigid and prone to aggregation of their molecules, which could lead to a decrease in the
501 bioaccessibility, similarly to what were found when comparing PT, PTF and lycopene
502 (Mapelli-Brahm et al., 2017; Meléndez-Martínez, Paulino, Stinco, Mapelli-Brahm, &
503 Wang, 2014; Rodrigo, Cilla, Barberá, et al., 2015). In all the samples, the
504 bioaccessibility of the colourless carotenoids was even higher than that of the
505 xanthophylls antheraxanthin and violaxanthin. These epoxy-carotenoids are not found at
506 detectable levels in human fluids or tissues (Khachik, 2006), while PT and PTF are
507 common circulating carotenoids and have been found in several organs (reviewed in
508 Meléndez-Martínez et al., 2015). The only carotenoid that showed greater
509 bioaccessibility than the colourless carotenoids was zeaxanthin, although in all cases the
510 difference with PT was not statistically significant. The high relative bioaccessibility of
511 PT and PTF compared to other carotenoids is consistent with previous investigations.
512 The colourless carotenoids showed bioaccessibilities higher than those of β -carotene, α -
513 carotene and/or lycopene in carrot, papaya, different types of salads, apricot juice,
514 grapefruit, melon, watermelon, tomato, and tomato juice (Jeffery et al., 2012; Mapelli-
515 Brahm et al., 2017; Rodrigues et al., 2017), and it was found to be even higher than that
516 of other xanthophylls such as β -cryptoxanthin, lutein and violaxanthin in certain
517 matrices (Jeffery et al., 2012; Mapelli-Brahm et al., 2017).

518 Total carotenoid bioaccessibility followed the order: PAST (26%) > UF-MW
519 (18%) > UF-RT (14%) > UF-FG (12%) > FRESH (8%). Thus, the pasteurization
520 increased such parameter over 3-fold and the freezing followed by thawing with
521 microwave over 2-fold. The same order was followed by all the carotenoids with the

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522 exception of violaxanthin and antheraxanthin (Table 4). Thus, PAST was the juice with
523 the highest bioaccessibility of PT and PTF, i.e. 32 and 27%, respectively, while the
524 lowest was FRESH with 10 and 8% (Table 4), that is over 3-fold less for both carotenes.
525 The differences in the bioaccessibility of each carotenoid and the total bioaccessibility
526 between PAST and FRESH were statistically significant ($P < 0.05$). These results
527 highlight the importance of the food matrix effect on the bioaccessibility. Since
528 carotenoids must release from the food matrix in order to be incorporated into the mixed
529 micelles, it seems reasonable to expect that the more the matrix is disrupted, the higher
530 the bioaccessibility will be. In order to test this hypothesis, the particle size and the cell
531 structures degradation of the samples were studied, and the results of both analyses
532 confirmed this assumption. Thus, TEM images (Figure 2) showed the same increasing
533 order in the degradation of the plastids among the samples (section 3.2) than the order
534 found in the bioaccessibility. PSD results also agree with those of TEM and
535 bioaccessibility, indicating that FRESH was the juice with the highest percentage of
536 particles with a big size and the lowest percentage of particles with a low size, followed
537 by UF-FG, UF-RT, UF-MW and PAST (section 3.3). Taken together, it could be
538 concluded that the higher bioaccessibility found in PAST is due to the fact that the
539 effect of heat during pasteurization causes greater matrix degradation than the
540 ultrafreezing followed by thawing. On the other hand, it is considered that the higher the
541 concentration of a given carotenoid in a matrix, the lower bioaccessibility, among other
542 reasons because high concentrations could increase the possibility of molecular
543 aggregation (reviewed in Borel, 2003; Mapelli-Brahm et al., 2017). Consistent with the
544 above, it was found that, probably due in part to the degradation, the carotenoid
545 concentration in PAST was significantly lower than that of FRESH (Table 1). Other
546 authors have found increases in carotenoids bioaccessibility in orange juices with the

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547 pasteurization (Aschoff et al., 2015). On the other hand, Stinco et al. (2013), also found
548 higher carotenoid bioaccessibility when orange juices were subjected to ultrafreezing
549 following to thawing in the microwave oven but this increase was not found when the
550 juices were defrosting at room temperature or in the fridge.

