

1 **Traditional agricultural practices enable sustainable remediation of highly**  
2 **polluted soils in Southern Spain for cultivation of food crops.**

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20

21 **Abstract**

22 This study relates elemental content of a range of edible crops grown in soils severely  
23 polluted by metals and metalloids as affected by traditional smallholder management  
24 practices. Five agricultural plots close to a sulfidic waste dump were monitored. Soil  
25 analysis demonstrated elevated concentrations of As, Cu, Pb and Zn that were greatly in  
26 excess of maximum statutory limits for agricultural soils in this region. The main  
27 vegetables (lettuce, chard, onion, potatoes) and lemon, together with their associated  
28 soils, were measured for elemental content. Extractable soil element concentrations  
29 were very low. There were differences in elemental accumulation between crops, but  
30 none exceeded statutory concentrations in edible parts. Soil-plant transfer factors were  
31 uniformly low for all elements and crops. It is concluded that traditional soil  
32 management practices (annual liming and application of animal manures) have created  
33 conditions for sustainable long-term safety use, with potential for multiple end-use, of  
34 these highly polluted soils

35  
36 **Keywords:** Trace elements, edible plants, soil pollution, Tharsis mines, Spain

37

38 **1. Introduction**

39 Plants are the first compartment of the terrestrial food chain. Due to their  
40 capacity to accumulate potentially toxic trace elements, they can represent a threat to  
41 animals and humans that consume them (Intawongse and Dean, 2006; Liu et al., 2006).

42 The risk that trace elements pose to the environment and human health is a  
43 function of their speciation in soils and subsequent accumulation and partitioning within  
44 plants (Kabata-Pendias, 2004). The solubility of trace elements varies widely because  
45 many factors influence their concentration in the soil solution: edaphic characteristics  
46 (pH, texture and organic matter content), climatic conditions, and agronomic

47 management (Alloway, 1995, Chojnacka et al., 2005; Tokalioğlu and Kartal 2006). The  
48 rate of metal translocation from the soil to edible and harvested parts of cultivated plants  
49 depends, in addition, on vegetation type and metal involved, besides the soil and  
50 climatic factors (Alloway, 1995; McLaughlin et al., 1999).

51 Chaney (1980) introduced the concept of the ‘soil-plant barrier’ and classified  
52 trace elements in four groups with respect to their potential for food chain transfer:  
53 Group 1 (Ag, Cr, Sn, Ti, Y and Zr) with low risk to human health because they are not  
54 usually taken up by plants due to their low solubility in soil. Group 2 (As, Hg and Pb)  
55 which are strongly sorbed by soil colloids, or may be immobilized in plant roots with  
56 very limited transfer to edible shoot tissues, and therefore pose marginal risks. Group 3  
57 (B, Cu, Mn, Mo, Ni and Zn) are phytotoxic at concentrations that pose little risk,  
58 although ‘the soil-plant barrier’ protects the food chain from these elements. Group 4  
59 consists of Cd, Co, Mo and Se, which pose human or animal health risks at plant tissue  
60 concentrations that are not phytotoxic.

61 Soil pollution with trace elements, mainly those from Groups 2 and 3 described  
62 above, represents one of the most prominent environmental hazards from abandoned  
63 mine sites in the Iberian Pyrite Belt (Fernández-Caliani et al. 2009a, b; Fernández-  
64 Caliani and Barba-Brioso, 2010). Past mining and ore processing activities resulted in  
65 deforestation and destruction of the natural vegetation, transforming soils into marginal  
66 lands unable to sustain commercial agriculture. However, traditional small-scale  
67 agriculture has persisted on limited areas of arable land adjacent to the mining villages  
68 in this region.

69 The goals of the study reported here were to: 1) determine total concentrations of trace  
70 elements in different parts of fruit and vegetables, and in the soil in which they grew,  
71 focusing on their edible parts; 2) assess the mobile and potentially-mobile fractions of

72 the soil trace element pools and the soil-plant transfer factors for the crops grown here;  
73 and 3) evaluate the long-term sustainability of the traditional soil management practices  
74 used in this region for marginal or soils of low fertility in relation to soil-plant transfer  
75 of inorganic pollutants.

76

## 77 **2. Site description**

78 The Tharsis Mines (UTM coordinates: X= 666,746 and Y= 4,162,779) are  
79 located in one of the oldest and best-known mining districts in the Iberian Pyrite Belt  
80 (South-West Spain), with a lengthy history of exploitation that dates back to pre-Roman  
81 times (Checkland, 1967). The most intensive period of mining operations occurred  
82 between the end of the 19th century and the middle of the 20th century, when pyrite was  
83 the main raw material used for sulphuric acid manufacture by the European chemical  
84 industry. Past mining and smelting activities were carried out without concern or even  
85 awareness of their negative environmental impact, resulting a present-day landscape of  
86 large opencast mines, waste rock piles and tailings dams that encompass an area of  
87 about 350 ha (Figure 1a).

88 During mining and after closure, pollutants have been transferred from the mine  
89 wastes to nearby soils by acid mine drainage and/or atmospheric deposition of wind-  
90 blown dust. Soils and vegetation show elevated trace element concentrations up to 2-3  
91 km away from the mining area (Chopin and Alloway, 2007a,b). There are also a number  
92 of small holdings, none of which exceeds one hectare in extent, located in the  
93 immediate vicinity of the mine waste dumps, all at risk of contamination with metals.

94 Five agricultural plots, close to the sulfidic waste dumps of the “Filón Norte”  
95 open pit (Figure 1 b, c) were selected for this study. The alluvial soils of these plots are  
96 devoted to traditional horticulture, despite the fact that they are occasionally flooded by

97 acid waters emanating from nearby waste dumps and surface mine workings. To correct  
98 soil acidity, farmers routinely amend soils with lime, together with regular additions of  
99 animal manure. These practices have taken place for at least 25 years in the investigated  
100 plots, a fact established by interviews with each plot owner. The main crops cultivated  
101 by the smallholders are vegetables, typically lettuce (*Lactuca sativa* L.), chard (*Beta*  
102 *vulgaris* (L.) var. *cicla* K. Koch), onion (*Allium fistulosum* L.) and potato (*Solanum*  
103 *tuberosum* L.). Some plots also contain mature Lemon trees (*Citrus limon* L. Brum).

104

### 105 **3. Methodology**

#### 106 **3.1 Materials and methods**

107 A suite of vegetable samples, comprising leaf, bulb, tuber and fruit crops, along  
108 with their associated soils were randomly collected from each plot in April 2008 at the  
109 indicated stage of maturity (Table 1).

