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5 A COMPARISON OF MICROFLUIDIZATION AND SONICATION TO OBTAIN NANOMETRIC  
6 ECOLOGICAL EMULSIONS. EFFECT OF DIUTAN GUM CONCENTRATION AS STABILIZER.

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## 12 Abstract

13 The objective of this work was to develop emulsions formulated with natural ingredients such  
14 as lemongrass essential oil and Appyclean 6552 (emulsifier), aiming to reach nanometric size  
15 droplets. The emulsions were prepared using two different processing techniques:  
16 microfluidization and sonication. Sonication demonstrated to be a powerful technique to  
17 produce nanoemulsions since even an excess of energy input did not promote recalescence,  
18 conversely to microfluidization. Despite the fact microfluidization led to recalescence  
19 phenomenon, the results obtained in terms of droplet sizes and stability were better at lower  
20 homogenization pressures than those obtained by sonication. The best nanoemulsion  
21 concerning physical stability resulting from microfluidization process was used as a reference  
22 to incorporate diutan gum, which reduces the creaming destabilization mechanism. However,  
23 an excess above 0.3 wt% diutan gum proved counterproductive for droplet size reaching  
24 values out of nanometric scale.

25 **Keywords:** Microfluidization, sonication, lemongrass essential oil, long-term stability, diutan  
26 gum, rheology.

## 27 1. Introduction

28 Droplet size distribution (DSD) of emulsions is a key parameter for the stability and rheology of  
29 these dispersed systems. For example, nanoemulsions present enhanced physical stability

30 compared to emulsions with higher droplet size. DSD is mainly governed by homogenization  
31 method used although formulation also influences. There are several high-energy methods to  
32 develop emulsions such as rotor-stators, sonicators, high-pressure pump homogenizers,  
33 colloids mills and microfluidizers. On the one hand, in the last ten years, Microfluidizers have  
34 attracted a lot of attention due to its ability to reduce droplet size <sup>1-3</sup>. This device is based on  
35 forcing the sample by high pressure (up to 150 MPa) through microchannels toward an  
36 impingement area. It creates an enormous shearing action, which can produce very fine  
37 emulsions. Some researchers claim that microfluidization is superior than traditional  
38 emulsifying homogenizers because the DSD obtained are narrower and smaller <sup>4,5</sup>. However,  
39 the use of microfluidizers is sometimes related to the over-processing phenomenon, which  
40 provokes the occurrence of re-coalescence <sup>6</sup>. On the other hand, the use of sonication  
41 technique in order to reduce the droplet size of emulsions presents some advantages such as  
42 the minimum recoalescence and the easiness for cleaning and servicing <sup>7,8</sup>. An increase of  
43 ultrasonic power or amplitude applied is directly related to the reduction of the droplet size  
44 since high amplitudes generates strong shear forces. This technique has been previously used  
45 in the production of nanoemulsions formulated with triglycerides, diazino or mucilage <sup>9-11</sup>.

46 The use of essential oils as natural food preservatives has attracting attention nowadays due to  
47 their properties such as strong antioxidant, anti-inflammatory, diuretic, analgesic and  
48 antimicrobial activities among others <sup>12,13</sup>. Lemongrass essential oil derived from *Cymbopogon*  
49 *citratus* possesses several applications in food, biomedical and pharmaceutical fields <sup>14-16</sup>.  
50 Recent studies reveal that the antimicrobial activity of lemongrass essential oil increases with  
51 emulsion droplet size decreases <sup>17</sup>. This fact highlights the importance of the droplet size  
52 obtained by processing and formulation control.

53 Other ingredients that are contributing to development of natural products are the surfactants  
54 derived from biomass. Appyclean 6552, which belongs to the alkyl poly pentosides family, is a  
55 novel surfactant that come from a renewable raw material (wheat biomass). In addition, it is a  
56 sustainable solution to an agricultural waste such as wheat straw and possesses the ECOCERT  
57 certification. As a consequence, this emulsifier has recently incorporated in green formulations  
58 <sup>18</sup>.

59 Emulsions are thermodynamically unstable systems that can suffer different destabilization  
60 mechanisms such as creaming, coalescence, Ostwald ripening and/or flocculation. The  
61 incorporation of polysaccharides as stabilizers and thickeners is very common in emulsion field  
62 in order to develop stable emulsions. Diutan gum, secreted by *Sphingomonas sp.*, is an

63 aqueous microbial polysaccharide that is considered biodegradable and biocompatible. This  
 64 novel thickener has been used recently to avoid droplet size increment in alkane  
 65 nanoemulsions<sup>19</sup>. However, to the best of our knowledge, there is a lack of information about  
 66 the reduction of creaming in nanoemulsions containing essential oils by using diutan gum.

67 The principal aim of this study was to develop stable emulsions using materials derived from  
 68 renewable resources. Furthermore, other important objective was to compare two different  
 69 emulsification methods (microfluidization and sonication) on the basis of physical stability and  
 70 droplet size distributions for lemongrass-in-water emulsions. In addition, the role of diutan  
 71 gum in order to reduce creaming was analysed using rheological and laser diffraction  
 72 measurements as well as multiple light scattering technique.

## 73 **2. Materials and methods**

### 74 2.1. Materials

75 The dispersed phase used was lemongrass essential oil, which was provided by Sigma Chemical  
 76 Company. Appyclean 6552, an ecological surfactant, was supplied by Wheatoleo. All emulsions  
 77 were prepared with deionized water.

### 78 2.2. Methods

#### 79 *Emulsions preparation*

80 Emulsions containing 5 wt% of lemongrass essential oil and 0.5 wt% of appyclean 6552 were  
 81 prepared via different emulsification methods. However, the formation of the coarse emulsion  
 82 was the same: this emulsion was prepared using a Silverson L5M at 4000 rpm for 1 minute.  
 83 Subsequently, different emulsions were developed as described in table 1.

84 Table 1. Preparation methodology for lemongrass nanoemulsions.

Samples	Technique	Homogenization pressure	Number of cycles
		(psi)	
1	Microfluidization	2500	I
2		2500	II
3		5000	I
4		5000	II
5		10000	I
6		10000	II

7		15000	I
8		15000	II
		<b>Amplitude (W)</b>	<b>time (min)</b>
9	<b>Sonication</b>	32	3
10		44	3
11		57	3
12		75	3
13		85	3
14		75	1
15		75	4
16		75	5
17		75	7

85

86 *Diutan gum preparation and incorporation to the nanoemulsion*

87 A stock of 1 wt% of diutan gum solution was prepared using a IKA-Visc at 700 rpm for 3 h at  
88 room temperature. Subsequently, the dispersion was stirred at 300 rpm for 2 h in order to  
89 remove bubbles. The incorporation of diutan gum solution to the nanoemulsion selected was  
90 carried out by mixing both systems using an IKA-Visc homogenizer at 300 rpm until complete  
91 homogenization.

92 *Laser diffraction measurements*

93 Malvern Mastersizer 2000 was used in order to analyse the droplet size distributions for the  
94 nanoemulsions developed. The measurements were made by triplicate. In order to evaluate  
95 the results obtained, Sauter diameter and span parameter were calculated as follows:

96 
$$D_{3,2} = \frac{\sum_{i=1}^N n_i d_i^3}{\sum_{i=1}^N n_i d_i^2} \quad \text{Eq. (1)}$$

97 
$$\text{span} = \frac{D_{90} - D_{10}}{D_{50}} \quad \text{Eq. (2)}$$

98 Where  $d_i$  is the droplet diameter,  $N$  is the total number of droplets,  $n_i$  is the number of  
99 droplets having a diameter  $d_i$ , and  $D_{90}$ ,  $D_{50}$ ,  $D_{10}$  are the diameters at 90%, 50% and 10%  
100 cumulative volume.

101 *Physical stability study*

102 Backscattering measurements (Turbiscan Lab Expert) were carried out with aging time in order  
103 to analyse and quantify the destabilization mechanisms for the emulsions developed. These  
104 measurements were performed for at least 25 days at 25°C. Some researchers have quantified  
105 the physical stability of emulsions using the Turbiscan Stability Index (TSI)<sup>20,21</sup>.