551 In order to study whether the thermal treatments affect to the same extent the
552 bioaccessibility of the different carotenoids, the increase in the bioaccessibility for each
553 carotenoid with the thermal treatments, taking as the reference the respective
554 bioaccessibility in the FRESH, was calculated (Supplementary Table 3). The
555 percentages of increase in the bioaccessibility among the carotenes were very similar
556 while in the xanthophylls these values were less homogeneous. This may be due in part
557 to the greater difference with respect to the number and type of functional groups, i.e.
558 epoxides and hydroxyls, and with this the greater difference in the polarity, found
559 among the xanthophylls as compared to carotenes. Also the variety in the degree of
560 esterification and in the fatty acids that may be involved in each ester could be one of
561 the causes of this difference (Borel, 2003). The average increase in the bioaccessibility
562 of the carotenes with the pasteurization was 238% while in the UF-FG the
563 bioaccessibility was increased by only 49%. Interestingly, for all treatments, the
564 bioaccessibility increase was higher in PTF than in PT.

565 The pH of the samples was measured as it is known that the pH can affect the
566 transfer of carotenoids to mixed micelles during the digestion (reviewed in Reboul &
567 Borel, 2011). However, no significant changes in the pH were found with the thermal
568 treatments, being 3.8 the pH average (data not shown). In any case, the effect of
569 possible differences in the acidity of the juices was expected to be counteracted by the
570 pH reached during the gastric and intestinal digestion phases, which was homogeneous
571 across samples.

572 3.8. *Bioaccessible carotenoid content*

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2 573 Information on the bioaccessible carotenoid contents can be more meaningful as
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4 574 they represent the potentially absorbable amount of carotenoids. These data are
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7 575 summarized in Table 5. In all samples, ζ -carotene was the carotenoid with the highest
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9 576 bioaccessible content, followed by PT and PTF which showed virtually the same
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11 577 bioaccessible content. The intake of approximately one glass of FRESH, i.e. 250 mL,
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13 578 provided 0.7 mg of ζ -carotene and 0.4 mg of PT and PTF. With the exception of ζ -
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15 579 carotene, the amount of potentially absorbable PT and PTF that is provided with the
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17 580 intake of any of the juices is much higher than that of the rest of carotenoids. Thus, for
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19 581 example, the bioaccessible amount of PT or PTF is more than 100 times greater than
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21 582 that of violaxanthin.
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26 583 Noteworthy is that neither the PAST, which presented the highest total carotenoid
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28 584 bioaccessibility, nor the FRESH, which exhibited the highest initial carotenoid
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30 585 concentration were the juices with highest content of bioaccessible carotenoids. Thus,
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32 586 the source that provided the highest quantities of bioaccessible PT, PTF, and ζ -carotene
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34 587 and the highest total carotenoid bioaccessible content was UF-MW, while the lowest
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36 588 was UF-FG. This clearly demonstrates the importance of using the bioaccessible
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38 589 content rather than the percentage of bioaccessibility and the concentration in the
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40 590 matrix, when it comes to evaluate the goodness of a food to raise the carotenoid status
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42 591 in humans. UF-FG was the only juice with an bioaccessible content of PT, PTF and
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44 592 total carotenoid lower than that of FRESH. In all the cases, the increase in the
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46 593 bioaccessible content with the thermal treatments was higher for PTF compared to PT,
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48 594 and this could mainly due to the higher increase in the bioaccessibility of PTF with the
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50 595 different thermal treatments.
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597 4. Conclusions

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5 599 carotenoids, i.e. PT, PTF and ζ -carotene that, despite having been found in human
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7 600 plasma, have been very little studied. These carotenes are particularly abundant in the
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10 601 sweet orange cultivar Pinalate, being then an excellent system to analyse their
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12 602 bioaccessibility in orange juices, a common carotenoid source in the diet, and also to
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14 603 investigate the effects of the preserving juice treatments on these carotenes. These three
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17 604 compounds have a similar behaviour upon thermal treatments, that is, they suffered a
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19 605 similar degradation and presented similar bioaccessibility in the samples. However, the
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22 606 highest bioaccessibility in all the samples was that of the PT and the lowest was that of
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24 607 the ζ -carotene, and this could be mainly due to the differences in the number of
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27 608 conjugated double bonds. On the other hand, considering just the treatments and not the
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29 609 digestion, it seems that PT and PTF compared to xanthophylls are less prone to
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32 610 degradation due to the combination of oxygen, enzymes and acid, but less heat stable.
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34 611 Possible explanations for this are the lack of terminal ring, the lower number of
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37 612 conjugated double bonds, the lower polarity, and the sites and patterns of accumulation
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39 613 in the cells of the colourless carotenoids.