110 Composite soil samples were taken to a depth of 20 cm (Ap horizon) from each  
111 plot where sampled vegetables were growing. These were transported to the laboratory  
112 in polyethylene bags, air dried, disaggregated with a wooden roller, passed through a 2  
113 mm stainless steel sieve, and homogenized prior to analysis. The particle-size  
114 distribution was determined by a combination of sieving and laser diffraction (Malvern  
115 MasterSizer instrument) methods. The pH, Eh and electrical conductivity values were  
116 measured with calibrated glass electrodes in a 1:2.5 (w/v) soil to water suspension. The  
117 content of total organic matter was determined by oxidation with potassium dichromate  
118 in a strong acid medium (Walkley-Black method), as described by Pansu and  
119 Gautheyrou (2006), and the carbonate content was measured by the Bernard calcimetry  
120 method.

121 Soil mineralogy was investigated in both bulk samples (<2 mm) and clay-size

122 fractions ( $<2 \mu\text{m}$ ) by powder X-ray diffraction (XRD) on a Bruker-AXS D8-Advance  
123 diffractometer, using monochromatic  $\text{CuK}\alpha$  radiation at 40 kV and 30 mA. The fine  
124 fraction was separated by sedimentation and analyzed in oriented aggregates of air-  
125 dried, ethylene glycol-treated and thermal-treated samples for clay mineral  
126 identification. Selected soil samples were examined by scanning electron microscopy  
127 using a JEOL JSM-5410 instrument coupled with an energy dispersive X-ray  
128 spectrometer (SEM-EDS).

129 Total concentrations of potentially toxic trace elements (As, Cu, Pb and Zn) in  
130 soil samples were determined by inductively coupled plasma-optical emission  
131 spectrometry (ICP-OES), (Jobin Yvon ULTIMA 2), after 4-acid ( $\text{HF-HClO}_4\text{-HNO}_3\text{-}$   
132  $\text{HCl}$ ) digestion of 0.1 g soil sample oven dried for 24 hours at  $110^\circ\text{C}$ . Quality control  
133 included the use of a method reagent blank and several certified reference materials  
134 (SARM-1 and SARM-4 for soil analysis) to check accuracy and precision of the  
135 analytical data (relative standard deviation below 10%).

136 In order to determine the most labile metal pools, an aliquot of each soil sample  
137 (2 g) was subjected to single extractions by shaking for one hour with deionised water, a  
138 mild neutral salt solution (0.01 M  $\text{CaCl}_2$ ) or a complexing agent (0.05 M EDTA at pH  
139 7) at a soil:solution ratio of 1:10 (w/v) (e.g. Houba et al. 1996; Ure et al., 1996). The  
140 first two soil extract solutions were analyzed after centrifugation (for 10 min at 4500  
141 rpm) by inductively coupled plasma mass spectrometry (ICP-MS), using a Hewlett  
142 Packard 4500 instrument with detection limits of  $0.01 \mu\text{gL}^{-1}$ , whilst EDTA-extractable  
143 metals were determined by ICP-OES. All trace element concentrations were reported on  
144 an oven-dry basis.

145 All plant samples (leaves, roots, bulbs, peel and seeds) were washed (for 10 s  
146 approximately) with a solution of phosphate-free detergent, then with a 0.1 N HCl

147 solution and finally with distilled water. Plant material was oven-dried at 70°C, ground  
148 and passed through a 500 µm stainless steel sieve. A 0.5 g aliquot was digested by wet  
149 oxidation with concentrated HNO<sub>3</sub> under pressure in a microwave digester (Jones and  
150 Case, 1990). Three consecutive steps (5 min. each) of power (250 W, 450 W and 600  
151 W) were applied, and then these extracts were diluted to 50 ml volume with deionised  
152 water of 18 mΩ quality. The analysis of trace metals in the digests was performed by  
153 ICP-MS. The accuracy and precision of the analytical method were assessed by routine  
154 analyses of the reference sample CRM-279 (Sea lettuce) and CS DC73350 (poplar  
155 leaves). Recovery rates for reference plant samples were between 90 and 110%.

156

### 157 ***3.2 Treatment of analytical data***

158 Quantitative assessment of overall soil pollution was based on the pollution load  
159 index (PLI), as defined by Tomlinson et al., (1980), taking into consideration the  
160 concentration factor (CF), which is the ratio between each trace element in the soil and  
161 its background value. The PLI of each soil sample was calculated by deriving the *n*-th  
162 root of the *n* factors (CF<sub>1</sub>×CF<sub>2</sub>×CF<sub>3</sub>×...×CF<sub>n</sub>). Thus, values of PLI close to one indicate  
163 heavy metal loads near the background level, while values >1 indicate soil pollution  
164 (Cabrera et al., 1999).

165 In order to find out what proportion of the total soil metal concentration was  
166 available and transferred to different organs of the vegetables grown by the  
167 smallholders, the transfer coefficient (TC) was calculated. This is defined as the ratio of  
168 metal concentration in the plant, [M]<sub>plant</sub>, to the total metal concentration in the soil,  
169 [M]<sub>soil</sub> (Adriano, 2001).

170  $TC = [M]_{\text{plant}} / [M]_{\text{soil}}$

171 The data was statistically analysed using StatSoft Statistica 7.0 to recognize  
172 variables trends and groupings. Kolmogorov-Smirnov and Shapiro-Wilk's normality  
173 tests were carried out for all variables, refusing the normal distribution for almost all of  
174 them. Because the variables required a non-parametric analyses, a Spearman correlation  
175 matrix (significance level  $p < 0.01$ ) was obtained, correlating total trace element contents  
176 in soils with those measured in soil extracts and plants.

177

## 178 **4. Results and discussion**

### 179 *4.1 Soil constituents and properties*

180 The upper part of the soil profile shows a well-developed organic horizon with  
181 granular structure and had similar edaphic properties in all the sampled soils (Table 2).  
182 The soil is brown (10YR 5/3, 10YR 6/3 dry) in colour and has a silty loam texture with  
183 <10% of clay-sized particles. Some soil samples contain significant amounts of coarse-  
184 grained components (greater than 2 mm), consisting of sub-angular lithic fragments of  
185 heterogeneous waste rocks and slag residues.

186 The soil had an average pH value of 7.4, and electrical conductivity ranged  
187 between 0.15 and 0.95 mS cm<sup>-1</sup>, indicating a low salinity level in the soil solution. The  
188 soil appears to be well drained and aerated spanning a narrow range of positive Eh  
189 values (457-505 mV) that reflect moderately oxidizing conditions.

190 The soil has a high content of total organic matter (8.1-14.9%) and carbonates  
191 (up to 14%) resulting from the application of organic amendments and lime to improve  
192 soil fertility and to prevent soil acidity.