106 
$$TSI = \sum_j |scan_{ref}(h_j) - scan_i(h_j)| \text{ Eq. (3)}$$

107 where  $scan_{ref}$  and  $scan_i$  are the initial transmission value and the transmission value at a  
108 specific time, respectively and  $h_j$  is a specific height in the measuring cell.

### 109 *Rheological tests*

110 All rheological measurements were performed using a controlled-stress rheometer AR2000 (TA  
111 Instruments) equipped by a serrated plate-plate geometry (60 mm of diameter). Small  
112 Amplitude Oscillatory Shear tests (SAOS) were conducted from 20 to 0.05 rad/s at a stress in  
113 the Linear Viscoelastic Range (LVR). The LVR was obtained by means of stress sweeps at 0.1, 1  
114 and 3 Hz. On the other hand, flow curves were carried out by a stress-based multistep protocol  
115 (3 min/point) at 20°C. The possible loss of water was avoided using a solvent trap.

### 116 *Cryo-Scanning Electron Microscopy*

117 Cryo Scanning Electron microscope Zeiss EVO was used in order to observe the microstructure  
118 of some selected emulsions with and without diutan gum. Samples were prepared following  
119 the protocol reported by Santos et al.<sup>22</sup> and observed at 10 kV.

## 120 **3. Results and discussion**

### 121 *Comparison between microfluidization and sonication technique for development of* 122 *lemongrass-in-water emulsions*

123 Figure 1 shows the influence of homogenization pressure and number of cycles on Sauter  
124 diameter and span values for lemongrass emulsions processed in Microfluidizer. First at all, it is  
125 important to note that the Sauter diameter and the span parameter for the pre-emulsion was  
126  $950 \pm 50$  nm and 0.97, respectively (data not shown). Taking this into account, the reduction of  
127 Sauter diameter and span values is very noticeable using microfluidization. The application of  
128 2500 and 5000 psi of homogenization pressure (lower homogenization pressures) in  
129 Microfluidizer provoked a decrease of Sauter diameter and span parameter reaching values of  
130 185 nm and 0.7, respectively. However, an increase of both parameter was observed above  
131 5000 psi. This fact is related to the over-processing phenomenon that provokes recalescence,

132 which is well-known in microfluidizer devices <sup>6,23</sup>. In addition, the increase of number of cycles  
133 from one to two cycles at lower pressures did not produce any change in span parameter or  
134 Sauter diameter. However, an increment of Sauter diameter with the second cycle was  
135 observed at higher pressures. This also points out a recoalescence phenomenon  
136 aforementioned. Hence, the application of low homogenization pressures showed the best  
137 results concerning Sauter diameter and span parameter, obtaining emulsions with a mean  
138 diameter of nanometric size (<200 nm).

139 Figure 2A illustrates the influence of ultrasonic power used on the Sauter diameter and span  
140 parameter for lemongrass-in-water emulsions developed using a sonicator for 3 minutes. The  
141 increase of ultrasonic power provoked a decrease in Sauter diameter and span parameter. This  
142 fact is due to that higher amplitudes produced more powerful shock waves, which results in  
143 smaller Sauter diameters <sup>24</sup>. However, a tendency to reach constant values of both parameters  
144 at higher amplitudes was detected (Sauter diameter=195 nm; span parameter= 0.8). This  
145 tendency has been previously reported by other authors for coconut-in-water emulsions <sup>25</sup>.  
146 These results are slight higher than those obtained by microfluidization technique. This same  
147 trend was observed for the influence of sonication time on Sauter and span parameter for  
148 emulsions developed at 75 W (figure 2B). In conclusion, once the necessary energy to obtain  
149 the smallest mean diameter is reached, an increase of energy does not provoke any change.  
150 Thus, emulsions prepared using sonicator did not show re-coalescence, conversely to  
151 microfluidizer. This fact was previously pointed out by Jafari et al. <sup>8</sup>.

152 Figure 3 shows the Sauter diameter as a function of energy density for all microfluidized and  
153 sonicated emulsions. The results obtained demonstrated that the recoalescence phenomenon  
154 detected in microfluidized emulsions is not directly related to the energy density. This is  
155 supported by the fact that the sonicated emulsion processed at the highest energy density  
156 tested did not show over-processing. The lack of over-processing in sonicator is probably due  
157 to the higher residence times during emulsification. On the other hand, an increase in energy  
158 density in microfluidizer involves a decrease in mean residence time of the emulsions in the  
159 interaction chamber (in the range of milliseconds) <sup>8</sup>. This could lead to a recoalescence  
160 phenomenon because of the possible lack of time in order to get the optimal adsorption of the  
161 emulsifier in the interface. Thus, residence time could be the crucial parameter for over-  
162 processing. It is also interesting to mark that emulsions produced by sonicator at different  
163 amplitudes and times (72 W, 1 min and 32W, 3 min) with similar energy densities (23 and 30  
164 MJ/m<sup>3</sup>, respectively) show significant differences in Sauter diameter. This seems to prove that  
165 higher amplitudes promote the creation of smaller droplets at similar energy densities. In

166 other words, higher amplitudes with smaller residence times are more efficient than lower  
167 amplitudes with higher residence times.

168 Figure 4 shows the Turbiscan Stability Index (TSI) for emulsions developed using A) microfluidizer at different pressure, B) sonicator at different ultrasonic power and C) sonicator at different sonication time. TSI was calculated in the low zone of the measuring cell, which is intimately related to a destabilization process by creaming. Interestingly, all emulsions studied underwent the same destabilization process (creaming). In figure 4A, all emulsions studied showed a linear increase in TSI values with aging time. The emulsion that exhibited the lowest slope was processed at the lowest energy density as expected taking into account that its lowest Sauter diameter and span parameter.

176 Figure 4B shows the influence of ultrasonic power applied in sonicator for 3 minutes on the TSI values with aging time. Firstly, these emulsions processed at higher ultrasonic power exhibited better stabilities against creaming. However, emulsions processed by microfluidizer showed better stability as demonstrated their lower TSI values. Figure 4C presents the influence of sonication time on the TSI values with aging time. TSI values are lower at higher sonication times, showing an enhanced stability in this way. However, this result does not improve the best result obtained for the emulsion processed in Microfluidizer at the lowest energy density. For this reason, this emulsion was selected for a further study that analyses the addition of diutan gum.

#### 185 *Influence of diutan gum concentration on the selected lemongrass-in-water nanoemulsion*

186 Figure 5 illustrates the frequency sweep for microfluidized nanoemulsion previously selected as a function of diutan gum concentration. Firstly, it is important to note that the nanoemulsion without diutan gum did not present measurable viscoelastic properties. A predominance of the elastic modulus,  $G'$ , over the viscous modulus,  $G''$ , at higher frequencies was presented for emulsions containing diutan gum. Nevertheless, a crossover point was observed at lower frequencies for 0.2wt% diutan gum emulsion. This point tends to shift progressively to lower frequencies with increasing gum concentration. These systems showed weak gel-like properties in all studied frequency range. This gel-like behaviour could reduce the droplets movement and therefore, the aforementioned creaming process. In addition, an increase in diutan gum concentration provoked an increase in viscoelastic functions. This fact suggests the formation of a network composed by diutan gum, similarly to other gums behaviour<sup>26</sup>. Interestingly, the values and the tendency of the viscoelastic functions for diutan gum emulsions are quite similar to those for diutan gum solutions with 0.5 wt% NaCl<sup>27</sup>.

199 Figure 6 exhibits the flow properties for the selected microfluidized emulsions as a function of  
 200 diutan gum concentration. First at all, it is important to mention that the emulsion formulated  
 201 without diutan gum showed Newtonian flow behaviour. This fact has previously reported by  
 202 other authors for emulsions containing different essential oils <sup>1</sup>. The addition of diutan gum to  
 203 the selected emulsion provoked the occurrence of shear-thinning behaviour. This flow  
 204 behaviour is fitted fairly well to Cross model ( $R^2 > 0.99$ ; Equation 4).

205 
$$\eta = \frac{\eta_0 - \eta_\infty}{1 + (k \cdot \dot{\gamma})^{1-n}} \text{ Eq.(4)}$$

206 where k in the inverse of critical shear rate for the onset of shear-thinning response,  $\eta_0$  is the  
 207 zero-shear viscosity and n is the so-called flow index.