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41 614 In summary, this study shows how pasteurization or ultrafreezing followed by
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44 615 thawing have a great impact on the colour of the orange juice and on the concentration
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46 616 and bioaccessibility of the carotenoids. Taking into account the particle size distribution
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49 617 and the cell structures degradation, it can be concluded that the increase in the
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51 618 bioaccessibility is mainly due to the degree of disruption that the matrix suffered with
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54 619 the thermal treatments. Although the thermal treatments tested generated carotenoids
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56 620 losses, the ultrafrozen juices which were thawed at room temperature or in the
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621 microwave oven were better sources of bioaccessible phytoene, phytofluene, and total
622 carotenoids than the fresh juice.

623 Thus, this study can be of interest to the functional food and nutricosmetics
624 industries as phytoene and phytofluene are raising increasing awareness as evidence is
625 accumulating that they may provide diverse health and cosmetic benefits.

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TABLES

Table 1. Carotenoid concentration (mg/100 g juice) of Pinalate fresh orange juice and Pinalate orange juices subjected to different thermal treatments

	Juice samples				
	FRESH	UF-RT	UF-FG	UF-MW	PAST
Phytoene	1.39 ± 0.14 ^{Ac}	1.00 ± 0.05 ^{Bb}	0.83 ± 0.06 ^{Bc}	0.79 ± 0.08 ^{BCb}	0.45 ± 0.10 ^{Cbc}
Phytofluene	1.81 ± 0.14 ^{Ab}	1.24 ± 0.07 ^{Bb}	1.08 ± 0.08 ^{Bb}	1.04 ± 0.10 ^{Bb}	0.59 ± 0.11 ^{Cb}
ζ-Carotene	4.35 ± 0.26 ^{Aa}	2.90 ± 0.19 ^{Ba}	2.59 ± 0.19 ^{Ba}	2.55 ± 0.25 ^{Ba}	1.47 ± 0.25 ^{Ca}

1	9- <i>cis</i> -Violaxanthin	0.05 ± 0.01 ^{Ad}	0.01 ± 0.00 ^{Bc}	0.01 ± 0.00 ^{Bd}	0.01 ± 0.00 ^{Bc}	0.02 ± 0.00 ^{Bc}
2	Zeaxanthin	0.02 ± 0.00 ^{Ad}	0.01 ± 0.00 ^{Bc}	0.01 ± 0.00 ^{Bd}	0.01 ± 0.00 ^{Bc}	0.01 ± 0.00 ^{Bc}
3						
4	9- <i>cis</i> -Antheraxanthin	0.02 ± 0.00 ^{Ad}	0.01 ± 0.00 ^{Cc}	0.01 ± 0.00 ^{Cd}	0.01 ± 0.00 ^{Cc}	0.01 ± 0.00 ^{Cc}
5						
6	β-Carotene	0.09 ± 0.00 ^{Ad}	0.07 ± 0.00 ^{Bc}	0.06 ± 0.00 ^{BCDd}	0.06 ± 0.01 ^{BCDc}	0.03 ± 0.00 ^{Ec}
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9	Total	7.73 ± 0.55 ^A	5.24 ± 0.32 ^B	4.59 ± 0.33 ^B	4.47 ± 0.44 ^B	2.57 ± 0.47 ^C

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12 660 FRESH, Fresh orange juice; UF-RT, ultrafrozen juice with thawing at room temperature; UF-FG,
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14 661 ultrafrozen juice with thawing in the fridge; UF-MW, ultrafrozen juice with thawing in the microwave
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16 662 oven; PAST, pasteurized juice. Values are the mean ± SD of 3 independent measures. Values within a
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18 663 column and within a row with different lowercase and uppercase letters respectively indicate statistically
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20 664 significant differences ($P < 0.05$).