193 Soil minerals, identified by XRD, were composed of phyllosilicates (50-60%),  
194 quartz (30-40%), feldspars (5-10%), calcite (<5%) and dolomite (<5%). The clay  
195 mineral assemblage was dominated by illite and kaolinite, with minor vermiculite



196 and/or poorly defined mixed-layer phases. In addition, SEM-EDS analysis revealed the  
197 occurrence of amorphous or poorly crystalline Fe oxyhydroxides and a number of  
198 accessory minerals, such as baryte, apatite and monazite.

199

#### 200 ***4.2 Total concentrations of trace elements in soil***

201 The sampled soils contained high levels of As, Cu, Pb and Zn (Table 3),  
202 although total concentrations varied, depending on the location of the sampling plot  
203 (Figure 1). The highest concentrations of trace elements (up to 621 mg kg<sup>-1</sup> As, 752 mg  
204 kg<sup>-1</sup> Cu, 2395 mg kg<sup>-1</sup> Pb, and 593 mg kg<sup>-1</sup> Zn) were found in plots 4 and 5, located in  
205 the immediate vicinity of the mine wastes. Similar total soil element concentrations  
206 have been reported for cultivated soils adjacent to other abandoned mine sites in the  
207 Iberian Pyrite Belt (López et al., 2008; Fernández-Caliani et al. 2009a).

208 The total concentrations of As, Cu, Pb and Zn in these soils are between one and  
209 two orders of magnitude above both the regional geochemical baseline (Galán et al.  
210 2008) and normal levels found in Spanish agricultural soils (López-Arias and Grau-  
211 Corbí, 2005). Furthermore, these values greatly exceed the maximum allowable  
212 concentrations for agricultural soils established by the Regional Government of  
213 Andalusia (Aguilar et al., 1999). This indicates potential health risks associated with  
214 consuming edible crops grown in these soils.

215 The Concentration Factor, CF, defined as the ratio between each trace element in  
216 the soil sample and its background value (50<sup>th</sup> percentile), was particularly elevated for  
217 As (up to 24.8), Cu (up to 23.5), and Pb (up to 63.0), indicating soil pollution. The  
218 lowest CF values were found for Zn, although the ratio was higher than 1 for all  
219 samples. Pollution load indices (PLI) at each plot are shown in Table 3. The PLI values

220 varied between 9.4 (plot 1) and 22.8 (plot 5), reflecting the high pollution loads of As,  
221 Cu, Pb and Zn, especially in plots near the mine wastes.

222

### 223 ***4.3 Trace element concentrations extracted with water, CaCl<sub>2</sub> and EDTA***

224 Water, CaCl<sub>2</sub> and EDTA remove trace elements from different compartments  
225 within the overall soil matrix. The water-soluble fraction and the fraction assessed by  
226 CaCl<sub>2</sub>-extraction are considered to simulate the proportion of the total soil metal pool  
227 that may be available for uptake by plants and play a key role in many transfer pathways  
228 (Gupta et al. 1996; Houba et al. 1996). The EDTA-extractable fraction is composed of  
229 those ions present in the soil solution, as well as those that are loosely bound to sites in  
230 the solid phase of the soil and potentially able to move into the plant root system  
231 (Madejón et al., 2009). In some cases, EDTA-extractable metal concentrations have  
232 been reported to be closely correlated with metal concentrations in plants (e.g.  
233 Sahuquillo et al., 2003), although methodology for assessing the bioavailability of  
234 metals in soil is a controversial issue (Menzies et al., 2007).

235 The mean water-soluble and CaCl<sub>2</sub>-extractable concentrations of As, Cu, Pb and  
236 Zn were found to be less than 1 mg kg<sup>-1</sup> in all the agricultural plots (Table 3), whereas  
237 the EDTA-extracted mean concentrations were higher, and varied widely depending on  
238 the trace element involved (2.9-7.2 mg kg<sup>-1</sup> for As, 22.3-32.6 mg kg<sup>-1</sup> for Cu, 46.4-95.8  
239 mg kg<sup>-1</sup> for Pb, and 17.8-20.3 mg kg<sup>-1</sup> for Zn). Accordingly, the mobile and very active  
240 metal fraction was practically negligible (lower than 0.22%), and the EDTA-extractable  
241 fraction was below 10% (Figure 2), indicating a limited potential mobility of all  
242 elements despite their high total concentrations in soil. No significant correlation was  
243 observed between the extractable concentrations and the total concentrations in soil for  
244 all the investigated elements, suggesting that the pollution load did not have a

245 noticeable effect on the proportion of water soluble, exchangeable and complexed  
246 fractions. These results are in good agreement with previous studies on soils from this  
247 area (Barba-Brioso et al., 2007; Chopin and Alloway, 2007a,b).

248 In general, trace elements showed the following order of relative abundance in  
249 the mobile fraction: As > Cu > Zn > Pb. Therefore, As seems to be the most easily  
250 extractable trace element in soil. This can be explained by the fact that, under  
251 circumneutral pH conditions, As mobility can be increased in soils (Hartley et al.,  
252 2004). Arsenic usually forms water-soluble oxyanion species which are repelled by the  
253 negatively-charged surfaces of soil particles, thus preventing the adsorption of As  
254 oxyanions. A further finding supports our assumption: As concentration up to 138  $\mu\text{g L}^{-1}$   
255 compared to much lower to Cu and Zn concentrations ( $< 5 \mu\text{g L}^{-1}$ ) were measured in  
256 groundwater from a nearby well (unpublished data); the guideline value for As  
257 irrigation water is 0.1  $\text{mg L}^{-1}$  (FAO, 1985). On the other hand, the low extractability of  
258 Pb with water and  $\text{CaCl}_2$  indicated that this trace element was more strongly bound to  
259 constituents of the soil than As, Zn, and Cu.

260 Differences between elemental concentrations removed by the 3 extractants  
261 show that each provides different information on the metal status of the tested soils. As  
262 EDTA is a chelating agent that extracts elements by forming complexes with cations, it  
263 can access trace elements associated with stable binding sites within the soil matrix. The  
264 higher extractability of trace elements with EDTA (Table 3) compared with water and  
265  $\text{CaCl}_2$  extraction could be due to EDTA removing trace elements bound to organic  
266 matter (soil OM contents were between 8 and 15%, Table 2).