208 Fitting parameters for this model are shown in table 1. An increase of zero shear viscosity with  
 209 diutan gum concentration was observed, which is consistent with the mechanical spectra for  
 210 these emulsions. In addition, there is a tendency of flow index reduction with diutan gum  
 211 concentration. This fact is also related to a higher structuration grade. Furthermore, emulsions  
 212 containing diutan gum present a lack of shear rates information of at least two decades, which  
 213 reveals a relatively subtle very shear-thinning behaviour. This is normally related to the  
 214 occurrence of a yield point. Although the presence of a yield point is not sufficiently clear in  
 215 this case, other authors have previously observed it for liquid paraffin nanoemulsions  
 216 containing diutan gum <sup>19</sup>. These authors pointed out that the occurrence of the yield point of  
 217 diutan gum emulsions is related to the network structure of the continuous phase.

218 Table 2. Flow curves fitting parameters for the Cross model for studied emulsions as a function  
 219 of diutan gum concentration.

wt% Diutan gum	$\eta_0$ (Pa · s)	$\eta_\infty$ (Pa · s)	k (s)	n
0.2	28.1	0	28	0.15
0.3	118	0	66	0.12
0.4	253	0.002	58	0.10

220

221 Figure 7A shows the increment of Backscattering (BS) at 25 days of aging time for both the  
 222 selected microfluidized emulsion and the emulsion that contains the highest concentration of  
 223 diutan gum studied. The microfluidized emulsion illustrates a great drop of BS in the low zone  
 224 and a marked increase in the top zone of the measuring cell, which are related to the  
 225 processes of creaming and oiling-off, respectively. The addition of diutan gum reduced



226 considerably the creaming process. However, a bit decrease in BS in the top zone was  
227 observed. This could be attributed to a oiling off.

228 The global Turbiscan Stability Index parameter is shown in figure 7B. This parameter allows all  
229 those destabilization processes involved to be globally quantified and compared. As previously  
230 explained, emulsion without diutan gum presented a linear increase in TSI with aging time.  
231 Conversely, TSI values of emulsions with diutan gum showed a tendency to reach constant  
232 values above 20 days of aging time. Thus, emulsions formulated with diutan gum presented an  
233 enhanced physical stability. 0.2 wt% and 0.3 wt% diutan gum emulsions presented similar  
234 behaviours concerning stability and emulsion with the highest diutan gum concentration  
235 showed the best result of stability.

236 Figure 8 illustrates the droplet size distributions for lemongrass-in-water emulsions as a  
237 function of diutan gum concentration. The addition of diutan gum (0.2 and 0.3 wt%) provoked  
238 the occurrence of a second peak in the DSD (above 1  $\mu\text{m}$ ) related to the widely known partial  
239 recoalescence phenomenon during the processing of the gum incorporation. It is important to  
240 note that the emulsion containing 0.4 wt% of diutan gum showed a movement of the mean  
241 peak in the DSD to higher values of diameters centred at 0.9  $\mu\text{m}$ . This fact could be explained  
242 by the displacement of the surfactant by the diutan gum from the interface to the bulk of the  
243 continuous phase. This movement would provoke a lack of interface protection, which leads to  
244 the total recoalescence phenomenon and therefore, an increase of droplet sizes.

245 The microstructures observed by Cryo-SEM technique for A) the selected microfluidized  
246 emulsion without diutan gum, B,C) 0.2 wt% diutan gum emulsion and D) 0.4 wt% diutan gum  
247 emulsion are shown in figure 9. Figure 9A shows the occurrence of isolated droplets of  
248 lemongrass essential oil in the continuous phase. The lack of interaction between droplets and  
249 the fact of the continuous phase is water and surfactant support the Newtonian flow  
250 behaviour of the emulsion. Figures 9B and C present the microstructure for emulsions  
251 containing 0.2 wt% of diutan gum at different magnifications. A 3D network formed by diutan  
252 gum with two different population of droplets embedded is observed. The higher size of  
253 droplets (figure 9C) corresponds to the second peak detected by laser diffraction. 0.4 wt%  
254 diutan gum emulsion shows a very similar microstructure than 0.2 wt% diutan gum emulsion.  
255 However, the network presented is more compacted and the droplets are bigger, which  
256 supports laser diffraction results.

257 **Conclusions**

258 The main aim of this work was to make a comparison of two different techniques such as  
259 microfluidization and sonication for producing nanoemulsions formulated with lemongrass  
260 essential oil and applyclean 6552. Results obtained by using microfluidization revealed a  
261 recoalescence phenomenon at higher pressures tested. However, this destabilization  
262 mechanism was not detected in emulsions prepared by sonication despite the high energies  
263 input reached. Further analysis of the results demonstrated that at similar values of energy  
264 input supplied by both preparation methods provoked recoalescence in microfluidized  
265 emulsions, conversely to sonicated emulsions. This fact has been explained in terms of the  
266 higher residence times achieved in sonication. All emulsions, regardless of the homogenization  
267 method used, underwent creaming with aging time. Nevertheless, the emulsion processed at  
268 the lowest homogenization pressure in microfluidizer showed the lowest creaming rate. As  
269 strategy to enhance the physical stability of this emulsion, diutan gum was incorporated. The  
270 addition of diutan gum provoked the occurrence of viscoelastic properties, showing a weak  
271 gel-like behaviour in all the cases. A trend to cross-over point at lower frequencies was  
272 detected for 0.3 and 0.4 wt% diutan gum and this point was reached for 0.2 wt% diutan gum at  
273 0.65 rad/s. This indicates a higher grade of structuration in more gum concentrated emulsions.  
274 The incorporation of diutan gum to the formulation promoted a substantial improvement of  
275 physical stability against creaming. Nevertheless, the emulsion formulated with diutan gum  
276 above 0.3 wt% showed coalescence just after preparation with the subsequent increase of  
277 droplet size (above 1  $\mu\text{m}$ ). Therefore, it is important to reach a balanced compromise on both  
278 droplet size and physical stability. These differences of droplet sizes were also evident in the  
279 microstructure observed by Cryo-SEM technique, which as well as revealing the existence of a  
280 3D network developed by diutan gum.