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40 673 **Table 2. Particle size characteristics in Pinalate fresh orange juice and Pinalate**
41
42 674 **orange juices subjected to different thermal treatments.**

	Juice samples					
	FRESH	UF-RT	UF-FG	UF-MW	PAST	
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50	A _s	91.7 ± 2.5 ^c	105.1 ± 1.6 ^b	93.2 ± 3.1 ^c	110.0 ± 4.4 ^{ab}	115.4 ± 1.8 ^a
51						
52	D _[3,2]	62.3 ± 1.7 ^a	54.4 ± 0.8 ^b	61.3 ± 2.0 ^a	52.0 ± 2.0 ^{bc}	49.5 ± 0.7 ^c
53						
54	D _[4,3]	446.3 ± 8.3 ^a	382.0 ± 6.6 ^c	413.0 ± 4.6 ^b	351.7 ± 8.5 ^d	325.3 ± 5.5 ^e
55						
56	D _v (10)	27.7 ± 1.4 ^a	22.6 ± 0.5 ^b	25.5 ± 0.9 ^a	22.1 ± 1.0 ^b	21.6 ± 0.3 ^b
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$D_v(50)$	467.7 ± 10.2^a	365.3 ± 14.0^c	421.0 ± 6.6^b	296.3 ± 18.9^d	224.7 ± 11.7^e
$D_v(90)$	844.3 ± 5.5^a	812.0 ± 5.0^c	829.3 ± 2.3^b	780.3 ± 7.4^d	764.0 ± 6.6^e

675 As, specific surface area (m^2/kg); $D_{[3,2]}$, surface area-based mean diameter (μm); $D_{[4,3]}$, volume-based
676 mean diameter (μm); $D_v(10)$, $D_v(50)$ and $D_v(90)$, values of particle size below which there is 10%, 50%
677 and 90% of sample volume, respectively (μm). FRESH, Fresh orange juice; UF-RT, ultrafrozen juice
678 with thawing at room temperature; UF-FG, ultrafrozen juice with thawing in the fridge; UF-MW,
679 ultrafrozen juice with thawing in the microwave oven; PAST, pasteurized juice. Values are the mean \pm
680 SD of 3 independent measures. Values within a row with different lowercase letters indicate statistically
681 significant differences ($P < 0.05$).

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689 **Table 3**
690 **CIELAB colour parameters of Pinalate fresh orange juice and Pinalate orange**
691 **juices that were subjected to different thermal treatments.**

	Colour parameters			
	L^*	C_{ab}^*	h_{ab}	ΔE^*
FRESH	86.6 ± 1.8^a	31.83 ± 1.90^a	88.78 ± 1.69^a	-
UF-RT	85.1 ± 0.6^{ab}	32.36 ± 1.42^a	87.60 ± 1.90^a	1.17 ± 0.15^c
UF-FG	84.1 ± 0.6^{ab}	34.47 ± 1.43^a	87.26 ± 1.75^a	2.93 ± 0.51^b

UF-MW	84.6 ± 0.4 ^{ab}	33.87 ± 0.82 ^a	87.75 ± 1.57 ^a	3.55 ± 0.60 ^{ab}
PAST	82.4 ± 0.5 ^b	31.24 ± 1.34 ^a	86.47 ± 1.84 ^a	4.15 ± 0.12 ^a

692 FRESH, Fresh orange juice; UF-RT, ultrafrozen juice with thawing at room temperature; UF-FG,
693 ultrafrozen juice with thawing in the fridge; UF-MW, ultrafrozen juice with thawing in the microwave
694 oven; PAST, pasteurized juice. Values are the mean ± SD of 5 independent measures. *L*^{*}, lightness; *C*^{*}_{ab},
695 chroma; *h*_{ab}, hue angle; ΔE^* , colour difference. ΔE^* was calculated taken the fresh juices as reference
696 sample. Values within a column with different lowercase letters indicate statistically significant
697 differences ($P < 0.05$).

