- 1 Influence of high pressure homogenization and pasteurization on the
- 2 in vitro bioaccessibility of carotenoids and flavonoids in orange juice
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#### Abstract

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- Production of high-quality healthy foods through sustainable methodologies is an 18 urgent necessity. High pressure homogenization (HPH) is an interesting alternative to 19 20 obtain premium citrus juices, but its effects on bioactive compounds are unclear. There was studied the influence of HPH (150 MPa) and pasteurization (92°C for 30 s and 21 22 85°C for 15 s) processing on physicochemical properties and in vitro bioaccessibility 23 of carotenoids and flavonoids in orange juices. Regarding fresh juice, physicochemical properties of samples remained unchanged although cloudiness was improved by 24 homogenization. Pasteurization did not affect total carotenoids content and retinol 25 activity equivalents (RAE) of juices whereas homogenization yielded a significant 26 reduction (1.37 and 1.35-fold, respectively). Interestingly, particle size reduction from 27 homogenization drastically enhanced (about 5-fold) bioaccessibility of carotenoids 28 including hardly bioaccessible epoxycarotenoids, finding unaltered rates in pasteurized 29 30 samples. Bioaccessibility of flavonoids was constant in all cases.
- Results can promote HPH as an efficient option to obtain health-enhanced foods.

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**Keywords:** Citrus juices; carotenoids; health-promoting foods; high pressure homogenization (HPH); sustainable processing; phytonutrients bioaccessibility.

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#### Chemical compounds

- 38 -Z-Violaxanthin (PubChem CID: 6442428)
- 39 -Luteolin (PubChem CID: 5280445)
- 40 -Zeaxanthin (PubChem CID: 5280899)
- -Zeinoxanthin (PubChem CID: 5281234)

- 42 -β-Cryptoxanthin (PubChem CID: 5281235)
- 43 -α-Carotene (PubChem CID: 4369188)
- -β-Carotene (PubChem CID: 5280489)
- -Phytoene (PubChem CID: 5280784)
- -Phytofluene (PubChem CID: 6436722)
- 47 -Narirutin (PubChem CID: 442431)

## 1. Introduction

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49 Orange juice is the most consumed fruit juice worldwide greatly appreciated by consumers as a delicious natural source of vitamin C, fibre and nutritionally relevant 50 51 phytochemicals (mainly carotenoids and flavonoids). Epidemiological studies reported that a diet rich in fruit phytochemicals prevents apparition of degenerative diseases, 52 finding that in the particular case of citrus carotenoids and flavonoids such benefits 53 54 include the protection against oxidizing agents, inflammation and modulation of gene expression (Meléndez-Martínez, 2019; Tripoli, Guardia, Giammanco, Majo, & 55 Giammanco, 2007). Health effects of such phytochemicals depend on the amount 56 consumed and their bioavailability. Intrinsic and extrinsic factors such as the 57 metabolite profile of citrus varieties, human metabolism and food/human genetics, 58 59 interactions within and between food matrices and industrial processing greatly affect bioaccessibility of these bioactive compounds (Rodríguez-Concepcion et al., 2018). In 60 this line, bioaccessibility is considered a measure of potential bioavailability and can 61 62 be defined as the amount of a compound released from the food matrix during digestion and potentially available for absorption (Rodríguez-Concepcion et al., 2018). 63 Interest of in vitro digestion methods to assess the bioaccessibility of food 64 65 phytochemicals is undeniable since they can facilitate the understanding of complex interactions naturally occurring under in vivo conditions, facilitating the design of 66 commercial products with improved nutritional properties (Etcheverry, Grusak, & 67 Fleige, 2012). 68 Nowadays, consumers demand minimally altered and more natural products hence the 69 70 interest of the industry to implement innovative non-thermal stabilization technologies such as dynamic high pressure homogenization (HPH) (Martínez-Monteagudo, Yan, 71 & Balasubramaniam, 2017). Similarly, there is an increasing interest to implement 72

innovative and environmentally friendly technologies for the sustainable production of foods alternatively to traditional stabilization processes mainly ruled by thermal treatments (Pereira & Vicente, 2010). Compared to heat processing, a significant reduction of energy (about 15%) was stated for HPH preservation of milk (Valsasina et al., 2017). Moreover, homogenization was also proposed to valorise biomass from different sources such as milk whey and pistachio shell, among others (Marciniak, Suwal, Britten, Pouliot, & Doyen, 2018; Özbek, Fockink, Yanık, Göğüş, & Lukasik, 2018).

The effect of processing on bioaccessibility of carotenoids and flavonoids from foodstuffs seems to be intimately associated to the severity of treatments applied and the food matrix composition (Mapelli-Brahm, Stinco, Rodrigo, Zacarías, & Meléndez-Martínez, 2018; Stinco, Fernández-Vázquez, Escudero-Gilete, Heredia, Meléndez-Martínez, & Vicario, 2012). Regarding the influence exerted by minimally processing HPH treatment on in vitro bioaccessibility of carotenoids from foodstuffs, studies were mainly addressed to tomato and tomato-based products with dissenting results (Knockaert et al., 2012; Svelander et al., 2011). Disparate results can be understood by factors outside those induced by the matrix disruption caused by HPH such as the carotenoid type (Panozzo et al., 2013), rheological properties modified by processing (Leite, Augusto, & Cristianini, 2014), the chromoplast substructures and cell wall barriers (Palmero, Panozzo, Colle, Chigwedere, Hendrickx, & Van Loey, 2016). This issue was assessed in some beverages made from the combination of fruit juices with water, milk and soymilk suggesting how in vitro bioaccessibility of carotenoids was generally increased by HPH and High Intensity Pulsed Electric fields (HIPEF) compared to traditional thermal treatment (Rodríguez-Roque et al., 2016). To the best of our knowledge, no studies were carried out to evaluate the effect of HPH processing on *in vitro* bioaccessibility of carotenoids from orange juice.

Considering phenolics from orange juice, HPH processing does not seem to affect their *in vitro* bioaccessibility (He et al., 2016). Moreover, human studies suggested that *in vivo* bioavailability of orange flavanones was mainly ruled by individual factors rather than technological treatments (Tomás-Navarro et al., 2014).

Main objective of this study was to assess the effect of HPH and a traditional pasteurization on the *in vitro* bioaccessibility of carotenoids and flavonoids of Lane Late orange juices. Furthermore, influence of treatments on physicochemical properties of juices was also studied. Results achieved can promote the implementation of more sustainable methodologies by food industry to produce high-quality health-promoting products to cater for the new trends in the demands of consumers.

### 2. Experimental

#### 2.1. Materials

Analytical-grade dichloromethane, acetone and methanol were from Carlo Erba (CARLO ERBA Reagents S.r.l. Milan, Italy). HPLC-grade acetonitrile (ACN) and formic acid (FA) were from Scharlab (Scharlab S.L., Barcelona, Spain). Methanol (MeOH) and methyl *tert*-butyl ether (MTBE) of HPLC-grade were from Merck (Merck KGaA, Darmstadt, Germany). Ultra-pure water was produced by a NANOpureDiamond system (Barnsted Inc.). Mineral salts (KCl, NaCl), sodium bicarbonate, monopotassium-phosphate, magnesium chloride hexahydrate, chlorhydric acid, ascorbic acid (L-AAH, 99% purity grade), ammonium formate, pepsin (porcine gastric mucosa), pancreatin (porcine pancreas), bile salt, α-carotene, β-carotene, β-carotene,

cryptoxanthin, lutein and zeaxanthin were from Sigma-Aldrich (Merck KGaA, Darmstadt, Germany). Phytoene and Phytofluene standards were extracted from tomato fruits and purified following the procedure of Mapelli-Brahm, Corte-Real, Meléndez-Martínez and Bohn (2017).