267 Speciation of trace elements in soils is mainly related to pH (McLaughlin et al.,  
268 2000). Although there were no correlation patterns between pH and extractable trace  
269 elements, the low extractability of Cu and Zn compared to their total concentrations

270 could also be related to the neutral pH of the soil (Madejón et al., 2009). Correlations  
271 between pH and both water and CaCl<sub>2</sub> extraction were significant (P<0.01) for As  
272 (r<sub>H<sub>2</sub>O</sub>= 0.59 and r<sub>CaCl<sub>2</sub></sub>= 0.62), Cu (r<sub>H<sub>2</sub>O</sub>= 0.49 and r<sub>CaCl<sub>2</sub></sub>= 0.52) and Pb (r<sub>H<sub>2</sub>O</sub>= 0.60 and  
273 r<sub>CaCl<sub>2</sub></sub>= 0.82). Significant correlations between pH and EDTA extraction were only found  
274 for Zn (r= 0.34).

275

#### 276 ***4.4 Trace element content of plants***

277 The elements that most commonly produce concerns about food safety are Cd,  
278 Hg, Pb, As and Se (Reilly, 1991). In addition, some micronutrients (e.g. Cu, Cr, Ni, Zn)  
279 may be toxic to both plants and animals when present at high concentrations  
280 (McLaughlin et al., 1999). Of the trace elements investigated here, As is considered to  
281 pose the major risk for human food-chain contamination (Kabata-Pendias and Pendias,  
282 1999, 2001).

283 Trace element content of onion bulb differed; this could be related depending on  
284 the stage of development when bulbs were sampled. In young onions bulbs, As  
285 concentration reached values up to 23.5 mg kg<sup>-1</sup> (above the statutory limits, Table 4).  
286 When the onion ripened and increased its biomass, this concentration decreased,  
287 analysis showing values below the statutory limit. This could be related to a ‘dilution  
288 effect’ of elements due to biomass increase with no concomitant increase in trace  
289 element uptake; this has been previously considered in the literature (e.g. Jarrel and  
290 Beverly 1981). Leafy vegetables such lettuce and chard are classified as crops with a  
291 high potential for trace element transfer from soil to the edible foliage (Pillay and  
292 Jonnalagadda, 2007), however Juhasz et al. (2008) found that such vegetables were poor  
293 As accumulators. In the case of the lettuces and chard sampled here, As contents in the  
294 edible leaves were within normal and statutory levels for vegetables. It is interesting to

295 note that the highest As concentrations were found in roots and outer leaves of lettuce  
296 (the most external dark green leaves), both non-edible parts of this vegetable (Figure 4).

297         Plants showed similar patterns of Pb uptake and partitioning as found for As;  
298 maximum contents in roots, up to 51.8 mg kg<sup>-1</sup> in onions (Table 4) and 13 mg kg<sup>-1</sup> in  
299 lettuce (Figure 4). This is in agreement with other observations on Pb uptake and  
300 distribution in plants (Adriano, 2001). In these soils, the soil–plant barrier may act to  
301 protect the human food chain against Pb toxicity (Chaney, 1989). Similar results for As  
302 and Pb uptake in onion and lettuce were reported by Lim et al. (2008). The effect of  
303 biomass increase in onions was also evident, bulbs of young onions showed Pb contents  
304 up to 8 mg kg<sup>-1</sup> in the edible part, but in ripe onions Pb content decreased to values  
305 below statutory levels (Table 4). This reflects the nature of bulb development, with  
306 biomass added as a result of carbohydrate transport from leaves. The Pb content of  
307 lettuce and chard foliage was also below statutory limits. There were also no age-  
308 dependant differences in Pb distribution in mature lettuce heads: the older, outer leaves  
309 had very similar Pb content to their less mature counterparts in the centre of the head  
310 (Figure 4).

311         The concentrations of As and Pb detected in whole potatoes, peel and tuber,  
312 were negligible (0.1 mg kg<sup>-1</sup>). There was no evidence of As contamination in potato  
313 tubers, which is in agreement with other findings (Dahal et al. 2008).

314         Copper and Zn have important physiological roles as micronutrients in plants,  
315 however excess concentrations in edible plant parts may pose a risk to both humans and  
316 animals. Maximum values for Cu and Zn were found in onion roots (up to 80 mg kg<sup>-1</sup>  
317 Cu and 135 mg kg<sup>-1</sup> Zn, Table 4), although that was not the case for lettuce roots. In  
318 case of Cu, concentrations in onion leaves (maximum of 30 mg kg<sup>-1</sup>, above normal  
319 levels in plants) were higher than in bulbs in contrast to data for Zn. The dilution effect

320 of trace elements (bulbs of mature vs young onions) was also observed for Cu and Zn.  
321 In general, concentrations in onions were above normal concentrations found in this  
322 plant according to Mohamed et al. 2003 (2.81 mg kg<sup>-1</sup> for Cu and 17.6 mg kg<sup>-1</sup> for Zn)  
323 and Kabata-Pendias and Pendias 1999, 2001 (Table 4). Likewise the contents of Cu and  
324 Zn in lettuce and chard were within the normal range reported in plants, although for Cu  
325 these contents were higher than contents in the same species from non-polluted sites  
326 (lettuce 3.81 mg kg<sup>-1</sup> Cu and 81.5 mg kg<sup>-1</sup> Zn and chard 5.49 mg kg<sup>-1</sup> Cu and 150 mg  
327 kg<sup>-1</sup> Zn, Pillay and Jonnalagadda, 2007). The maximum concentrations of Cu and Zn in  
328 lettuce were detected in the inner leaves (Figure 4), so the distribution patterns of both  
329 micronutrients were different from those found for As and Pb.

330         The relatively high concentrations of Cu in onion leaves, lettuce and chard could  
331 have been influenced by foliar absorption of Cu-based agrochemicals, which are applied  
332 by the local farmers. Plants growing on Cu-polluted sites tend to accumulate increased  
333 amounts of this metal, especially near industrial areas and in soils treated with Cu-  
334 bearing pesticides (Kabata-Pendias and Mukherjee, 2007).

335         Elemental concentrations in different organs of lemon trees are shown in Figure  
336 5. Arsenic and Pb reached the highest contents in leaves (although within the normal  
337 range in plants) whereas concentrations of these potentially toxic elements in different  
338 parts of the fruit were very low, especially in seed (0.1 mg kg<sup>-1</sup>). Both trace elements in  
339 lemon leaves and fruits followed the same pattern found in other trees growing on  
340 polluted soils: high content in leaves and much lower in fruits and seeds (Madejón et al.,  
341 2006). Ernst et al. (1992) reported that plants do not totally exclude trace elements (with  
342 no essential function in the plant) from their reproductive organs although the content in  
343 fruits and seed is usually very low. On the other hand, Cu and Zn showed a similar

344 distribution pattern, with high concentrations in leaves and in seeds when compared to  
345 As and Pb.