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706 **Table 4. Carotenoid bioaccessibility (in percentage) of Pinalate fresh orange juice**

707 **and Pinalate orange juices subjected to different thermal treatments.**

	Juice samples				
	FRESH	UF-RT	UF-FG	UF-MW	PAST
Phytoene	10.26 ± 0.27 ^{Ca}	17.06 ± 1.54 ^{BCa}	14.18 ± 0.75 ^{BCab}	21.51 ± 2.42 ^{Bab}	31.97 ± 4.82 ^{Aab}
Phytofluene	7.76 ± 0.24 ^{Cbc}	14.74 ± 1.00 ^{Bab}	11.89 ± 0.74 ^{BCbc}	18.55 ± 1.96 ^{Bbc}	26.70 ± 4.51 ^{Ab}
ζ-Carotene	6.72 ± 0.31 ^{Dc}	13.41 ± 0.67 ^{BCab}	10.65 ± 0.74 ^{CDcd}	17.19 ± 1.71 ^{Bbc}	24.16 ± 4.26 ^{Ab}
9- <i>cis</i> -Violaxanthin	2.72 ± 0.39 ^{CDd}	16.65 ± 2.96 ^{Aa}	6.88 ± 1.30 ^{BCef}	9.74 ± 2.79 ^{Bc}	7.88 ± 0.80 ^{Bc}

Zeaxanthin	9.24 ± 1.22 ^{Bab}	15.50 ± 1.70 ^{Bab}	15.10 ± 0.81 ^{Ba}	30.73 ± 4.67 ^{Aa}	39.57 ± 5.32 ^{Aa}
9- <i>cis</i> -Antheraxanthin	1.56 ± 0.18 ^{Dd}	6.14 ± 0.83 ^{Bc}	4.49 ± 0.88 ^{BCf}	8.91 ± 1.68 ^{Ac}	3.57 ± 0.52 ^{CDc}
β-Carotene	6.21 ± 0.41 ^{Cc}	11.72 ± 0.60 ^{BCb}	8.98 ± 0.81 ^{BCde}	14.57 ± 1.28 ^{Bbc}	21.01 ± 3.87 ^{Ab}
Total	7.57 ± 0.28 ^C	14.44 ± 0.99 ^B	11.54 ± 0.75 ^B	18.22 ± 1.89 ^B	25.93 ± 4.35 ^A

708 FRESH, Fresh orange juice; UF-RT, ultrafrozen juice with thawing at room temperature; UF-FG,
709 ultrafrozen juice with thawing in the fridge; UF-MW, ultrafrozen juice with thawing in the microwave
710 oven; PAST, pasteurized juice. Values are the mean ± SD of 5 independent measures. Values within a
711 column and within a row with different lowercase and uppercase letters respectively indicate statistically
712 significant differences ($P < 0.05$).

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719 **Table 5. Bioaccessible carotenoid content (mg/250 mL) of Pinalate fresh orange**
720 **juice and Pinalate orange juices subjected to different thermal treatments.**

	Juice samples			
	UF-RT	UF-FG	UF-MW	PAST
Phytoene	28.22 ± 3.60	40.53 ± 4.33	43.32 ± 5.97	67.85 ± 7.25
Phytofluene	31.22 ± 4.00	40.23 ± 4.45	42.58 ± 5.57	67.57 ± 5.95
ζ-Carotene	33.27 ± 4.32	40.59 ± 4.31	41.38 ± 5.82	66.21 ± 5.74
9- <i>cis</i> -Violaxanthin	85.71 ± 4.35	75.50 ± 7.37	68.33 ± 3.82	62.51 ± 4.38

Zeaxanthin	37.80 ± 5.46	49.35 ± 5.11	57.29 ± 0.47	58.46 ± 2.97
9-cis-Antheraxanthin	54.95 ± 5.63	55.52 ± 2.39	59.12 ± 1.60	55.98 ± 5.37
β-Carotene	22.01 ± 4.48	28.70 ± 3.30	31.02 ± 7.49	63.70 ± 4.62
Total	32.18 ± 4.12	40.67 ± 4.22	42.18 ± 5.76	66.74 ± 6.03

721 FRESH, Fresh orange juice; UF-RT, ultrafrozen juice with thawing at room temperature; UF-FG,
722 ultrafrozen juice with thawing in the fridge; UF-MW, ultrafrozen juice with thawing in the microwave
723 oven; PAST, pasteurized juice. Values are the mean ± SD of 5 independent measures. Values within a
724 column and within a row with different lowercase and uppercase letters respectively indicate statistically
725 significant differences ($P < 0.05$).