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#### 2.2. Preparation of orange juice assayed

Treatments assayed were decided on the basis of results from previous studies 129 addressing the obtaining of high quality orange juices with extended shelf life through 130 HPH (Carbonell, Navarro, Izquierdo, & Sentandreu, 2013) and pasteurization 131 132 (Sentandreu, Carbonell, Carbonell, & Izquierdo, 2005) processing. Lane Late (Citrus sinensis L. Osb.) orange fruits were harvested in Lliria (Valencia, 133 Spain), collected in May 2017, washed with tap water, drained, sized and finally 134 135 squeezed in an Exzel industrial extractor from Luzzysa (Luzzysa, El Puig, Valencia, Spain). A paddle finisher (model EPF 06, Luzzysa, El Puig, Valencia, Spain) with 136 137 holes of 0.4 mm Ø sieved raw juice to obtain 600 L of freshly prepared juice (FJ) that was divided into four different batches (Fig. S1). One batch was homogenized at 150 138 MPa in a continuous high pressure system (NS3015H model, GEA Niro Soavi S.p.A., 139 Parma, Italy) with a digital thermometer (probe PT-100, range -10/110 °C) installed in 140 141 the outlet section to monitor the temperatures reached by the samples. Tap water was 142 used to warm up the HPH system for 30 min operating at 150 MPa at a flow rate of 100 L/h. Tempered juice substituted water for 5 min to reach the stationary conditions 143 of 15 s of residence time and 68 °C at the outlet of the HPH device. Then, the juice 144 was quickly cooled at 7-9 °C in a plate heat exchanger (Junior model, APV Ibérica 145 S.A., Madrid, Spain) to obtain the H150 sample. A second batch was centrifuged in a 146 Westfalia system (model SAOH 205, GEA Westfalia, Nuremberg, Germany) to 147

separate low pulp (serum) and pulpy (Pu) fractions. The pulpy fraction was 148 149 homogenized at 150 MPa under the same conditions described above and blended with serum in the appropriate 85/15 proportion to reconstitute the juice that was finally 150 pasteurized at 85 °C for 15 s to obtain sample CHPuRP8515. Relevance of the 151 CHPuRP8515 sample concerns the capacity of the production line of a juice plant in 152 where the flow rate (volume/time) of reconstituted HPH juice is much higher than the 153 154 achieved by homogenization of the whole juice (H150). Finally, FJ was pasteurized under strong (92 °C for 30 s) and mild (85 °C for 15 s) 155 thermal conditions to yield P9230 and P8515 samples, respectively. 156

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#### 2.3. Physicochemical parameters

- 159 2.3.1. Total soluble solids and acidity analyses
- 160 Total soluble solids were determined as <sup>o</sup>Brix with a digital refractometer (Pal-1,
- Atago Co. Ltd, Tokyo, Japan) and total acidity was assessed by titration with 0.1 N 161
- 162 NaOH (using phenolphthalein 0.1% in 20% EtOH) and expressed as % citric acid
- 163 (w/v). The Brix to acid ratio and pH of fresh juice were 13.8 and 14.9, respectively

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- 165 2.3.2. Suspended pulp and cloudiness assessment
- 166 Determinations followed the methodology proposed by Cheng (2002) and samples
- were analysed in triplicate expressing the results as average (mean  $\pm$  Std. dev.). 167

- 2.3.3. HPLC-DAD analysis of ascorbic acid 169
- It was determined using a liquid chromatograph-diode array (HPLC-DAD) detector 170
- Agilent 1100 (Agilent Technologies, Palo Alto, CA, USA) equipped with a binary 171
- 172 pump, vacuum degasser and temperature-controlled autosampler. Separation was

achieved on a 250 mm x 4.6 mm i.d., 5 µm Phenomenex Luna (2) C18 column from 173 174 Phenomenex Inc. (Torrance, CA, USA) with the following conditions: solvent A, water/ACN (95:5) containing 0.1% FA and ammonium formate (25 mM); solvent B, 175 ACN; initially 0% B hold for 10 min, 90% B at 10.5 min and purging up to 20 min, 176 column equilibration with 0% B from 20.5 min up to 35 min; injection volume, 3 µL; 177 flow rate, 1 mL/min; autosampler and column temperatures were 10 °C and 23 °C, 178 respectively; DAD detection wavelength,  $\lambda = 260$  nm. Quantification of L-AAH in 179 samples was through external calibration using a preliminary calibration curve of a 180 commercial standard solution with a correlation coefficient ( $\mathbb{R}^2$ ) of 0.999. 181

Samples were analysed in triplicate and results expressed as average (mean  $\pm$  Std.

183 dev.).

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2.3.4. Pectin methyl esterase (PME) activity assay

186 It was determined following the methodology proposed by Carbonell, Navarro,

Izquierdo and Sentandreu (2013) and samples were analysed in triplicate expressing

results as average (mean  $\pm$  Std. dev.).

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2.3.5 Particle size distribution analysis

191 It was determined using a Malvern Mastersizer 2000 system (Malvern Instruments

Limited, Worcestershire, UK) with a short wavelength blue light source with forward

and backscatter detection for enhanced performance in the 0.02-2000 lm range.

Values at 1.73 and 1.33 were used as refractive indexes of the juice and the dispersant

(water), respectively, using 0.1 for absorption index of cloud particles (Corredig, Kerr,

& Wicker, 2001). The equivalent volume mean diameter  $D_{[4,3]}$  was calculated as

197 follows:

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$$D_{[4,3]} = \sum_{i} n_{i} d_{i}^{4} / \sum_{i} n_{i} d_{i}^{3}$$
 (1)

where  $n_i$  is the number of particles of diameter  $d_i$ .

Samples were analysed in triplicate and results expressed as average (mean  $\pm$  Std.

201 dev.).

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203 2.3.6. Colour measurement

204 It was measured with a PC controlled Labscan II Hunter colourimeter (Hunter

205 Associates Lab., Reston, Vi, USA) following the indications from Cerdán-Calero,

Izquierdo and Sentandreu (2013). An optical glass cell (3.8 cm high and 6 cm of

diameter) containing a 3.5 cm thick layer of sample was covered with the white

standard plate (X 78.50; Y 83.32; Z 87.94) to determine diffused reflected light from

the cell bottom using a 13 mm diaphragm aperture.

210 The colour parameters of the CIELAB space L\*, a\* and b\* were obtained. L\* is an

estimation of the lightness. a\* takes positive values for reddish colours and negative

values for the greenish ones. b\* takes positive values for yellowish colours and

213 negative values for the bluish ones. Chroma (C\*<sub>ab</sub>) and hue-angle (h<sub>ab</sub>) are calculated

from a\* and b\* and are considered the quantitative and qualitative attribute of

colourfulness, respectively. The illuminant D<sub>65</sub> and a 10° angle of vision were

considered The total colour difference among samples assayed was calculated as

217 follows:

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$$\Delta E^*_{ab} = [(\Delta L^*)^2 + (\Delta a^*)^2 + (\Delta b^*)^2]^{1/2}$$
 (2)

Samples were analyzed in triplicate and results expressed as average (mean  $\pm$  Std.

220 dev.).

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#### 2.4. Simulated in vitro digestion of juices assayed

223 It was followed the methodology originally proposed by Minekus et al. (2014) with

minor modifications implemented by Stinco et al.(2019) skipping the mastication step.