346 Finally, the concentrations of As and Pb detected in whole potatoes, peel and  
347 tuber, were negligible ( $0.1 \text{ mg kg}^{-1}$ ), whereas the concentrations of Cu in tubers (up to  
348  $12.0 \text{ mg kg}^{-1}$ ) and Zn (up to  $28.8 \text{ mg kg}^{-1}$ ) were greater than those compiled by Kabata-  
349 Pendias and Pendias (1999, 2001) as possible background values.

350

#### 351 ***4.5 Trace element correlations between soil and plants and soil-plant transfer*** 352 ***coefficients***

353 Significant correlations between soils and plants (taking together all species and  
354 plant parts) were found for As with water and  $\text{CaCl}_2$  extraction ( $r_{\text{H}_2\text{O}} = 0.45$  and  $r_{\text{CaCl}_2} =$   
355  $0.46$ ). Correlation coefficients for each species could be only calculated for young  
356 onions, lettuce and chard. There were significant correlations between As in lettuce  
357 leaves and EDTA extractable As ( $r = 0.86$ ) and chard leaves and pH ( $r = -0.87$ ). For Pb,  
358 significant negative correlations were found between lettuce and chard and total soil Pb  
359 content; this may be indicative of a non-soil pollution source (wind-blown dust).

360 Significant soil-plant correlations for Cu were found for lettuce and chard and for total,  
361 EDTA and  $\text{CaCl}_2$  extraction concentrations, but these are less robust due to the low  
362 number of samples.

363 The soil-plant transfer coefficients were calculated for As, Cu, Pb and Zn (Table  
364 5). The transfer factors may depend not only on plant species, but also on the element  
365 concentration and bioavailability in soil (Huang et al., 2006). The values calculated  
366 were, in general, very low, especially in the case of As and Pb, and in the different  
367 tissues of lemon and potato that were analyzed. It has been observed that transfer factors  
368 tended to decrease with increasing soil concentrations (Alan et al., 2003). In the case of

369 onions, the high As TC in roots, (TC= 3.03) coupled with the greatly reduced transfer to  
370 aerial parts (TC= 0.03) and bulb (TC= 0.001) would indicate that the As accumulated in  
371 root tissues. In case of Cu and Zn, TC were always low, with maximum values found in  
372 onion roots (TC= 0.15 for Cu and 0.18 for Zn). Similar results were found for other  
373 plants growing in multi-element polluted soils by Madejón et al. (2007). The present  
374 results imply that there is a low transfer of pollutants from these soils to primary  
375 producers.

376

## 377 **5. Conclusions**

378 There are several conclusions that can be drawn from this study:

379 First: There is limited soil-plant transfer of potentially hazardous trace elements (As,  
380 Cu, Pb, Zn) to selected vegetables and fruit cultivated on a mine-polluted soil. One  
381 reason may be the current management practices employed on the agricultural small  
382 holdings under investigation such as regular inputs of organic matter and lime to  
383 maintain soil pH close to neutrality.

384 Second: The vegetables investigated are typical of those grown in this region, and no  
385 edible part showed element accumulation that could be considered as a significant risk  
386 to human health, based on statutory limits for food crops. The vegetables showed  
387 different patterns of trace element accumulation: highest concentrations in non-edible  
388 parts of onions and lowest concentrations in potatoes tubers.

389 Third: The data from this survey is a clear illustration of the use of simple, routine  
390 agronomic procedures as a means of a) cultivating crops on potentially phytotoxic soils  
391 and b) reducing the concentrations of potentially toxic elements (As, Pb) in their edible  
392 tissues. Routine application of a combination of livestock manure and agricultural lime  
393 has created a soil environment where the labile pools of As and Pb are very low, in



394 contrast to the very elevated total content of these elements in the soil. In addition, this  
395 treatment has regulated plant uptake of both Cu and Zn, reducing potential phytotoxicity  
396 whilst still providing a source of trace concentrations of both micronutrients to maintain  
397 normal growth. The *ad hoc* treatments applied by the smallholders appear to have  
398 successfully contributed to the safe use of the polluted soils where it can sustain crop  
399 production over an extended period of time. There are clear lessons for future  
400 development of field scale remediation of soils with mixed metal/metalloid pollution  
401 from the successful and sustainable outcome of the ‘in situ’ treatments applied to these  
402 highly polluted soils by the smallholders who use traditional practices to farm these  
403 sites. However it is important to consider that the success obtained in this case cannot  
404 be generally extended to other situations where soil and climate conditions, as well as  
405 type of crop species, are different.

406

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537

538 **FIGURE AND TABLE CAPTIONS**

539 **Figure 1.** Panoramic view (a) of the Tharsis mining area (photo courtesy: E. Romero)  
540 showing the location of the agricultural plots selected for this study (b,c).

541 **Figure 2.** Percentages of elements extracted with (a) deionised water, (b) CaCl<sub>2</sub>, and (c)  
542 EDTA in soil growing young onion (C1, C3, C4, C5), ripe onion (C2), lettuce (L1, L2,  
543 L3), chard (A1, A2, A3), lemon (S5), and potato (S12).

544 **Figure 3.** Distribution of trace elements (mean values in mg kg<sup>-1</sup>) in different parts of  
545 the onion.

546 **Figure 4.** Distribution of trace elements (mean values in mg kg<sup>-1</sup>) in different parts of  
547 the lettuce.

548 **Figure 5.** Distribution of trace elements (mean values in mg kg<sup>-1</sup>) in different parts of  
549 the lemon tree.