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Supplementary Table 1. Degradation in percentage of carotenoid in Pinalate orange juices subjected to different thermal treatments.

	Juice samples			
	UF-RT	UF-FG	UF-MW	PAST
Phytoene	28.22 ± 3.60	40.53 ± 4.33	43.32 ± 5.97	67.85 ± 7.25
Phytofluene	31.22 ± 4.00	40.23 ± 4.45	42.58 ± 5.57	67.57 ± 5.95
ζ-Carotene	33.27 ± 4.32	40.59 ± 4.31	41.38 ± 5.82	66.21 ± 5.74
9- <i>cis</i> -Violaxanthin	85.71 ± 4.35	75.50 ± 7.37	68.33 ± 3.82	62.51 ± 4.38
Zeaxanthin	37.80 ± 5.46	49.35 ± 5.11	57.29 ± 0.47	58.46 ± 2.97
9- <i>cis</i> -Antheraxanthin	54.95 ± 5.63	55.52 ± 2.39	59.12 ± 1.60	55.98 ± 5.37
β-Carotene	22.01 ± 4.48	28.70 ± 3.30	31.02 ± 7.49	63.70 ± 4.62
Total or mean?	32.18 ± 4.12	40.67 ± 4.22	42.18 ± 5.76	66.74 ± 6.03

FRESH, Fresh orange juice; UF-RT, ultrafrozen juice with thawing at room temperature; UF-FG, ultrafrozen juice with thawing in the fridge; UF-MW; ultrafrozen juice with thawing with the microwave; PAST, pasteurized juice. Values are the mean ± SD of 3 independent measures.

Supplementary Table 2. Retention times in minutes and absorption maxima in nm of some compounds of the Pinalate orange juices.

Compound	Retention time	Absorption maxima
Compound 1	21.87	406, 430, 455
Compound 2	22.65	416, 439, 467
Compound 3	23.00	417, 440, 469
Compound 4	23.16	440, 463, 491

Supplementary Table 3. Increase in percentage of the carotenoid bioaccessibility in Pinalate orange juices subjected to different thermal treatments.

	Juice samples			
	UF-RT	UF-FG	UF-MW	PAST
Phytoene	66.28	38.21	109.65	211.60
Phytofluene	89.95	53.22	139.05	244.07
ζ-Carotene	99.55	58.48	155.80	259.52
9- <i>cis</i> -Violaxanthin	512.13	152.94	258.09	189.71
Zeaxanthin	67.75	63.42	232.58	328.25
9- <i>cis</i> -Antheraxanthin	293.59	187.82	471.15	128.85
β-Carotene	88.73	44.61	134.62	238.33
Total	90.75	52.44	140.69	242.54

FRESH, Fresh orange juice; UF-RT, ultrafrozen juice with thawing at room temperature; UF-FG, ultrafrozen juice with thawing in the fridge; UF-MW, ultrafrozen juice with thawing with the microwave; PAST, pasteurized juice. Values are the mean ± SD of 5 independent measures. To calculate the increase of the bioaccessibility the corresponding bioaccessibility found in the FRESH was taken as control.

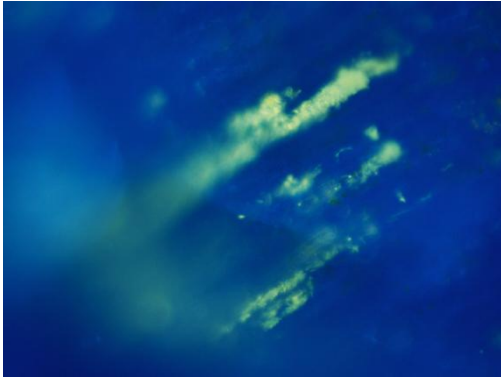
FIGURE CAPTIONS

Figure 1. Representative micrographs of Pinalate orange pulp (A) and juice (B) and Navelate pulp (C) and juice (D) obtained by epifluorescence microscopy.