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- 2.4.1. Gastric phase
- 227 Five mL of samples, 3.7 mL of simulated gastric fluid (SGF) and 2.5 μL of
- 228 CaCl<sub>2</sub>(H<sub>2</sub>O)<sub>2</sub> (588 g/L, w/v) were mixed into centrifuge tubes. The pH was adjusted to
- $3 \pm 0.1$  by adding 0.1 M HCl. Then, 500  $\mu$ L of porcine pepsin solution (40 mg/mL in
- SGF) was incorporated into the mixture. The final volume was adjusted to 10 mL with
- water and it was incubated in a shaker Max Q5000 (Thermofisher scientific Inc.,
- Waltham, MA) at 37 °C at 150 rpm for 2 h.

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- 234 *2.4.2. Duodenal phase*
- To stop the gastric digestion, the mixtures were placed in an ice bath. Subsequently, 7
- mL of simulated duodenal fluid (SDF) and 20 μL of CaCl<sub>2</sub>(H<sub>2</sub>O)<sub>2</sub> (0.3 M) were added.
- 237 After adjusting pH to 7, additional volumes of pancreatin (1 mL) and porcine bile
- extract in SDF (1 mL) were added. The final volume was made up to 20 mL with
- water to reach final concentrations of 390 mg/L and 2.1 g/mL of pancreatin and bile
- extract respectively in the reaction mixture. Samples were incubated for 2 h at 37 °C in
- 241 the shaker (Max Q5000 shaker, Thermofisher scientific Inc., Waltham, MA) at 150
- 242 rpm to mimicking the intestinal digestion. Once completed the digestion, digests were
- centrifuged for 20 min at 3900 x g at 4 °C in an Allegra X-12R centrifuge (Beckman
- Coulter, USA). The supernatant was filtered through a 0.22 μm nylon membrane filter
- 245 (Agilent Technologies, USA) and the aqueous micellar phase were stored at -20 °C in
- a nitrogen atmosphere until extraction.

#### 2.5. Calculation of bioaccessibility

249 It was calculated as percentage (%) of carotenoids or flavonoids as follows:

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$$\%Bioaccessibility_{BC} = \frac{M_d}{M_s} \times 100$$
 (3)

- in where BC was the considered bioactive compound,  $M_d$  was the mg of BC in the
- 252 digestate and M<sub>s</sub> was the mg of BC in the sample. In the case of carotenoids, the
- 253 micellar fraction of the digestate was considered.

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# 2.6. Carotenoid analysis by HPLC-DAD assay

- 256 The sampling protocol (extraction from juices and micelles followed by
- saponification) and carotenoid determination by HPLC-DAD analysis followed the
- 258 methodology described by Stinco, Fernández-Vázquez, Escudero-Gilete, Heredia,
- 259 Meléndez-Martínez and Vicario, (2012). Standards were used for identification and
- external calibration addressing quantification achieving a correlation coefficient  $(R^2)$
- of 0.999 in all cases. Total carotenoid contents were determined as the sum of
- 262 individual pigments.
- 263 The vitamin A activity of juices was expressed in terms of retinol activity equivalents
- 264 (RAE) to 1 L of sample (Institute of Medicine, 2001) and it was calculated as follows:

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$$RAE = \frac{\mu g \ \beta \ carotene}{12} + \frac{\mu g \ \beta \ cryptoxanthin + \mu g \ \alpha \ carotene}{24}$$
 (4)

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#### 2.7. Flavonoid analysis by HPLC-DAD assay

- 268 It was followed the methodology proposed by Stinco et al. (2015). Before HPLC
- analysis, undigested juice and digestate were centrifuged at 18000 g for 15 min at 4 °C
- and subsequently filtered through a 0.22-μm pore size membrane filter.

Standards were used for identification and external calibration addressing quantification achieving a correlation coefficient (R<sup>2</sup>) of 0.999 in all cases. Total flavonoid contents were calculated as the sum of individual phenolics.

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#### 2.8. Statistical Analysis.

Results from flavonoid and carotenoid analyses were given as mean and standard deviation of six independent determinations. One-way analysis of variance (ANOVA) was used to compare the means. Statistically significant differences (p<0.05) were determined using the Tukey's multiple comparison procedure. Correlations between particle size parameters and bioaccessibility of phytochemicals assayed were assessed by simple regressions. Statistical analyses were performed with Statistica v.8.0 software.

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#### 3. Results and discussion

#### 3.1. Physicochemical parameters 285

Table S1 summarizes physicochemical attributes of the juices assayed. Interestingly, 287 there were not significant differences in the ascorbic acid levels across samples. Residual PME activity of H150 sample was about 40% (0.614 nkat/mL) than initially showed by FJ juice (1.501 nkat/mL) whereas the rest of samples studied achieved 289 approximately a 25-fold reduction (0.045-0.067 nkat/mL). The decrease in the activity 290 291 of PME in orange juice after treatment by HPH and pasteurization was previously 292 observed in other studies (Carbonell, Navarro, Izquierdo, & Sentandreu, 2013) (Table 293 S1). In this line, the transmittance of the samples decreased as a result of the 294 treatments indicating how turbidity can be preserved by the particle size reduction 295 (Fig. S2) caused by the stabilization treatments as previously suggested by Carbonell, Navarro, Izquierdo and Sentandreu (2013). Thus, while the transmittance of the fresh sample was approximately 13%, those of P9230, P8515 and H150 ranged from 9.5 to 10% and the one exhibited by the CHPuRP8515 sample was by far the lowest ( $\sim$ 3.3%). This can be understood considering the average diameter of particles (D<sub>[4,3]</sub>, Table S1) that was significantly reduced in all treated samples compared to FJ but with special relevance in homogenized juices that showed a 10-fold reduction. Moreover, homogenization yielded samples with the lower mean surface  $(D_{[3,2]})$  and higher specific surface areas (SSA) as consequence of having a less coarse and more compactable pulpy material that can affect the distribution of their carotenoid and fiber contents. That is, fresh and pasteurized juices exhibited higher mean particle diameters and so they may tend to settle down more rapidly during storage than in the H150 sample as stated by previous results (Leite, Augusto, & Cristianini, 2014). Fig. S3 illustrates the representation of a\* and b\* colour parameters (Table S1) of juices assayed evidencing their clear differentiation (p<0.05) according to the technological treatments applied. The FJ was darker (lower L\*) and more reddish (higher a\*) than the treated samples. Moreover, it was found how HPH at 150 MPa significantly increased L\* and b\* but decreased a\* values in the same line than previous results found in orange juices (Velázquez-Estrada et al., 2019). Thus, appearance of H150 sample was lighter, less reddish and more yellowish. In general, colour parameters of CHPuRP8515 sample were similar to those of the H150 juice. On the other hand, although the colour parameters of the P9230 and P8515 samples were in general significantly different compared to the FJ, the colour of such samples were closer to the control juice than to the CHPuRP8515 and H150 samples (Table S1). Overall, the highest colour differences (>7 CIELAB units) were observed between CHPuRP8515 and H150 samples and the FJ, P9230 and P8515 samples. Since 2.8

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CIELAB unit has been proposed as a threshold over which untrained people with normal vision are expected to differentiate colour of orange juices visually (Fernández-Vázquez, Stinco, Hernanz, Heredia, & Vicario, 2013), it can be concluded that consumers with normal vision could differentiate some of the juices on the basis of their colours.

Variation of colour attributes of samples can be explained to a large extent by the alteration of their carotenoids (pulp-related phytochemicals) profile by processing and changes in the size and structure of pulp particles. In this context, the apparition of aggregative effects from the increased compressibility of smaller particles in HPH juices has been suggested elsewhere (Arena, Fallico, & Maccarone, 2000), which could help explain their lower observable pulp content (Table S1). According to Cerdán-Calero, Izquierdo and Sentandreu (2013), the difference between the color of a fresh juice and that of a juice treated by HPH at 150 MPa could be reduced by

#### 3.2. Carotenoid content of juices

including a pre-homogenization at 20 MPa in the process.