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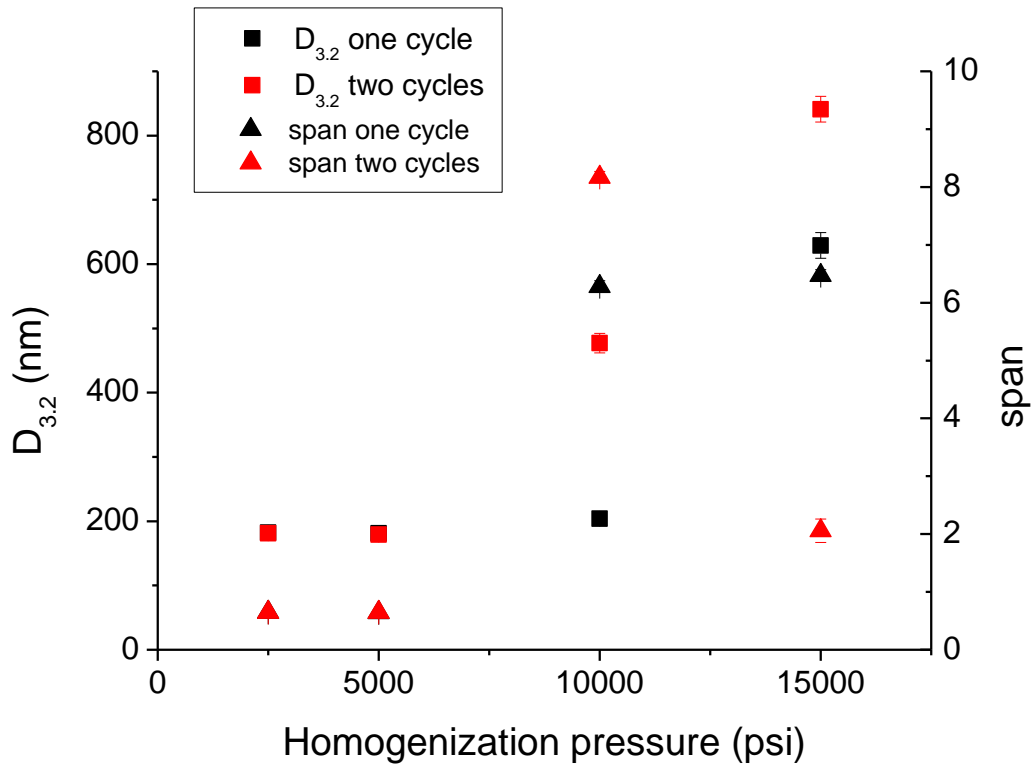
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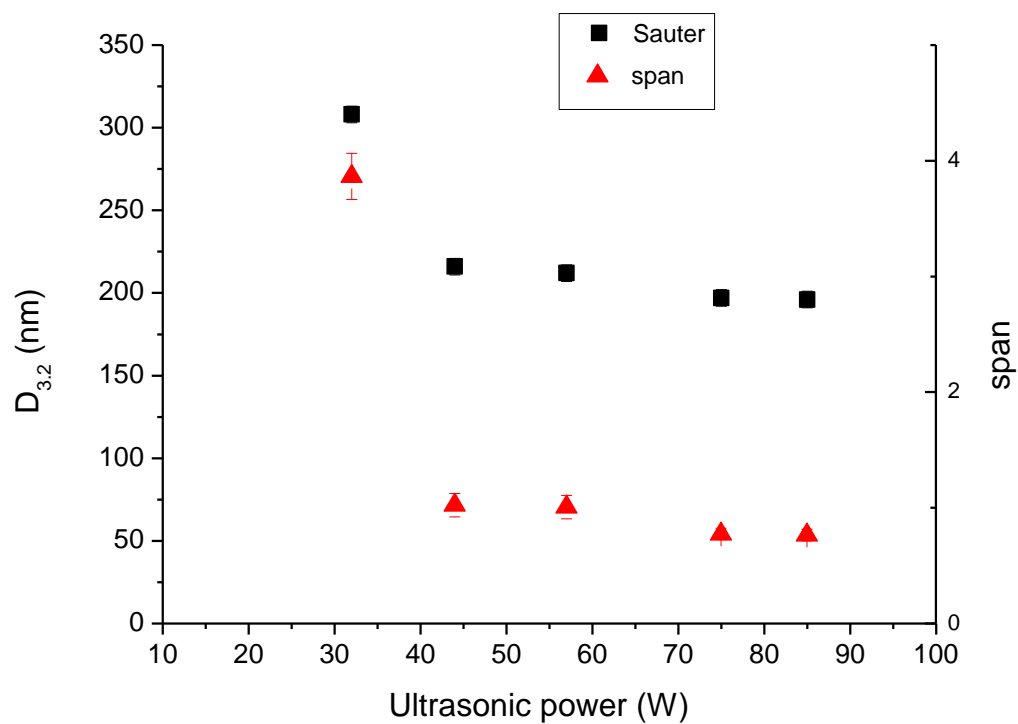
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376 Figure 1. Influence of homogenization pressure and number of cycles on Sauter diameter and  
377 span for lemongrass emulsions.

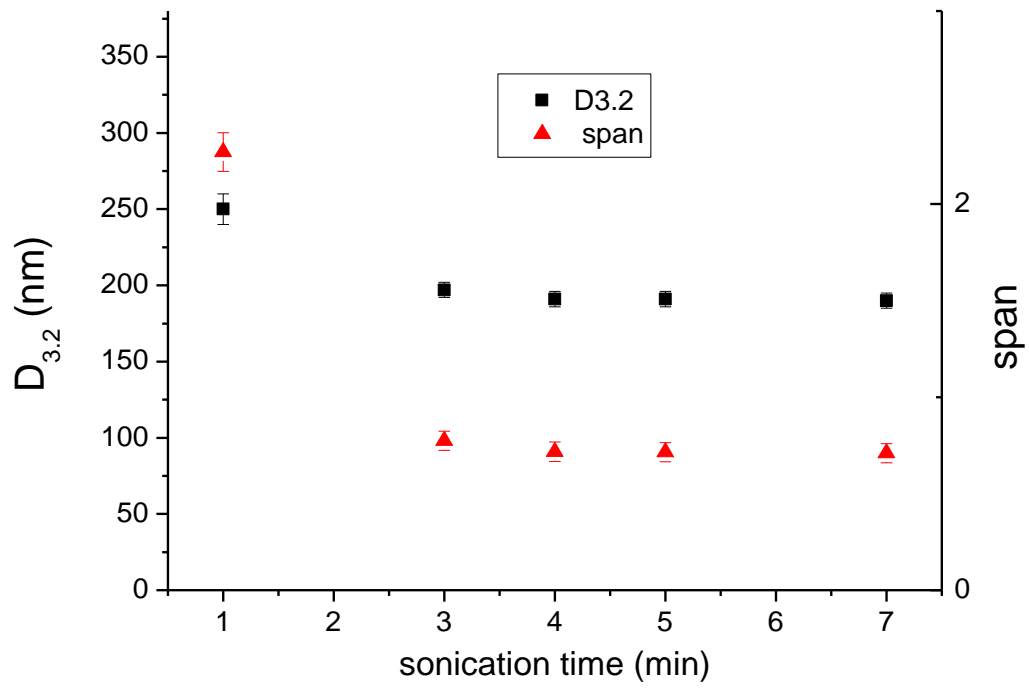
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380 Figure 2A. Influence of ultrasonic power on Sauter diameter and span values for lemongrass  
 381 emulsions processed in sonicator for 3 minutes.

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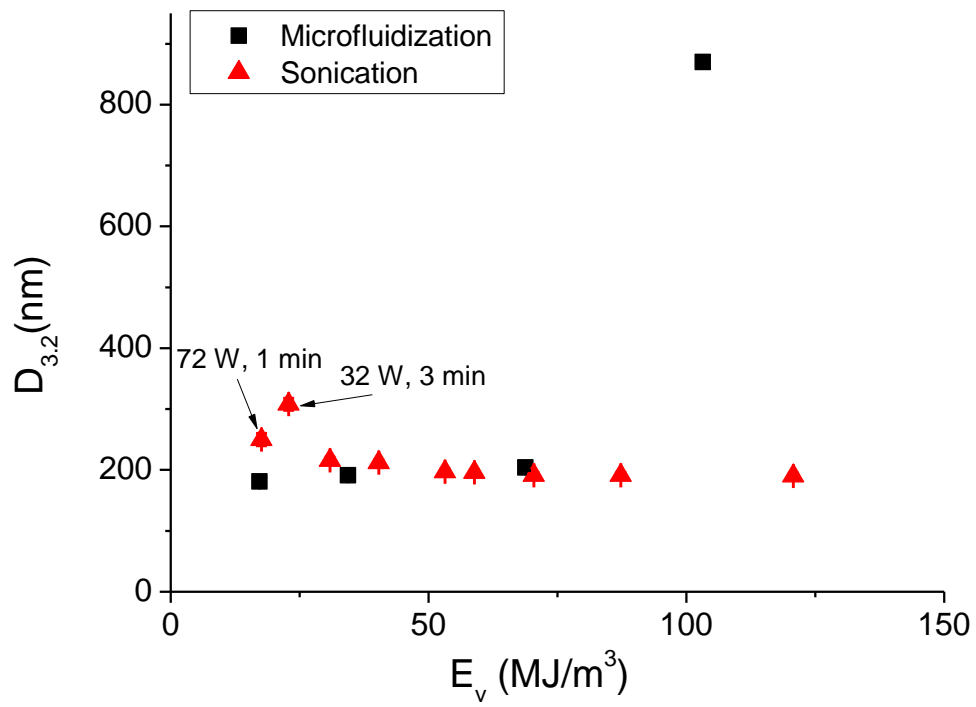


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386 Figure 2B. Influence of sonication time on Sauter diameter and span values for lemongrass  
387 emulsions processed in sonicator at 75 W.