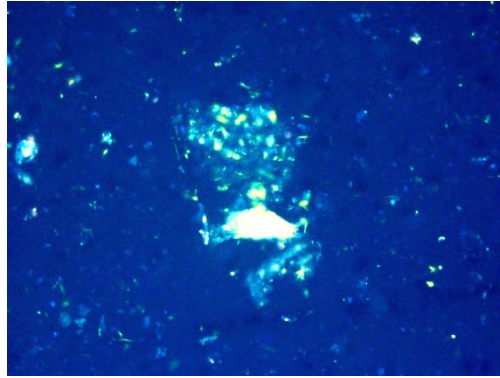
Figure 2. Representative micrographs of Pinalate orange pulp and juices obtained by Transmission Electron Microscopy. (A) Pulp (bar 1 μm); (B) Fresh orange juice (FRESH) (bar 1 μm); (C) Ultrafrozen juice with thawing in the fridge (bar 0.5 μm); (D) Ultrafrozen juice with thawing at room temperature (UF-RT) (bar 1 μm); (E) Ultrafrozen juice with thawing with the microwave oven (UF-MW) (bar 1 μm); (F) Pasteurized juice (PAST) (bar 0.5 μm). CW, cell wall; Mm, achlorophyllous membranes; Mt, mitochondria; Pg, plastoglobuli; s, starch grains; V, vesicles.

Figure 3. Particle size distribution of Pinalate fresh orange juice and Pinalate orange juices subjected to different thermal treatments.

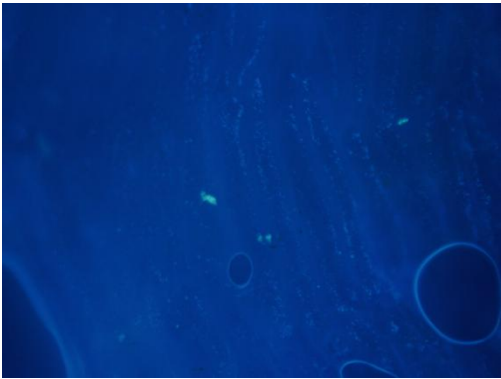
A (Pinalate pulp)



B (Pinalate juice)



C (Navelate pulp)



D (Navelate juice)