550

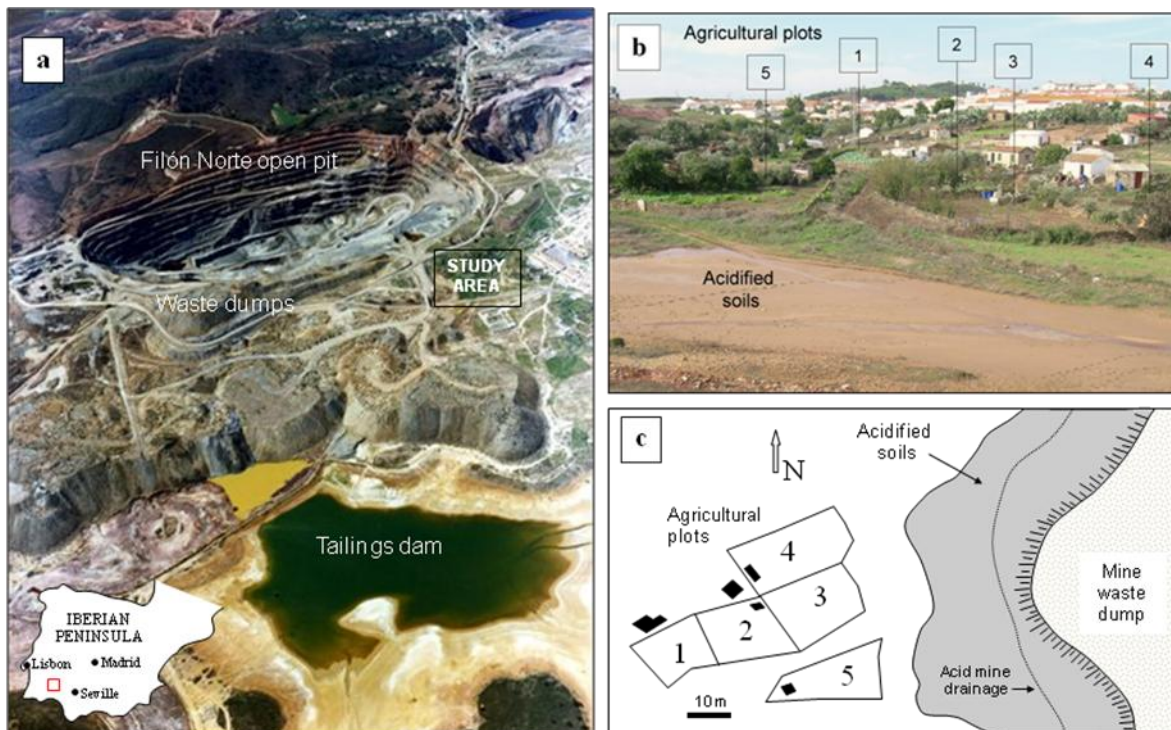
551 **Table 1.** Plant species, analyzed organs (number of samples in brackets) and their  
552 associated soil samples.

553 **Table 2.** Physico-chemical parameters of the soils where vegetables are cultivated.

554 **Table 3.** Mean and standard deviation of the total trace element concentrations and  
555 pollution load index (PLI) and trace elements extracted with deionized water, CaCl<sub>2</sub> and  
556 EDTA. All values are expressed in mg kg<sup>-1</sup>. Number of samples (n) appears in brackets

557 **Table 4.** Trace element content (mean and range values in mg kg<sup>-1</sup>) in different organs  
558 of vegetables growing in the agricultural plots.

559 **Table 5.** Transfer coefficients (TC) for As, Cu, Pb and Zn in the different organs of the  
560 studied vegetables.