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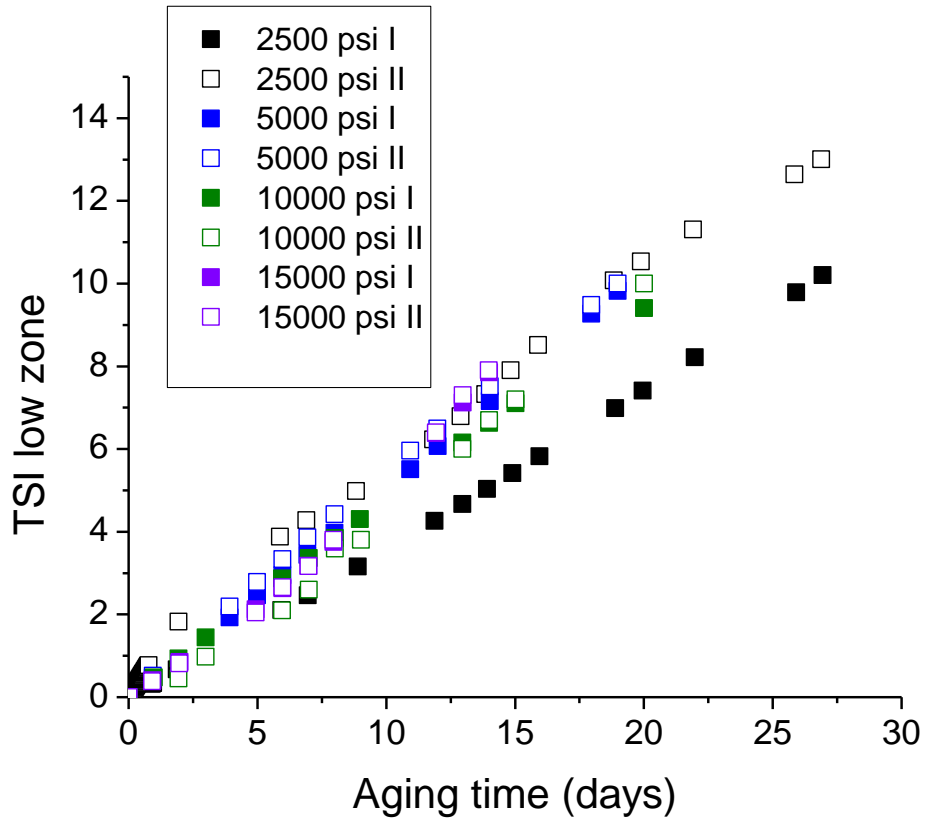




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391 Figure 3. Sauter diameter obtained as a function of energy input for lemongrass-in-water  
392 emulsions.

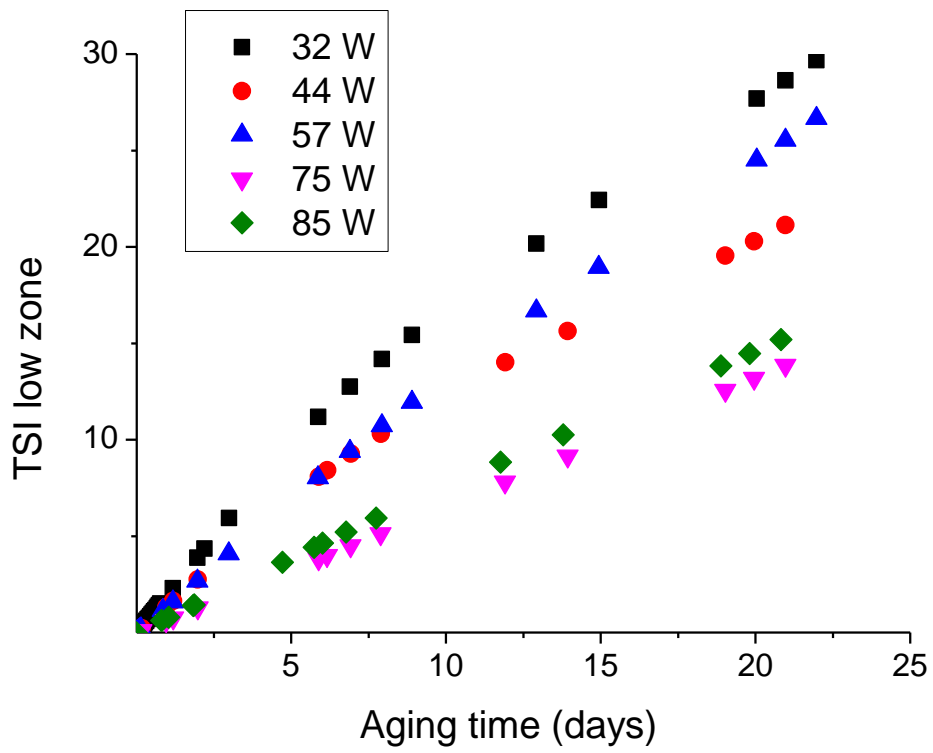
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395 Figure 4A. TSI values in the low zone with aging time as a function of homogenization pressure  
 396 and number of cycles applied in Microfluidizer.

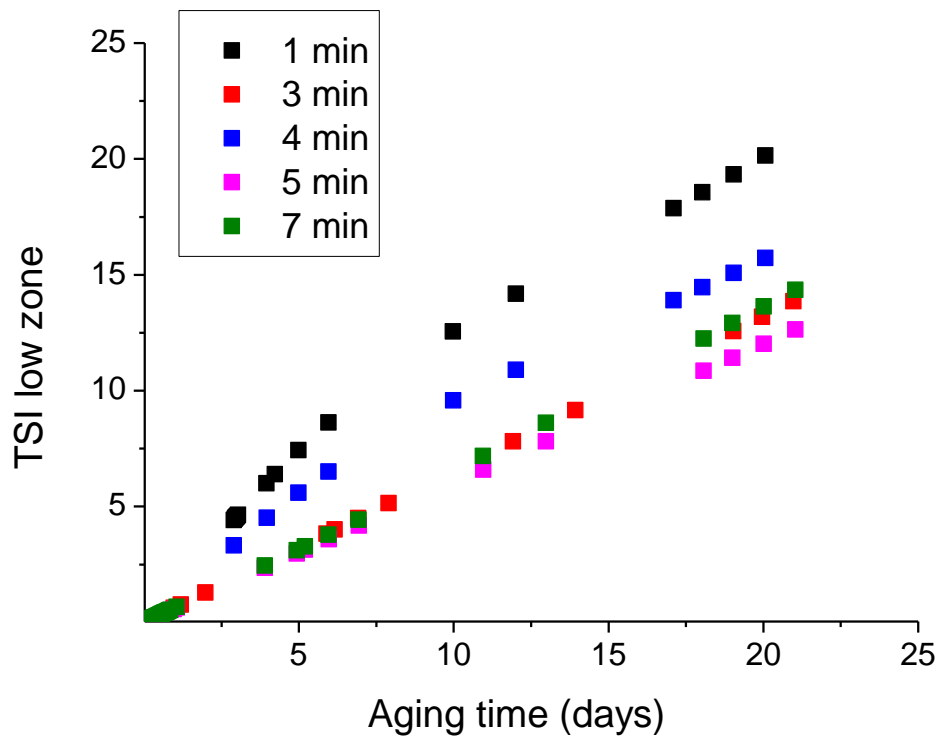
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399 Figure 4B. TSI values in the low zone with aging time as a function of ultrasonic power applied  
 400 in sonicator for 3 minutes.