Figure 1

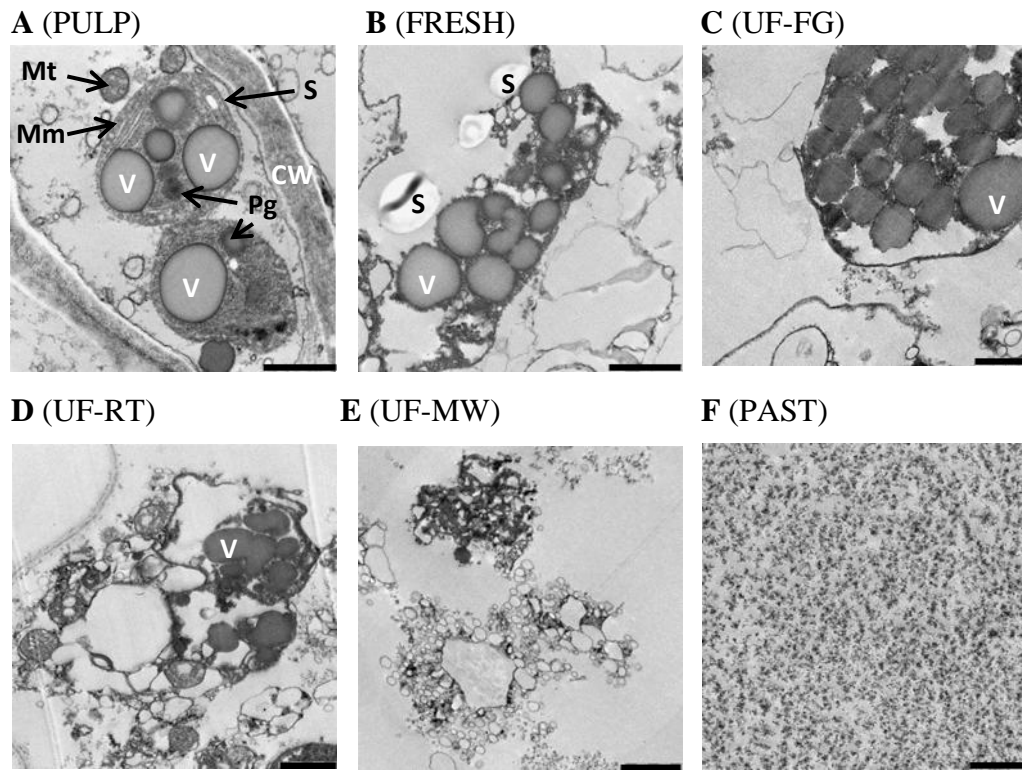


Figure 2

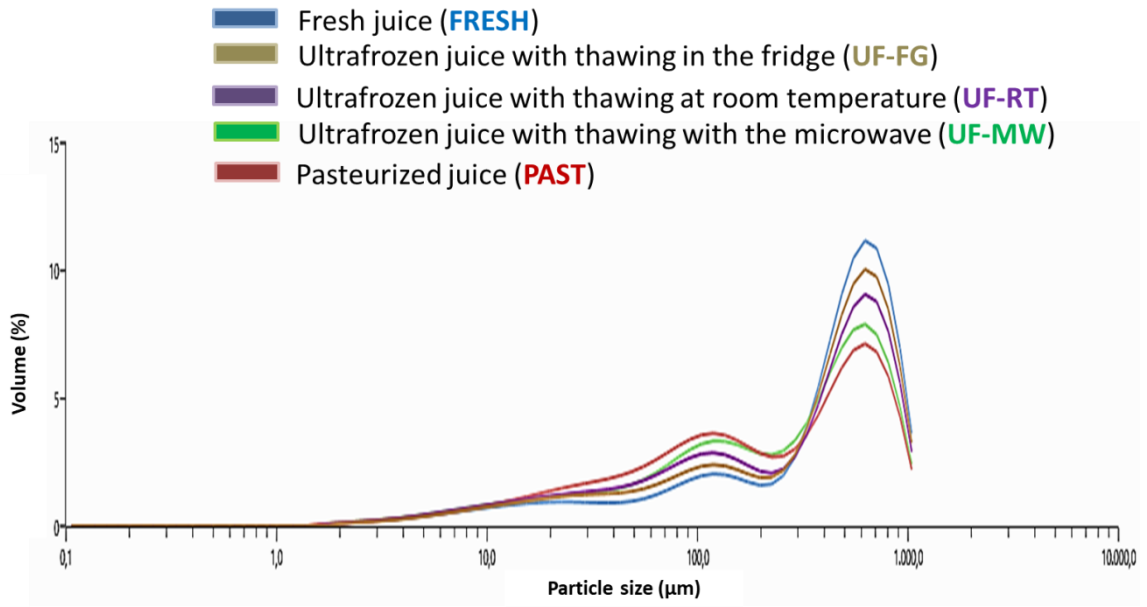


Figure 3

Graphical abstract

[Click here to download Figure: Graphical Abstract_Thermal Treatments PT P](#)

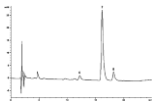


MINERAL ORANGE JUICE SAMPLES

1. Fresh juice (FRESH)
2. Ultrafrozen juices thawed at room temperature (RT)
3. Ultrafrozen juices thawed in microwave oven (MW)
4. Ultrafrozen juices thawed in the fridge (FG)
5. Pasteurized juice (PAST)

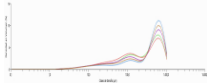


Carotenoid analysis (HPLC)

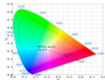


- PHYTOENE (PT)
- PHYTOFLUENE (PTF)
- ζ-Carotene
- β-Carotene
- Violaxanthin
- Antheraxanthin
- Zeaxanthin

Particle Size Distribution (PSD)



Colour analysis



TEM images



BIOACCESSIBILITY:

PT > PTF > ζ-Carotene

CAROTENOID DEGRADATION:

Pasteurized juices: PT and PTF > Xanthophylls
Ultrafrozen juices: PT and PTF < Xanthophylls

PSD:

FRESH > FG > RT > MW > PAST

DEGRADATION OF SUBSTRUCTURES (TEM):

FRESH < FG < RT < MW < PAST

BIOACCESSIBILITY:

FRESH < FG < RT < MW < PAST