Table 1 shows levels of carotenoids before and after the digestion process. Theoretical vitamin A activity was expressed as Retinol Activity Equivalents (RAE). An ANOVA analysis was carried out to evaluate whether the treatments led to significant changes. As expected, the highest level of total carotenoids was detected in FJ sample finding lower results, but not significantly different, in pasteurized juices according to the temperature assayed. These results are in agreement with those obtained by other authors (Aschoff et al., 2015), that reported a slight albeit not significant decrease in total carotenoid levels during the processing of fresh oranges into pasteurized orange juices.

Homogenization decreased the total carotenoid content and RAE significantly (1.37 and 1.35-fold, that is, 27% and 26%, respectively) compared to FJ. Similarly, Suárez-Jacobo, Rüfer, Gervilla, Guamis, Roig-Sagués and Saldo (2011) described a decrease in the vitamin A content of 18%, 20% and 33% when increasing the homogenization pressure at 100, 200 and 300 MPa. These results suggested that degradation of carotenoids could be mainly due to the cellular disruption promoted by HPH treatments that enhanced their exposure to gastric enzymes and acidic pH, decreasing their stability and favouring their oxidation (Meléndez-Martínez, Vicario, & Heredia, 2007). As expected, such negative effects were more pronounced in CHPuRP8515 sample since both HPH and pasteurization treatments were applied and led to a reduction of 2.63- and 2.12-fold of its total carotenoid content and RAE, respectively.

#### 3.3. Bioaccessibility of carotenoids from juices

Fig. 1 and Table 2 summarize results achieved regarding bioaccessibility of juice carotenoids in percentage and the theoretical amount of carotenoids that could be incorporated into micelles after the ingestion of 1L of juice (referred to as carotenoid bioaccessible content, CBC), respectively. Comparable results were found for FJ and pasteurized juices with the exception of β-carotene, (Z)-antheraxanthin isomers and (Z)-luteoxanthin isomers that had a higher bioaccessibility by thermal treatment (Fig. 1). These findings followed, in general, previous results about bioaccessibility of carotenoids from orange juices and discrepancies can be explained through the different experimental conditions considered (Mapelli-Brahm et al., 2018; Stinco, Fernández-Vázquez, Escudero-Gilete, Heredia, Meléndez-Martínez, & Vicario, 2012). 

Very interestingly, homogenization in H150 sample had a very significant (p<0.05) positive impact on bioaccessibility (about 80%) and CBC (from 3- to 5-fold) of carotenoids lutein, zeaxanthin, β-cryptoxanthin, α-carotene, β-carotene, lycopene, phytoene and phytofluene compared to FJ (Fig. 1 and Table 2). In this line, RAE activity of H150 sample had a 4.4-fold increase and suggested the relevance of HPH processing to improve the vitamin A status of populations with high consumption of orange juice. Epoxycarotenoids were also found in the micellar fractions from digested H150 sample having a 2- (violaxanthin isomers) to 7.4-fold ((9Z)-violaxanthin + Zisomer of antheraxanthin) CBC enhancement with a 3.86-fold increasing of total carotenoid content compared to FJ (Table 2). Relevance of this finding deserves to be highlighted since bioavailability of epoxycarotenoids used to be considered negligible in humans (Meléndez-Martínez et al., 2017). Similarly, bioaccessibility and CBC of carotenoids in CHPuRP8515 sample were significantly (p<0.05) higher than in FJ and pasteurized juices, but lower than that of H150 juice as consequence of the complementary pasteurization treatment assayed (Fig. 1 and Table 2). These firstly reported findings in orange juice demonstrated the relevance of disruptive processes to enhance in vitro bioaccessibility of pulp-related phytochemicals such as carotenoids initially hypothesized by Sánchez-Moreno, Plaza, De Ancos, and Cano (2003) in the same food matrix. Since homogenization had a negative effect on carotenoids content from samples assayed (**Table 1**), it is plausible to confer betterment of bioaccessibility and CBC of carotenoids to changes induced by processing on size and structure of pulp particles of juices. Results suggested that particle size reduction featured by homogenization (Table S1) greatly increased SSA and boosted extractability of carotenoids. To assess correlations between particle size parameters ( $D_{[4,3]}$ ,  $D_{[3,2]}$  and SSA) and the

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bioaccessibility of carotenoids, simple linear regression analyses were carried out (Table S2). Good correlations between the particle size parameters and bioaccessibility of all carotenoids were found when all the samples were considered altogether. For the correlations between  $D_{[4,3]}$  and  $D_{[3,2]}$  and bioaccessibility, the correlation coefficients were negative and significant for all cases (r ranging from 0.62 to 0.98), while the correlations between SSA and bioaccessibility were positive and significant in all cases (r ranging from 0.55 to 0.86) except for bioaccessibility of αcarotene. According to the signs of the correlation coefficients, the lower the particle size, (ie. lower  $D_{[4,3]}$  and  $D_{[3,2]}$ ), the higher the bioaccessibility of carotenoids. On the contrary, the bioaccessibility was directly proportional to SSA. These results were to be expected, since  $D_{[4,3]}$  and  $D_{[3,2]}$  indicates the diameter of the average volume and the diameter of the average area of a particle, respectively, SSA is related to the density of each particle. Altogether, it can be argued that the reduction in the size of the particles and the subsequent increase in SSA facilitates the release of carotenoids during digestion. Thus, treatments having greater impact on particle size (CHPuRP8515 and H150) can confer better bioaccessibility to carotenoids than nondisruptive thermal treatments. As it can be observed in Fig. S2, processing of CHPuRP8515 sample yielded particles of lower size than that shown in H150 sample although, in general, the volume in percentage of particles in CHPuRP8515 juice was lower, explaining to a large extent the differences in bioaccessibility observed between HPH samples assayed. Results suggested that particle size of the food matrix plays a determinant role in the bioaccessibility in of carotenoids. Importantly, the net result of treatments on the actual amount of carotenoids that can be absorbed (referred to as CBC in this work) also depends on how such treatments affect carotenoid retention in juices. In other

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words, technological treatments that lead to an increased release of carotenoids during digestion can result in an actual increase or decrease CBC depending on how the treatments affected the total content of the carotenoids in the matrix.

# 3.4. Flavonoid content of juices

Table 3 shows the mean levels of flavonoids found in juices assayed. Results from H150 and CHPuMP8515 samples were significantly (*p*<0.05) lower (about 18%) than those reached by fresh and pasteurized juices. Controversial results were stated by Velázquez-Estrada et al. (2013) that observed an increase in the flavonoid content (hesperidin and kaempferol) of orange juices treated at 200 and 300 MPa. The total flavonoid content determined in P9230 and P8515 samples were not significantly affected, in agreement to previous results from Sentandreu, Navarro and Sendra (2007) reporting that thermal pasteurization at 90°C-30 s had a negligible effect on the phenolic content of citrus juices.