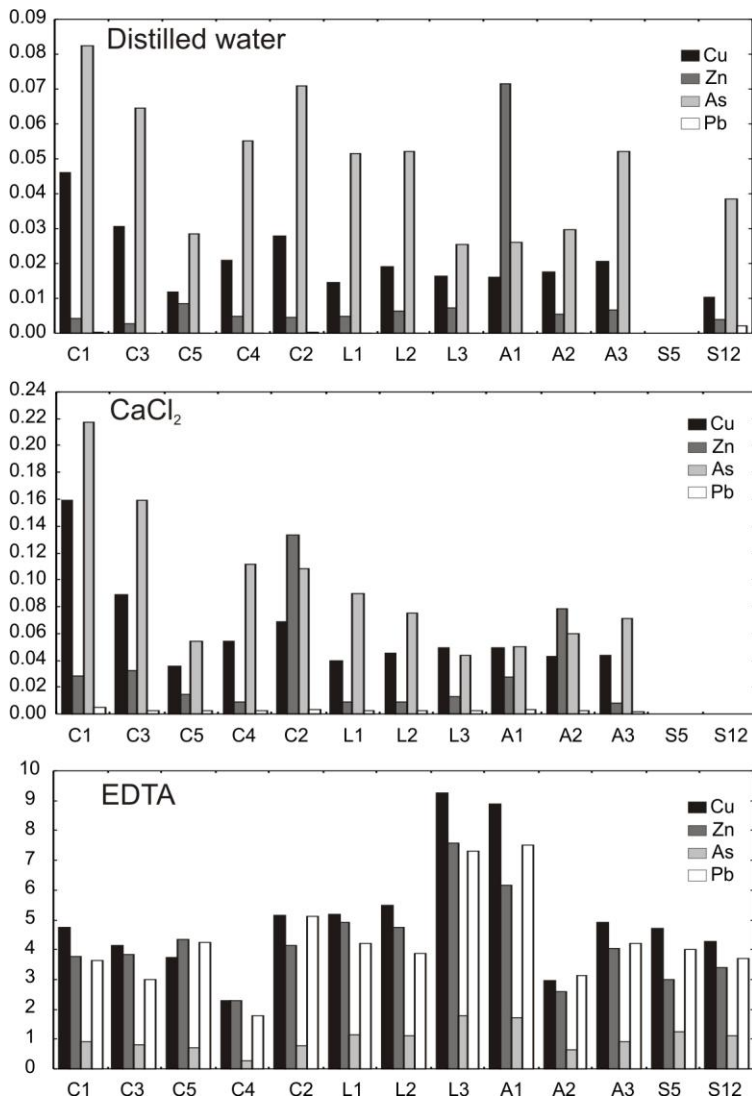


Figure 2

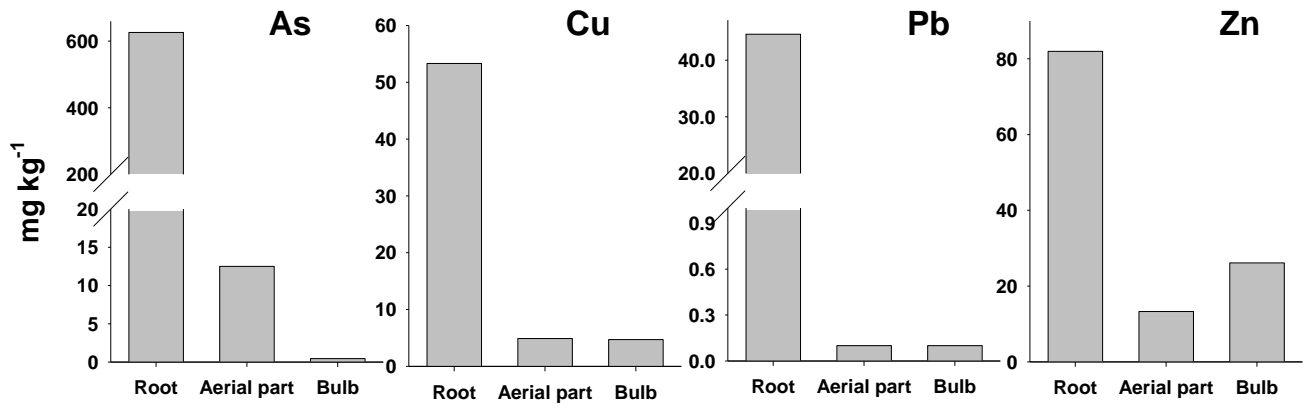


Figure 3

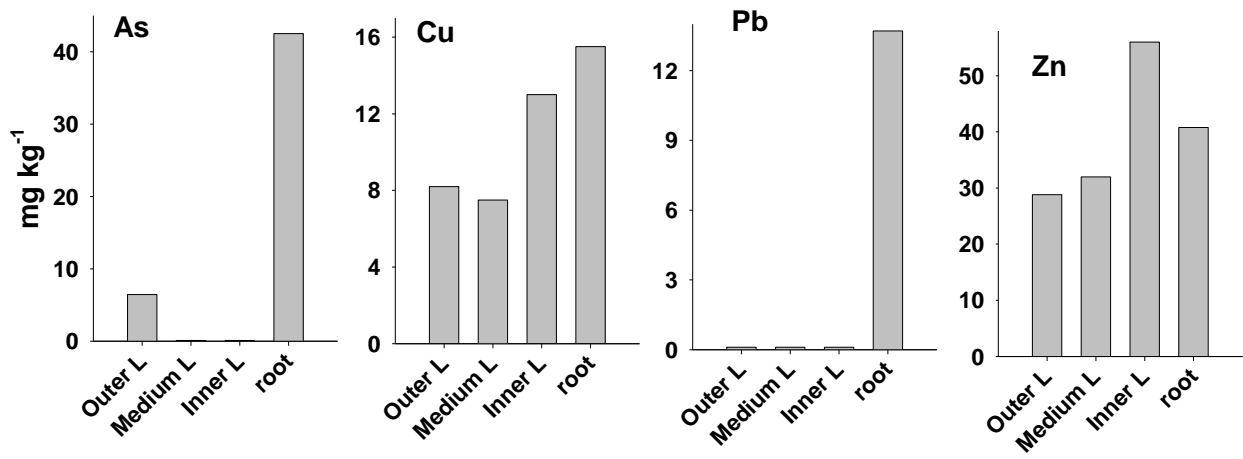


Figure 4

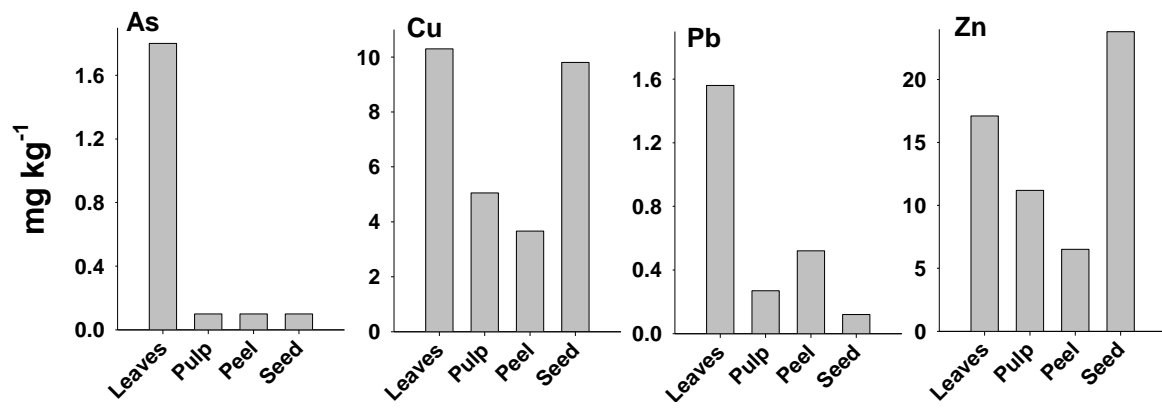


Figure 5

**Table 1.** Plant species, analyzed organs (number of samples in brackets) and their rhizospheric soil samples.

<b>Plant species</b>	<b>Analyzed organs</b>	<b>Rhizosphere samples</b>
Young onions	Bulb (7) and leaves (7)	C1, C3, C4,C5
Ripe onions	Bulb (6), leaves (6) and roots (3)	C2
Lettuce	Leaves (6) and roots (3) Inner, medium and outer leaves	L1, L2, L3
Chard	Leaves (6)	A1, A2, A3
Potato	Peel (5) and inner tissues (edible part) (5)	S12
Lemon	Leaves (3), peel (3), inner tissue (edible part) (3) and seeds (3)	S5

**Table 2..** Physico-chemical parameters of the soils where vegetables are cultivated.

Plot	Soil sample	Color (Munsell)	Coarse fragments (%)	Fine earths (<2 mm)			pH (H <sub>2</sub> O)	Eh (mV)	Electrical conductivity (mS cm <sup>-1</sup> )	Organic matter (%)	Carbonates(%)
				Sand (%)	Silt (%)	Clay (%)					
1	A1	10YR 5/3	15.3	14.2	76.8	9.0	7.5	491	0.20	13.7	9.0
1	L3	10YR 6/3	23.8	17.8	76.0	6.2	7.4	498	0.15	14.9	14.0
1	C5	10YR 6/3	19.3	30.0	57.7	4.3	7.1	461	0.28	14.8	8.1
2	A2	10YR 6/3	30.2	15.0	76.5	8.5	7.3	487	0.22	9.9	7.3
2	A3	10YR 6/3	20.0	18.2	75.3	6.5	7.1	486	0.14	11.8	11.8
2	L1	10YR 5/3	18.6	19.9	74.1	6.0	7.5	494	0.13	11.9	12.9
2	L2	10YR 6/3	24.8	20.5	73.6	5.9	7.1	505	0.20	13.9	10.1
2	C2	10YR 5/2	21.8	23.8	68.6	7.6	7.1	485	0.63	11.9	5.6
3	C1	10YR 5/3	16.1	18.3	74.4	7.3	8.2	457	0.43	12.8	9.3
3	C3	10YR 5/3	16.6	23.1	70.0	6.9	7.6	471	0.95	13.8	7.6
3	C4	10YR 6/2	18.5	28.8	71.0	7.2	7.5	468	0.56	11.8	8.7
4	S5	10YR 4/2	13.2	31.0	59.1	9.9	7.3	n.d.	0.69	10.4	n.d.
5	S12	10YR 5/3	9.9	33.1	59.0	7.9	7.5	n.d.	0.45	8.1	n.d.

**Table 3.** Mean and standard deviation of the total trace element concentrations and pollution load index (PLI) and trace elements extracted with deionised water, CaCl<sub>2</sub> and EDTA. All values are expressed in mg kg<sup>-1</sup>. Number of samples (n) appears in brackets.