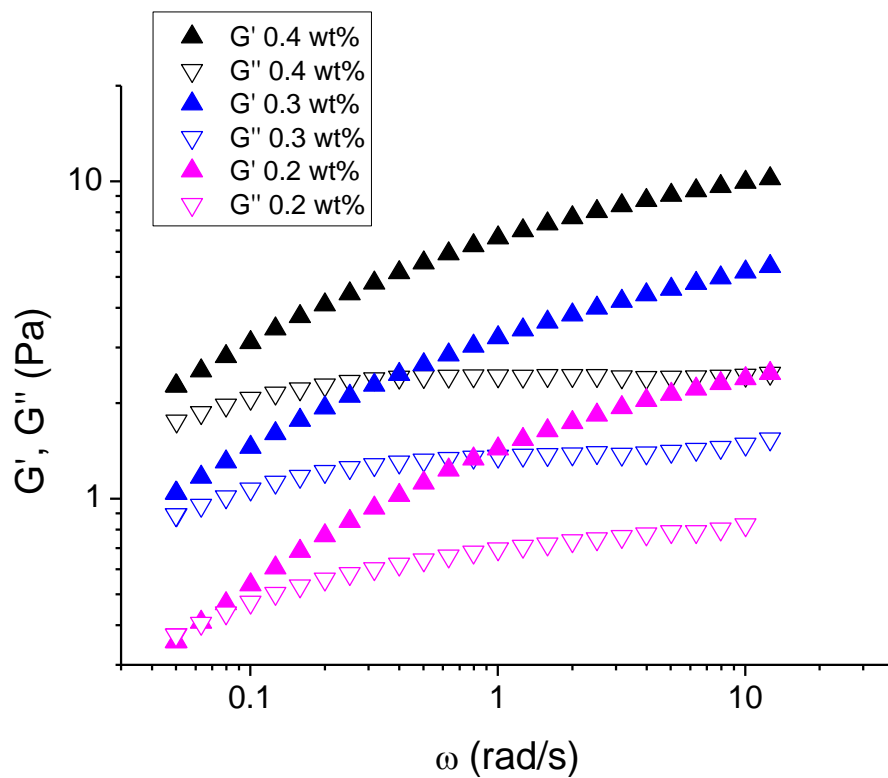
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403 Figure 4C. TSI values in the low zone with aging time as a function of ultrasonic power applied  
 404 in sonicator for 3 minutes.

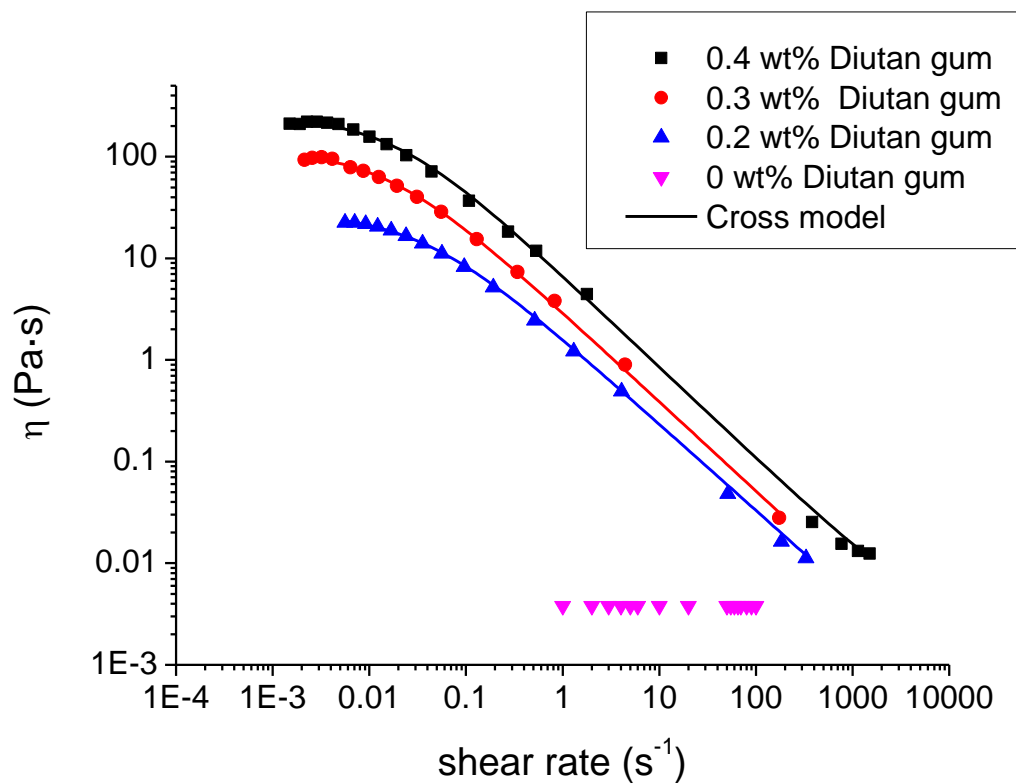
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407 Figure 5. Mechanical spectra for lemongrass-in-water emulsions as a function of diutan gum  
 408 concentration at 20°C.

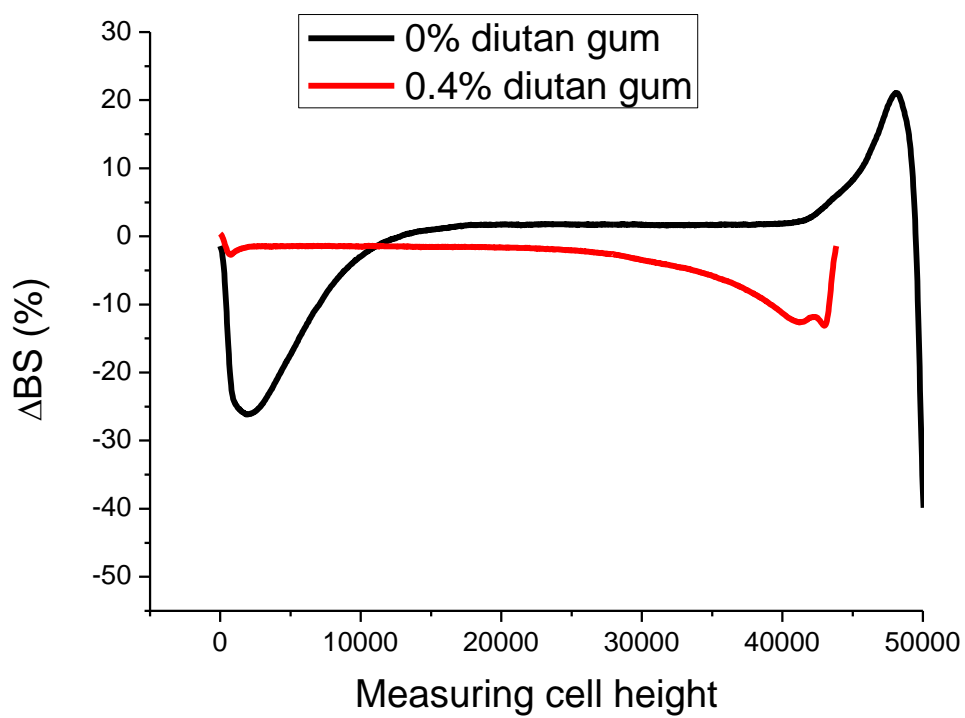
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411 Figure 6. Flow curve for lemongrass-in-water emulsions as a function of diutan gum  
 412 concentration.

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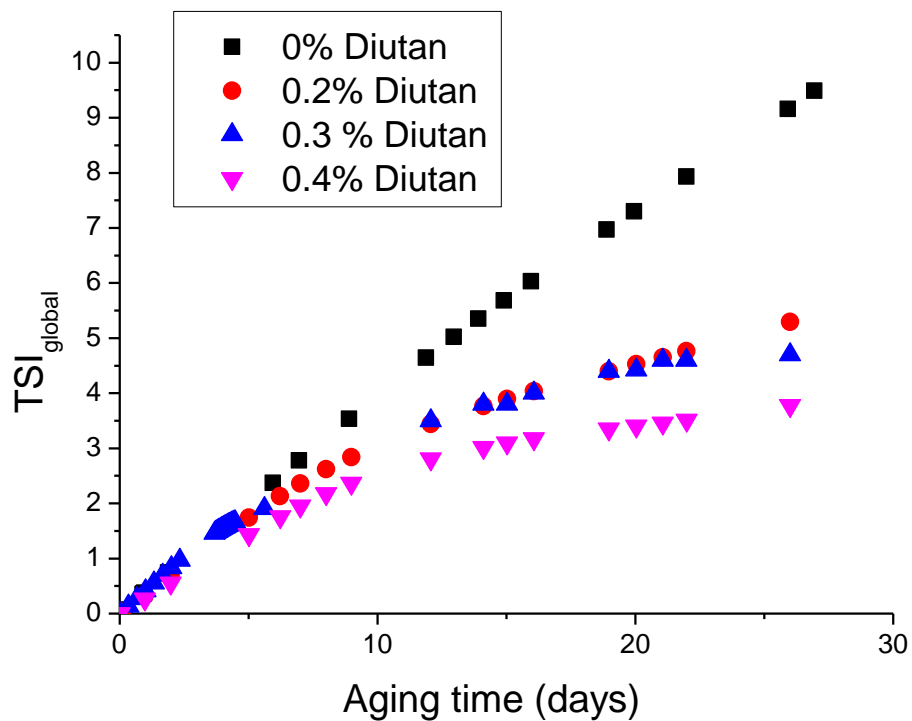
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415 Figure 7A. Backscattering variation as a function of measuring cell height for emulsions without  
416 and with 0.4 wt% of diutan gum at 25 days of aging time.