# 3.5. Bioaccessibility of flavonoids from juices

As shown by **Fig. 2**, bioaccessibility of flavonoids from the juices assayed was in the range of 3.1% to 11.2%, lower than the bioaccessibility of carotenoids. Overall, bioaccessibility of flavones (**Fig. 2A**) and flavanones (**Fig. 2B**) were not significantly affected by the pasteurization treatments. In the same way, HPH treatment at 150 MPa did not induce significant differences in the bioaccessibility of flavonoids. These results were in agreement with those reported by He et al. (2016) who found that the bioaccessibility of phenolic compounds was not affected in orange juices treated with HPH relative to the untreated juice. It was noticeable how sample CHPuMP8515 showed an increased bioaccessibility of total flavones (**Fig. 2A**) and flavanones (**Fig.** 

2B) of 2.5 and 3.3%, respectively, compared to FJ. Bioaccessibility of individual flavonoids were significantly higher than in pasteurized and fresh juices. Findings suggested that after pulp homogenization, these compounds were not protected by cell wall components that could improve extractability of flavonoids, and therefore, their bioaccessibility. Several studies have reported that the processing of plant foods can influence the bioaccessibility of phenolic compounds, mainly through changes in the cell wall structure (Dutra et al., 2017). Likewise, Martínez-Huélamo et al. (2015) suggested that mechanical and thermal treatments of tomato sauce increased bioaccessibility of phenolics. As far as flavonoid bioaccessible contents (FBC) (that is, the amount of flavonoid present in the digestates per liter of juice) the value corresponding to total flavonoids was not significantly affected by the pasteurization (Table 4). It was observed a slight, but not significant, increase in juices pasteurized at 92 °C. Regarding H150 sample, only vicenin-2 was significantly increased compared with untreated juices, as a result of which the bioaccessibility of total flavones was significantly higher too. However, CHPuRP8515 sample exhibited a significant increase in the bioaccessibility of flavanones narigin-d and nariturin compared to FJ. Bioaccessibility of total flavonoids in this juice, albeit not significantly different, was 1.32-fold higher compared to reference (Table 4).

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#### 4. Conclusions

Although HPH processing (150 MPa) altered colour of juices, no negative effects were found regarding their ascorbic acid content finding in addition, how the particle size reduction induced by homogenization improved cloudiness of samples. Moreover, diminution of the particle size drastically increased (from 4 to 5-fold) bioaccessibility

and RAE value of carotenoids from homogenized juices. Very interestingly, epoxycarotenoids, which are not perceived in human plasma or tissues, were readily detected in the micellar fraction of digested homogenized samples. Thus, there was evidenced the relevance of HPH processing to improve incorporation into organism of pulp-linked bioactive nutrients such as carotenoids from natural extracts.

In contrast to traditional pasteurization treatments, results of this study can help food industry to adopt greener HPH stabilization technologies for trading high-quality products with enhanced health properties that can satisfy current trends demanded by consumers. Reduction of costs from the use of energy saving homogenization alternatives in addition to their high output rate can favour implementation of HPH processing in a wide range of industrial applications.

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# **Declaration of competing interest**

Authors declare no conflict of interest.

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# Figure captions

- 670 Fig. 1. Bioaccessibility (%) of carotenoids (A) and epoxycarotenoids (B) from juices
- assayed: FJ, fresh juice; P9230, pasteurized juice at 92 °C 30s; P8515, pasteurized
- 672 juice at 85 °C 15s; H150, homogenized juice at 150 MPa; CHPuRP8515, centrifuged
- juice, pulp fraction homogenized at 150 MPa, reconstituted with the serum fraction
- and pasteurized at 85 °C 15s. Nomenclature of bioactive compounds: LUT, lutein;
- ZEA, zeaxanthin; ZEINO, zeinoxanthin; BCR, β-cryptoxanthin; ACAR, α-carotene;
- 676 BCAR, β-carotene; PT, phytoene; PF, phytofluene; VIO, violaxanthin; ANT,
- antheraxanthin; LUTEO, luteoxanthin; MUT, mutatoxanthin.
- Different letters indicate statistically significant differences (p<0.05). Number of
- 679 technical replicates, n=6.
- 680
- Fig. 2. Bioaccessibility (%) of flavones (A) and flavanones (B) from juices assayed:
- FJ, fresh juice; P9230, pasteurized juice at 92 °C 30s; P8515, pasteurized juice at 85
- 683 °C 15s; H150, homogenized juice at 150 MPa; CHPuRP8515, centrifuged juice, pulp
- 684 fraction homogenized at 150 MPa, reconstituted with the serum fraction and
- pasteurized at 85 °C 15s.
- Different letters indicate statistically significant differences (p<0.05). Number of
- 687 technical replicates, n=6.
- 688
- 689 Fig. S1. Scheme of the operations carried out to obtain orange juices studied: FJ,
- fresh juice; P9230, pasteurized juice at 92 °C 30s; P8515, pasteurized juice at 85 °C
- 691 15s; H150, homogenized juice at 150 MPa; CHPuRP8515, centrifuged juice, pulp
- 692 fraction homogenized at 150 MPa, reconstituted with the serum fraction and
- pasteurized at 85 °C 15s.

Fig. S2. Particle size distribution of the different samples assayed: FJ, fresh juice; P9230, pasteurized juice at 92 °C 30s; P8515, pasteurized juice at 85 °C 15s; H150, homogenized juice at 150 MPa; CHPuRP8515, centrifuged juice, pulp fraction homogenized at 150 MPa, reconstituted with the serum fraction and pasteurized at 85 °C 15s. Number of technical replicates, n=3.

**Fig. S3.** Representation of the colorimetric a\* and b\* parameters found in samples assayed: FJ, fresh juice; P9230, pasteurized juice at 92 °C 30s; P8515, pasteurized juice at 85 °C 15s; H150, homogenized juice at 150 MPa; CHPuRP8515, centrifuged juice, pulp fraction homogenized at 150 MPa, reconstituted with the serum fraction and pasteurized at 85 °C 15s. Number of technical replicates, n=3.

**Table 1.** Carotenoid levels (mg/L) and Retinol Activity Equivalents (RAE, in  $\mu$ g/L) found in juices assayed  $^{\Psi}$ .

Carotenoids	Orange juice samples						
Carotenolas	FJ	P9230	P8515	H150	CHPuMP8515		
(Z)-ANT isomers	1.65±0.17a	1.32±0.18ab	1.43±0.17ab	1.12±0.07b	0.59±0.08c		
all-(E)-VIO + (Z)-VIO isomers	0.35±0.05a	0.24±0.05b	0.28±0.03ab	0.22±0.03b	0.10±0.02c		
(Z)-LUTEO isomer	0.76±0.11a	0.76±0.12a	0.78±0.05a	0.64±0.06a	0.33±0.04b		
(9Z)-VIO + (Z)-ANT isomer	3.88±0.64a	3.22±0.57a	3.37±0.62a	2.80±0.30a	1.14±0.17b		
(Z)-LUTEO isomer	0.59±0.06a	0.45±0.06b	0.50±0.04ab	0.38±0.03bc	0.26±0.03c		
MUT- epimer A	0.48±0.06a	0.47±0.05a	0.50±0.01a	0.41±0.04a	0.27±0.01b		
LUT	0.51±0.06a	0.50±0.05a	0.53±0.01a	0.43±0.05a	0.28±0.01b		
MUT- epimer B	0.86±0.06a	0.78±0.11a	0.89±0.04a	0.76±0.02a	0.41±0.02b		
ZEA	1.00±0.06a	0.76±0.10bc	0.82±0.07b	0.62±0.05c	0.41±0.01d		
(9Z)- or (9´Z)-ANT	1.80±0.26a	1.43±0.26ab	1.58±0.21ab	1.21±0.15bc	0.68±0.03c		
ZEI	0.37±0.04a	0.29±0.05a	0.31±0.04a	0.28±0.02a	0.16±0.01b		
BCR	1.01±0.11a	0.86±0.12ab	0.93±0.06ab	0.71±0.06b	0.48±0.03c		
ACAR	0.12±0.01a	0.10±0.01ab	0.11±0.01ab	0.10±0.01b	0.09±0.01b		
BCAR	0.23±0.02a	0.20±0.03a	0.22±0.01a	0.18±0.01a	0.09±0.02b		
PF	0.09±0.01ab	0.08±0.01ac	0.09±0.01ab	0.10±0.01b	0.06±0.01c		
PT	0.57±0.06a	0.47±0.06ab	0.51±0.04ab	0.43±0.04b	0.15±0.02c		
Total Carotenoids	14.29±1.68a	11.92±1.83ab	12.84±1.29ab	10.38±0.83b	5.50±0.33c		
RAE*	66.50±6.06a	56.82±8.04ab	61.42±3.03ab	49.02±3.09b	31.37±3.07c		