<b>Total contents</b>	<b>As</b>	<b>Cu</b>	<b>Pb</b>	<b>Zn</b>
Plot 1 ( PLI 9.4)	203 ± 26.7	326 ± 51.4	864 ± 74.3	314 ± 25.1
Plot 2 (PLI 14.5)	332 ± 70.9	486 ± 84.6	1281 ± 241	499 ± 39.9
Plot 3 (PLI 18.6)	469 ± 85.1	603 ± 20.2	1715 ± 274	570 ± 15.7
Plot 4 (PLI 22.2)	575	692	2395	593
Plot 5 (PLI 22.8)	621	752	2260	589
<b>Water soluble</b>	<b>As</b>	<b>Cu</b>	<b>Pb</b>	<b>Zn</b>
Plot 1 (3)	0.054 ± 0.004	0.047 ± 0.002	0.001 ± 0.0001	0.097 ± 0.128
Plot 2 (5)	0.164 ± 0.035	0.094 ± 0.011	0.001 ± 0.0005	0.029 ± 0.007
Plot 3 (3)	0.309 ± 0.015	0.195 ± 0.069	0.003 ± 0.0011	0.023 ± 0.007
Plot 4 (1)	n.d.	n.d.	n.d.	n.d.
Plot 5 (1)	0.240	0.078	0.045	0.024
<b>CaCl<sub>2</sub>-extractable</b>	<b>As</b>	<b>Cu</b>	<b>Pb</b>	<b>Zn</b>
Plot 1 (3)	0.099 ± 0.007	0.144 ± 0.007	0.024 ± 0.005	0.058 ± 0.030
Plot 2 (5)	0.259 ± 0.040	0.227 ± 0.016	0.031 ± 0.003	0.225 ± 0.259
Plot 3 (3)	0.734 ± 0.116	0.602 ± 0.299	0.057 ± 0.016	0.131 ± 0.070
Plot 4 (1)	n.d.	n.d.	n.d.	n.d.
Plot 5 (1)	n.d.	n.d.	n.d.	n.d.
<b>EDTA-extractable</b>	<b>As</b>	<b>Cu</b>	<b>Pb</b>	<b>Zn</b>
Plot 1 (3)	2.95 ± 1.51	22.7 ± 7.22	55.8 ± 19.7	19.1 ± 5.56
Plot 2 (5)	3.06 ± 1.07	23.1 ± 6.70	51.7 ± 8.79	20.3 ± 4.43
Plot 3 (3)	2.88 ± 1.25	22.3 ± 7.31	46.4 ± 9.99	18.8 ± 4.89
Plot 4 (1)	7.18	32.6	95.7	17.7
Plot 5 (1)	6.86	32.1	83.9	20.1
Regional background*	25	32	38	76
Normal levels in agricultural soils**	n.d.	13	16	47
Threshold values for agricultural soils of Andalusia***	20	100	200	300

\* Galán et al., (2008); \*\* López-Arias and Grau-Corbí, (2004); \*\*\* Aguilar et al., (1999). All these values are referred to total concentration

**Table 4.** Trace elements contents (mean and range values) in different organs of vegetables growing in the agricultural plots under study.

<b>Plant species</b>	<b>Organ</b>	<b>As</b>	<b>Cu</b>	<b>Pb</b>	<b>Zn</b>
Ripe Onion (n=6)	Leaves	7.14 (0.75-15.2)	12.1 (3.5-31.4)	0.65 (0.10-1.92)	14.1 (12.8-16.2)
	Bulbs	0.31 (0.1-1.2)	4.15 (2.9-5.0)	0.33 (0.1-0.83)	23.9 (14.9-29.1)
	Roots	627 (267-905)	53.3 (39.5-80.5)	44.5 (35.9-51.8)	82.0 (43.2-135)
<i>Onion bulbs*</i>		-	4.0-6.0	1.1-2.0	22-32
Young Onion (n=6)	Leaves	9.56 (0.10-15.2)	20.4 (14.1-20.5)	3.44 (0.1-8.8)	24 (19.2-28.0)
	Bulbs	8.81 (1.95-23.5)	13.3 (9.0-18.0)	2.65 (3.01-8.3)	62.4 (34.9-101)
Lettuce (plot 1, n=3)	Leaves	1.00 (0.86-1.1)	13.5 (13.3-13.6)	0.76 (0.75-0.77)	64 (61.6-66.0)
Lettuce (plot 2, n=3)	Leaves	0.93 (0.2-1.6)	9.48 (7.6-11.7)	0.45 (0.35-0.65)	49 (37.6-60.3)
<i>Lettuce leaves*</i>		-	6.0-8.0	0.7-3.6	44-73
Chard (plot 1, n=3)	Leaves	0.45 (0.20-0.70)	14.8 (13.6-15.9)	0.90 (0.75-1.1)	43.9 (41.4-46.4)
Chard (plot 2, n=3)	Leaves	1.07 (0.77-1.30)	18.4 (16.0-19.9)	1.50 (1.2-1.76)	79.8 (54.2-110)
Potato (n=5)	Peel	0.1	11.5 (8.9-12.6)	0.1	21.1 (19.2-24.0)
	Tuber	0.1	10.3 (8.2-12.0)	0.1	25.0 (20.8-28.8)
<i>Potato tubers*</i>		-	3.0-6.6	0.5	10-26
* Normal levels in vegetables**		0.0001-0.46	0.04-2.4	0.0004-0.8	0.5-118
Statutory limits***		1.00	-	1.00	

\* Possible background values compiled by Kabata-Pendias and Mukherjee (2007); \*\*Szefer and Nriagu, (2007); \*\*\* Davis and White (1981)



**Table 5.** Transfer coefficients (TC) for As, Cu, Pb and Zn at the different organs of the studied vegetables.

<b>Vegetable</b>	<b>organ</b>	<b>As</b>	<b>Cu</b>	<b>Pb</b>	<b>Zn</b>
Ripe onion	Leaves (n=6)	0.034	0.034	0.001	0.03
	Bulbs (n=6)	0.001	0.011	0.0004	0.05
	Roots (n=3)	3.03	0.150	0.052	0.18
Young onion	Leaves (n=7)	0.020	0.03	0.002	0.04
	Bulbs (n=7)	0.017	0.02	0.001	0.11
Lettuce	Leaves (n=6)	0.003	0.03	0.0005	0.13
Chard	Leaves (n=6)	0.003	0.04	0.001	0.14
Lemon	Leaves (n=3)	0.003	0.015	0.0007	0.03
	Pulp (n=3)	0.0002	0.010	0.0001	0.02
	Peel (n=3)	0.0002	0.005	0.0002	0.01
	Seed (n=3)	0.0002	0.014	0.00005	0.04
Potato	Peel (n=5)	0.0002	0.015	0.00004	0.035
	Tuber (n=5)	0.0002	0.014	0.00004	0.04