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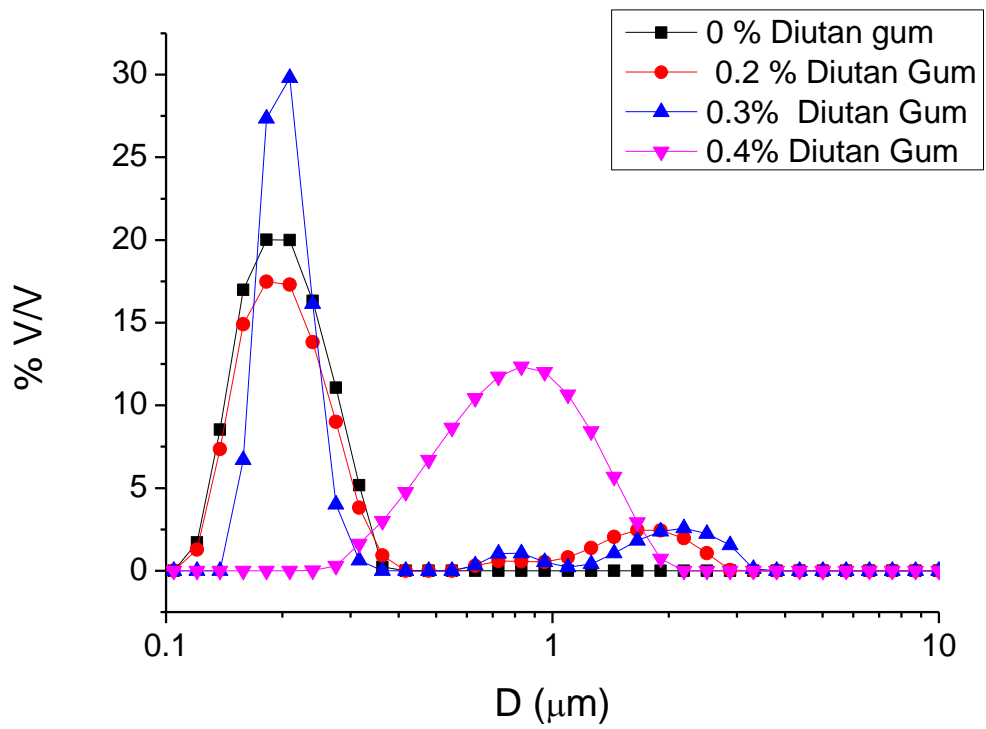


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421 Figure 7B. Global Turbiscan Stability Index (TSI) with aging time as a function of diutan gum  
422 concentration for lemongrass-in-water emulsions.

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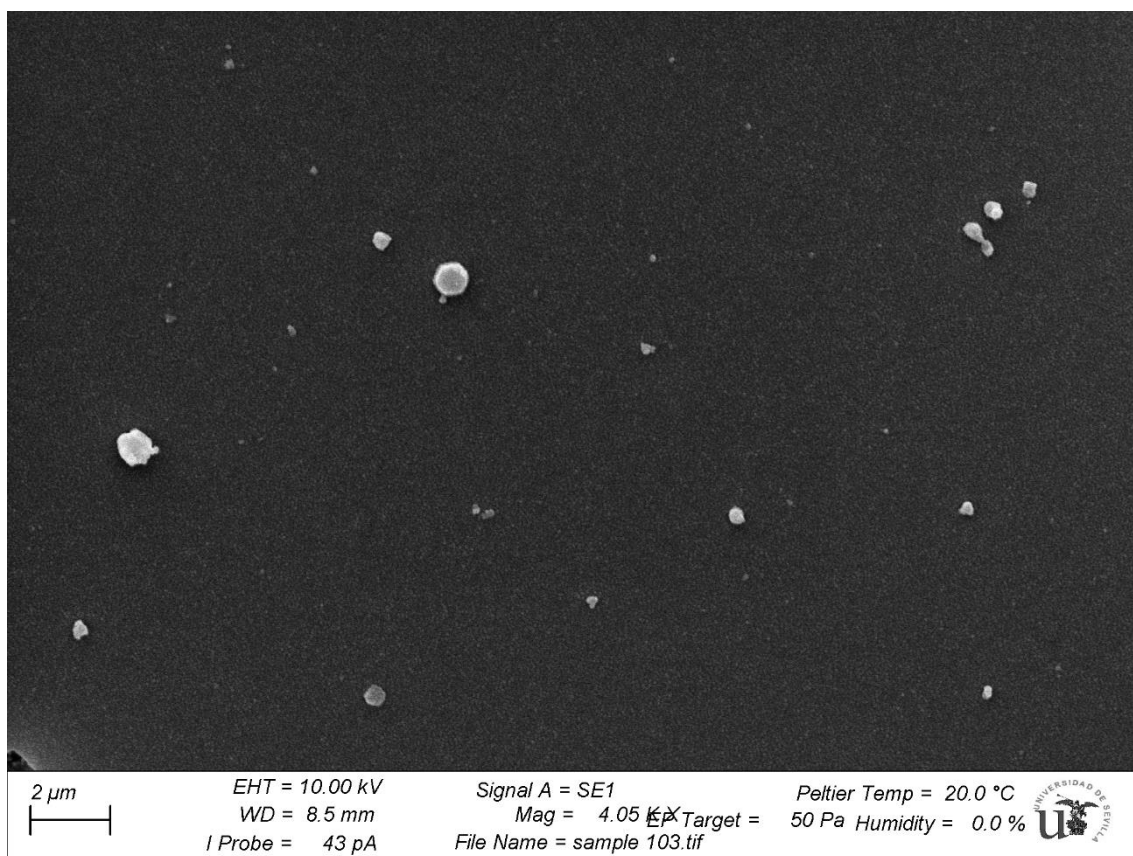




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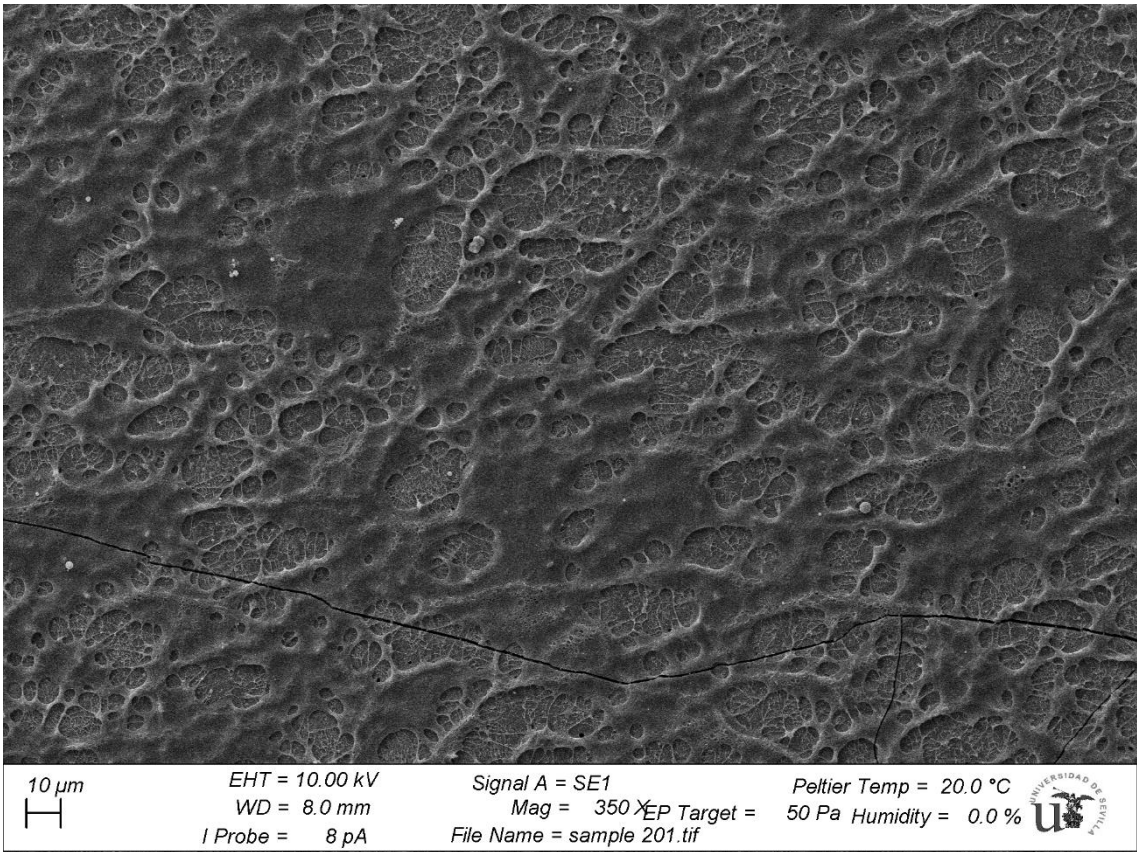
426 Figure 8. Droplet size distributions for lemongrass-in-water emulsions as a function of diutan  
 427 gum.