 $<sup>^{\</sup>Psi}$ FJ, fresh juice; P9230, pasteurized juice at 92  $^{\Omega}$ C 30s; P8515, pasteurized juice at 85  $^{\Omega}$ C 15s; H20, homogenized juice at 20 MPa; H150, homogenized juice at 150 MPa; CHPuRP8515, centrifuged juice, pulp fraction homogenized at 150 MPa, reconstituted with the serum fraction and pasteurized at 85  $^{\Omega}$ C 15s.

Nomenclature used for carotenoids: ANT= (antheraxanthin), VIO = (violaxanthin), LUTEO = (luteoxanthin), MUT = (mutatoxanthin), LUT= (lutein), ZEA = (zeaxanthin), ZEINO = (zeinoxanthin), BCR = ( $\beta$ -cryptoxanthin), ACAR= ( $\alpha$ -carotene), BCAR= ( $\beta$ -carotene), PT= (phytoene), PF= (phytofluene)

<sup>\*</sup>Retinol activity equivalent: calculated as RAE ( $\mu$ g/L) = ( $\mu$ g of  $\beta$ -carotene)/12 + ( $\mu$ g  $\beta$ -cryptoxanthin +  $\mu$   $\alpha$ -carotene)/24.

**Table 2.** Carotenoid Bioaccesible Content (CBC, theoretical amount of carotenoids in mg that could be incorporated into micelles after the ingestion of 1L of juice) of juices assayed $^{\Psi}$ .

Country state	Carotenoid Bioaccesible Content						
Carotenoids	FJ	P9230	P8515	H150	CHPuMP8515		
(Z)-ANT isomers	0.33±0.03a	0.42±0.07a	0.40±0.03a	0.89±0.03b	0.43±0.07a		
all-(E)-VIO + (Z)-VIO isomers	0.09±0.01a	0.09±0.02a	0.11±0.01a	0.18±0.02b	0.08±0.01a		
(Z)-LUTEO isomer	0.15±0.02a	0.19±0.04a	0.16±0.01a	0.51±0.05b	0.22±0.02a		
(9Z)-VIO + (Z)-ANT isomer	0.24±0.06a	0.33±0.07a	0.23±0.06a	1.77±0.09b	0.78±0.10c		
(Z)-LUTEO isomer	0.06±0.01a	0.07±0.02ab	0.06±0.01a	0.23±0.03c	0.11±0.01b		
MUT- epimer A	0.10±0.01a	0.13±0.03a	0.11±0.01a	0.32±0.03b	0.15±0.01a		
LUT	0.11±0.01a	0.14±0.03ab	0.12±0.01a	0.37±0.02c	0.17±0.02b		
MUT- epimer B	0.23±0.01a	0.25±0.04a	0.25±0.04a	0.69±0.05b	0.34±0.01c		
ZEA	0.16±0.01a	0.19±0.06a	0.16±0.03a	0.49±0.02b	0.24±0.02a		
(9Z)- or (9´Z)-ANT	0.15±0.02a	0.17±0.06a	0.15±0.01a	0.77±0.02b	0.36±0.05c		
ZEI	0.04±0.01a	0.05±0.01a	0.04±0.01a	0.18±0.01b	0.09±0.01c		
BCR	0.13±0.02a	0.17±0.03a	0.15±0.01a	0.56±0.01b	0.25±0.03c		
ACAR	0.01±0.01a	0.02±0.01a	0.02±0.01a	0.06±0.01b	0.02±0.01a		
BCAR	0.03±0.01a	0.07±0.03ab	0.05±0.01ab	0.140±0.003c	0.07±0.01b		
PF	0.02±0.01a	0.02±0.01ab	0.02±0.01ab	0.08±0.01c	0.03±0.01b		
PT	0.07±0.01a	0.09±0.02a	0.08±0.01a	0.29±0.01b	0.13±0.02c		
Total Carotenoids	1.95±0.16a	2.41±0.52a	2.11±0.10a	7.52±0.32b	3.48±0.36c		
RAE*	8.48±1.29a	13.44±3.68ab	11.19±1.20a	37.33±0.22c	17.20±1.72b		

<sup>&</sup>lt;sup>Ψ</sup>FJ, fresh juice; P9230, pasteurized juice at 92 <sup>Q</sup>C 30s; P8515, pasteurized juice at 85 <sup>Q</sup>C 15s; H20, homogenized juice at 20 MPa; H150, homogenized juice at 150 MPa; CHPuRP8515, centrifuged juice, pulp fraction homogenized at 150 MPa, reconstituted with the serum fraction and pasteurized at 85 <sup>Q</sup>C 15s.

Nomenclature used for carotenoids: ANT= (antheraxanthin), VIO = (violaxanthin), LUTEO = (luteoxanthin), MUT = (mutatoxanthin), LUT= (lutein), ZEA = (zeaxanthin), ZEINO = (zeinoxanthin), BCR = ( $\beta$ -cryptoxanthin), ACAR= ( $\alpha$ -carotene), BCAR= ( $\beta$ -carotene), PT= (phytoene), PF= (phytofluene)

<sup>\*</sup>Retinol activity equivalent: calculated as RAE ( $\mu$ g/L) = ( $\mu$ g of  $\beta$ -carotene)/12 + ( $\mu$ g  $\beta$ -cryptoxanthin +  $\mu$   $\alpha$ -carotene)/24.

**Table 3.** Flavonoid levels (mg/L) found in juices assayed $^{\Psi}$ .

Flanca at da	Orange juice samples						
Flavonoids	FJ	P9230	P8515	H150	CHPuMP8515		
Vicenin-2	10.07±0.19a	10.27±0.13ab	10.256±0.12ab	10.57±0.12bc	10.91±0.26c		
Apigenin-d	7.82±0.04a	7.64±0.13a	7.89±0.208a	7.17±0.20b	6.08±0.10c		
∑Flavones	18.65±0.25a	18.60±0.27a	18.68±0.34a	18.33±0.21ab	17.54±0.40b		
Naringin-d	13.81±0.39a	14.60±0.25a	14.40±0.37a	14.03±1.04a	13.11±0.53a		
Narirutin	16.61±0.05ab	16.68±0.26ab	17.33±0.53b	15.68±0.43ac	15.03±0.66c		
Hesperidin	251.28±9.30a	271.02±4.27a	262.64±5.24a	207.66±5.49b	205.03±19.40b		
Didymin	6.42±0.42abc	7.14±0.39b	7.01±0.63ab	5.87±0.106bc	5.55±0.45c		
∑Flavanones	288.12±9.51a	309.44±4.83a	301.37±5.08a	243.24±6.32b	238.72±20.99b		
∑Flavonoids	306.77±9.76a	328.04±5.10a	320.06±5.42a	261.57±6.32b	256.26±21.35b		

 $<sup>^{\</sup>Psi}$ FJ, fresh juice; P9230, pasteurized juice at 92  $^{\Omega}$ C 30s; P8515, pasteurized juice at 85  $^{\Omega}$ C 15s; H20, homogenized juice at 20 MPa; H150, homogenized juice at 150 MPa; CHPuRP8515, centrifuged juice, pulp fraction homogenized at 150 MPa, reconstituted with the serum fraction and pasteurized at 85  $^{\Omega}$ C 15s.

**Table 4.** Flavonoid Bioaccesible Content (FBC, theoretical amount of flavonoids in mg present in the digestates after the ingestion of 1L of juice) of juices assayed $^{\Psi}$ .

Flavonoids	Flavonoid Bioaccessible Content						
riavonoius	FJ	P9230	P8515	H150	CHPuMP8515		
vicenin-2	0.82±0.05a	0.89±0.05ab	0.79±0.07a	1.01±0.06bc	1.15±0.09c		
apigenin-d	0.25±0.02a	0.26±0.02a	0.25±0.01a	0.29±0.03a	0.29±0.03a		
∑Flavones	1.07±0.07ab	1.15±0.07ab	1.04±0.08a	1.30±0.09bc	1.44±0.12c		
naringin-d	0.78±0.10a	0.91±0.20a	0.90±0.09a	0.91±0.11a	1.46±0.23b		
narirutin	1.01±0.06a	1.05±0.13a	1.01±0.05a	1.19±0.09ab	1.38±0.16b		
hesperidin	16.33±1.16a	18.04±2.13a	16.17±1.32a	17.09±2.16a	19.69±1.44a		
didymin	0.46±0.05a	0.56±0.06a	0.47±0.05a	0.56±0.06a	0.59±0.07a		
∑Flavanones	18.58±1.36a	20.55±2.50a	18.55±1.51a	19.74±2.41a	23.11±1.81a		
∑Flavonoids	19.65±1.44a	21.70±2.57a	19.595±1.59a	21.04±2.49a	24.55±1.92a		

 $<sup>^{\</sup>Psi}$ FJ, fresh juice; P9230, pasteurized juice at 92  $^{\circ}$ C 30s; P8515, pasteurized juice at 85  $^{\circ}$ C 15s; H20, homogenized juice at 20 MPa; H150, homogenized juice at 150 MPa; CHPuRP8515, centrifuged juice, pulp fraction homogenized at 150 MPa, reconstituted with the serum fraction and pasteurized at 85  $^{\circ}$ C 15s.

Figure 1. Carla M. Stinco

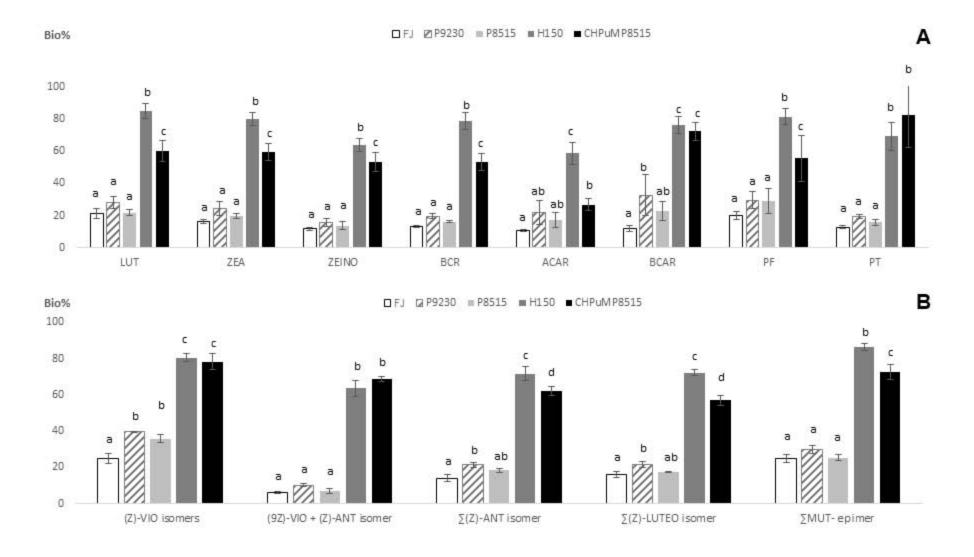
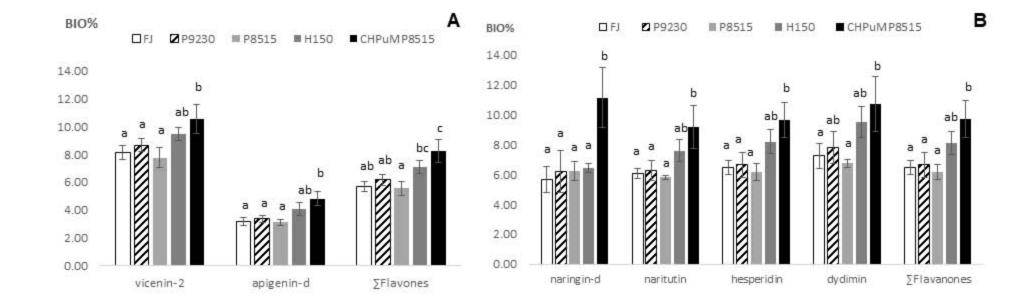


Figure 2. Carla M. Stinco



**Table S1.** Physicochemical parameters  $^{\Psi}$  of orange juices assayed.

Parameters	FJ	P9230	P8515	H150	CHPuRP8515
Pulp (%)	8.83±0.59 <sup>a</sup>	7.56±0.31 <sup>a</sup>	8.01±0.01 <sup>a</sup>	2.01±0.35°	2.33±0.23°
Transmitance (%)	13.02±0.91 <sup>a</sup>	9.65±0.49 <sup>b</sup>	9.50±1.34 <sup>b</sup>	9.97±0.39 <sup>b</sup>	3.32±0.08°
Ascorbic acid (mg/L)	428.70±0.35 <sup>a</sup>	413.07±0.73 <sup>a</sup>	414.17±0.02 <sup>a</sup>	425.91±0.78 <sup>a</sup>	388.07±0.66 <sup>a</sup>
PME activity (nkat/mL)	1.501±0.023 <sup>a</sup>	0.045±0.001 <sup>d</sup>	$0.067\pm0.001^{d}$	0.614±0.001°	$0.061\pm0.001^{d}$
L*	44.69±0.01 <sup>f</sup>	45.73±0.07 <sup>e</sup>	46.71±0.01 <sup>d</sup>	50.27±0.01 <sup>a</sup>	49.91±0.04 <sup>f</sup>
a*	5.33±0.01 <sup>a</sup>	4.95±0.01 <sup>d</sup>	5.07±0.01 <sup>b</sup>	3.24±0.01 <sup>f</sup>	3.52±0.02 <sup>d</sup>
b*	54.06±0.01°	51.06±0.27 <sup>d</sup>	50.91±0.11 <sup>d</sup>	58.49±0.01 <sup>a</sup>	58.50±0.01 <sup>a</sup>
C* <sub>ab</sub>	54.32±0.01°	51.30±0.027 <sup>d</sup>	51.17±0.11 <sup>d</sup>	58.58±0.01 <sup>a</sup>	58.61±0.01 <sup>a</sup>
$h_{ab}$	84.36±0.01 <sup>e</sup>	84.46±0.03 <sup>d</sup>	84.31±0.01 <sup>e</sup>	86.83±0.01 <sup>a</sup>	86.55±0.02 <sup>b</sup>
$\Delta E^*_{ab} FJ$	_	3.20°	3.75 <sup>b</sup>	7.42 <sup>a</sup>	7.09 <sup>a</sup>
ΔE* <sub>ab</sub> P9230	3.20°	_	1.02 <sup>d</sup>	8.87ª	8.65 <sup>a</sup>
ΔE* <sub>ab</sub> P8515	3.75°	1.02 <sup>d</sup>	_	8.56°	8.38 <sup>a</sup>
ΔE* <sub>ab</sub> H150	7.42 <sup>b</sup>	8.87ª	8.56ª	<del>-</del>	0.45 <sup>d</sup>
ΔE* <sub>ab</sub> CHPuRP8515	7.09 <sup>a</sup>	8.65 <sup>a</sup>	8.38 <sup>a</sup>	0.45 <sup>d</sup>	_
<sup>¥</sup> D <sub>[4,3]</sub> (μm)	540.89±2.97 <sup>a</sup>	369.50±4.10 <sup>b</sup>	355.13±7.13 <sup>b</sup>	70.36±3.52°	52.05±3.14°
<sup>¥</sup> D <sub>[3,2]</sub> (μm)	108.84±1.53 <sup>a</sup>	67.82±2.87 <sup>b</sup>	64.14±0.85 <sup>b</sup>	31.18±2.70°	14.54±0.96 <sup>d</sup>
<sup>¥</sup> SSA (m²/g)	0.06±0.01 <sup>d</sup>	0.09±0.01 <sup>d</sup>	0.09±0.01 <sup>d</sup>	0.19±0.02°	0.42±0.03°
¥d(0.1) (μm)	70.16±1.70 <sup>a</sup>	35.11±1.86 <sup>b</sup>	28.86±0.83 <sup>b</sup>	18.96±3.29°	9.79±1.96 <sup>d</sup>
¥d(0.5) (μm)	480.13±2.06 <sup>a</sup>	325.16±4.46 <sup>b</sup>	320.28±7.28 <sup>b</sup>	60.28±3.29°	39.81±1.92°
$^{4}d(0.9) \text{ (µm)}$	1096.51±4.16 <sup>a</sup>	768.54±6.37 <sup>b</sup>	732.53±12.74 <sup>b</sup>	137.55±6.47°	114.68±7.47°