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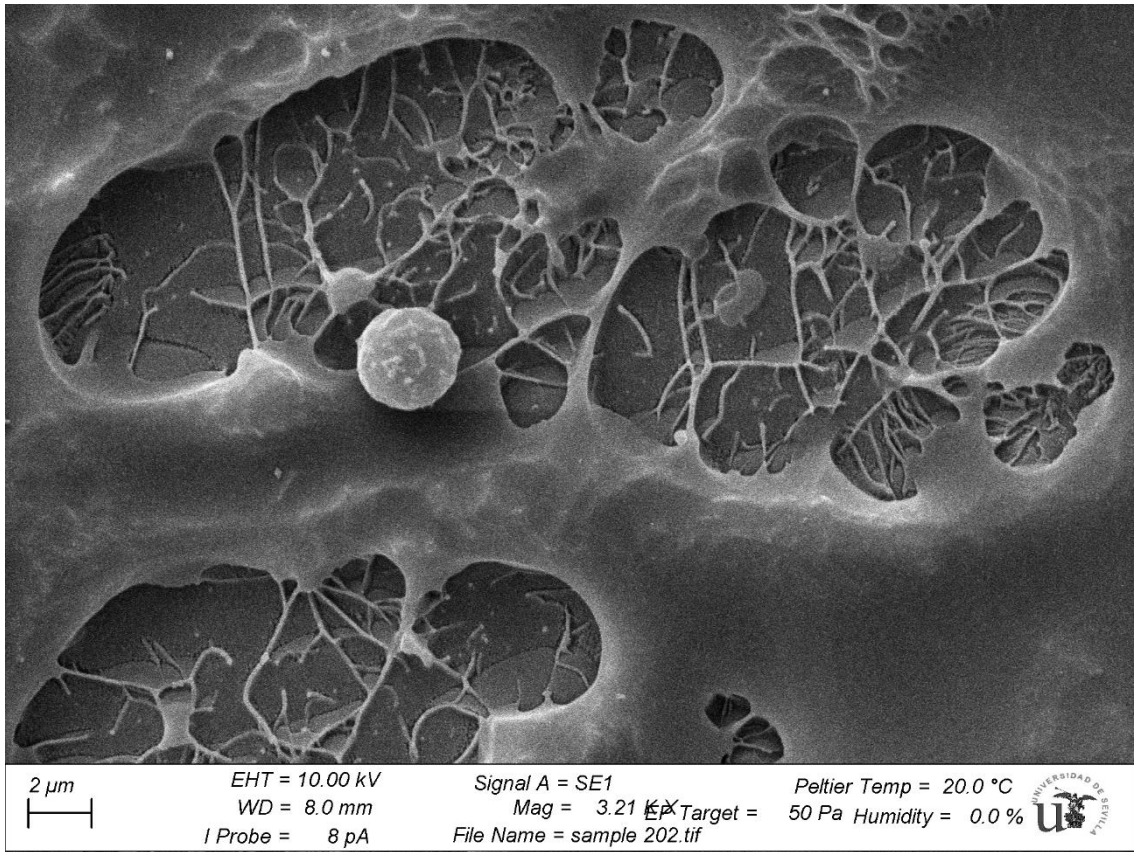
431 Figure 9A. Microstructure of the selected microfluidized lemongrass-in-water emulsion  
432 observed by Cryo-SEM technique.



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434 Figure 9B. Microstructure of lemongrass-in-water emulsion containing 0.2wt% diutan gum  
435 observed by Cryo-SEM technique at 350X.

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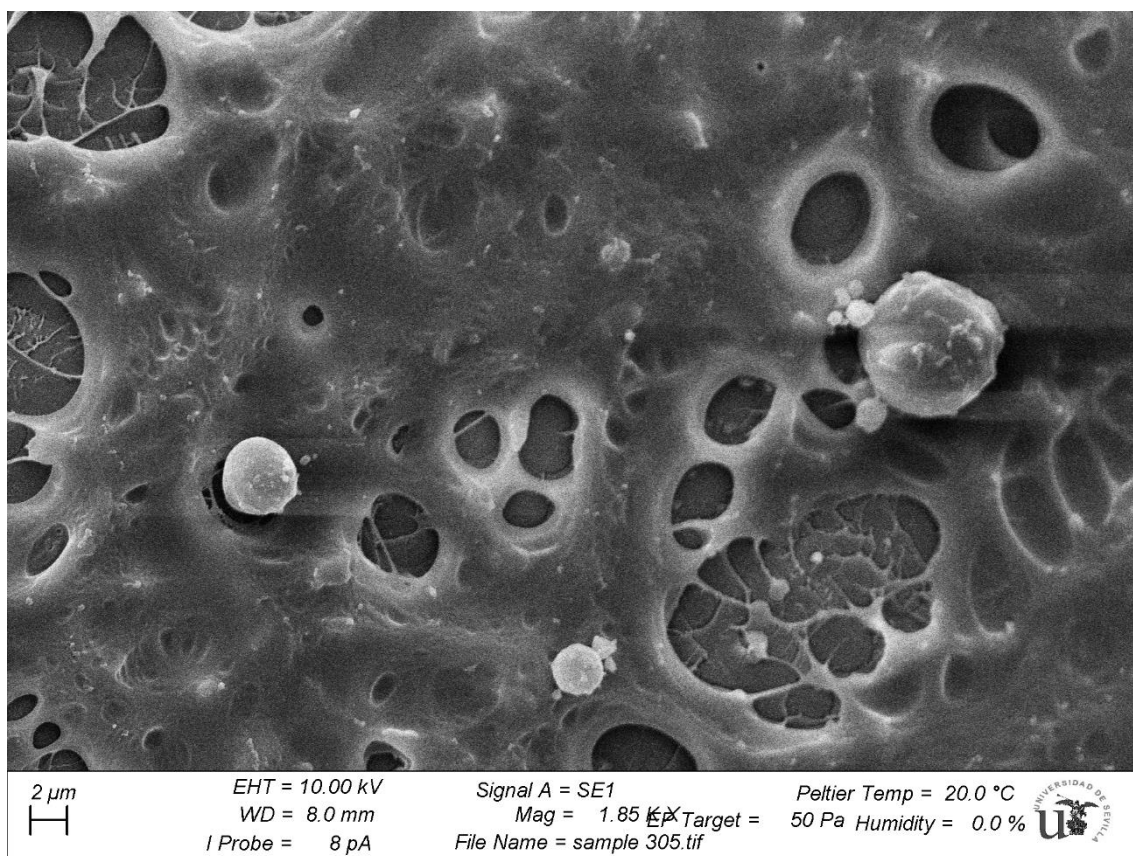


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438 Figure 9C. Microstructure of lemongrass-in-water emulsion containing 0.2 wt% diutan gum  
439 observed by Cryo-SEM technique at 3.21 KX.

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443 Figure 9D. Microstructure of lemongrass-in-water emulsion containing 0.4 wt% diutan gum  
444 observed by Cryo-SEM technique at 1.85KX.

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