<sup>&</sup>lt;sup>Ψ</sup> FJ, fresh juice; P9230, pasteurized juice at 92  $^{\circ}$ C 30s; P8515, pasteurized juice at 85  $^{\circ}$ C 15s; H150, homogenized juice at 150 MPa; CHPuRP8515, centrifuged juice, pulp fraction homogenized at 150 MPa, reconstituted with the serum fraction and pasteurized at 85  $^{\circ}$ C 15s. Different letters within the same row indicate statistically significant differences (p < 0.05).

<sup>\*</sup> D<sub>[4,3]</sub>, volume mean diameter; D<sub>[3,2]</sub>, surface area mean diameter; A<sub>s</sub>, specific surface area, d(0.1), d(0.5), and d(0.9), standard percentile reading.

**Table S2.** Summary of the simple regression analysis\* between the particle size parameters and the carotenoids bioaccessibility found in orange juices assayed.

Bioaccessibility(%)	D[4,3]	D [3, 2]	As	d (0.1)	d (0.5)	d (0.9)
(Z)-VIO isomers	-0.975	-0.918	0.813	-0.817	-0.977	-0.978
(9Z)-VIO + (Z)-ANT isomer	<i>-0.942</i>	<i>-0.873</i>	0.859	-0.734	-0.946	-0.947
$\Sigma(Z)$ -LUTEO isomer	-0.919	-0.821	0.705	-0.694	-0.921	-0.926
∑MUT- epimer	-0.922	-0.827	0.736	-0.694	-0.924	-0.929
∑(Z)-ANT isomer	-0.943	-0.856	0.759	-0.734	-0.946	- <b>0.949</b>
LUT	<i>-0.878</i>	<i>-0.765</i>	0.620	-0.643	-0.880	- <b>0.886</b>
ZEA	-0.908	-0.805	0.666	- <b>0.688</b>	-0.908	-0.915
ZEINO	-0.924	-0.829	0.743	<i>-0.701</i>	-0.925	-0.930
BCR	-0.892	<i>-0.783</i>	0.628	- <b>0.667</b>	-0.892	-0.900
ACAR	-0.739	-0.628	0.336	- <b>0.</b> 565	-0.736	<i>-0.748</i>
BCAR	-0.966	-0.915	0.796	-0.824	-0.968	-0.968
PF	-0.861	<i>-0.758</i>	0.557	-0.668	-0.860	-0.868
PT	-0.909	<i>-0.852</i>	0.849	-0.723	-0.913	-0.912
TC	-0.940	-0.852	0.765	-0.724	-0.942	-0.946
RAE	-0.913	-0.816	0.656	-0.708	-0.914	-0.920

<sup>\*</sup>Numbers in italic and bold indicate the existence of significant correlations (p<0.05).

Figure S1. Carla M. Stinco

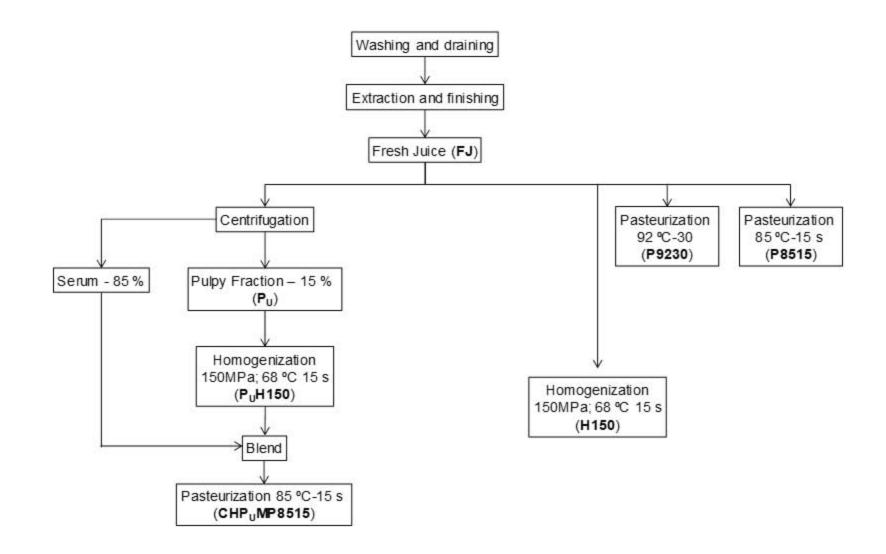


Figure S2. Carla M. Stinco

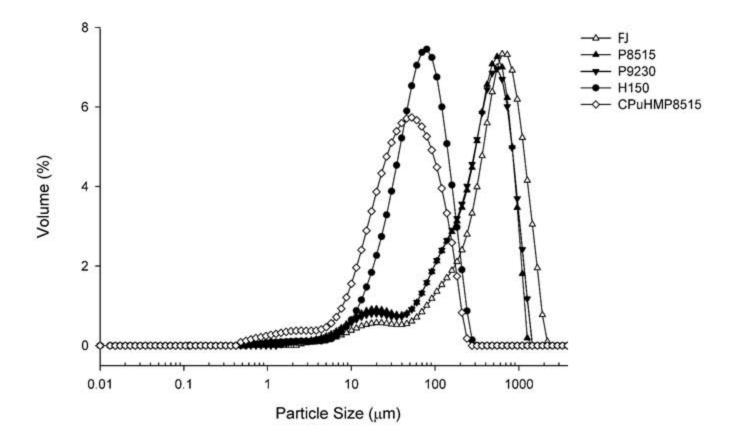


Figure S3. Carla M. Stinco

